

Fig. 1A-10-001. BaTiO₃. Pseudocubic cells of the phases I, II, III and IV [66Car]. The arrows show the direction of P_s . The thin lines show the original cubic cell and the bold lines show the pseudocubic cell.

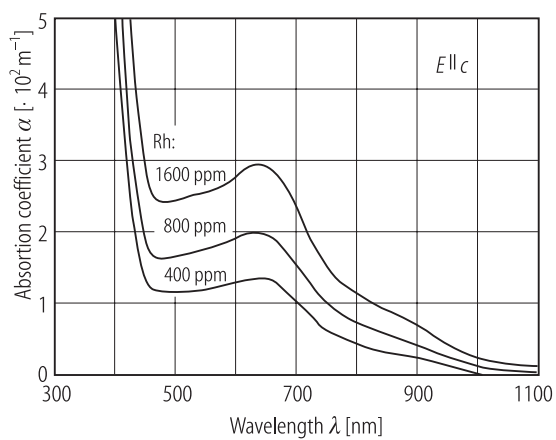


Fig. 1A-10-120. BaTiO₃ (Rh doped). α vs. λ at RT [94Wec]. α : optical absorption coefficient. Polarized light: E parallel to the c -axis. Parameter: Rh concentration (with respect to Rh/Ba ratio in the melt). The crystal was grown by the top-seeded solution growth method.

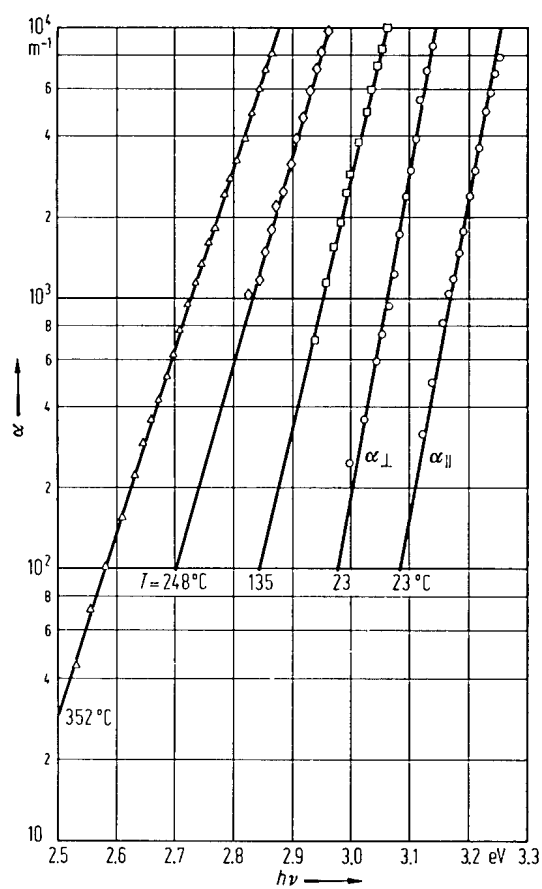


Fig. 1A-10-121. BaTiO₃. α vs. $h\nu$ [70Wem]. Parameter: T . Below $\Theta_f = 132\text{ }^{\circ}\text{C}$, two absorption coefficients α_{\parallel} and α_{\perp} correspond to light polarized parallel and perpendicular to the tetragonal c -axis, respectively. The crystal was grown by the top-seeded solution technique.

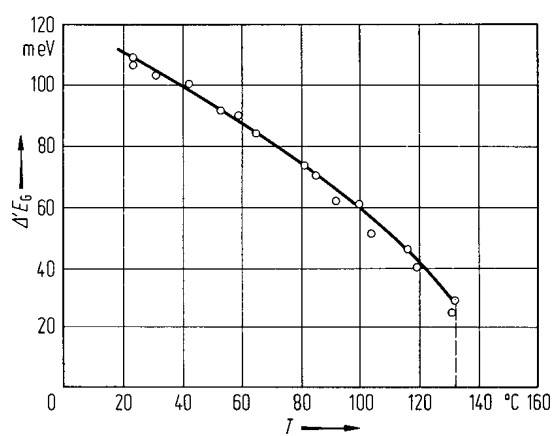


Fig. 1A-10-122. BaTiO_3 , $\Delta'E_G$ vs. T [70Wem]. $\Delta'E_G$: energy separation between the interband absorption edges for light polarized parallel and perpendicular to the tetragonal c -axis. The crystal was grown by the top-seeded solution technique. $\Theta_t = 132$ $^\circ\text{C}$.

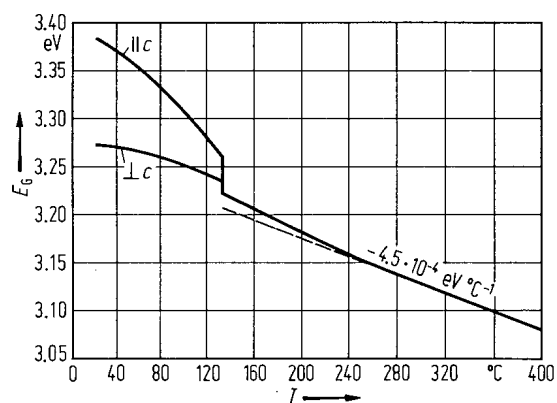


Fig. 1A-10-123. BaTiO₃. E_G vs. T [70Wem]. E_G : band gap energy estimated from extrapolation of the Urbach edge to the absorption coefficient $\alpha = 3 \cdot 10^5 \text{ m}^{-1}$. The crystal was grown by the top-seeded solution technique. $\Theta_t = 132 \text{ }^\circ\text{C}$. See also [67Gah].

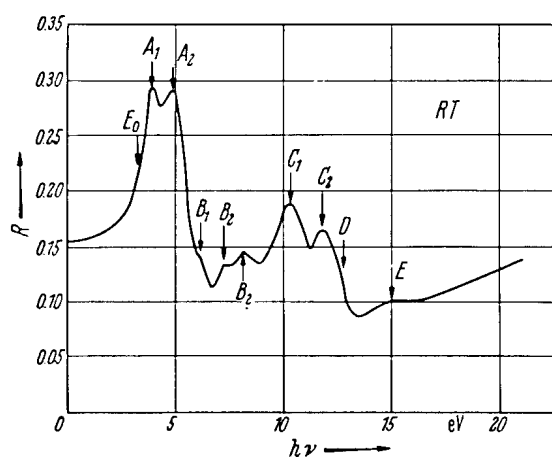


Fig. 1A-10-124. BaTiO_3 . R vs. $h\nu$ [65Car]. R : reflectivity. For energies of fundamental absorption edges E_0 , A etc., see Table 1A-10-022. See also [78Bau].

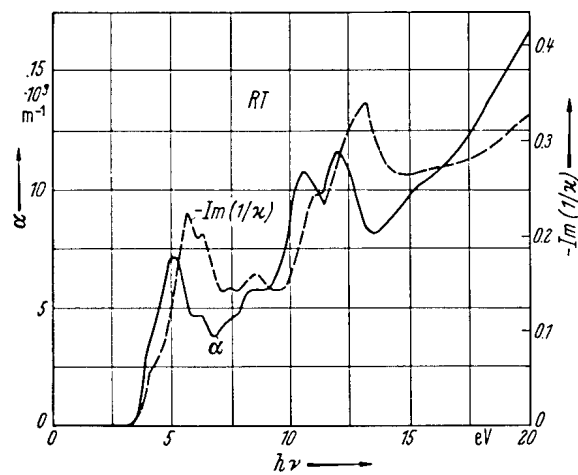


Fig. 1A-10-125. BaTiO₃. α and $-\text{Im}(1/\kappa)$ vs. $h\nu$ [65Car].
 α : absorption coefficient. See also [78Bau].

