

Chapter 1

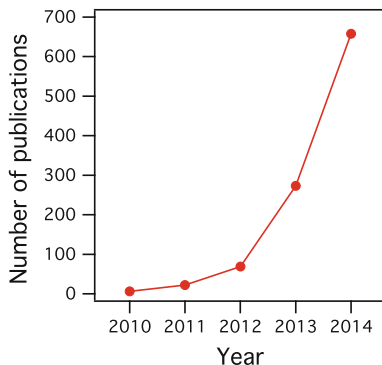
Introduction

Transition metal dichalcogenides (TMDCs) are old; the oldest known samples of MoS₂ date over 2.9 billion years [1]. A comprehensive, even though somewhat outdated, review about structure and properties of three-dimensional TMDCs can be found in [2]. Despite rather similar structure, TMDCs cover a wide spectrum of properties ranging from insulators, to semiconductors, to metals. This diversity of properties is a consequence of the existence of non-bonding d bands and the degree to which they are filled with electrons. It is also interesting to note that one of the first reports on monolayer TMDC was published in the mid 1980s [3] while few-layer-thick MoS₂ single crystals were reported even earlier [4] but these publications remained largely unnoticed.

These old materials experienced a renaissance after the discovery of unique electronic properties of graphene, for which K.S. Novoselov and A.K. Geim were awarded the Nobel Prize in 2010. In single-layer graphene's band structure, the linear dispersion at the K points gives rise to novel phenomena, such as the anomalous room-temperature quantum Hall effect, and opened up a new category of Fermi-Dirac physics. Graphene is a fantastic electronic and thermal conductor, and graphene-based materials have been proposed for a host of applications ranging from high-speed electronic and optical devices, energy generation and storage, hybrid materials, chemical sensors, and even DNA sequencing, and a variety of proof-of-concept devices have been demonstrated.

The success of graphene generated explosive interest in other two-dimensional materials, where use of different elements opens novel opportunities for the exciting new physics and ultimately thin devices. Two-dimensional TMDCs [5], which can be easily exfoliated [4] and present very interesting electrical and optical properties, became one of the most intensely studied areas of solid state physics and technology. Among these materials, semiconducting TMDCs are of special interest since the possibilities of gap engineering by varying the number of layers makes them exciting candidates for device applications. Figure 1.1 shows the number of publications per year found in the Web of Knowledge with the key words “MoS₂” and “monolayer”, which clearly indicates an exponential increase in interest in the past several years.

Fig. 1.1 The number of publications found by the Web of Knowledge with the search key words “MoS₂” and “monolayer”



The band structure of those compounds dramatically changes from bulk to single-layer samples, going from indirect gap in bulk materials to direct gap in monolayers, underscoring the important role of interlayer coupling. In addition, their electronic properties are very sensitive to external conditions such as temperature, pressure or strain.

The presence of boundaries, vacancies and/or adatoms in the samples can lead to interesting magnetic properties. Strong spin-orbit interaction in TMDCs alongside with the coupling of the spin, valley and layer degrees of freedom open unprecedented possibilities from both fundamental and applied perspectives. This possibility is especially interesting in single layers where the spin-orbit coupling lifts the spin degeneracy of the energy bands due to the absence of inversion symmetry. Furthermore, reduced dielectric screening in monolayer and few-layer samples of TMDCs makes excitonic effects exceptionally strong.

These and other properties of two-dimensional TMDCs are the subject of this volume. There have been numerous reviews published on this topic [6–21] as well as an edited volume on MoS₂ [22] and the interested readers can check them for details that may be missing here.

