

Preface

Understanding the size and shape dependence of physical properties in nanoscale particles is a fundamental step towards the design, the fabrication, and the assembly of materials and devices with predictable behavior. In recent years, there has been a remarkable advancement in the ability to fabricate shape-controlled nanoparticles, for example rods, wires, and nanoparticles with branched shapes, especially via synthetic approaches in solution. Shape-controlled inorganic nanoparticles are among the most promising candidates as building blocks in nanoscale materials and devices, both because their physical properties are modified considerably compared to those of spherical nanoparticles and because their intrinsic geometry opens many new opportunities for their assembly into organized super-structures. In this book, we have decided to review the physical properties of elongated inorganic nanoparticles, with particular emphasis on the transition in these properties when the shape of the nanoparticles evolves from a sphere to a rod, but we will consider in many cases also nanowires. From the point of view of specific properties and materials, we have decided to cover the optical properties of semiconductors and noble metals, the electrical properties of semiconductors, the magnetic properties of various metals and metal oxides, the catalytic properties of various classes of materials, and the mechanical properties of metals and metal alloys.

[Chapter 1](#) will give an introduction into some basic quantum physics concepts, specifically tailored to the following [Chaps. 2 and 3](#) that are devoted to the optical and electrical properties of semiconductor nanorods. Semiconductor nanocrystals are among the most studied materials in nanoscience nowadays, due to the large number of potential applications employing these materials, for example, in optical devices (lasers [1–3], light emitting diodes [4, 5], photo-detectors [6], solar cells [7–9]), or biological labeling [10, 11], to cite a few. Elongated, rod-shaped semiconductor nanocrystals possess interesting physical properties which depend on their size, aspect ratio, and chemical composition, and these nanoparticles have been proposed as active materials in light emitting devices [12], photocatalysis [13], optically induced light modulation [14], photovoltaics, [7–9, 15] wave-function engineering [16–18], and optical memory elements exploiting the exciton

storage process [19]. More in general, these nanoparticles have been considered as replacement for spherical nanocrystals (the so-called “quantum dots”) in all those studies in which the elongated shape could in principle add new or improved properties.

Chapter 4 will deal with optical properties of elongated metal nanocrystals. Metallic nanocrystals have been proposed in a wide range of applications in various fields, among them sensing, biosensing, photodynamic therapy, photovoltaics, optics (light emitting diodes, photo detectors, lasers, imaging techniques beyond the diffraction limit), nano-optics, and nano-electronics (for example plasmonic waveguides) [20–32]. Metal nanostructures can interact strongly with light in the visible and near infrared region of the spectrum, due to the presence of free electrons, which can be promoted both to empty energy levels in the same band or to levels of an empty overlapping band. An incident electromagnetic field can elicit collective oscillations of these free electrons [20–23], which cause a displacement of the electrons from the nuclei, leading to the formation of various possible distributions in the surface charges. This creates Coulomb interactions between positive and negative charges, which induce restoring oscillating forces acting on free electrons. Each type of surface charge distribution is characterized by a collective oscillation mode, also termed as localized surface plasmon resonance. Various factors influence the possible types of SPRs in nanostructures and the frequencies at which they are observed and the shape of metal nanoparticles is certainly one of them. As an example, in rod-shaped nanoparticles the plasmon mode is split into two modes, a longitudinal one and a transverse one. There are many other physical effects connected with an elongated shape which differ from the spherical case, and these will be covered in detail in **Chap. 4**.

In **Chap. 5** we will review the magnetic properties of elongated nanoparticles. Many of the applications of magnetic nanoparticles are in life sciences and biomedicine [33–35]. Superparamagnetism is the term used for describing the absence of coercivity and remanent magnetization in particles that still maintain a considerable amount of polarizable spins under the effect of an external magnetic field [36, 37]. These magnetic nanocrystals, also known as superparamagnets, combine their reduced sizes with their magnetic field responsive character even if no residual magnetization is observed in the absence of an external magnetic field. For this reason they have been proposed as vectors for both in vitro and in vivo transport of different drugs or biomolecules attached to their surfaces, thereby providing selective access to cellular or molecular levels which are inaccessible to conventional therapeutic approaches. In the same way, they can also be used in biodetection and bioseparation techniques since once the target molecule has been attached to the nanocrystals, the application of an external magnetic field will allow their recovery [38, 39]. Iron oxide is clearly the most suitable material for such purposes due to its high chemical stability, biocompatibility, and superparamagnetic properties and iron oxide nanoparticles are being already used in several diagnostic and therapeutic techniques [40].