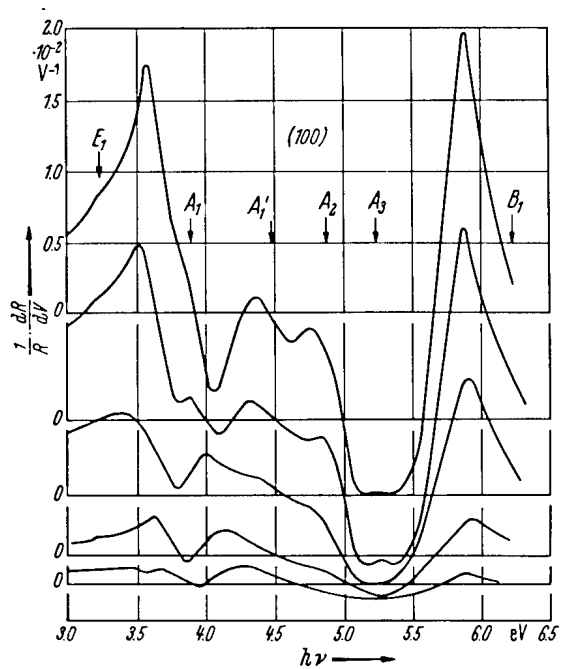


Fig. 1A-10-126. BaTiO₃ (Ta doped). $(1/R) \cdot (dR/dV)$ vs. $h\nu$.
 Parameter: V , increasing from top to bottom [67Fro].
 R : reflectivity. V : applied voltage. For energies of
 fundamental absorption edges, see Table 1A-10-022
 [65Car] in which E_0 corresponds to E_1 of this figure.

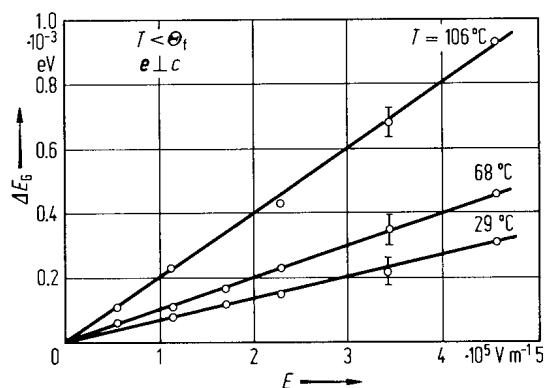


Fig. 1A-10-127. BaTiO₃. ΔE_G vs. E in the ferroelectric state [67Gah]. Parameter: T . ΔE_G : increase of the absorption edge energy E_G (defined at $\alpha = 2 \cdot 10^4$ m⁻¹) due to a dc field along the tetragonal c -axis. e : electric field vector of the incident polarized light. $\Theta_t = 115$ °C.

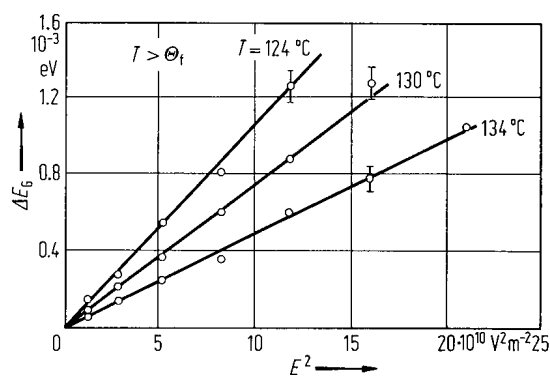


Fig. 1A-10-128. BaTiO₃, ΔE_G vs. E^2 in the paraelectric state [67Gah]. Parameter: T . ΔE_G : increase of the absorption edge energy E_G (defined at $\alpha = 2 \cdot 10^4 \text{ m}^{-1}$) due to a dc field along the tetragonal c -axis. $\Theta_f = 115^\circ\text{C}$.

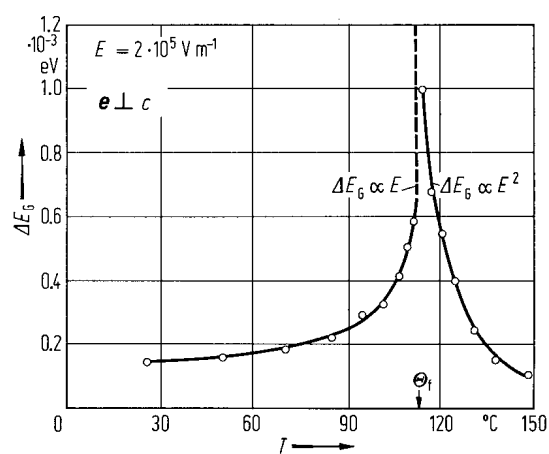


Fig. 1A-10-129. BaTiO₃. ΔE_G vs. T [67Gah]. ΔE_G : increase of the absorption edge energy E_G (defined at $\alpha = 2 \cdot 10^4 \text{ m}^{-1}$) due to a dc field along the tetragonal c -axis. e : electric field vector of the incident polarized light. $\Theta_f = 115^{\circ}\text{C}$.

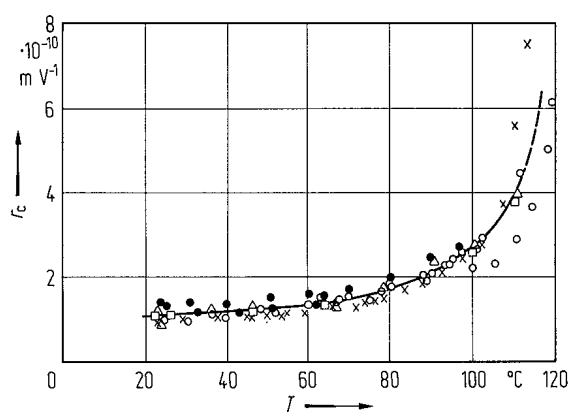


Fig. 1A-10-130. BaTiO_3 . Electrooptic constant r_c vs. T [65Joh2]. $r_c = r_{33} - (n_d/n_c)^3 r_{13}$. $\lambda = 546.1 \text{ nm}$.

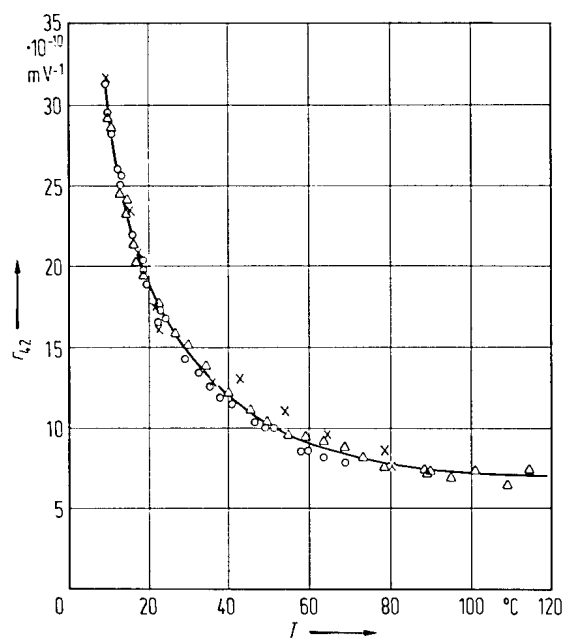


Fig. 1A-10-131. BaTiO₃. Electrooptic constant r_{42} vs. T [65Joh2]. $\lambda = 546.1 \text{ nm}$.

