

substance: MoTe_{2-x}

property: crystal structure, physical properties

2H-MoTe_{2-x} (0.01 ≤ x ≤ 0.1)

(S: structure (space group), CG: crystal growth (the numbers in parentheses correspond to T_1 and T_2 , the temperatures (in °C) of the hot and cold end of the crystal growth tube, respectively)).

(The references in the last column refer to all data of this document)

lattice parameters

a	3.519 Å	hexagonal	S: semiconductor-metal	66B,
c	13.964 Å	modification	transition at T_{tr} , where	70V,
a	6.33 Å	monoclinic	$T_{tr} = 820^\circ\text{C}$ for Te-rich	70W,
b	3.469 Å	modification	MoTe _{2-x} and $T_{tr} = 880^\circ\text{C}$	72A,
c	13.86 Å		for Mo-rich MoTe _{2-x} .	72B,
β	93°55'		$T < T_{tr}$: hexagonal,	72D,
			$D_{6h}^4 - P6_3/mmc$	75G,
			$T > T_{tr}$: monoclinic, $C_{2h}^2 - P2_1/m$	79C
			CG: halogen transport (900/700);	
			high-temperature form	
			halogen transport (1000/900)	
			and water quenched	

resistivity, Seebeck coefficient, Hall coefficient, electron concentration

ρ_{\perp}	0.5...8.5 Ω cm	n-type,
S_{\perp}	- 422 μV K ⁻¹	synthetic
$R_H (B \parallel c)$	- 24.17 cm ³ /C	single
n	2.6·10 ¹⁷ cm ⁻³	crystal

energy gap

E_g	1.1 eV		optical gap
	1.2 eV		direct energy gap, calculated
$E_{g,th}$	1.08 eV	$T > 660 \text{ K}$	

Figures to this document:

band structure: Fig. 1

Brillouin zone: Fig. 2

electrical conductivity, resistivity: Figs. 3, 5

Seebeck coefficient: Figs. 4, 5

References:

- 66B Brown, B. E.: Acta Crystallogr. 20 (1966) 268.
- 70V Vellinga, M. B., de Jonge, R., Haas, C.: J. Solid State Chem. 2 (1970) 299.
- 70W Wieting, T. J.: J. Phys. Chem. Solids 31 (1970) 2148.
- 72A Al-Hilli, A. A., Evans, B. L.: J. Cryst. Growth 15 (1972) 93.
- 72B Bromley, R. A., Murray, R. B., Yoffe, A. D.: J. Phys. CS (1972) 759.
- 72D Davey, B., Evans, H. L.: Phys. Status. Solidi (a) 13 (1972) 483.
- 75G Grant, A. J., Griffiths, T. M., Pitt, G. D., Yoffe, A. D.: J. Phys. C 8 (1975) L17.
- 79C Conan, A., Delaunay, D., Bonnet, A., Moustafa, A. G., Spiesser, M.: Phys. Status. Solidi (b) 94 (1979) 279.

Fig. 1.

MoTe₂. Band structure [72B]; see Fig. 2 for Brillouin zone, and for the meaning of P and Q. 1 Ry = 13.6 eV.

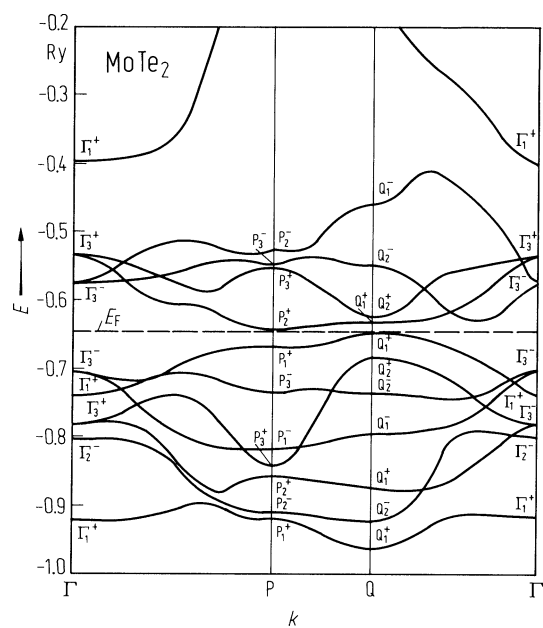


Fig. 2.

MoSe₂. (a) Band structure and (b) Brillouin zone for 2H-MoSe₂ [72B]. Q and P are the middle of each edge and the vertex of the hexagon in the two-dimensional approximation used for the calculations.

