

**substance: FeO (Fe<sub>1-x</sub>O)**

**property: band structure**

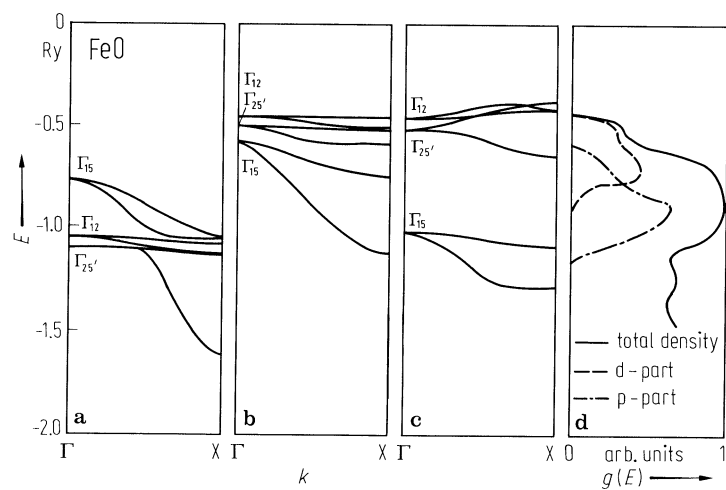
Band structure calculations; [72M, 78K]. Results compared to photoemission data [75E]: Fig. 1. Substantial overlap between the d-orbital energies and the O 2p band is evident. A local cluster calculation on FeO<sub>6</sub><sup>10-</sup> has also been reported [76T] (see Fig. 2). The qualitative conclusions are similar to the later band calculation [78K] with the t<sub>2g</sub>↑ (Γ'<sub>25</sub>) lying below the O 2p valence band edge.

## References:

- 72M     Mattheiss, L. F.: Phys. Rev. B5 (1972) 290.
- 72S     Slater, J. C., Johnson, K. H.: Phys. Rev. B5 (1972) 844.
- 75E     Eastman, D. E., Freeouf, J. L.: Phys. Rev. Lett. 34 (1975) 395.
- 76T     Tossell, J. A.: J. Electron Spectrosc. Relat. Phenom. 8 (1976) 1.
- 78K     Kunz, A. B., Surratt, G. T.: Solid State Commun. 25 (1978) 9.

**Fig. 1.**

FeO. Band structure showing (a) Hartree-Fock data, (b) self-consistent calculation with correlation, (c) non-self consistent calculation, (d) experimental density of states from photoemission spectroscopy data [75E].



**Fig. 2.**

FeO. One-electron unrestricted X $\alpha$  cluster calculation on FeO<sub>6</sub><sup>10-</sup> [76T]. Shown are the electronic levels of an FeO<sub>6</sub><sup>10-</sup> cluster and the free-atom energy levels.  $E_{x\alpha}$ : orbital energy. For discussion of cluster method, see [72S].