The achievement of higher coercivity values in particles with reduced size for information storage devices, or faster magnetic responses for smaller biomedical vectors, could be possible if one considers not only the finite size effects of spherical nanocrystals but also the additional phenomena arising from the shape anisotropy of particles such as nanorods. Nanorods or other one-dimensional nanosystems could also be capable of widening the temperature range of applications of a certain magnetic material compared to its bulk form (as will be shown later). The uniaxial shape anisotropy of metallic and oxide nano-objects will probably become a key factor for the development of improved devices. This chapter will also present an overview of various classes of magnetic materials that have been synthesized in nanorod shapes.

Chapter 6 will deal with the catalytic properties of elongated nanoparticles. Today, there is an increasing demand for catalytic materials, in terms of catalytic efficiency, cost of production, specificity, durability, and environmental sustainability [41–44]. This demand is driving research towards the exploitation of new nanoscale catalyst particles, in which the individual components have specific size, shape, exposure of specific reactive surfaces, and suitable combination of materials [45, 46]. Micro- and nanoparticles of various materials have been used as catalysts for many years [47–50], and experimental evidence has been collected so far demonstrating that the catalytic activity of particles is strongly related to their size, and in particular that nanosized particles exhibit increased catalytic activity with respect to larger particles, due to their higher surface to volume ratio [51, 52]. With recent advances in the synthesis of inorganic nanoparticles with controlled size and morphology [53–56], interest has grown towards the understanding of how the catalytic performance of these materials is dependent on shape. In terms of catalytic properties, there are several reasons why an elongated morphology is often preferable over a spherical morphology, and these will be described in Chap. 6, along with several case studies of nanorod-shaped catalysts.

Chapter 7 deals mainly with the mechanical properties of elongated nanoparticles. The miniaturization of micro electromechanical devices and the fabrication of thin films in the electronic industry have started to raise questions already decades ago about the mechanical behavior of confined systems. Early experiments on tensile testing of metal whiskers with micrometer transverse sides have evidenced strengths much higher than the bulk value [57], and recently pure metals and alloys with at least one dimension in the micro- and nanoscale range have been investigated, thanks to advances in the fabrication of new generations of samples suitable for mechanical testing (for example micro pillars prepared by focused ion beam) and in various techniques for studying their stress and deformation properties. Those studies have revealed a marked deviation in the mechanical properties of samples from bulk-like behavior already when their size is of the order of a few micrometers, which is comparable to the length scale of many plasticity mechanisms based on dislocation nucleation and propagation. The increased

strength of single nanocrystals could be useful for applications of these materials as active probes in nano-indentation, scanning probe microscopy, and field emission [58–60], to cite a few. [Chapter 7](#) ends with a paragraph on melting studies on nanorods.

Finally, we conclude this book with some remarks and an outlook on the future directions in this field.

Genoa	Roman Krahne, Liberato Manna, Chandramohan George
Lecce	Giovanni Morello
Barcelona	Albert Figuerola
Delhi	Sasanka Dekka