The TMDCs are about sixty in number; two-thirds of these assume layered structures. Most of these layered materials are synthetic but some exist naturally, e.g. natural MoS₂ crystals with the *2H* and *3R* phases (see Sect. 3.1.2 below for the explanation of the symbols) are quite common. The structure of this mineral, molybdenite, was first determined in 1923 [23]. Bulk TMDC crystals are conventionally grown using the chemical vapor transport method [24–26], where purified dichalcogenide material in the form of powder is mixed with the transport agent, usually bromine or iodine and sealed in a quartz ampoule. The quartz ampoule is introduced into a zone electric furnace, with a temperature gradient formed along the tube [27, 28]. Pure component materials, for example W and Se for the growth of WSe₂ can also be used.

18 VIIIA 8A																		2																	
1 IA H Hydrogen 1.008																		He Helium 4.003																	
3 Li Lithium 6.941	4 IIA Be Beryllium 9.012																	19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.833	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80
11 Na Sodium 22.990	12 Mg Magnesium 24.305	3 IIIB Sc Scandium 44.956	4 IVB Ti Titanium 47.88	5 VB V Vanadium 50.942	6 VIB Cr Chromium 51.996	7 VIIB Mn Manganese 54.938	8 VII Fe Iron 55.833		9 VIII Co Cobalt 58.933	10 VIII Ni Nickel 58.693	11 IIB Cu Copper 63.546	12 IIB Zn Zinc 65.38	13 IIIA Al Aluminum 26.982	14 IVA Si Silicon 28.086	15 VA P Phosphorus 30.974	16 VIA S Sulfur 32.06	17 VIIA Cl Chlorine 35.453	18 VIIIA Ar Argon 39.948																	
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.833	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80																		
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.71	51 Sb Antimony 121.760	52 Te Tellurium 127.6	53 I Iodine 126.904	54 Xe Xenon 131.29																		
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71 Lanthanide Series		72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.85	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.383	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium 209	85 At Astatine 210	86 Rn Radon 222																	
87 Fr Francium 223.020	88 Ra Radium 226.025	89-103 Actinide Series		104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [266]	107 Bh Bohrium [264]	108 Hs Hassium [277]	109 Mt Meitnerium [268]	110 Ds Darmstadtium [271]	111 Rg Roentgenium [272]	112 Cn Copernicium [285]	113 Nh Nihonium [284]	114 Fl Flerovium [289]	115 Uup Ununpentium [288]	116 Lv Livermorium [293]	117 Uus Ununseptium [294]	118 Uuo Ununoctium [294]																	
Lanthanide Series																																			
57 La Lanthanum 138.905	58 Ce Cerium 140.115	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.966	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.259	69 Tm Thulium 168.934	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967																					
Actinide Series																																			
89 Ac Actinium 227.028	90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.080	99 Es Einsteinium 252.083	100 Fm Fermium 257.095	101 Md Mendelevium 258.1	102 No Nobelium 259.101	103 Lr Lawrencium 262																					

Fig. 1.2 The transition metals and the three chalcogen elements that crystallise into layered structures are highlighted in the Periodic Table. The transition metals that crystallise into layered structures with some chalcogens but not with others are colour-framed. The columns in the Periodic Table show both ‘old’ and new labels, i.e. chalcogens can be referred to as either group VIA or group 16 elements

In Fig. 1.2 the Periodic Table of elements is shown with the transition metals that form layered structures highlighted in different colours; those metals that form layered structures with some chalcogens but not with others are colour-framed, e.g. while NiTe_2 has a layered structures, NiS_2 possesses a three-dimensional pyrite structure. Chalcogen atoms are highlighted in orange. Also shown in the Figure are the “old” and “new” notations for the columns, thus chalcogen atoms will be referred to as either group VIA or group 16 elements. Later in this volume the old and new notations are used interchangeably, following the original publications.

As can be seen from Fig. 1.2, group 4–7 metals are predominantly layered while some of the group 8–10 metals form three-dimensional crystals. Recent interest is mainly associated with semiconducting TMDCs, which form the main body of this volume. At the same time, we would like to mention that there are also reports on other materials, such as noble-transition-metal (Pt and Pd) dichalcogenides [29, 30] and tin disulphide [31], the latter being indirect-gap semiconductors.

