

**substance: RhP<sub>2</sub>**

**property: physical properties**

Energy level scheme is similar to that given for CoAs<sub>2</sub> (Fig. 1).

**energy gap**

$E_g$	$\approx 1$ eV	RT	from diffuse reflectance of powder samples	63H
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**magnetic susceptibility**

(in  $10^{-6}$  cm<sup>3</sup> mol<sup>-1</sup>,  $\chi$  in CGS units)

$\chi_m$	- 37	$T = 295$ K	(temperature dependence probably	63H
	- 14	$T = 77$ K	due to paramagnetic impurities)	
	- 77.5	$T = 80 \dots 1000$ K	( $\chi_g = 0.47 \cdot 10^{-6}$ cm <sup>3</sup> g <sup>-1</sup> )	71K

**far infrared absorption:** for spectrum in the range 50...400 cm<sup>-1</sup>, see [83L].

**Comparative tables on structural data of transition metal dipnictides:**

**structure, chemical bond:** see document ,

**crystallographical data** of compounds with octahedrally coordinated cations, see document ,

**interatomic distances** in arsenopyrite-type phases, see document .

**References:**

- 63H     Hulliger, F.: Phys. Lett. 4 (1963) 282.  
71K     Kjekshus, A.: Acta Chem. Scand. 25 (1971) 411.  
72G     Goodenough, J. B.: J. Solid State Chem. 5 (1972) 144.  
83L     Lutz, H. D., Schneider, O., Kliche, G.: Phys. Chem. Minerals 9 (1983) 109.

**Fig. 1.**

$\text{CoAs}_2$ . Energy-level scheme for the valence electrons in the arsenopyrite-type phase with formal  $d^5$  configuration [72G].