References

1. Chan Y, Caruge JM, Snee PT, Bawendi MG (2004) Multiexcitonic two-state lasing in a CdSe nanocrystal laser. *Appl Phys Lett* 85(13):2460–2462
2. Klimov VI, Ivanov SA, Nanda J, Achermann M, Bezel I, McGuire JA, Piryatinski A (2007) Single-exciton optical gain in semiconductor nanocrystals. *Nature* 447(7143):441–446
3. Klimov VI, Mikhailovsky AA, Xu S, Malko A, Hollingsworth JA, Leatherdale CA, Eisler HJ, Bawendi MG (2000) Optical gain and stimulated emission in nanocrystal quantum dots. *Science* 290(5490):314–317
4. Caruge JM, Halpert JE, Wood V, Bulovic V, Bawendi MG (2008) Colloidal quantum-dot light-emitting diodes with metal-oxide charge transport layers. *Nat Phot* 2(4):247–250
5. Anikeeva PO, Halpert JE, Bawendi MG, Bulovic V (2009) Quantum dot light-emitting devices with electroluminescence tunable over the entire visible spectrum. *Nano Lett* 9(7):2532–2536
6. Oertel DC, Bawendi MG, Arango AC, Bulovic V (2005) Photodetectors based on treated CdSe quantum-dot films. *Appl Phys Lett* 87(21):art. no. 213505
7. Huynh WU, Dittmer JJ, Alivisatos AP (2002) Hybrid nanorod-polymer solar cells. *Science* 295(5564):2425–2427
8. Kim S, Fisher B, Eisler HJ, Bawendi M (2003) Type-II quantum dots: CdTe/CdSe(core/shell) and CdSe/ZnTe(core/shell) heterostructures. *J Am Chem Soc* 125(38):11466–11467
9. Gur I, Fromer NA, Geier ML, Alivisatos AP (2005) Air-stable all-inorganic nanocrystal solar cells processed from solution. *Science* 310(5747):462–465
10. Dekka S, Quarta A, Lupo MG, Falqui A, Boninelli S, Giannini C, Morello G, De Giorgi M, Lanzani G, Spinella C, Cingolani R, Pellegrino T, Manna L (2009) CdSe/CdS/ZnS Double shell nanorods with high photoluminescence efficiency and their exploitation as biolabeling probes. *J Am Chem Soc* 131(8):2948–2958
11. Michalet X, Pinaud F, Lacoste TD, Dahan M, Bruchez MP, Alivisatos AP, Weiss S (2001) Properties of fluorescent semiconductor nanocrystals and their application to biological labeling. *Single Mol* 2(4):261–276
12. Zhou RH, Chang HC, Protasenko V, Kuno M, Singh AK, Jena D, Xing H (2007) CdSe Nanowires with illumination-enhanced conductivity: Induced dipoles, dielectrophoretic assembly, and field-sensitive emission. *J Appl Phys* 101(7):art. no. 073704
13. Hewa-Kasakarage NN, El-Khoury PZ, Tarnovsky AN, Kirsanova M, Nemitz I, Nemchinov A, Zamkov M (2010) Ultrafast carrier dynamics in Type II ZnSe/CdS/ZnSe nanobarbells. *ACS Nano* 4(4):1837–1844