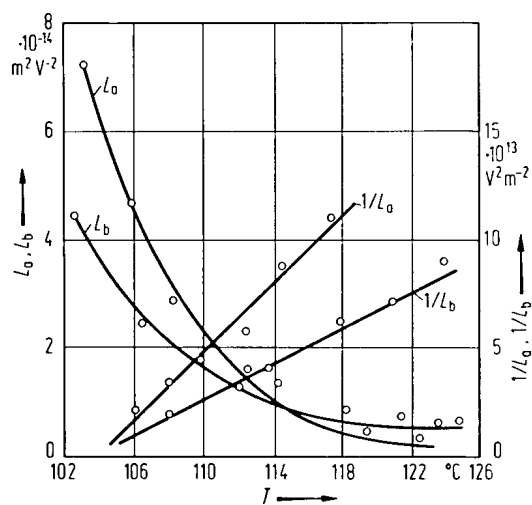


Fig. 1A-10-132. BaTiO₃. L_a , L_b vs. T [69Son].
 $L_a = n_0^3 (L_{11} - L_{12})$, $L_b = n_0^3 L_{44}$. $L_{i\lambda}$: quadratic electrooptic constant for E . $\lambda = 500.0$ nm.

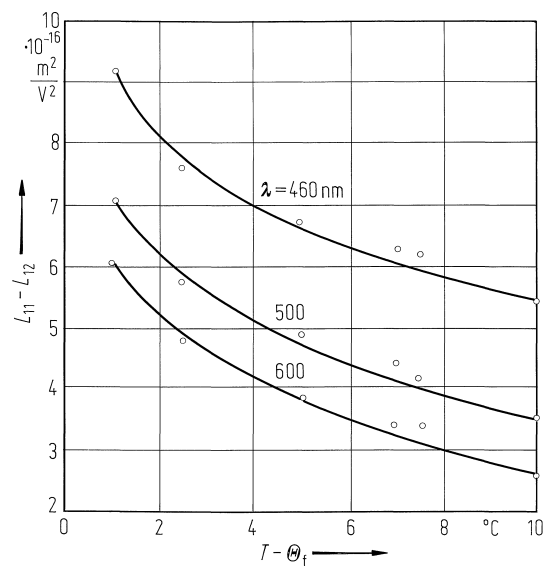


Fig. 1A-10-133. BaTiO₃. $L_{11} - L_{12}$ vs. $T - \Theta_f$ [79Ara].
 Parameter: λ . L_{11} , L_{12} : quadratic electrooptic constant for E .

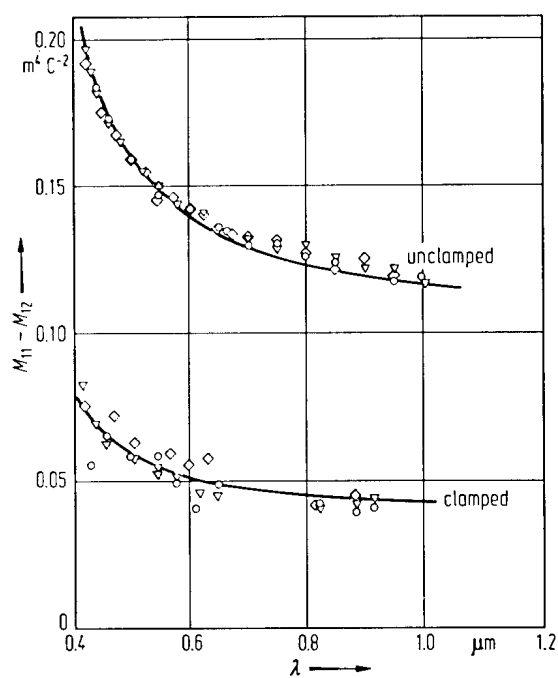


Fig. 1A-10-134. BaTiO₃. $M_{11} - M_{12}$ vs. λ [71Joh].
 $M_{i\lambda}$: quadratic electrooptic constant for P . $T = \Theta_f + 10^\circ\text{C}$.

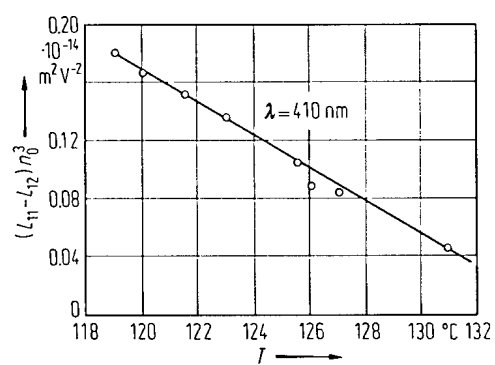


Fig. 1A-10-135. BaTiO₃. $(L_{11} - L_{12}) n_0^3$ vs. T [66Per].

$L_{i\lambda}$: quadratic electrooptic constant for E .

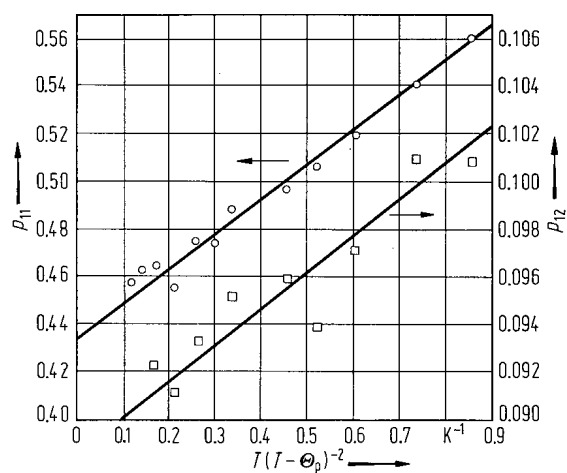


Fig. 1A-10-136. BaTiO_3 , $p_{i\lambda}$ vs. $T(T-\Theta_p)^{-2}$ [70Coh]. $p_{i\lambda}$: piezooptic constants for **S**. The crystal was grown by the top-seeded solution technique. $\Theta_p = 110^\circ\text{C}$, $\Theta_f = 131.2^\circ\text{C}$. $\lambda = 632.8\text{ nm}$.

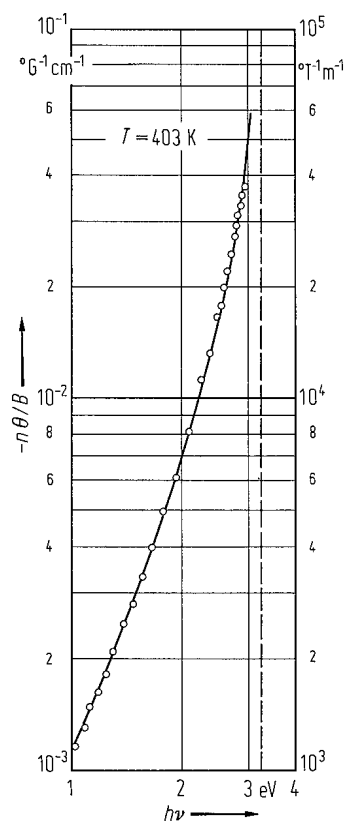


Fig. 1A-10-137. BaTiO₃. $-n\theta/B$ vs. $h\nu$ [67Bae].
 n : refractive index. θ : Faraday rotation per unit length.
 B : magnetic induction. See Table 1A-10-026. Solid line: fit
to dispersion function $F_1(\omega/\omega_b)$. Dashed line: band gap
energy $E_G = 3.25$ eV.

