

**substance:** OsAs<sub>2</sub>

**property:** physical properties

Energy level scheme is similar to that given for FeAs<sub>2</sub> (Fig. 1).

**energy gap**

$E_g$	$\approx 0.9$ eV	RT	from diffuse reflectance	63H
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**magnetic susceptibility**

(in  $10^{-6}$  cm<sup>3</sup> mol<sup>-1</sup>,  $\chi$  in CGS units)

$\chi_m$	- 78	$T = 85$ K	powder sample, Gouy method	63H
	- 100	$T = 295$ K		
	- 110	$T = 90$ K	$\chi(T)$ linear from 90 to 775 K,	63H
	- 115	$T = 775$ K	polycrystalline sample	

far infrared absorption: for spectrum in the range 50...400 cm<sup>-1</sup>, see [83L].

**Comparative tables on structural data of transition metal dipnictides:**

**structure, chemical bond:** see document ,

**crystallographical data** of compounds with octahedrally coordinated cations, see document ,

**interatomic distances** in marcasite- and loellingite-type compounds, see document .

**References:**

- 63H Hulliger, F.: Nature (London) 198 (1963) 1081.  
72F Fan, A. K. L., Rosenthal, G. H., McKinzie, H. L., Wold, A.: J. Solid State Chem. 5 (1972) 136.  
72G Goodenough, J. B.: J. Solid State Chem. 5 (1972) 144.  
83L Lutz, H. D., Schneider, O., Kliche, G.: Phys. Chem. Minerals 9 (1983) 109.

**Fig. 1.**

$\text{FeAs}_2$ . One-electron energy levels for the valence electrons in loellingite [72F, 72G].

