

substance: chromium sesquioxide (Cr₂O₃)

property: transport mechanism, conductivity

Cr₂O₃ is a p-type semiconductor, but equilibrium with oxygen is only sluggishly established at temperatures below 1500 K, so thermogravimetric analysis has been unable to reveal the nature of the defect. Early work [51H] suggested: $\text{Cr}_{\text{Cr}} \rightarrow \text{V}_{\text{Cr}}''' + \text{Cr}_i^{2\cdot} + e^+$ with substantial mobility for the interstitial $\text{Cr}_i^{2\cdot}$ only found at high temperatures [61H]. More complex defect structures have been proposed to account for the oxygen pressure dependence [72M]. Ionic transport numbers remain less than 0.1% even at 1800 K [61H]; ionic diffusion coefficient of vacancies is $\approx 2 \cdot 10^{-7} \text{ cm}^2 \text{ s}^{-1}$ at 1500 K with an activation energy of 0.25 eV [72M].

conductivity

At high temperatures (1150...1400°C), the conductivity becomes intrinsic and independent of p_{O_2} [64C, 72M] (Fig. 1). In this region $\sigma \propto \sigma_0 \exp(-U/kT)$ with

U	1.63(1) eV	unoriented sample	64C
σ_0	$1.48(10) 10^4 \Omega^{-1} \text{ cm}^{-1}$		
U	1.68(1) eV	a -direction	
σ_0	$1.96(16) 10^4 \Omega^{-1} \text{ cm}^{-1}$		
U	1.59 (2) eV	c -direction	
σ_0	$1.37(15) 10^4 \Omega^{-1} \text{ cm}^{-1}$		

At low temperatures ($T < 1100^\circ\text{C}$), an extrinsic region is found with

conductivity activation energies

E_A	0.26 eV	$T = 670 \dots 1370 \text{ K}$	see Fig. 2; hot pressed sample, $\sigma \propto CT^{-3/2} \exp(-0.26[\text{eV}]/kT)$	72M
	0.6 eV	$T = 400 \dots 1000 \text{ K}$	ceramic, $\sigma = 3.8 \cdot 10^{-6} T^{-1} \cdot 2x(1-2x) \exp(-0.6[\text{eV}]/kT)$ [$\Omega^{-1} \text{ cm}^{-1}$] for $\text{Cr}_2\text{O}_{3+x}$	69C
	0.41 eV	$T = 470 \dots 970 \text{ K}$	ceramic, from ac conductivity at 1 kHz	71R
	0.4 eV	RT		67V
	0.519 eV	$T = 370 \text{ K}$	polycrystalline sample	74C

In the low T range the defects created at high temperatures, presumably [V_{Cr}'''], are "frozen in" and the thermopower is constant [72M], or varies only slightly [69C] with T and p_{O_2} . The variation of σ with p_{O_2} is too small to be explained in terms of any simple defect equilibria. ac conductivity has an activation energy of 0.9 eV for $T = 300^\circ\text{C}$, $f = 100 \dots 10^6 \text{ Hz}$ [77R].

influence of non-stoichiometry: Fig. 3 (conductivity), Fig. 4 (Seebeck coefficient).

References:

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Fig. 1.

Cr_2O_3 . Conductivity vs. (reciprocal) temperature for unoriented single crystal in the intrinsic region for two values of p_{O_2} [64C].

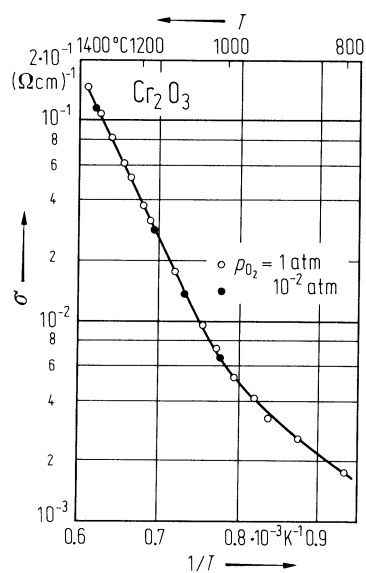


Fig. 2.

Cr_2O_3 . Conductivity (times $T^{3/2}$) vs. (reciprocal) temperature for hot-pressed ceramic samples under oxygen and argon atmosphere, respectively [72M].

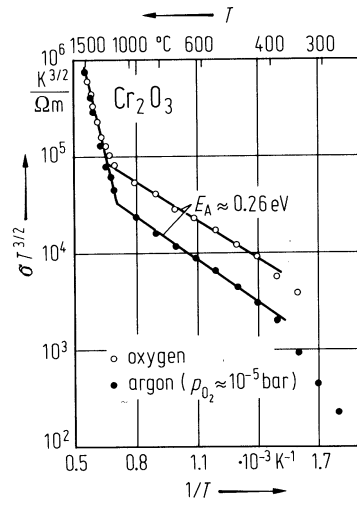


Fig. 3.

$\text{Cr}_2\text{O}_{3+\delta}$. Conductivity (times temperature) vs. reciprocal temperature for cold-pressed ceramic samples of different stoichiometry. $\delta = 0.06$ (A), 0.02 (B), 0.01 (C), 0.004 (D) [69C].

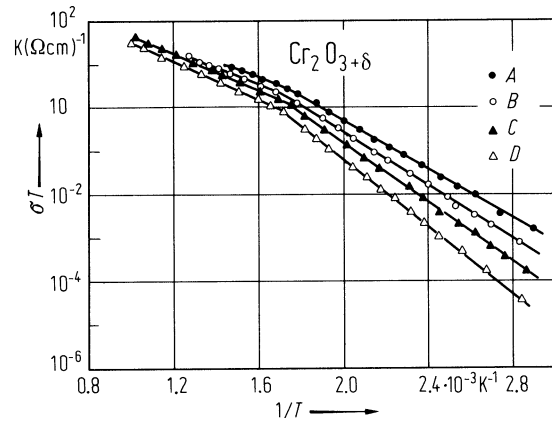


Fig. 4.

$\text{Cr}_2\text{O}_{3+\delta}$. Seebeck coefficient vs. temperature for the cold-pressed ceramic samples of Fig. 3 [69C].

