

substance: CoO

property: phase diagram, crystal structure, lattice parameters

phase diagram: Fig. 1. The spinel Co_3O_4 is unstable at high temperatures and the monoxide shows a wide range of stoichiometry [64F].

phase boundaries

$\log_{10} a_{\text{O}_2}$	$16.5 - 20300/T[\text{K}]$ for $\text{Co}_{1-\delta}\text{O}/\text{Co}_3\text{O}_4$	a_{O_2} is the oxygen activity	77D
	$7.2 - 24100/T[\text{K}]$ for $\text{Co}/\text{Co}_{1-\delta}\text{O}$		

crystal structure

$T > T_N$: NaCl cubic structure, space group $\text{O}_h^5 - \text{Fm}\bar{3}\text{m}$, $Z = 4$ [53G].

$T < T_N$: Jahn-Teller distortion leading to an approximately tetragonal structure that preserves orbital angular momentum (Fig. 2) [53G], space group $\text{C}_{2h}^3 - \text{C2}/\text{m}$, $Z = 2$ [66S].

lattice parameters

a	4.2581(5) Å 4.258 Å	high- temperature phase, $T = 295$ K	pressure dependence: Fig. 3	53G 66C
a	5.183 Å	low- temperature phase, $T = 77$ K	within the NaCl setting this can be described as $a = b = 4.265$ Å, $c = 4.217$ Å, $c/a = 0.9887$, $\alpha = \beta = \gamma = 89^\circ 58'$	66S
b	3.015 Å			
c	3.017 Å			
β	$125^\circ 33'$			

References:

- 53G Greenwald, S.: Acta Crystallogr. 6 (1953) 396.
- 64F Fisher, B., Tannhauser, D. S.: J. Electrochem. Soc. 111 (1964) 1194.
- 66C Clendenen, R. L., Drickamer, H. G.: J. Chem. Phys. 44 (1966) 4223.
- 66S Saito, S., Nakahiyashi, K., Shinomura, Y.: J. Phys. Soc. Jpn. 21 (1966) 850.
- 77D Dieckmann, R.: Z. Phys. Chem. (N. F.) 107 (1977) 189.

Fig. 1.

Co – O. Phase diagram [77D]. a_{O_2} : oxygen activity.

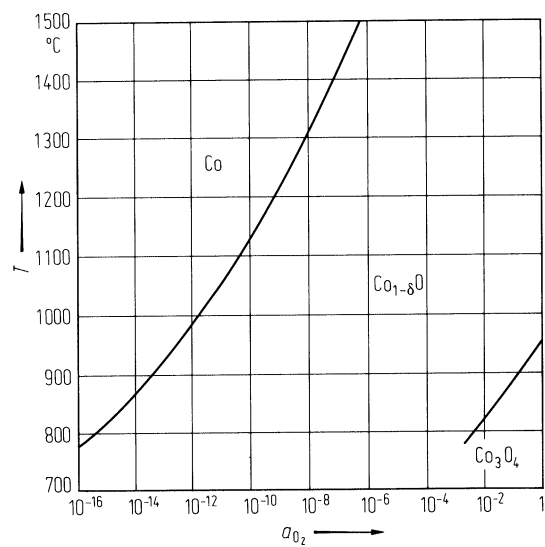


Fig. 2.

CoO. Unit cell dimensions above and below the transition temperature. The dashed vertical line shows the transition temperature as determined from magnetic measurements [53G].

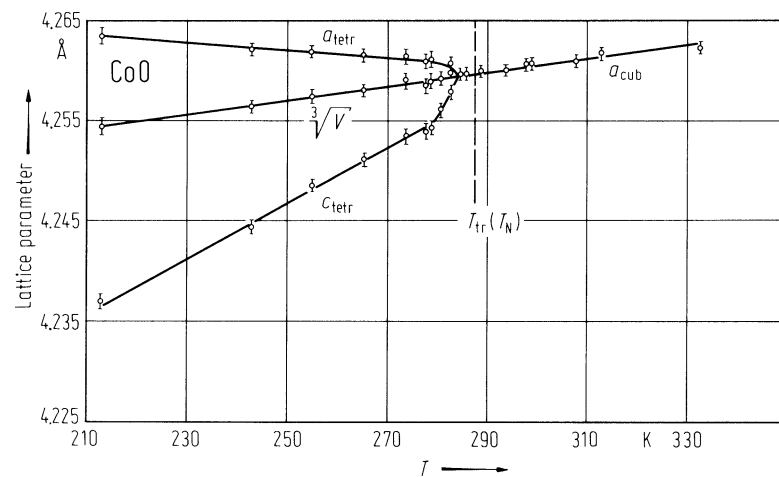


Fig. 3.

CoO. Unit cell length vs. pressure at RT [66C].