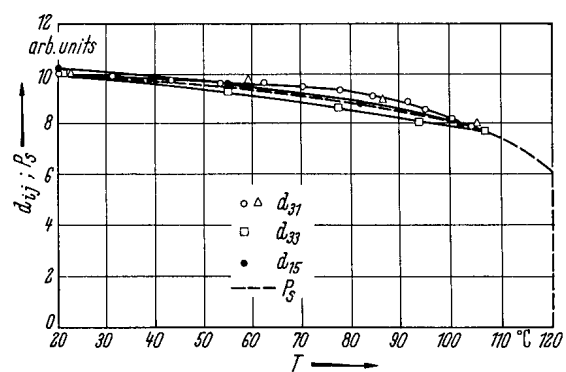


Fig. 1A-10-138. BaTiO₃. d_{ij} , P_s vs. T [64Mil].
 d_{ij} : nonlinear optical susceptibility.

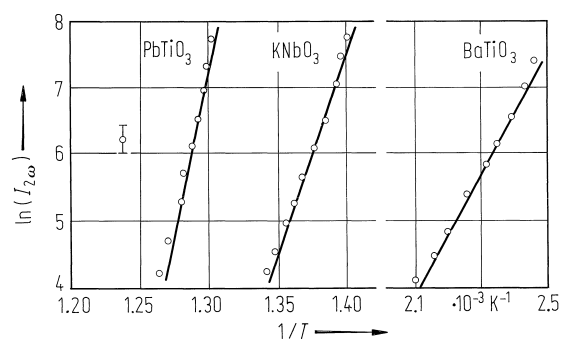


Fig. 1A-10-139. KNbO_3 , BaTiO_3 , PbTiO_3 . $\ln(I_{2\omega})$ vs. $1/T$ [81Lib]. $I_{2\omega}$: optical second harmonic intensity. $\lambda = 1.064 \mu\text{m}$.

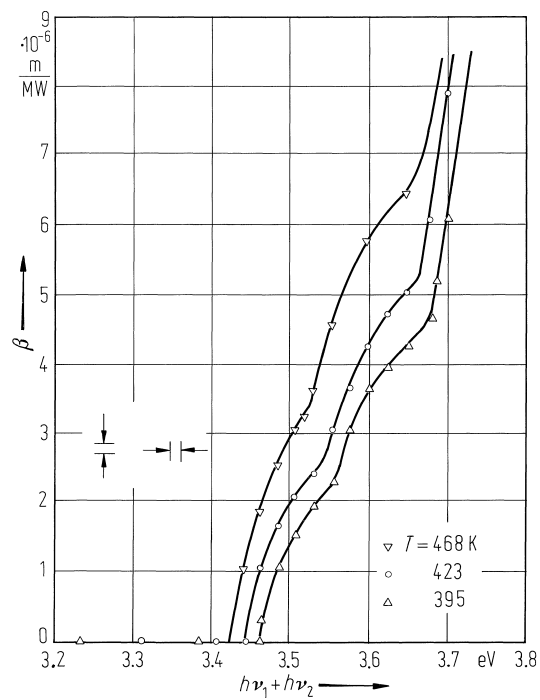


Fig. 1A-10-140. BaTiO_3 . β vs. $h\nu_1 + h\nu_2$ [84Sha].
 Parameter: T . β : two-photon absorption coefficient.
 $h\nu_1 = 1.17 \text{ eV}$.

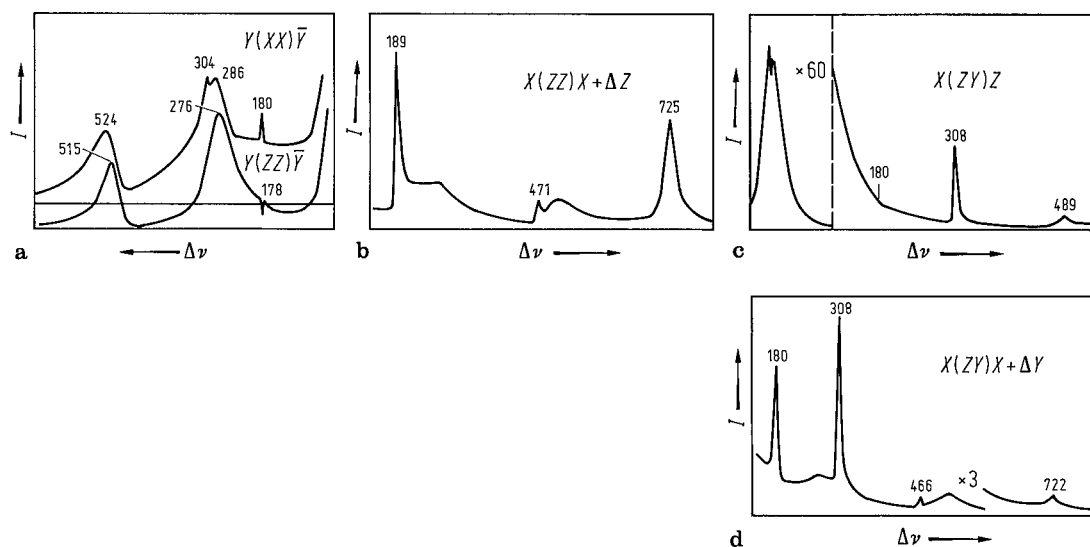


Fig. 1A-10-141. BaTiO₃. I vs. $\Delta\nu$ at RT [77Sca]. I : Raman scattering intensity. $\Delta\nu$: frequency shift. **(a)** A₁(TO) phonons, **(b)** A₁(LO) phonons, **(c)** E(TO) phonons, and **(d)** E(LO) phonons. Polarization configurations are written in the figures. Numbers in the figures are frequency shifts in units of cm⁻¹. 1 cm⁻¹ corresponds to $3 \cdot 10^{10}$ Hz. See also [68DiD].

