

**substance: Nd<sub>2</sub>S<sub>3</sub>**

**property: crystal structure, physical properties**

**α-Nd<sub>2</sub>S<sub>3</sub>**

**crystal structure** orthorhombic (D<sub>2h</sub><sup>16</sup> – Pnma)

**lattice parameters**

<i>a</i>	7.42 Å	limit of stability < 900°C; color black; thermal conductivity data [73S]	71B
<i>b</i>	15.95 Å		
<i>c</i>	4.22 Å		
<i>a</i>	7.442(2) Å		68S
<i>b</i>	15.519(3) Å		
<i>c</i>	4.029(2) Å		

**electrical conductivity**

$\sigma$	35 Ω <sup>-1</sup> cm <sup>-1</sup>	68S
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**β-Nd<sub>2</sub>S<sub>3</sub>**

**crystal structure** tetragonal (D<sub>4h</sub><sup>20</sup> – I4<sub>1</sub>/acd)

**lattice parameters**

<i>a</i>	14.99 Å	limit of stability 900...1300°C; thermal conductivity [73S]	71B
<i>c</i>	19.90 Å		

**γ-Nd<sub>2</sub>S<sub>3</sub>**

**crystal structure** cubic (Th<sub>3</sub>P<sub>4</sub>-defect structure, T<sub>d</sub><sup>6</sup> – I $\bar{4}$  3d)

**lattice parameters**

<i>a</i>	8.527 Å	60P, 81K 82N 85Z
	8.533(1) Å	
	8.522 Å	

**melting point**

<i>T<sub>m</sub></i>	2010°C	60P, 72G 81K
	1860°C	

**density**

<i>d</i>	5.50(1) g cm <sup>-3</sup>	82N
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**energy gap and other energy parameters**

<i>E<sub>g</sub></i>	2.50 ± 0.1 eV	optical gap	85Z
	2.4 ± 0.25 eV	thermodynamic	85Z
	1.9 eV	X-ray spectra	83S
	3 eV	optical absorption	70H
	2.5 eV	optical absorption	80S

$E_b$	5.6 eV	S 3p-level	MgK $\alpha$ XPS, Fig. 14 ( $E_b$ rel. to $E_F$ )	91K
	5.6 eV	Nd 4f-level	MgK $\alpha$ XPS, Fig. 14	
	14.5 eV	S 3s-level	MgK $\alpha$ XPS, Fig. 14	
	20.6 eV	Nd 5p $_{3/2}$ -level	MgK $\alpha$ XPS, Fig. 14	
	24.3 eV	Nd 5p $_{1/2}$ -level	MgK $\alpha$ XPS, Fig. 14	
	41 eV	Nd 5s-level	MgK $\alpha$ XPS, Fig. 14	
	121 eV	Nd 4d $_{5/2}$ -level	MgK $\alpha$ XPS, Fig. 15	
	123 eV	Nd 4d $_{3/2}$ -level	MgK $\alpha$ XPS, Fig. 15	
	983 eV	Nd 3d $_{5/2}$ -level	AlK $\alpha$ XPS, Fig. 16	
	1005 eV	Nd 3d $_{3/2}$ -level	AlK $\alpha$ XPS, Fig. 16	
$E$	6.3 eV	S 3p- $E_F$	ELS, Fig. 17	91K
	12.1 eV	S 3p-cond. band, surface plasmon	ELS, Fig. 17	
	16.7 eV	bulk plasmon	ELS, Fig. 17	
	22.5 eV	Nd 5p- $E_F$	ELS, Fig. 17	
	30.6 eV	Nd 5p-5d	ELS, Fig. 17	
	$\approx 44$ eV	Nd 5s- $E_F$	ELS, Fig. 17	
	135 eV	Nd 4d-4f	ELS, Fig. 18	
	<b>phonon wavenumbers</b>			
$(\nu/c)_{TO}$	192 cm $^{-1}$			79A
	235 cm $^{-1}$			
	243.9 cm $^{-1}$		IR measurements	80S
	270.2 cm $^{-1}$			
	298.5 cm $^{-1}$			
$(\nu/c)_{LO}$	305 cm $^{-1}$			79A
	250.9 cm $^{-1}$			80S
	292.3 cm $^{-1}$			
	338.4 cm $^{-1}$			
<b>dielectric constants</b>				
$\epsilon(0)$	13			80S
	18.35			79A
$\epsilon(\infty)$	7.0			80S
	7.1			79A, 79Z
<b>activation energy</b>				
$E_A$	2.7...3.8 meV		electrical measurement	77T
<b>electrical conductivity</b>				
$\sigma$	10 $^{-10}$ $\Omega^{-1}$ cm $^{-1}$	n-type		70H, 80S
	10 $^2$ $\Omega^{-1}$ cm $^{-1}$	S-excess		70H
	1...2 $\cdot$ 10 $^{-3}$ $\Omega^{-1}$ cm $^{-1}$			74T

