

substance: CoO

property: electrical conductivity, Seebeck coefficient

activation energies for conductivity

E_A	0.22...0.57 eV	$T = 220...290$ K	quenched samples in which δ of $\text{Co}_{1-\delta}\text{O}$ varies from 0.012...0.001	66F
	0.24...0.5 eV	$T = 200...300$ K	quenched samples: $\delta = 0.012...0.002$	67B
	0.29...0.43 eV	$T = 150...300$ K	Li doped samples; smaller energies correspond to larger lithium content	69B
	0.41...0.49 eV	$T = 290...420$ K	Li doped single crystal	67A
	0.58 eV	$T = 1270...1670$ K	region A	66F
	0.70 eV	$T = 1270...1670$ K	region B	66F
	0.48 eV	$T = 1270...1470$ K		68E
	0.742 eV	$T = 300...500$ K	single crystal	80J

Conductivity dependence on oxygen partial pressure: Fig. 1, on Li addition: Figs. 5, 6, 8. The deviation from linearity at low values of Li dopant is due to Co vacancies dominating the conductivity. This is borne out by the Seebeck data (Fig. 7).

activation energy for Seebeck coefficient

E_A	0.54 eV	$T = 1270...1670$ K	region A	66F
	0.94 eV	$T = 1270...1670$ K	region B	66F
	0.27...0.42 eV	$T = 290...420$ K	Li-doped single crystal	67A
	0.306 eV	$T = 300...1000$ K	single crystal	80J

Positive sign of Seebeck coefficient (Fig. 2, 3) supports conduction by holes. Sign inversion at high temperatures near Co/CoO boundary [70G, 72G]. For lithium doped samples the similarity of the slopes for the thermopower and conductivity suggests that conduction is of the band or large-polaron type [71B, 67A, 67B, 65Z] (see also Fig. 4), but results on single crystals that have been annealed at high temperature suggest the possibility of activated conductivity [77D, 66F, 80J]. For Seebeck coefficient of Li doped samples, see also Fig. 7.

References:

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Fig. 1.

$\text{Co}_{1-\delta}\text{O}$. Electrical conductivity vs. oxygen activity at various temperatures. Data from [68E, 66S, 36W, 75M, 72G, 66F]. [77D].

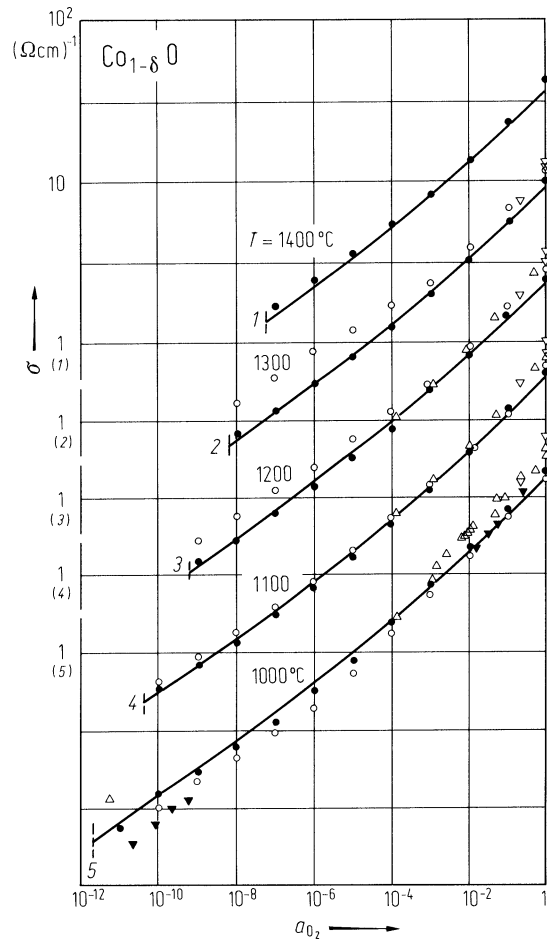


Fig. 2.

CoO. Conductance and reduced Seebeck coefficient vs. reciprocal temperature for a single crystal [71B].