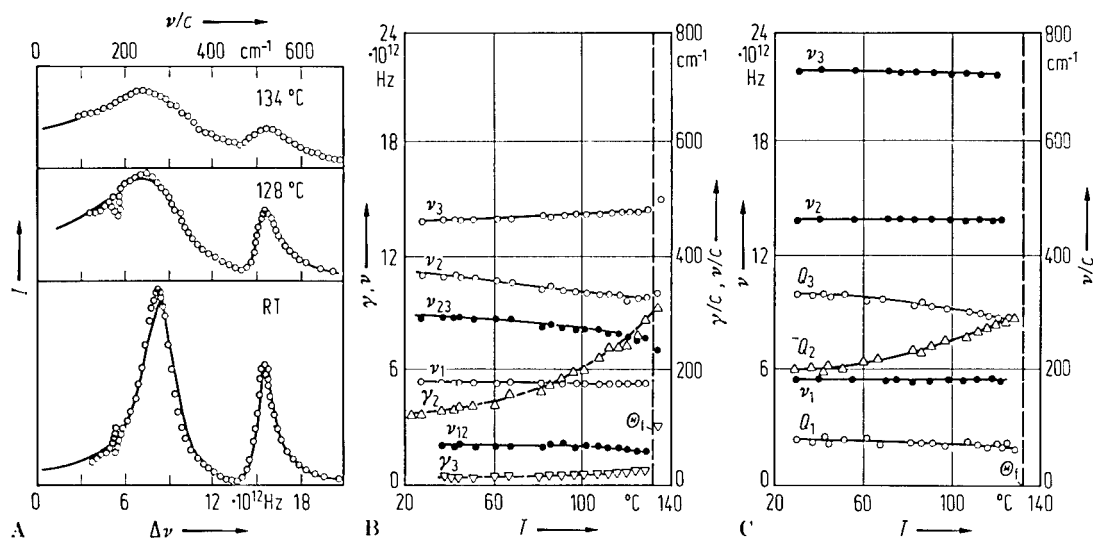


Fig. 1A-10-142. BaTiO₃. Temperature dependence of A₁ phonons [77Sca]. **(A):** I vs. $\Delta\nu$. I : Raman scattering intensity. $\Delta\nu$: frequency shift. Dots are experimental points and solid lines are fit of the data to the coupled oscillator model. **(B):** A₁(TO) mode parameters vs. T . ν_i and γ_i : the frequency and the damping parameters of the i -th mode, respectively. ν_{ij} : the coupling coefficient between i -th and j -th modes. **(C):** A₁(LO) mode parameters vs. T . Q_i is the effective charge of the i -th mode.

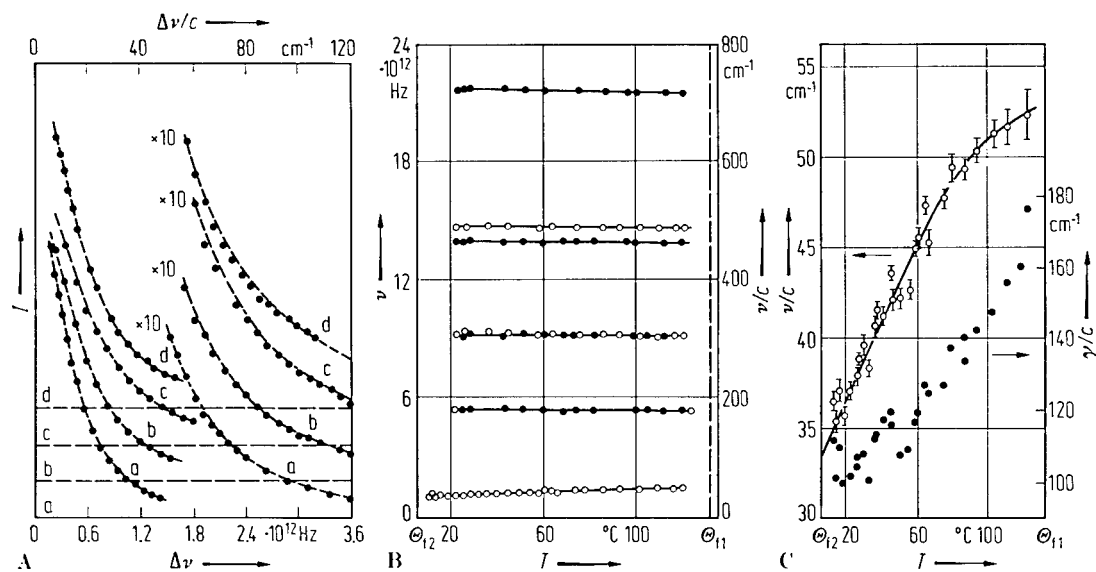


Fig. 1A-10-143. BaTiO₃. Temperature dependence of E mode [77Sca]. **(A):** I vs. $\Delta\nu$. I : Raman scattering intensity. $\Delta\nu$: frequency shift. Parameter: T ; a: 12.7 °C, b: 40.0 °C, c: 78.0 °C, d: 121 °C. The dots are the experimental points and the dashed lines are the fit to the damped oscillator model. Base lines are shown by the broken lines. **(B):** ν vs. T . ν : mode frequency. Open circles: E(TO) mode, full circles: E(LO) mode. **(C):** ν , γ vs. T for the overdamped E mode shown in (A). ν : mode frequency. γ : damping parameter of the model. 1 cm^{-1} is equal to $3 \cdot 10^{10}$ Hz.

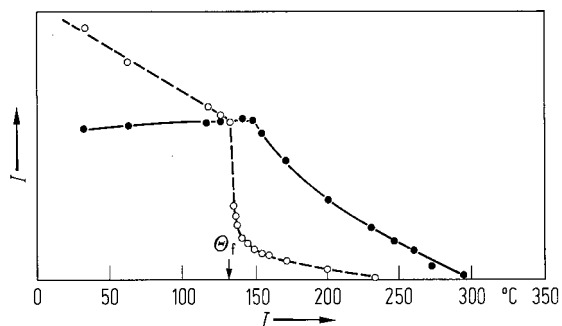


Fig. 1A-10-144. BaTiO₃. I vs. T [72Fon]. I : integrated Raman scattering intensity. The open circles and full circles represent the intensity integrated over the frequency range 10...50 cm⁻¹ and 50...1000 cm⁻¹, respectively. The crystal was grown by the top-seeded solution method.

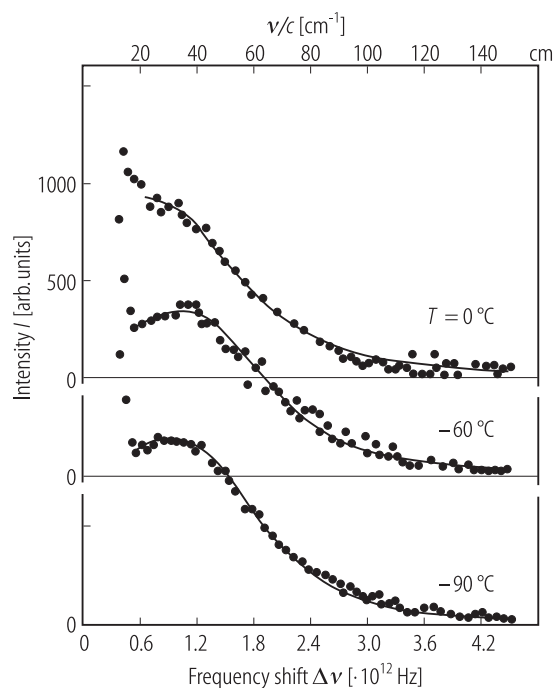


Fig. 1A-10-145. BaTiO₃. I vs. $\Delta\nu$ in the phase III [90Laa].
 I : Raman scattering intensity for the lowest frequency $B_2(\text{TO})$ mode. $\Delta\nu$: frequency shift. Parameter: T .

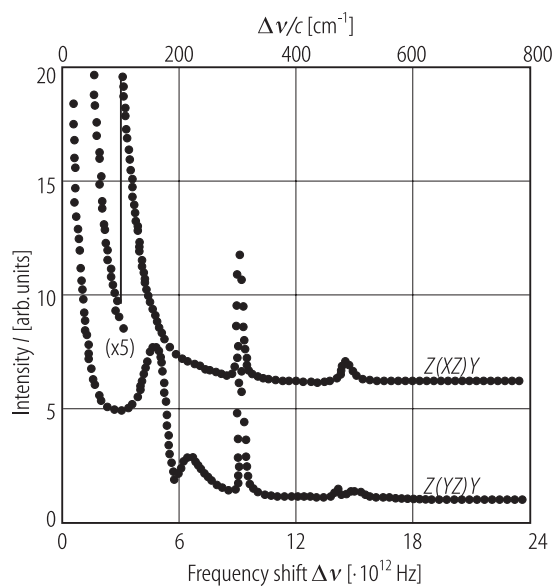


Fig. 1A-10-146. BaTiO₃. I vs. $\Delta\nu$ [94Fon]. I : Raman scattering intensity of single-domain crystal at 25 °C. $\Delta\nu$: frequency shift.