*Further figures and references:*

**coordination polyhedra:** Fig. 1

**heat capacity:** Fig. 4

**thermal conductivity:** Fig. 5 and [73S]

anomaly of temperature dependent **thermal expansion coefficient:** Fig. 6

**Raman spectra:** Fig. 7

**absorption and transmission spectra:** Figs. 8, 9, 10; see also Fig. 2

**reflectivity spectrum:** Fig. 3

temperature dependence of **electrical conductivity:** Fig. 11

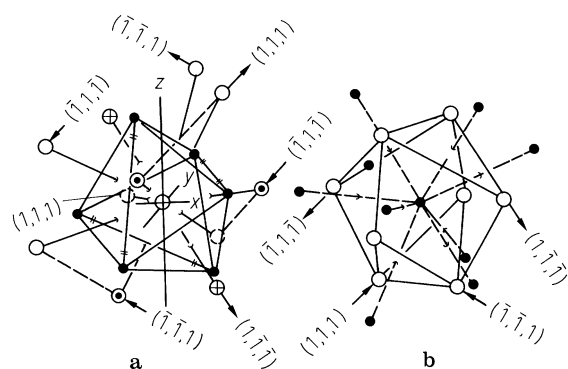
temperature dependence of **thermoelectric power** and **Hall coefficient:** Figs. 12, 13

## References:

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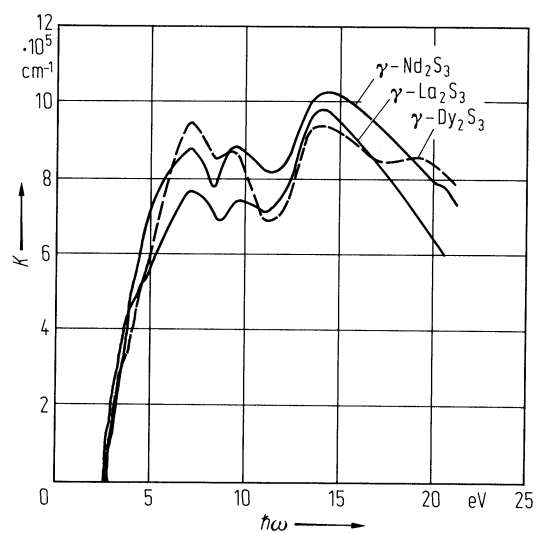
**Fig. 1.**

Th<sub>3</sub>P<sub>4</sub>-type compounds. The coordination polyhedra of the cations and the anions. Full circles: Th- atoms, other circles: P-atoms [66H].