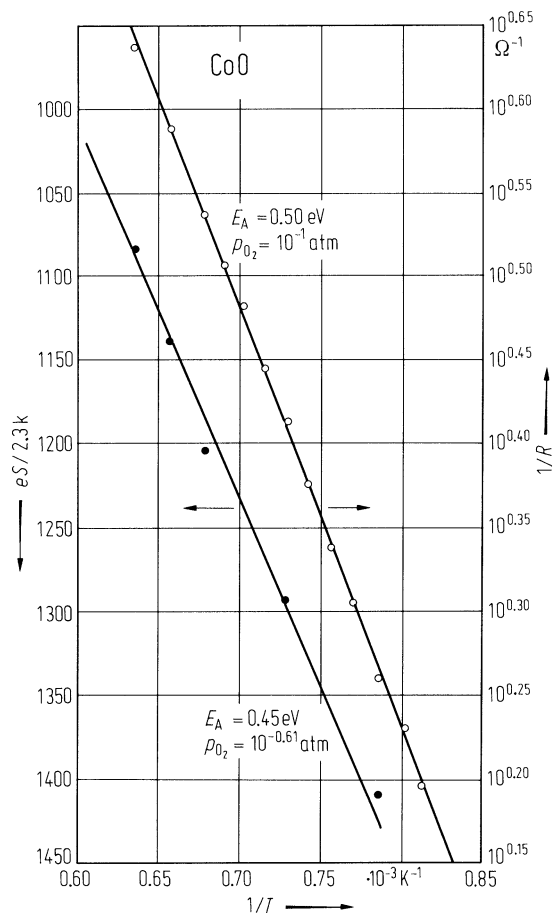


Fig. 3.

CoO. (a) Resistivity and (b) Seebeck coefficient vs. reciprocal temperature for various pure single crystals. 1 data from [80J], 2 data from [67B].

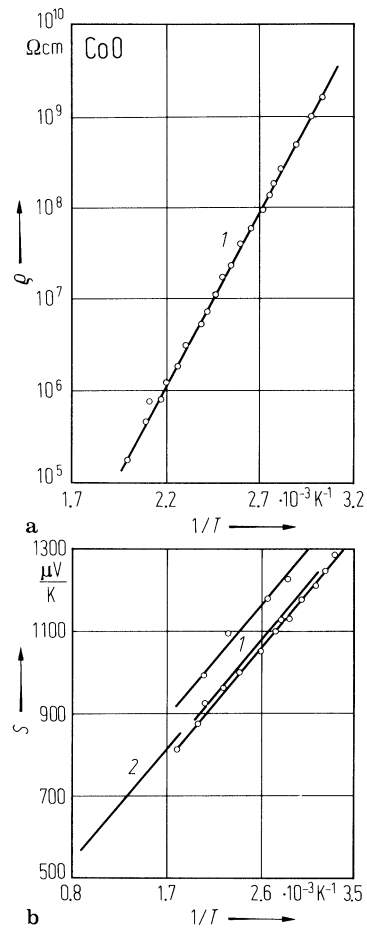


Fig. 4.

CoO. Hall mobility for holes vs. (reciprocal) temperature for a non-stoichiometric and a Li doped sample [67N].

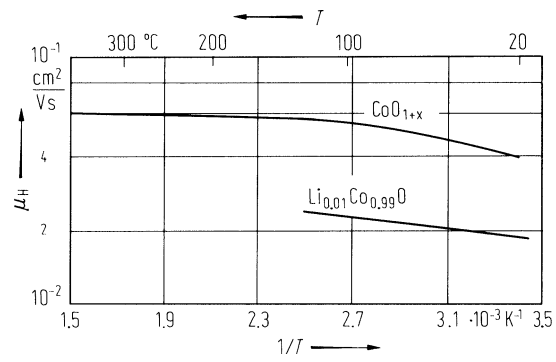


Fig. 5.

$\text{Li}_x\text{Co}_{1-x}\text{O}$. Resistivity vs. reciprocal temperature for various Li doped samples. Open symbols: from dc measurements, full symbols: from ac measurements [69B].

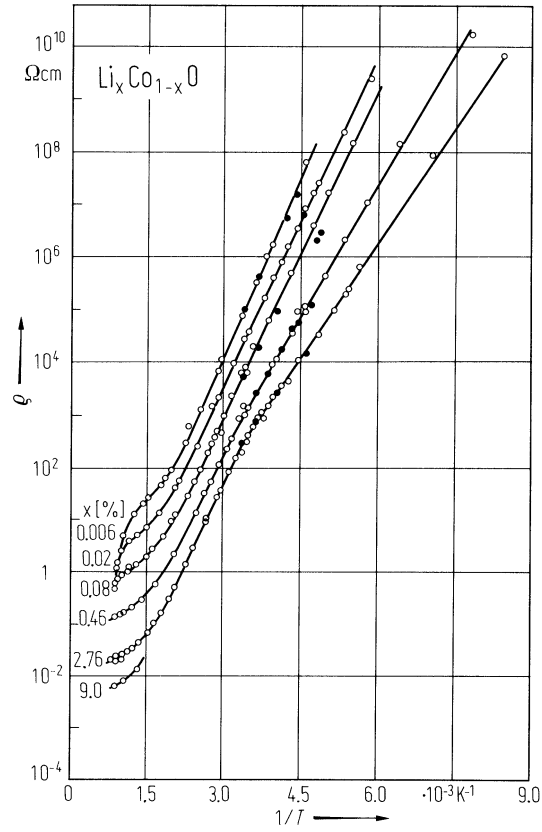


Fig. 6.