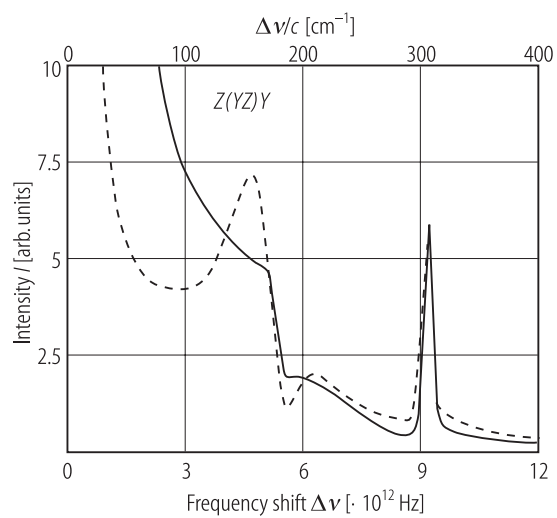


Fig. 1A-10-147. BaTiO₃. I vs. $\Delta\nu$ [94Fon]. I : Raman scattering intensity of the lowest mode at 25 °C (solid curve) and at 120 °C (broken curve). $\Delta\nu$: frequency shift.

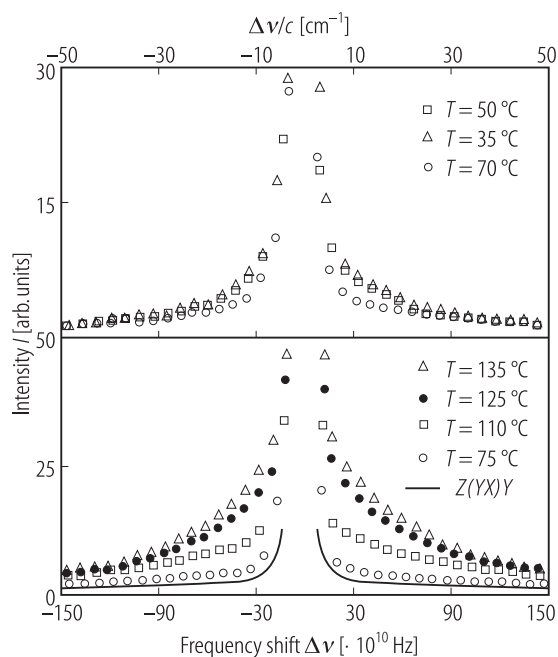


Fig. 1A-10-148. BaTiO_3 . I vs. $\Delta\nu$ [94Fon]. Parameter: T . I : scattering intensity of the central-peak in the $Z(YZ)Y$ geometry. $\Delta\nu$: frequency shift. Intensity measured in $Z(YX)Y$ geometry is shown by solid lines as a zero level reference for the pure elastic scattering.

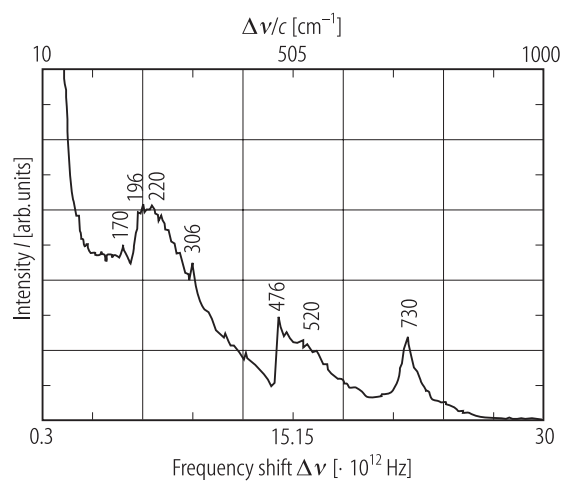


Fig. 1A-10-149. BaTiO₃ (ceramics). *I* vs. $\Delta\nu$ [94Zha1].
I: Raman scattering intensity at RT. $\Delta\nu$: frequency shift.

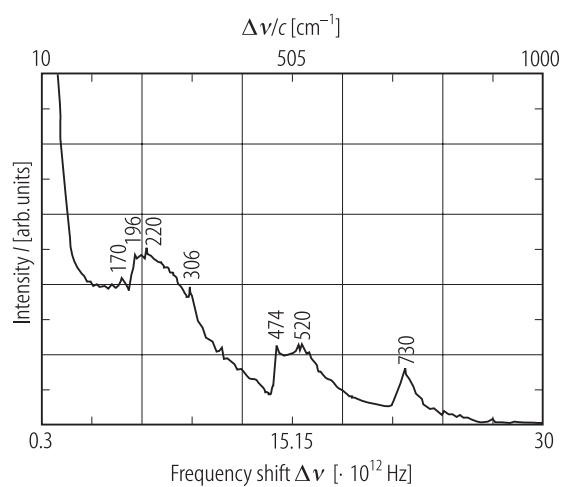


Fig. 1A-10-150. BaTiO₃ (reduced ceramics). I vs. $\Delta\nu$ [94Zha1]. I : Raman scattering intensity at RT. $\Delta\nu$: frequency shift.

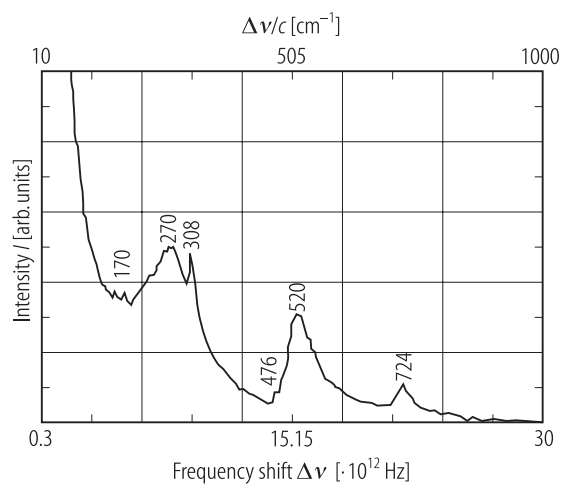


Fig. 1A-10-151. BaTiO₃ (Cu-doped ceramics). I vs. $\Delta\nu$ [94Zha1]. I : Raman scattering intensity at RT. $\Delta\nu$: frequency shift.

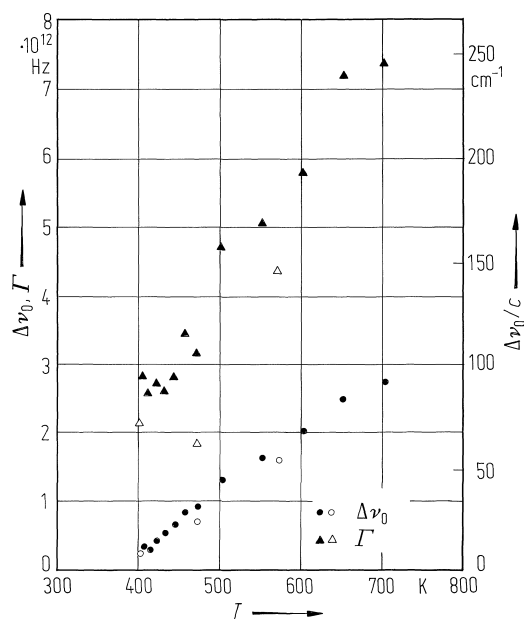


Fig. 1A-10-152. BaTiO_3 . $\Delta\nu_0$, Γ vs. T [83Pre]. $\Delta\nu_0$, Γ : frequency and damping constant of the soft mode in the phase I obtained by hyper-Raman scattering, respectively. Full symbols: from [82Vog], open symbols: from [83Ino].

