

**substance: V<sub>2</sub>O<sub>3</sub>**

**property: crystal structure, lattice parameters of pure material, room-temperature phase**

Corundum structure, trigonal space group  $D_{3d}^6 - R\bar{3}c$ ,  $Z = 6$ . This phase is normally V-deficient, phase diagram: Fig. 1. Detailed structure: Fig. 2.

**lattice parameters**

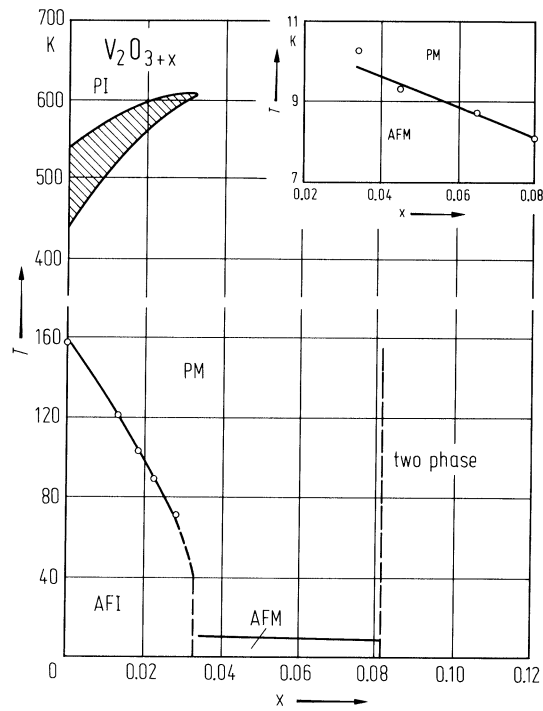
$a$	4.9717 Å	$T = 298$ K	lattice parameters as a function	78B1,
$c$	14.005 Å		of $x$ in $V_2O_{3+x}$ and as a	78B2
$c/a$	2.8283		function of pressure: Fig. 3	
$a$	$4.954 + 1.619 \cdot 10^{-4} T - 6.905 \cdot 10^{-8} T^2$	$T = 298 \dots 1273$ K	anomalous variation of lattice parameters above room temperature: Fig. 4	73E
$c$	$13.963 - 1.68 \cdot 10^{-4} T + 1.577 \cdot 10^{-7} T^2$			
$a$	4.9492 Å		variation of $c/a$ ratio with	75R
$c$	13.998 Å		temperature: Fig. 5	
$c/a$	2.828			
$V$	296.94 Å <sup>3</sup>			
$a$	4.951 Å			73P
$c$	14.007 Å			
$a_{rh}$	5.474 Å	$T = 300$ K	rhombohedral	78B1,
$\alpha_{rh}$	53.80°		cell parameters	78B2

## References:

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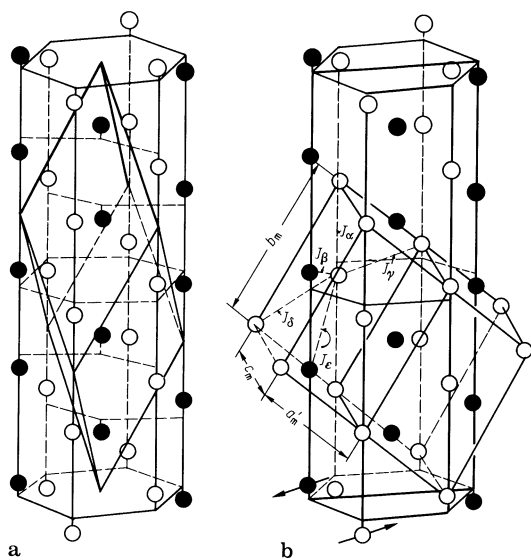
**Fig. 1.**

$V_2O_{3+x}$ . (Magnetic) phase diagram obtained from experimental results [80U]. PI: paramagnetic insulator, PM paramagnetic metal, AFI: antiferromagnetic insulator AFM: antiferromagnetic metal. Hatched area shows region where the upper transition is seen.



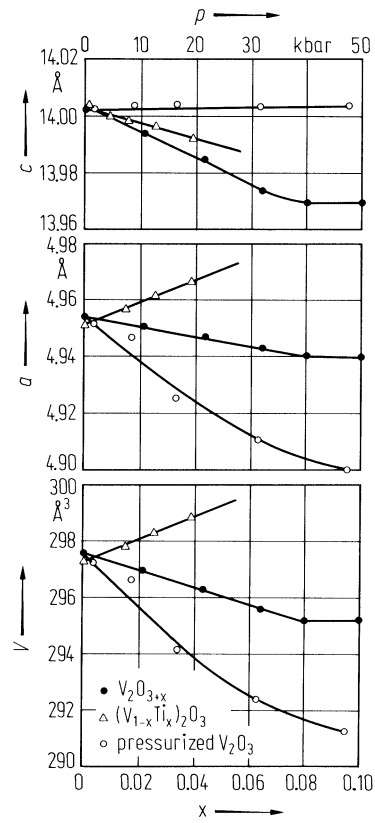
**Fig. 2.**

$V_2O_3$ . Relationships between V-atoms in hexagonal, rhombohedral and monoclinic unit cells. The primitive rhombohedral axes are shown as bold lines in (a). The magnetic and crystallographic unit cells are shown in (b). Spins on the filled circles are opposite to those on the open circles. Assuming equivalent metal atoms, the magnetic monoclinic cell is half ( $a'_m = 1/2 a_m$ ) of the crystallographic cell. [81W].  $J_\alpha \dots J_\epsilon$  are magnetic coupling constants.



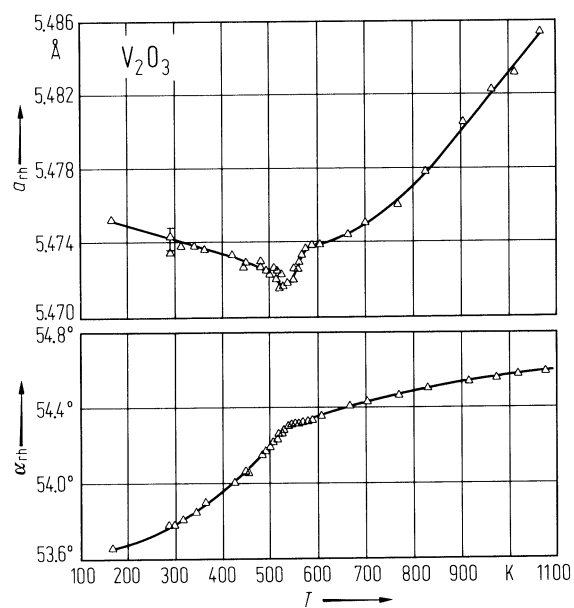
**Fig. 3.**

$V_2O_{3+x}$ . Comparison of the lattice parameters vs. composition  $x$  for  $V_2O_{3+x}$ ,  $(V_{1-x}Ti_x)_2O_3$  and vs. pressure for  $V_2O_3$  [70M].



**Fig. 4.**

$V_2O_3$ . Lattice constants  $a_{rh}$ ,  $\alpha_{rh}$  in the rhombohedral unit cell vs. temperature [78B1].



**Fig. 5.**

$V_2O_3$ ,  $(V_{0.96}Cr_{0.04})_2O_3$ .  $c/a$  ratios vs. temperature. Inset:  $c/a$  ratios for metal oxides of the corundum structure. Open circles: first heating cycle, triangles: second heating cycle, full circles: cooling cycle [69M].