The present monograph is organised as follows. In Chap. 2, the readers are introduced to chemistry of chalcogenides and transition metals, which is followed by Chap. 3, where the structure and properties of bulk TMDCs are briefly reviewed. Chapter 4 describes the major fabrication methods to produce two-dimensional TMDCs. In Chaps. 5 and 6, atomic and electronic structures of monolayer and few-layer TMDCs are discussed in detail. Raman scattering, which evolved into a major method of TMDCs characterisation, is the subject of Chap. 7. Luminescence from 2D TMDC and exciton behaviour are subsequently discussed in Chaps. 8 and 9, followed by Chaps. 10 and 11 dedicated to magnetism and spin-valley coupling in 2D TMDCs.

Chapter 12 is dedicated to miscellaneous phenomena observed in 2D TMDC, that are too short to be the subject of dedicated chapters, e.g. second-harmonic generation. Engineered heterostructures based on 2D TMDCs are described in Chap. 13, followed by the conclusive Chap. 14 that discusses emerging applications of 2D TMDCs in nanoelectronics.

References

1. J. Golden, M. McMillan, R.T. Downs, G. Hystad, I. Goldstein, H.J. Stein, A. Zimmerman, D.A. Sverjensky, J.T. Armstrong, R.M. Hazen, Rhenium variations in molybdenite (MoS_2): evidence for progressive subsurface oxidation. *Earth Planet. Sci. Lett.* **366**, 1 (2013)
2. J. Wilson, A. Yoffe, The transition metal dichalcogenides discussion and interpretation of the observed optical, electrical and structural properties. *Adv. Phys.* **18**(73), 193 (1969)
3. P. Joensen, R. Frindt, S.R. Morrison, Single-layer MoS_2 . *Mat. Res. Bull.* **21**(4), 457 (1986)
4. R. Frindt, Single crystals of MoS_2 several molecular layers thick. *J. Appl. Phys.* **37**(4), 1928 (1966)
5. H.S.S.R. Matte, A. Gomathi, A.K. Manna, D.J. Late, R. Datta, S.K. Pati, C.N.R. Rao, MoS_2 and WS_2 analogues of graphene. *Angew. Chem.* **122**(24), 4153 (2010)
6. M. Xu, T. Liang, M. Shi, H. Chen, Graphene-like two-dimensional materials. *Chem. Rev.* **113**(5), 3766 (2013)
7. S.Z. Butler, S.M. Hollen, L. Cao, Y. Cui, J.A. Gupta, H.R. Gutierrez, T.F. Heinz, S.S. Hong, J. Huang, A.F. Ismach et al., Progress, challenges, and opportunities in two-dimensional materials beyond graphene. *ACS Nano* **7**(4), 2898 (2013)
8. A. Kuc, Low-dimensional transition-metal dichalcogenides. *Chem. Modell.* **11**, 1 (2014)
9. M. Chhowalla, H.S. Shin, G. Eda, L.J. Li, K.P. Loh, H. Zhang, The chemistry of two-dimensional layered transition metal dichalcogenide nanosheets. *Nat. Chem.* **5**(4), 263 (2013)
10. W. Zhao, R.M. Ribeiro, G. Eda, Electronic structure and optical signatures of semiconducting transition metal dichalcogenide nanosheets. *Acc. Chem. Res.* **48**(1), 91 (2015)
11. H. Zeng, X. Cui, An optical spectroscopic study on two-dimensional group-vi transition metal dichalcogenides. *Chem. Soc. Rev.* **44**, 2629 (2015)
12. X. Zhang, X.F. Qiao, W. Shi, J.B. Wu, D.S. Jiang, P.H. Tan, Phonon and raman scattering of two-dimensional transition metal dichalcogenides from monolayer, multilayer to bulk material. *Chem. Soc. Rev.* **44**(9), 2757 (2015)
13. Q. Ji, Y. Zhang, Y. Zhang, Z. Liu, Chemical vapour deposition of group-VIB metal dichalcogenide monolayers: engineered substrates from amorphous to single crystalline. *Chem. Soc. Rev.* **44**, 2587 (2015)
14. D. Jariwala, V.K. Sangwan, L.J. Lauhon, T.J. Marks, M.C. Hersam, Emerging device applications for semiconducting two-dimensional transition metal dichalcogenides. *ACS Nano* **8**(2), 1102 (2014)
15. Q.H. Wang, K. Kalantar-Zadeh, A. Kis, J.N. Coleman, M.S. Strano, Electronics and optoelectronics of two-dimensional transition metal dichalcogenides. *Nat. Nanotech.* **7**(11), 699 (2012)
16. D. Voiry, A. Mohite, M. Chhowalla, Phase engineering of transition metal dichalcogenides. *Chem. Soc. Rev.* **44**(9), 2702 (2015)
17. H. Wang, H. Yuan, S.S. Hong, Y. Li, Y. Cui, Physical and chemical tuning of two-dimensional transition metal dichalcogenides. *Chem. Soc. Rev.* **44**(9), 2664 (2015)
18. C. Tan, H. Zhang, Two-dimensional transition metal dichalcogenide nanosheet-based composites. *Chem. Soc. Rev.* **44**(9), 2713 (2015)
19. N. Zibouche, A. Kuc, J. Musfeldt, T. Heine, Transition-metal dichalcogenides for spintronic applications. *Ann. Phys. (Berlin)* **526**(9–10), 395 (2014)