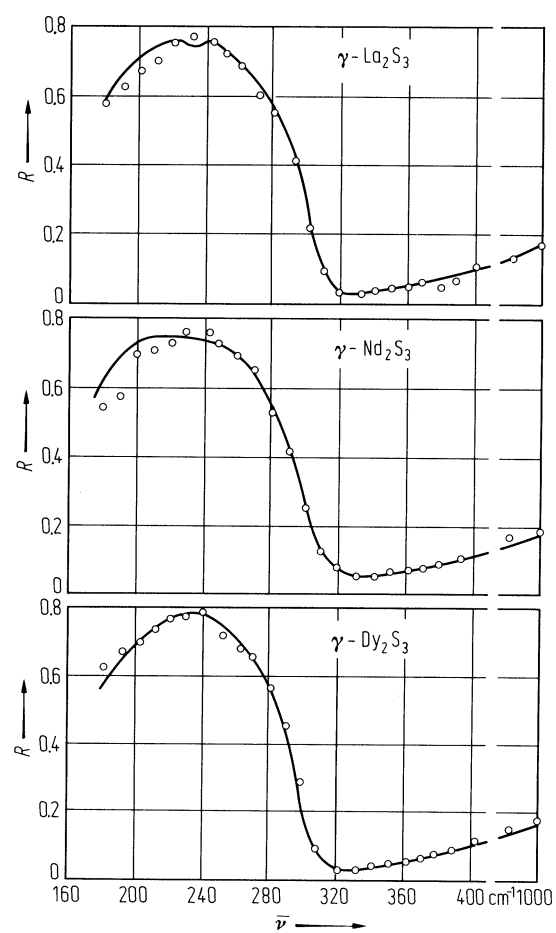
**Fig. 2.**

$\gamma$ - $\text{La}_2\text{S}_3$ ,  $\gamma$ - $\text{Nd}_2\text{S}_3$ ,  $\gamma$ - $\text{Dy}_2\text{S}_3$ . Absorption coefficient vs. photon energy [79Z].



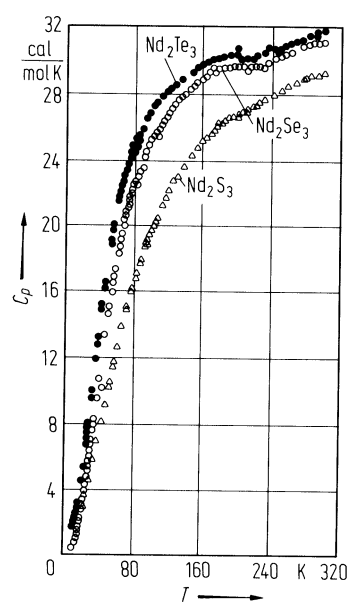
**Fig. 3.**

$\gamma$ - $\text{La}_2\text{S}_3$ ,  $\gamma$ - $\text{Nd}_2\text{S}_3$ ,  $\gamma$ - $\text{Dy}_2\text{S}_3$ . Reflectivity vs. wavenumber in the infrared range [79A].



**Fig. 4.**

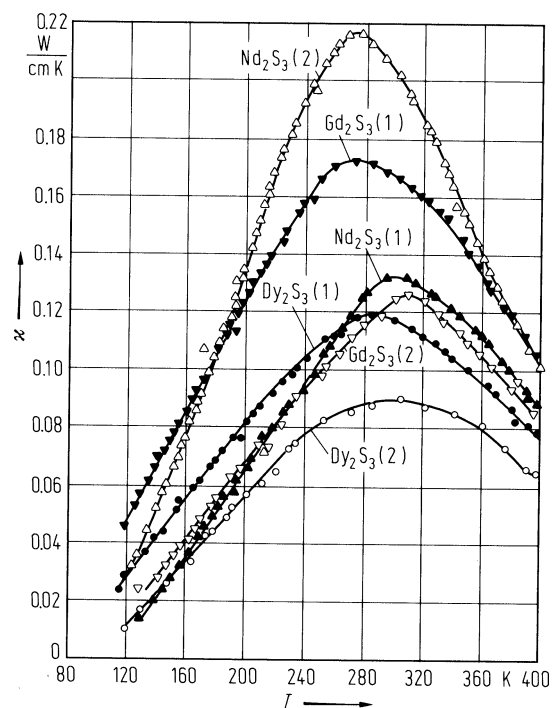
$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>, Nd<sub>2</sub>Se<sub>3</sub>, Nd<sub>2</sub>Te<sub>3</sub>. Molar heat capacity vs. temperature [72S].





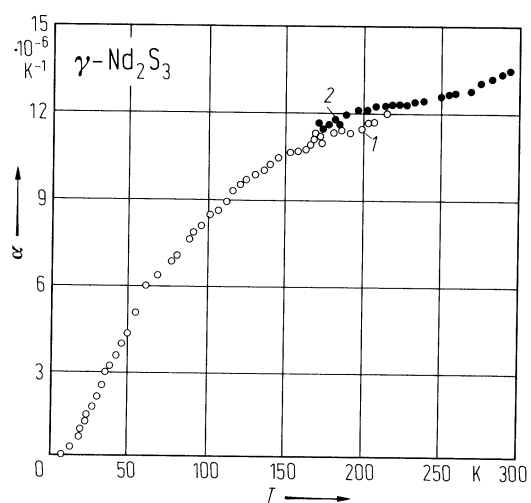
**Fig. 5.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>,  $\gamma$ -Gd<sub>2</sub>S<sub>3</sub>,  $\gamma$ -Dy<sub>2</sub>S<sub>3</sub>. Thermal conductivity vs. temperature of several single crystals. The samples show the following excess metal content: Nd<sub>2</sub>S<sub>3</sub> (1): 0.7% Nd, Nd<sub>2</sub>S<sub>3</sub> (2): 0.6% Nd, Gd<sub>2</sub>S<sub>3</sub> (1): 0.4% Gd, Gd<sub>2</sub>S<sub>3</sub> (2): 0.5% Gd, Dy<sub>2</sub>S<sub>3</sub> (1): 0.1% Dy, Dy<sub>2</sub>S<sub>3</sub> (2): 0.2% Dy [77T].