14. Petti L, Ripa M, Fiore A, Manna L, Mormile P (2010) Optically induced light modulation in an hybrid nanocomposite system of inorganic CdSe/CdS nanorods and nematic liquid crystals. *Opt Mater* 32(9):1011–1016
15. Farva U, Park C (2010) Colloidal synthesis and air-annealing of CdSe nanorods for the applications in hybrid bulk hetero-junction solar cells. *Mater Lett* 64(13):1415–1417
16. Muller J, Lupton JM, Rogach AL, Feldmann J, Talapin DV, Weller H (2004) Monitoring surface charge movement in single elongated semiconductor nanocrystals. *Phys Rev Lett* 93(16):art. no. 167402
17. Muller J, Lupton JM, Lagoudakis PG, Schindler F, Koeppel R, Rogach AL, Feldmann J, Talapin DV, Weller H (2005) Wave function engineering in elongated semiconductor nanocrystals with heterogeneous carrier confinement. *Nano Lett* 5(10):2044–2049
18. Muller J, Lupton JM, Rogach AL, Feldmann J, Talapin DV, Weller H (2005) Monitoring surface charge migration in the spectral dynamics of single CdSe/CdS nanodot/nanorod heterostructures. *Phys Rev B* 72(20):art. no. 205339
19. Kraus RM, Lagoudakis PG, Rogach AL, Talapin DV, Weller H, Lupton JM, Feldmann J (2007) Room-temperature exciton storage in elongated semiconductor nanocrystals. *Phys Rev Lett* 98(1):art. no. 017401
20. Klimov V (2004) Semiconductor and metal nanocrystals. Marcel Dekker, New York
21. Noguez C (2005) Optical properties of isolated and supported metal nanoparticles. *Opt Mater* 27(7):1204–1211
22. Link S, El-Sayed MA (2000) Shape and size dependence of radiative, non-radiative and photothermal properties of gold nanocrystals. *Intern Rev Phys Chem* 19(3):409–453
23. Perez-Juste J, Pastoriza-Santos I, Liz-Marzan LM, Mulvaney P (2005) Gold nanorods: Synthesis, characterization and applications. *Coord Chem Rev* 249(17–18):1870–1901
24. Gonzalez AL, Reyes-Esqueda JA, Noguez C (2008) Optical properties of elongated noble metal nanoparticles. *J Phys Chem C* 112(19):7356–7362
25. Myroshnychenko V, Rodriguez-Fernandez J, Pastoriza-Santos I, Funston AM, Novo C, Mulvaney P, Liz-Marzan LM, de Abajo FJG (2008) Modelling the optical response of gold nanoparticles. *Chem Soc Rev* 37(9):1792–1805
26. Zhang JZ, Noguez C (2008) Plasmonic optical properties and applications of metal nanostructures. *Plasmonics* 3(4):127–150
27. Kelly KL, Coronado E, Zhao LL, Schatz GC (2003) The optical properties of metal nanoparticles: The influence of size, shape, and dielectric environment. *J Phys Chem B* 107(3):668–677
28. Meier M (2007) Plasmonics: Fundamentals and applications. Springer, New York
29. Link S, El-Sayed MA (2003) Optical properties and ultrafast dynamics of metallic nanocrystals. *Ann Rev Phys Chem* 54:331–366
30. Huang X, Neretina S, El-Sayed MA (2009) Gold nanorods: From synthesis and properties to biological and biomedical applications. *Adv Mater* 21:4880–4910. doi:[10.1002/adma.200802789](https://doi.org/10.1002/adma.200802789)
31. Liz-Marzan LM (2006) Tailoring surface plasmons through the morphology and assembly of metal nanoparticles. *Langmuir* 22:32–41. doi:[10.1021/la0513353](https://doi.org/10.1021/la0513353)
32. Schwartzberg AM, Olson TY, Talley CE, Zhang JZ (2006) Synthesis, characterization, and tunable optical properties of hollow gold nanospheres. *J Phys Chem B* 110(40):19935–19944
33. Dobson J (2006) Magnetic nanoparticles for drug delivery. *Drug Dev Res* 67(1):55–60
34. Park K, Lee S, Kang E, Kim K, Choi K, Kwon IC (2009) New generation of multifunctional nanoparticles for cancer imaging and therapy. *Adv Funct Mater* 19(10):1553–1566
35. Wilhelm C, Lavalie F, P  choux C, Tatischeff I, Gazeau F (2008) Intracellular trafficking of magnetic nanoparticles to design multifunctional biovesicles. *Small* 4(5):577–582
36. Jeong U, Teng X, Wang Y, Yang H, Xia Y (2007) Superparamagnetic colloids: Controlled synthesis and niche applications. *Adv Mater* 19(1):33–60
37. Lin X-M, Samia ACS (2006) Synthesis, assembly and physical properties of magnetic nanoparticles. *J Magn Magn Mater* 305(1):100–109