20. D. Lembke, S. Bertolazzi, A. Kis, Single-layer MoS₂ electronics. *Acc. Chem. Res.* **48**(1), 100 (2015)
21. H. Yu, X. Cui, X. Xu, W. Yao, Valley excitons in two-dimensional semiconductors. *Nat. Sci. Rev.* **2**(1), 57 (2014)
22. Z.M. Wang, MoS₂: *Materials, Physics, and Devices* (Springer, Heidelberg, 2013)
23. R.G. Dickinson, L. Pauling, The crystal structure of molybdenite. *J. Am. Chem. Soc.* **45**(6), 1466 (1923)
24. A. Al-Hilli, B. Evans, The preparation and properties of transition metal dichalcogenide single crystals. *J. Cryst. Growth* **15**(2), 93 (1972)
25. H. Schäfer, *Chemical Transport Reactions* (Academic Press Inc., Cambridge, 1964)
26. A. Ubaldini, E. Giannini, Improved chemical vapor transport growth of transition metal dichalcogenides. *J. Cryst. Growth* **401**, 878 (2014)
27. L. Brixner, Preparation and properties of the single crystalline AB₂-type selenides and tellurides of niobium, tantalum, molybdenum and tungsten. *J. Inorg. Nucl. Chem.* **24**(3), 257 (1962)
28. R. Nitsche, The growth of single crystals of binary and ternary chalcogenides by chemical transport reactions. *J. Phys. Chem. Solid.* **17**(1), 163 (1960)
29. P. Miró, M. Ghorbani-Asl, T. Heine, Two dimensional materials beyond MoS₂: Noble-transition-metal dichalcogenides. *Angew. Chem. Int. Ed.* **53**(11), 3015 (2014)
30. N. Zibouche, A. Kuc, P. Miró, T. Heine, Noble-metal chalcogenide nanotubes. *Inorganics* **2**(4), 556 (2014)
31. Y. Huang, E. Sutter, J.T. Sadowski, M. Cotlet, O.L. Monti, D.A. Racke, M.R. Neupane, D. Wickramaratne, R.K. Lake, B.A. Parkinson et al., Tin disulfide an emerging layered metal dichalcogenide semiconductor: materials properties and device characteristics. *ACS Nano* **8**(10), 10743 (2014)

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