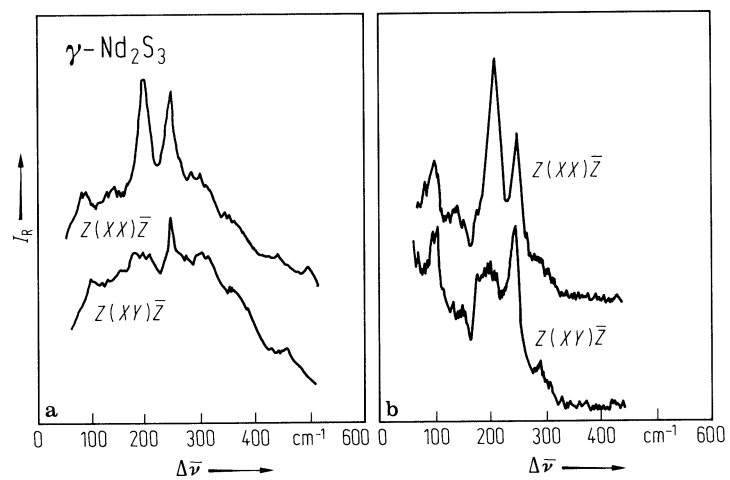
**Fig. 6.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>. Linear thermal expansion coefficient vs. temperature showing an anomaly. (1): values obtained after preliminary cooling below 169 K, (2): values obtained when the sample was not cooled below 169 K [82N].



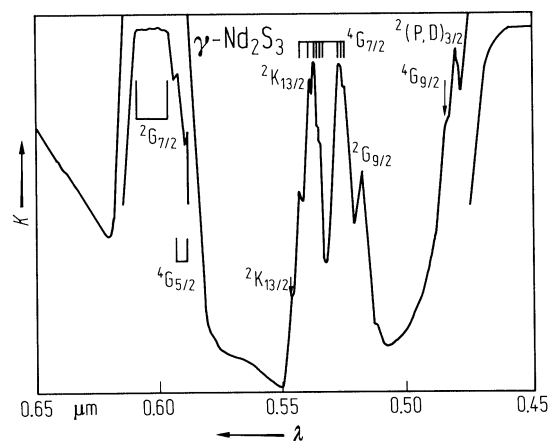
**Fig. 7.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>. Raman spectra (intensity vs. Raman shift) in the Stokes (a) and anti-Stokes (b) regions for different polarizations [80S].



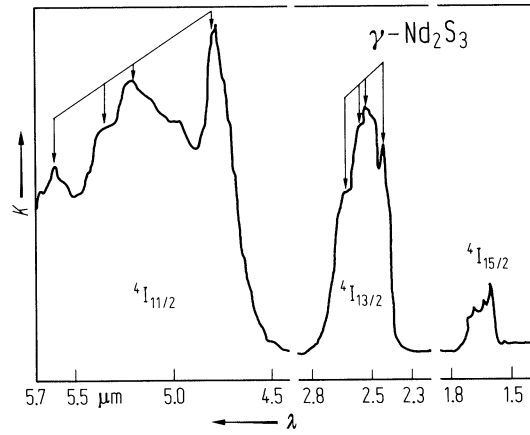
**Fig. 8.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>. Absorption coefficient vs. wavelength at 80 K, uv-region, Nd<sup>3+</sup>-ground state multiplets are indicated [70H].



