

Preface

When I first came into contact with EPR spectroscopy during my diploma thesis, I was skeptical. How should radicals be of any help in the structural determination of complex materials? After all, I was merely interested in the radicals themselves but in the environment surrounding them. I forgot that unlike nuclear spins, electrons experience a large spatial distribution enabling them to ‘see’ and ‘sense’ far into their neighborhood. Although other characterization methods I used at that time also provided quite insightful results, it was EPR spectroscopy that delivered an insight into the nanoscopic structural and dynamic arrangement of molecules. This triggered my decision to solely focus on EPR spectroscopy as a tool to probe complex macromolecular systems in my PhD research.

The unique sensing capabilities of electron spins have been known for a long time and led many researchers to exploit radicals as spectral sensors. The concept of spin probing and spin labeling has come of age and has been applied to many biological and synthetic systems. Yet, the recent advent and continuing development of pulse techniques keep expanding the potential and applicability of EPR spectroscopy. Although still a scientific niche, it is a fast growing field that nicely complements established scattering and spectroscopic techniques. Particularly, the possibility to reliably infer distances in the range of several nanometers between two spin labels fueled the expansion of EPR spectroscopy in the field of structural and molecular biology. Such distance constraints can be used to determine the 3D structure and assembly of macromolecular systems like proteins and nucleic acids. However, the synthetic effort for obtaining and labeling specific protein mutants is high and tedious. Moreover, only one distance constraint can be obtained by one spin labeled mutant and distance measurement in most cases. Focusing on transporting materials this triggered the idea to approach the problem differently. Instead of labeling the fatty acid transporter human serum albumin itself, I simply replaced the transported guests by EPR-active fatty acids. The insights into the functional structure of albumin from the viewpoint of the transported guest molecules and the implications of such self-assembled systems for the respective EPR method are presented in [Chap. 3](#). Yet, one does not always require state-of-the-art

technology to obtain novel and exciting findings. This is demonstrated in [Chaps. 5 and 7](#), where the thermal collapse of thermoresponsive polymeric systems is studied by one of the most elementary nitroxide spin probes and simple continuous wave EPR spectroscopy.

The broad applicability of modern EPR spectroscopy in structural biology and materials science is emphasized by the examined systems: With a transport protein, a biomimetic host system, and responsive hydrogels and dendronized polymers, three substantially different transporting agents were examined by a variety of EPR spectroscopic methods and probes. In either case, EPR spectroscopy delivered illuminating and sometimes surprising insights into the molecular transporters, leading to a detailed molecular understanding of the materials' functions.

This thesis would not have been possible without the support of many people. First of all, I wish to thank my supervisors Prof. Dr. Hans Wolfgang Spiess and Dr. Dariush Hinderberger for their continuing interest in my research, their excellent ideas and their comprehensive support. Many people answered my never ending questions. Representative for all these people, I want to thank Dr. Uli Jonas for introducing me into the scientific world and Prof. Dr. Gunnar Jeschke for many helpful suggestions related to orientation selective and flip angle dependent DEER. I thank my co-workers from the University of Montréal, from the ETH in Zurich, and from the AK Spiess for productive and fruitful collaborations. I also wish to acknowledge the excellent infrastructure of the MPI for Polymer Research, most notably Christian Bauer for invaluable technical support. Finally, I would like to say a big 'thank you' to my colleagues, friends, and family for making my life at the institute and beyond an enjoyable and unforgettable time.

Money is not important as long as you have it. I was in the lucky position to receive independent funding during my PhD thesis which enabled me to attend many conferences and workshops around the world. I gratefully acknowledge financial support from the Fonds of the German Chemical Industry through a Chemiefonds scholarship and from the Graduate School of Excellence "Materials Science in Mainz" funded by the German Research Foundation.

Santa Barbara, August 2011

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Assessing the Functional Structure of Molecular
Transporters by EPR Spectroscopy

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2012, XVI, 212 p., Hardcover

ISBN: 978-3-642-25134-4