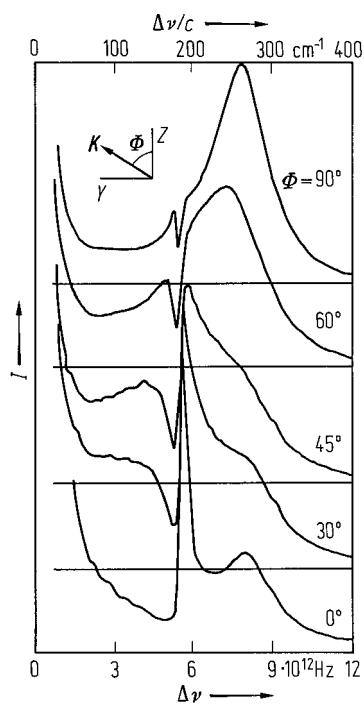


Fig. 1A-10-153. BaTiO₃. I vs. $\Delta\nu$ for the low frequency oblique polariton at RT [77Sca]. I : Raman scattering intensity. $\Delta\nu$: frequency shift. Scattering geometry is $X(ZZ)X + \Delta Z$. Scattering angle θ is 8° for all spectra. Parameter: Φ , the angle between Z axis and the propagation direction of the phonon as shown in the figure. See also [68Rou].

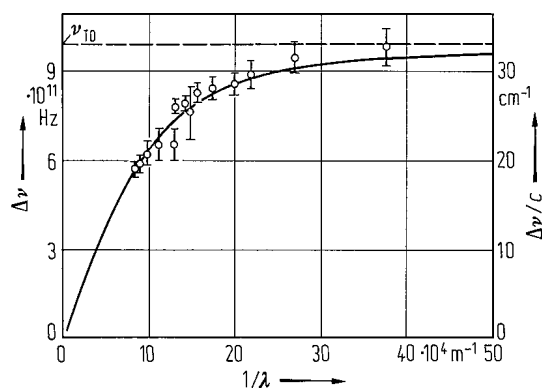


Fig. 1A-10-154. BaTiO₃. Polariton dispersion of the overdamped E-mode at RT. $\Delta\nu$ vs. $1/\lambda$ [74Hei]. $\Delta\nu$: Raman frequency shift. $1/\lambda$: wave number of the polariton. The solid curve is the theoretical fit.

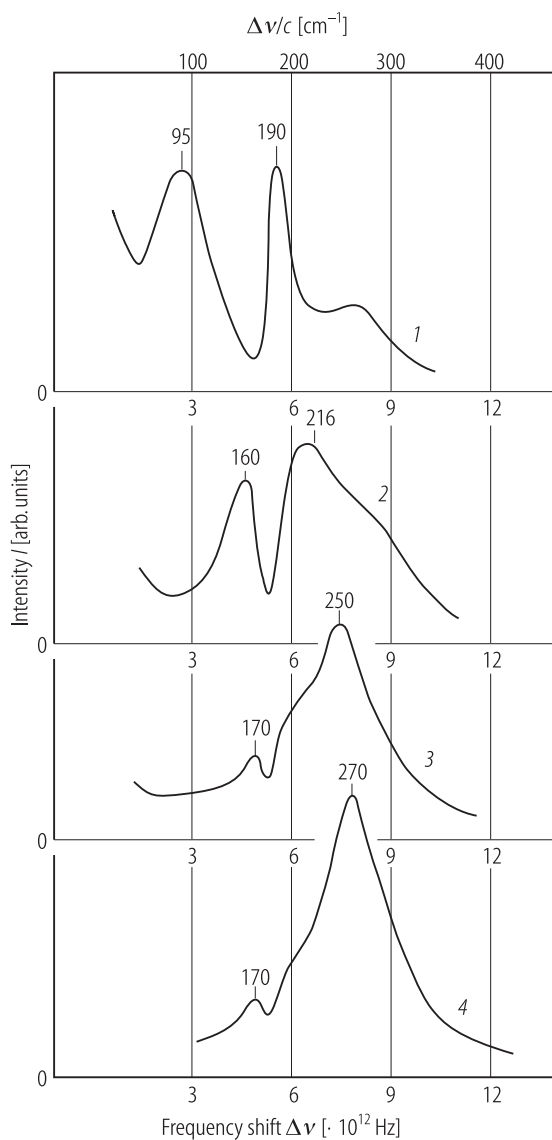


Fig. 1A-10-155. BaTiO₃. I vs. $\Delta\nu$ [95Yar]. I : Raman scattering intensity of polariton at RT. $\Delta\nu$: frequency shift. Parameter: scattering angle θ . Curve 1: $\theta = 0.85^\circ$, 2: $\theta = 2^\circ$, 3: $\theta = 4^\circ$, 4: $\theta = 8^\circ$.

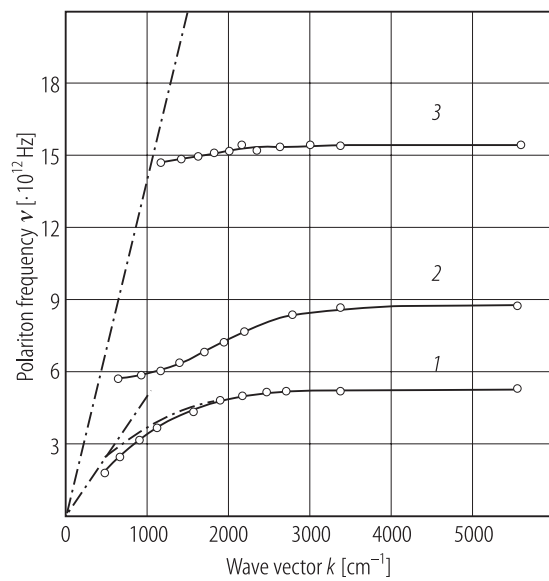


Fig. 1A-10-156. BaTiO_3 . ν vs. k [95Yar]. ν : polariton frequency, k : wave vector. Parameter: scattering angle θ . Curve 1: $\theta = 0.85^\circ$, 2: $\theta = 2^\circ$, 3: $\theta = 4^\circ$.