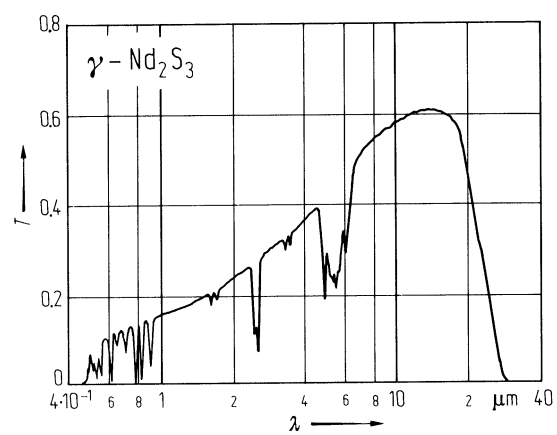
**Fig. 9.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>. Absorption coefficient vs. wavelength at 80K, ir-region, Nd<sup>3+</sup>-ground state multiplets are indicated [70H].



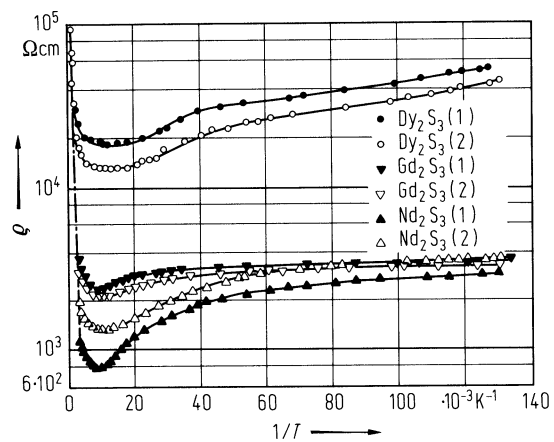
**Fig. 10.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>. Transmission vs. wavelength of a 0.07 mm thick crystal [80S].



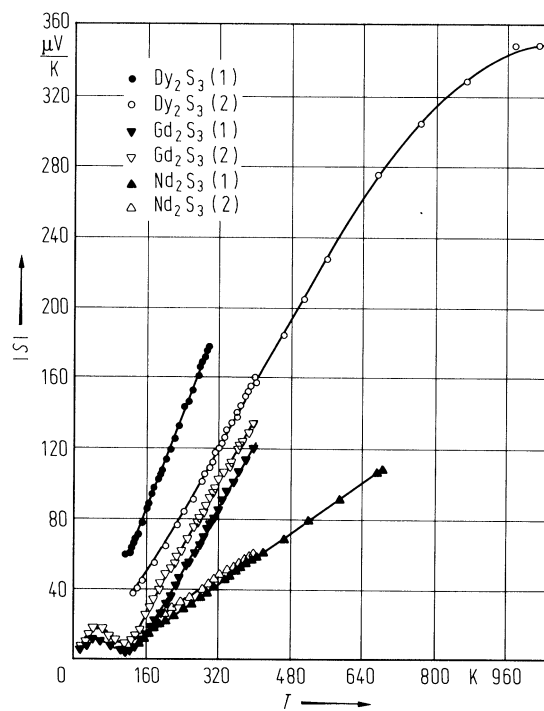
**Fig. 11.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>,  $\gamma$ -Gd<sub>2</sub>S<sub>3</sub>,  $\gamma$ -Dy<sub>2</sub>S<sub>3</sub>. Temperature dependence of the electrical conductivity. The sample compositions are given in Fig. 5 [77T].



**Fig. 12.**

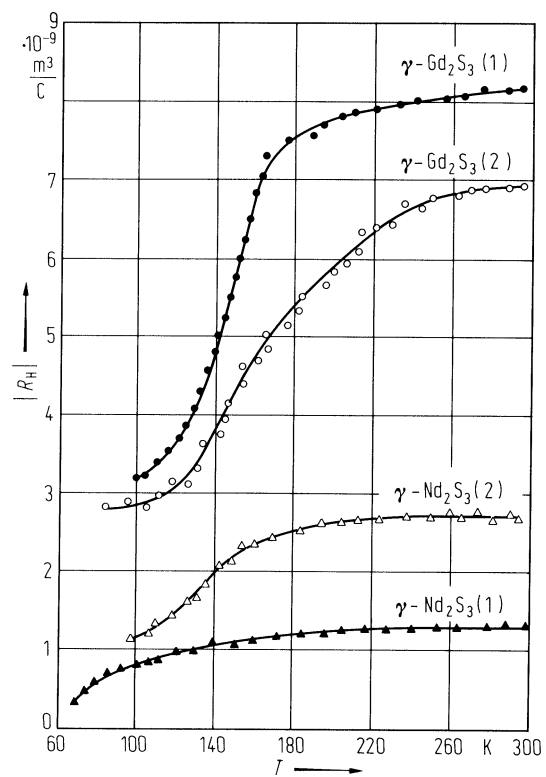
$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>,  $\gamma$ -Gd<sub>2</sub>S<sub>3</sub>,  $\gamma$ -Dy<sub>2</sub>S<sub>3</sub>. Thermoelectric power vs. temperature. The sample compositions are given in Fig. 5 [77T].