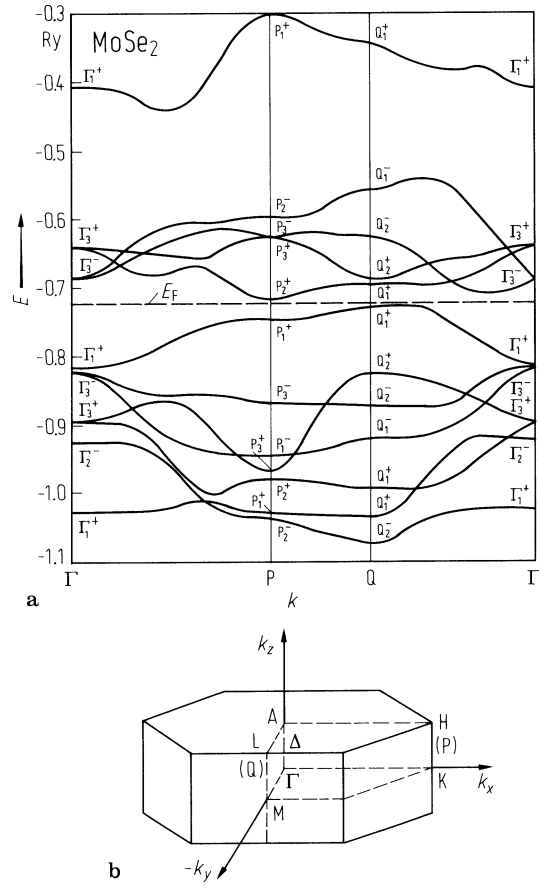


Fig. 3.

MoTe₂. Electrical conductivity vs. reciprocal temperature for 2H-modification. Theoretical results are drawn in full lines. 1: monocrystalline sample, 2: polycrystalline sample [79C]. σ in $\Omega^{-1} \text{ cm}^{-1}$, $\sigma \perp c$ (curve 1).

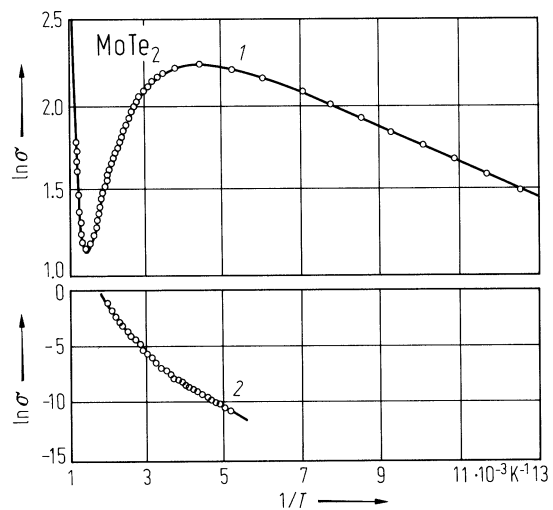


Fig. 4.

MoTe₂. Seebeck coefficient vs. reciprocal temperature for 2H-modification. 1: monocrystalline sample, 2: polycrystalline sample [79C]. $S \perp c$ (curve 1).

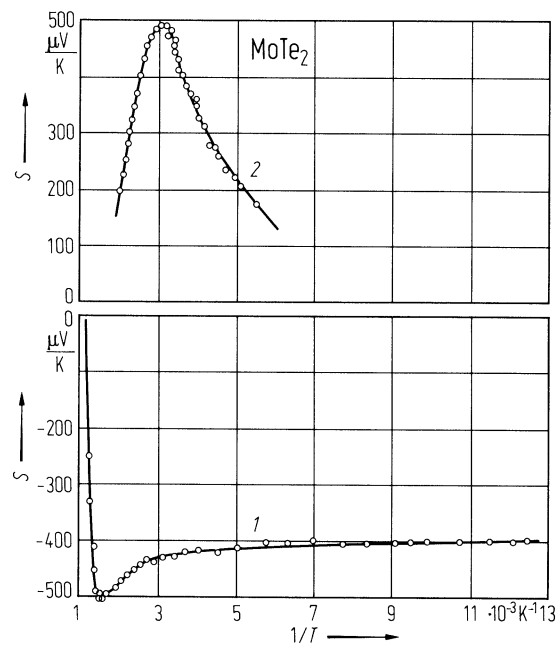


Fig. 5.

MoTe₂. Resistivity and thermoelectric power vs. temperature for metastable β -MoTe₂. The rapid increase of the resistivity at 750 K corresponds to a transition to the α -phase [70V]. Pressed-powder bar.

