

substance: Ta₂O₅

property: crystal structure and related parameters of L-Ta₂O₅

The Ta–O phase diagram appears to show only one thermodynamically stable phase, Ta₂O₅.

Ta₂O₅ exists in two forms, L- or β-Ta₂O₅, stable below 1320°C, and H- or α-Ta₂O₅, which is the stable form at high temperature. Kinetics of transformation are slow and H-Ta₂O₅ is therefore the form available as crystals at 300 K and for which the data reported below pertain. H-Ta₂O₅ melts at 1872°C [56R].

L-Ta₂O₅:

crystal structure

Indexing of the diffraction pattern of this oxide proved very difficult. The most intense lines correspond to an orthorhombic lattice, $a' = 6.20 \text{ \AA}$, $b' = 3.66 \text{ \AA}$, $c' = 3.89 \text{ \AA}$ [70R1], but numerous weak lines appear contingent on b' multiplied by a factor m which is dependent on impurity content and thermal history. A single phase L-form can be equilibrated by heating Ta₂O₅ to 1350°C for a long period, just below the upwards transition temperature [70R2]. This has $m = 11$, $a = 6.198 \text{ \AA}$, $b = 40.290 \text{ \AA}$, $c = 3.888 \text{ \AA}$, $V = 970.9 \text{ \AA}^3$, $Z = 11$, $d_{\text{calc}} = 8.31 \text{ g cm}^{-3}$ [71S].

Electron microscopy also shows lattice imaging with $m = 11$. Equilibration of L-Ta₂O₅ at 1000°C for long periods gives a structure with $m = 14$, and at intermediate temperatures an apparently infinite number of ordered, or partially ordered intergrowths with $m = 11$ to 14 are possible. The L-form is also stabilized by small amounts of cation impurity, provided the ionic radius of the impurity is small; examples are W, Ge, B and Al [70R2].

detailed structure: Fig. 1. There is some relationship to T–Nb₂O₅. The structure shown in Fig. 1 in (001) projection would contain 22 Ta and 58 O, so three O atoms have to be eliminated, which the structure achieves by distortion in the planes $d_1 \dots d_4$ of Fig. 1. Low temperature amorphous films appear to consist of edge-sharing pentagonal bipyramids, and it is the fusion of these, and concomitant elimination of OH and H₂O that leads to the structure of Fig. 1.

For a detailed table of interatomic distances, see [71S].

References:

- 56R Reisman, A., Holtzberg, F., Berkenblit, M., Berry, M.: J. Am. Chem. Soc. 78 (1956) 4514.
- 70R1 Roth, R. S., Waring, J. L., Parker, H. S.: J. Solid State Chem. 2 (1970) 445.
- 70R2 Roth, R. S., Waring, J. L., Brower, W.: J. Res. Nat. Bur. Stand. Sect. A 74 (1970) 485.
- 71S Stephenson, N. C., Roth, R. S.: Acta Crystallogr. B27 (1971) 1037.

Fig. 1.

$\text{L-Ta}_2\text{O}_5$. A (001) projection of the structure. There are three distortion planes in the unit cell located at d_1 , d_2 and d_3 . The fourth position at d_4 is related by symmetry but is not used in this unit cell. Black dots represent metal atoms and shaded areas oxygen coordination polyhedra [71S]. $11b'$: unit cell having 11 UO_3 type subcell structure with $b' =$ length of VO_3 type subcell.