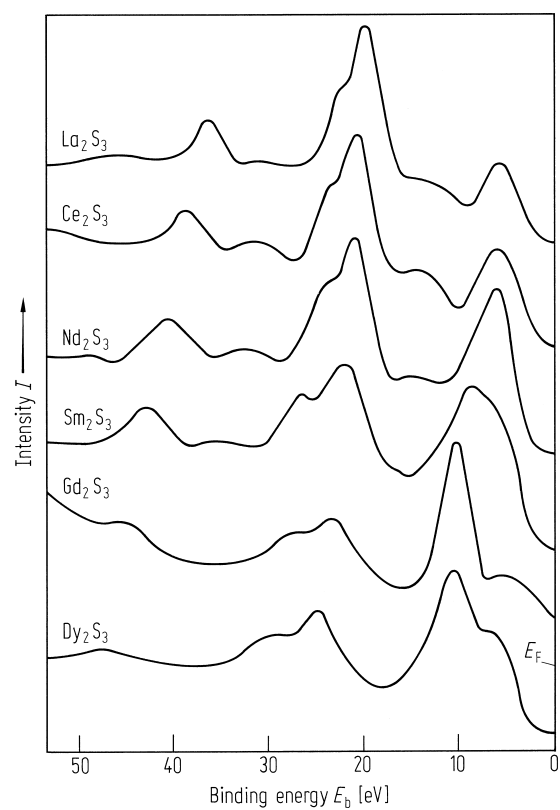
**Fig. 13.**

$\gamma$ -Nd<sub>2</sub>S<sub>3</sub>,  $\gamma$ -Gd<sub>2</sub>S<sub>3</sub>. Absolute value of the Hall coefficient vs. temperature. The excess metal content is as follows: Nd<sub>2</sub>S<sub>3</sub> (1): 0.7% Nd, Nd<sub>2</sub>S<sub>3</sub> (2): 0.65% Nd, Gd<sub>2</sub>S<sub>3</sub> (1): 0.6% Gd, Gd<sub>2</sub>S<sub>3</sub> (2): 0.6% Gd [74T].



