

substance: FeSi₂

property: phase transitions, Debye temperature, heat capacity, density

phase transition: Figs. 1, 2.

decomposition temperatures (peritectic temperature)

(compare phase diagram, Fig. 3)

T_{dec}	1223 K			59S,
				60S
	1200 K			68B
	1259 K			68P
	1250 K			71U
	1260 K		from resistivity jump	73N
	1253 K		for Fe _{0.97} Mn _{0.03} Si ₂	
	1218 K		for Fe _{0.94} Mn _{0.06} Si ₂	

Debye temperature

Θ_{D}	630 K	$T = 0$ K	from heat capacity $c_{\text{p}}(T)$	73W
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heat capacity: Fig. 4.

density

d	4.93 g cm ⁻³	RT	pycnometric	71D
	4.94 g cm ⁻³	RT	X-ray density	71D

References:

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- 68P Piton, J. P., Fay, M. F.: C. R. Acad. Sci., Paris 266C (1968) 514.
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Fig. 1.

FeSi₂. Electrical conductivity vs. reciprocal temperature [69B]. Semiconductor-metal transition in a nominally stoichiometric polycrystalline sample (50.15 wt% Si). Comparison with the decomposition product below T_{tr} .

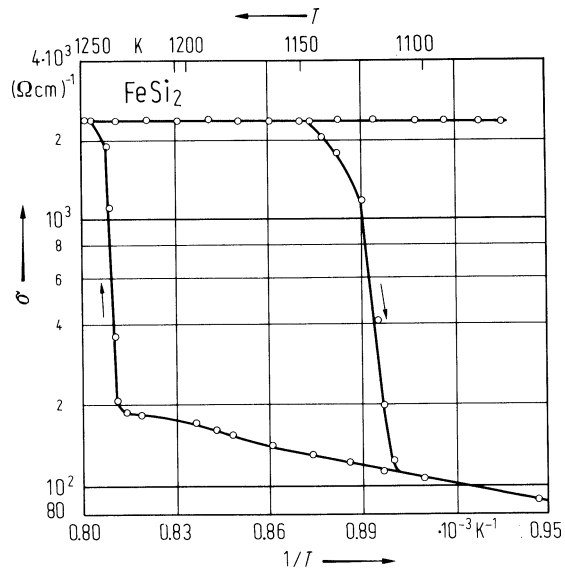


Fig. 2.

$\text{Fe}_{0.955}\text{Si}_2$. Electrical conductivity vs. reciprocal temperature [69B]. Semiconductor-metal transition in an iron-deficient polycrystalline sample (51.3 wt% Si). Comparison with the decomposition product below T_{tr} .

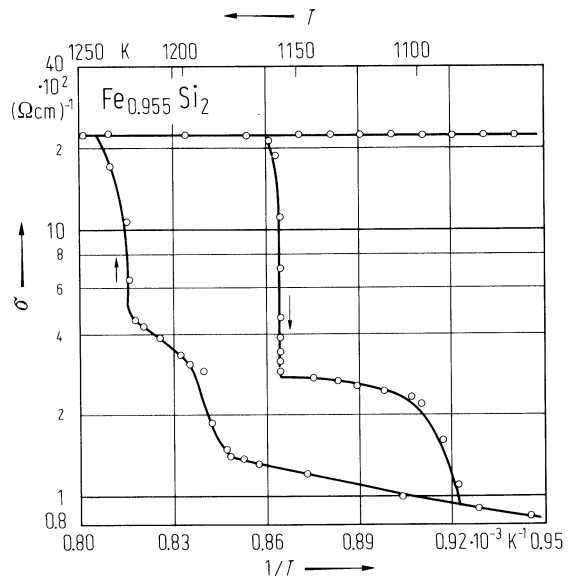


Fig. 3.

Fe-Si. Phase diagram near FeSi_2 [68P]. α refers to the high-temperature phase $\text{Fe}_{\approx 0.8}\text{Si}_2$ often designated $\alpha\text{-FeSi}_2$.

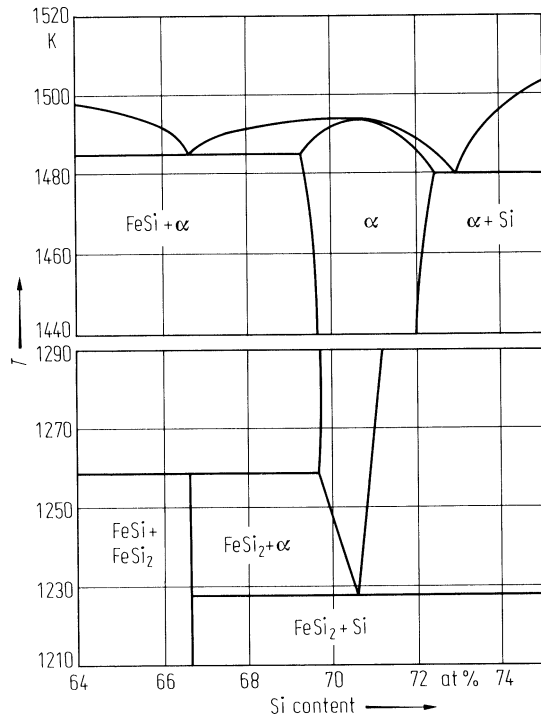


Fig. 4.

FeSi₂. Heat capacity vs. temperature [73W]. Full curve: measurements of Krentsis, Gel'd and Kalishevich (1963); Broken curve: measurements of Maglic and Parrot (1970).

