

substance: WO₃

property: band structure, energy gap

band structure: No calculation of monoclinic or triclinic WO₃ has been reported but a calculation of cubic NaWO₃ and idealized cubic WO₃ [77K] gives the band structure of Fig. 1. The NaWO₃ density of states is shown in Fig. 2. Strong covalent interaction is evident.

energy gap

(due to O(2p) – W(5d) transitions)

E_g	2.585 eV	$T = 300$ K, $E \parallel [001]$	absorption, see Fig. 3 for absorption coefficient, Fig. 4 for temperature shift (extrapolated)	74S, 76H
dE_g/dT	3.1 eV	$T = 0$ K	in the temperature range 0...700°C	60I
	$-9.0 \cdot 10^{-4}$ eV K ⁻¹	a -polarized light		
	$-6.5 \cdot 10^{-4}$ eV K ⁻¹	c -polarized light		

References:

- 60I Iwai, T.: J. Phys. Soc. Jpn. 15 (1960) 1596.
- 72S Schröder, F. A., Felser, H.: Z. Kristallogr. 135 (1972) 391.
- 74S Salje, E.: J. Appl. Crystallogr. 7 (1974) 615.
- 76H Hollinger, G., Trinh Minh Duc, Deneuveville, A.: Phys. Rev. Lett. 37 (1976) 1564.
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Fig. 1.

WO₃. Energy bands of ideal cubic WO₃ along the major symmetry axes [77K].

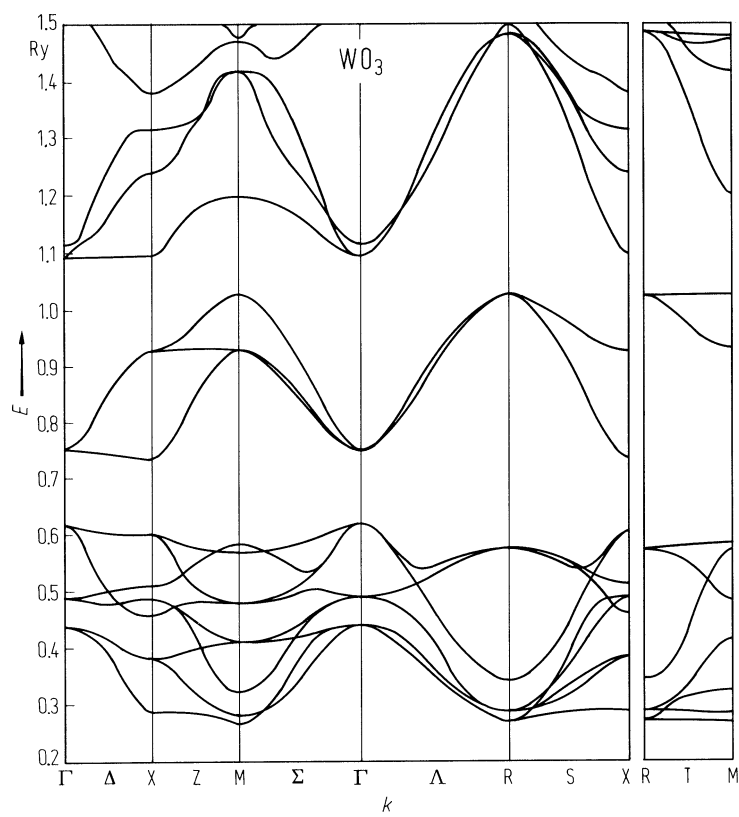


Fig. 2.

NaWO_3 . Orbital density of states together with their sum (a) and total density of state (b) [77K].

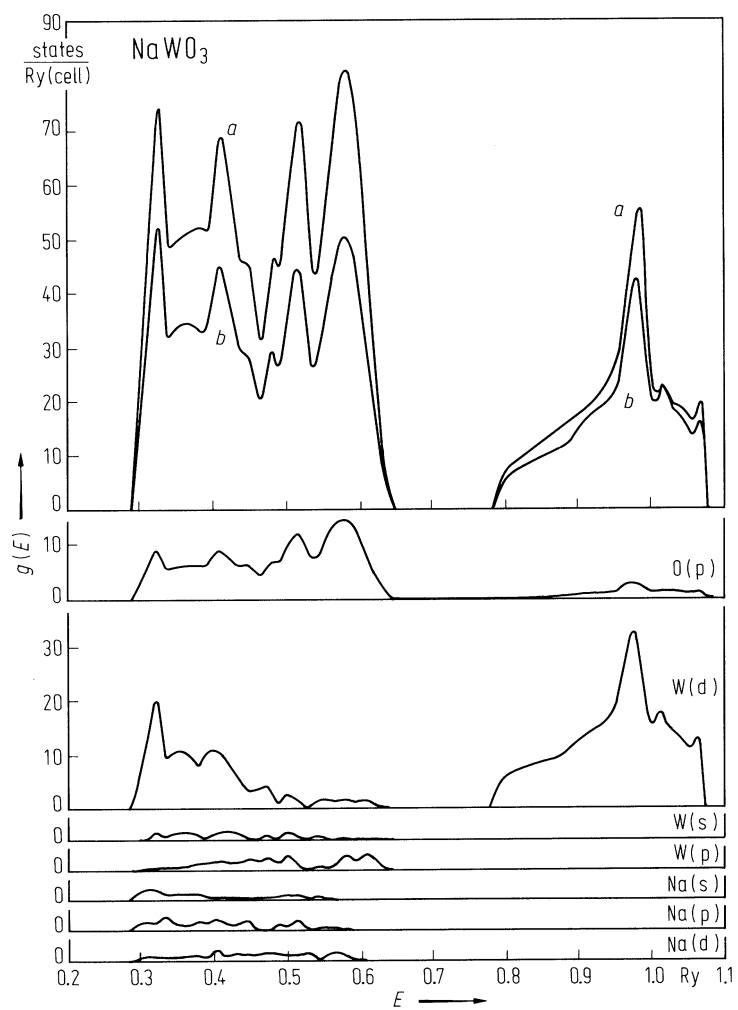


Fig. 3.

WO₃. Transmission vs. wavelength in [001]-direction at RT [72S].

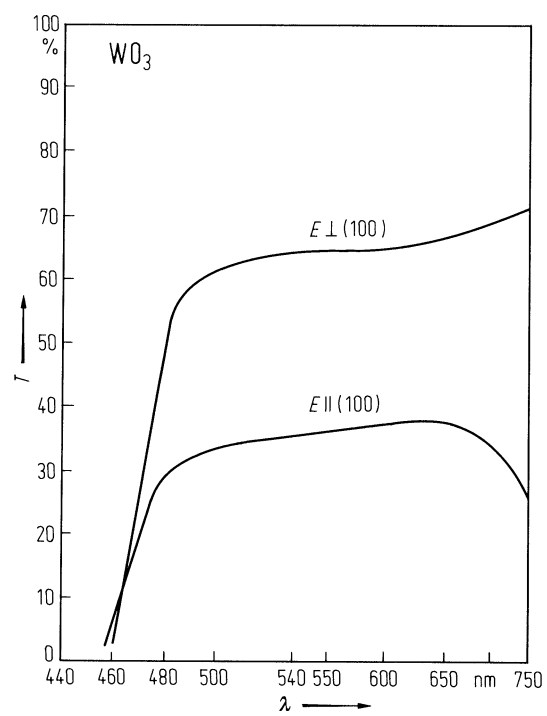


Fig. 4.

WO₃. Optical absorption edge vs. temperature in the *b* plane (a), in the *c* plane (b), in the *c* plane (c) at low temperature [60I].