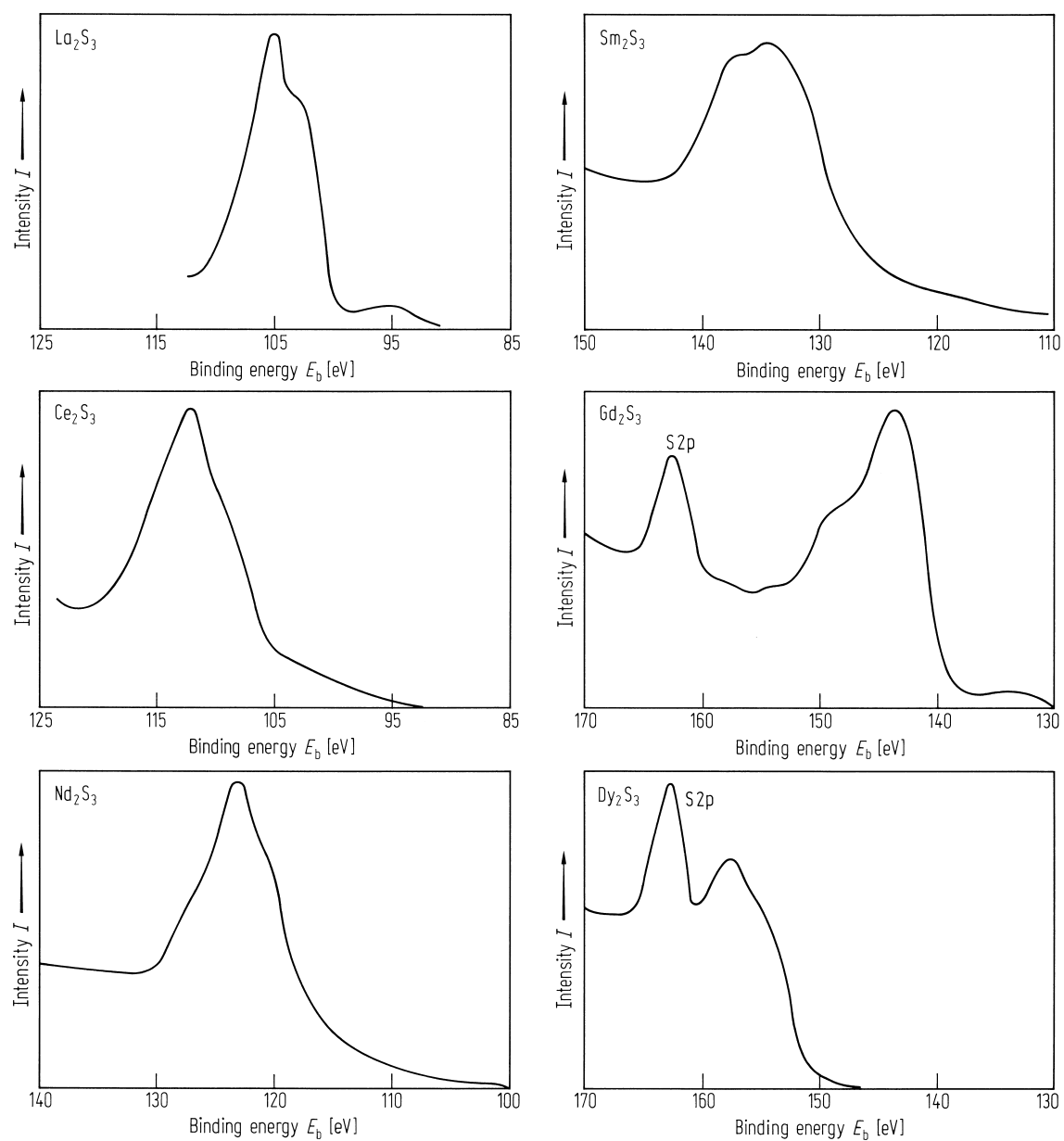
**Fig. 14.**

$\gamma$ -Ln<sub>2</sub>S<sub>3</sub>. MgK<sub>α</sub> X-ray photoelectron spectra of the rare-earth sesquisulfides (Ln = La, Ce, Nd, Sm, Gd, Dy) in the energy region below Fermi level down to Ln 5s core level [91K].