38. Lu A-H, Salabas EL, Schüth F (2007) Magnetic nanoparticles: Synthesis, protection, functionalization, and application. *Angew Chem Int Edit* 46(8):1222–1244
39. Dave SR, Gao X (2009) Monodisperse magnetic nanoparticles for biodetection, imaging, and drug delivery: A versatile and evolving technology. *Wiley Interdisc Rev Nanomed Nanobiotechnol* 1(6):583–609
40. Figuerola A, Di Corato R, Manna L, Pellegrino T (2010) From iron oxide nanoparticles towards advanced iron-based inorganic materials designed for biomedical applications. *Pharm Res* 62(2):126–143
41. Somorjai GA, Park JY (2008) Molecular factors of catalytic selectivity. *Angew Chem Int Edit* 47(48):9212–9228. doi:[10.1002/anie.200803181](https://doi.org/10.1002/anie.200803181)
42. Min BK, Friend CM (2007) Heterogeneous gold-based catalysis for green chemistry: Low-temperature CO oxidation and propene oxidation. *Chem Rev* 107(6):2709–2724. doi:[10.1021/cr050954d](https://doi.org/10.1021/cr050954d)
43. Heitbaum M, Glorius F, Escher I (2006) Asymmetric heterogeneous catalysis. *Angew Chem Int Edit* 45(29):4732–4762. doi:[10.1002/anie.200504212](https://doi.org/10.1002/anie.200504212)
44. Roucoux A, Schulz J, Patin H (2002) Reduced transition metal colloids: A novel family of reusable catalysts? *Chem Rev* 102(10):3757–3778. doi:[10.1021/cr010350j](https://doi.org/10.1021/cr010350j)
45. Norskov JK, Bligaard T, Rossmeisl J, Christensen CH (2009) Towards the computational design of solid catalysts. *Nat Chem* 1(1):37–46
46. Cuenya BR (2010) Synthesis and catalytic properties of metal nanoparticles: Size, shape, support, composition, and oxidation state effects. *Thin Solid Films* 518(12):3127–3150. doi:[10.1016/j.tsf.2010.01.018](https://doi.org/10.1016/j.tsf.2010.01.018)
47. Lewis LN (1993) Chemical catalysis by colloids and clusters. *Chem Rev* 93(8):2693–2730
48. Schmid G (1992) Large clusters and colloids—metals in the embryonic state. *Chem Rev* 92(8):1709–1727
49. Astruc D, Lu F, Aranzas JR (2005) Nanoparticles as recyclable catalysts: The frontier between homogeneous and heterogeneous catalysis. *Angew Chem Int Edit* 44(48):7852–7872. doi:[10.1002/anie.200500766](https://doi.org/10.1002/anie.200500766)
50. Gates BC (1995) Supported metal clusters: Synthesis, structure, and catalysis. *Chem Rev* 95(3):511–522
51. Wilson OM, Knecht MR, Garcia-Martinez JC, Crooks RM (2006) Effect of Pd nanoparticle size on the catalytic hydrogenation of allyl alcohol. *J Am Chem Soc* 128(14):4510–4511. doi:[10.1021/ja058217m](https://doi.org/10.1021/ja058217m)
52. Herzing AA, Kiely CJ, Carley AF, Landon P, Hutchings GJ (2008) Identification of active gold nanoclusters on iron oxide supports for CO oxidation. *Science* 321(5894):1331–1335. doi:[10.1126/science.1159639](https://doi.org/10.1126/science.1159639)
53. El-Sayed MA (2004) Small is different: Shape-, size-, and composition-dependent properties of some colloidal semiconductor nanocrystals. *Acc Chem Res* 37(5):326–333
54. Yin Y, Alivisatos AP (2005) Colloidal nanocrystal synthesis and the organic–inorganic interface. *Nature* 437(7059):664–670
55. Cozzoli PD, Pellegrino T, Manna L (2006) Synthesis, properties and perspectives of hybrid nanocrystal structures. *Chem Soc Rev* 35(11):1195–1208
56. Tao AR, Habas S, Yang PD (2008) Shape control of colloidal metal nanocrystals. *Small* 4(3):310–325
57. Brenner SS (1959) In: Doremus RH, Roberts BW, Turnbull D (eds) *Growth and perfection of crystals*. Wiley, New York
58. Greer JR, Nix WD (2005) Size dependence of mechanical properties of gold at the sub-micron scale. *Appl Phys A Mater Sci Process* 80(8):1625–1629
59. Chattopadhyay S, Chen LC, Chen KH (2006) Nanotips: Growth, model, and applications. *Crit Rev Solid State Mater Sci* 31(1–2):15–53
60. Mazilova TI, Mikhailovskij IM, Ksenofontov VA, Sadanov EV (2009) Field-Ion microscopy of quantum oscillations of linear carbon atomic chains. *Nano Lett* 9(2):774–778

Physical Properties of Nanorods

Krahne, R.; Manna, L.; Morello, G.; Figuerola, A.; George, C.; Deka, S.

2013, XVI, 282 p. 136 illus., 76 illus. in color., Hardcover

ISBN: 978-3-642-36429-7