$\text{Li}_x\text{Co}_{1-x}\text{O}$. Resistivity vs. Li content at 800 K and 1200 K [69B].

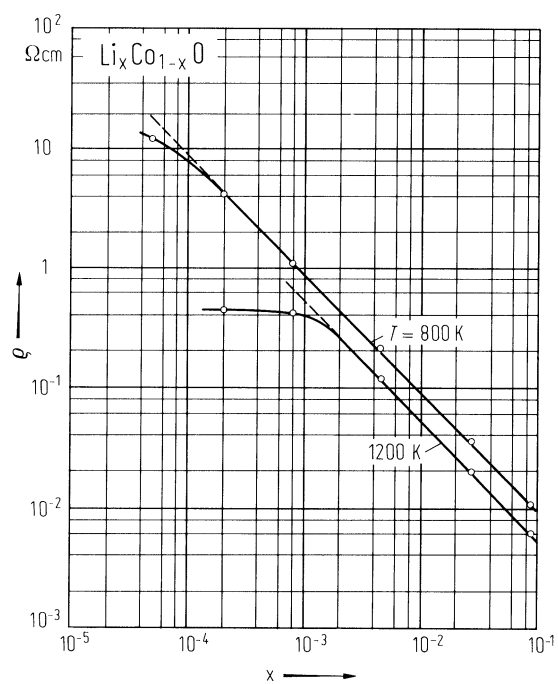


Fig. 7.

$\text{Li}_x\text{Co}_{1-x}\text{O}$. Seebeck coefficient vs. temperature for 0.08 at% Li doped CoO (a). (b) shows the high temperature region expanded together with results for 0.006 at% Li doped material [69B].

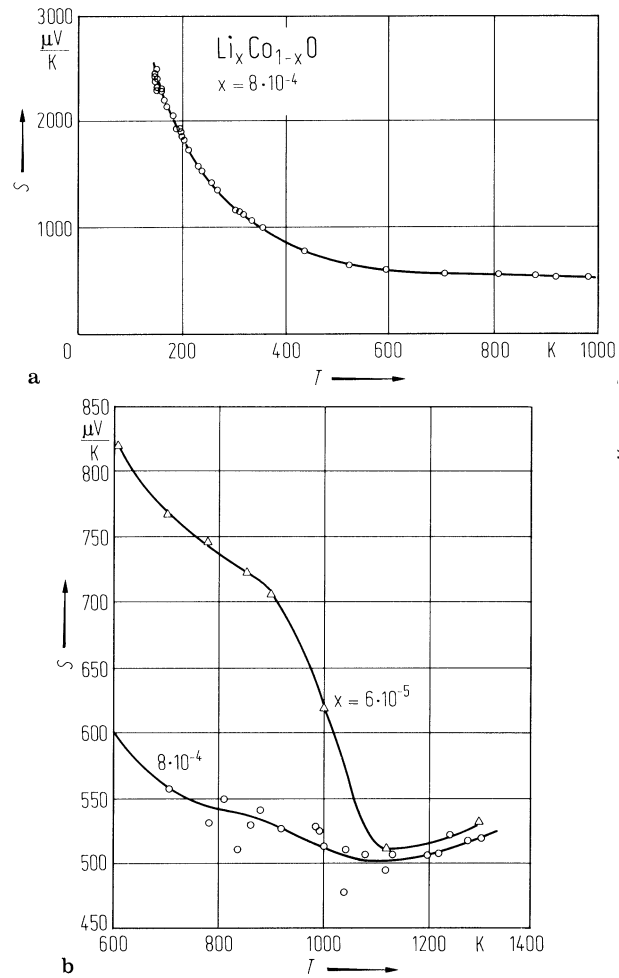


Fig. 8.

CoO:Li. Hall mobility, Hall coefficient and resistivity vs. (reciprocal) temperature for p-type Li doped ceramic samples measured in nitrogen (N_2) or oxygen (O_2) atmosphere [67D].