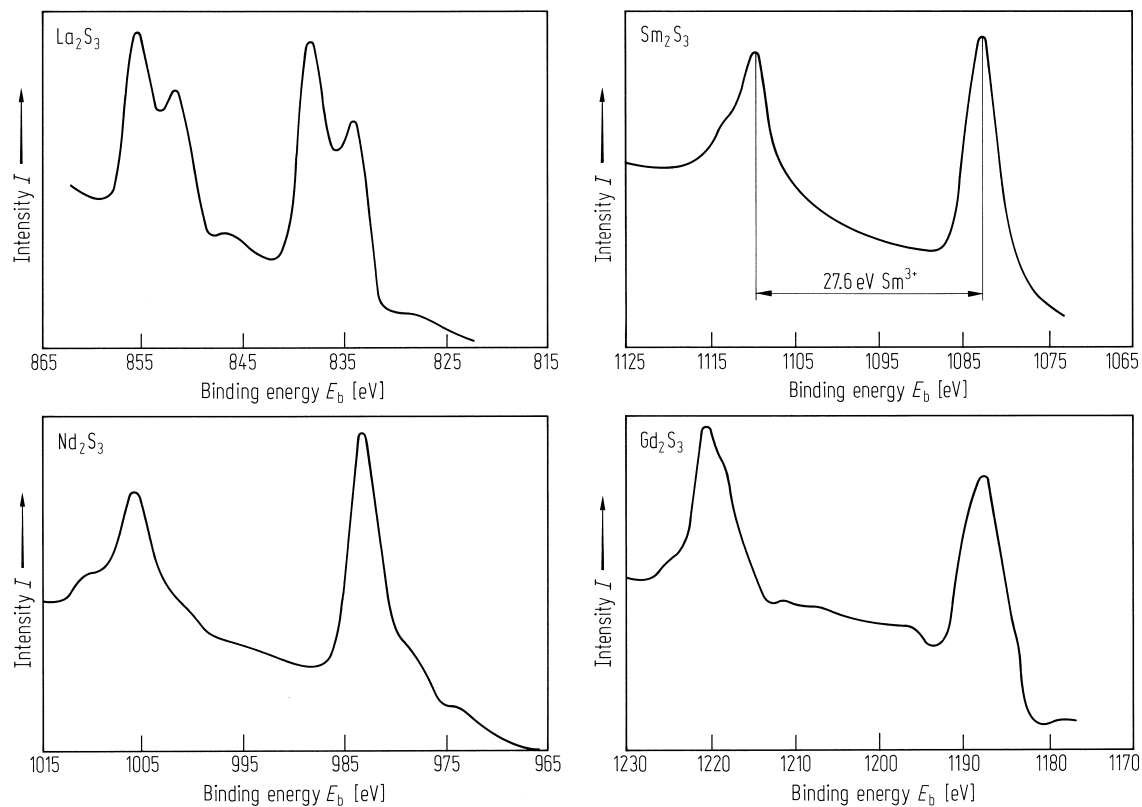
**Fig. 15.**

$\gamma$ - $\text{Ln}_2\text{S}_3$ .  $\text{MgK}_{\alpha}$  X-ray photoelectron spectra of the rare-earth sesquisulfides ( $\text{Ln} = \text{La}, \text{Ce}, \text{Nd}, \text{Sm}, \text{Gd}, \text{Dy}$ ) in the 4d core level region [91K].  $E_b$  relative to  $E_F$ .



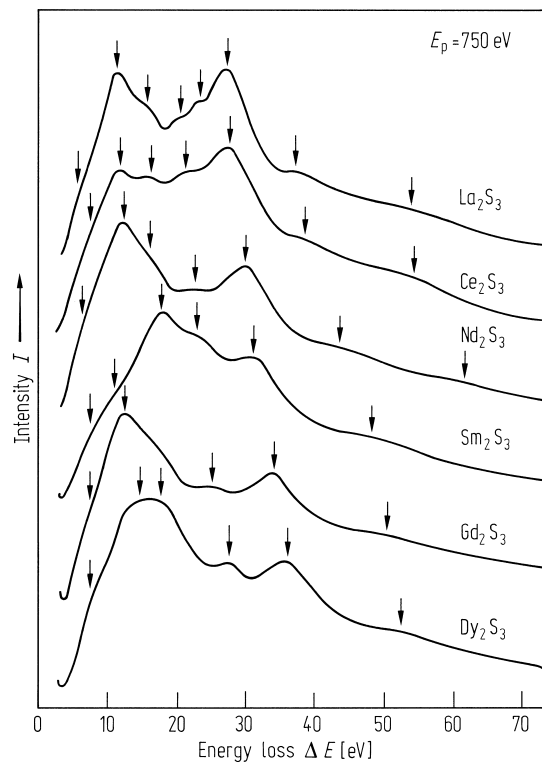
**Fig. 16.**

$\gamma$ - $\text{Ln}_2\text{S}_3$ .  $\text{AlK}_\alpha$  X-ray photoelectron spectra of the rare-earth sesquisulfides ( $\text{Ln} = \text{La, Nd, Sm, Gd}$ ) in the 3d core level region [91K]. The trivalence of Sm is testified by the value of the Sm 3d doublet line spin-orbit splitting ( $= 27.6 \text{ eV}$ ).  $E_b$  relative to  $E_F$ .



**Fig. 17.**

$\gamma$ - $\text{Ln}_2\text{S}_3$ . Electron loss spectra of the rare-earth sesquisulfides ( $\text{Ln} = \text{La, Ce, Nd, Sm, Gd, Dy}$ ) for primary electron beam energy  $E_p = 750 \text{ eV}$ . All the peaks, revealed from the second derivative  $d^2N/dE^2$  are indicated by arrows [91K].



**Fig. 18.**

$\gamma$ - $\text{Ln}_2\text{S}_3$ ,  $\text{Ln} = \text{La, Nd, Gd}$ . 4f-4f giant resonance in energy loss spectra at primary electron energy  $E_p = 750$  eV after Shirley background subtraction [91K].