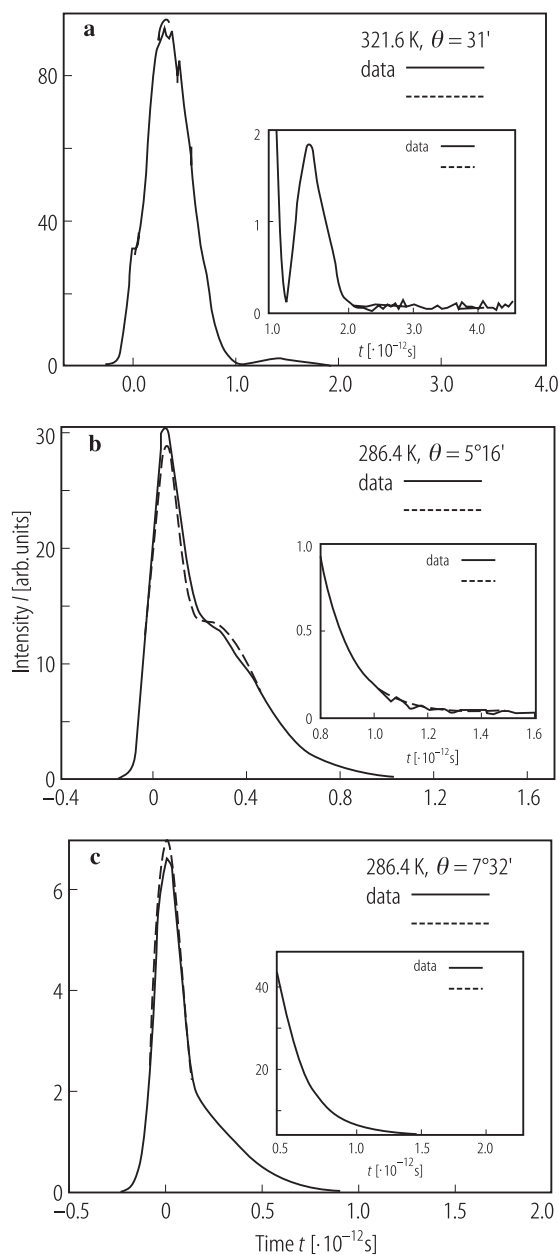


Fig. 1A-10-157. BaTiO₃. I vs. t at 321.6 K and 286.4 K at several crossing angles of the excitation pulses [94Dou]. $I(t)$: intensity of induced stimulated Raman scattering from E mode. t : time.

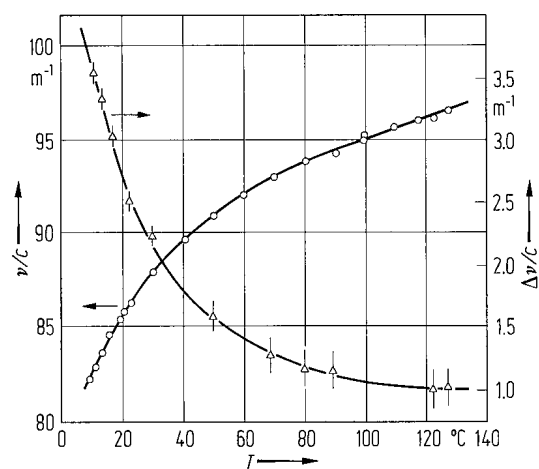


Fig. 1A-10-158. BaTiO_3 . ν/c and $\Delta\nu/c$ vs. T [71Lax]. ν/c : Brillouin frequency. $\Delta\nu/c$: Brillouin linewidth. The Fabry-Perot instrumental linewidth is 2.3 m^{-1} . The crystal was grown by the top-seeded solution technique.

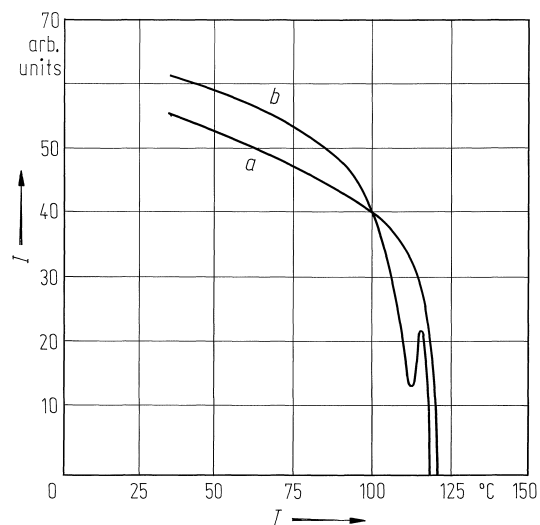


Fig. 1A-10-159. BaTiO₃, I vs. T [85Aga]. I : intensity of hyper Rayleigh scattering, $\lambda = 578.2$ nm. Curve a : sample without specially introduced impurity, b : 0.1% Bi₂O₃ doped crystal.

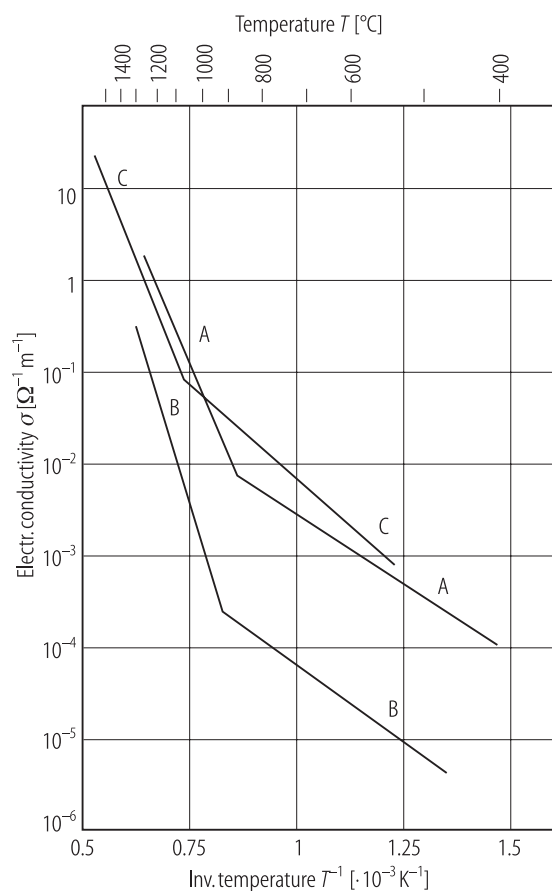


Fig. 1A-10-160. BaTiO_3 . σ vs. $1/T$ [88Jan]. σ : electrical conductivity. Parameter: three different samples, A, B and C. A and B were measured in argon ambience, C in open air. The samples were single crystals grown by the top-seeded solution method. The curves exhibit two distinct linear parts with different activation energies of 0.65 eV and 2.6 eV on average.

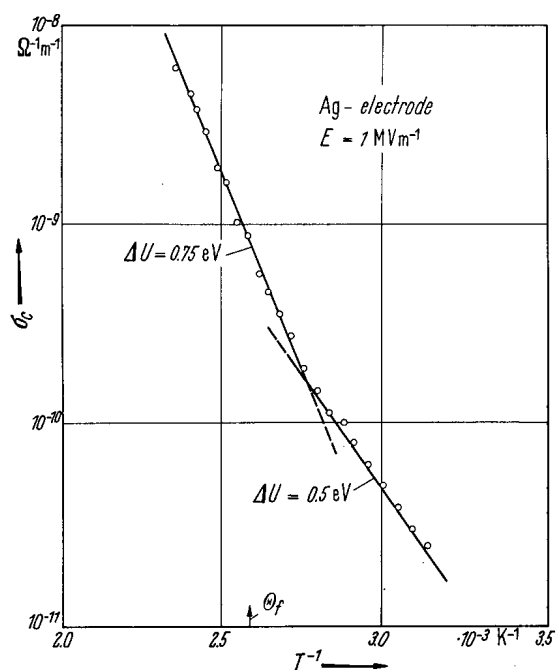


Fig. 1A-10-161. BaTiO₃. σ_c vs. T^{-1} [58Inu]. ΔU : activation energy.

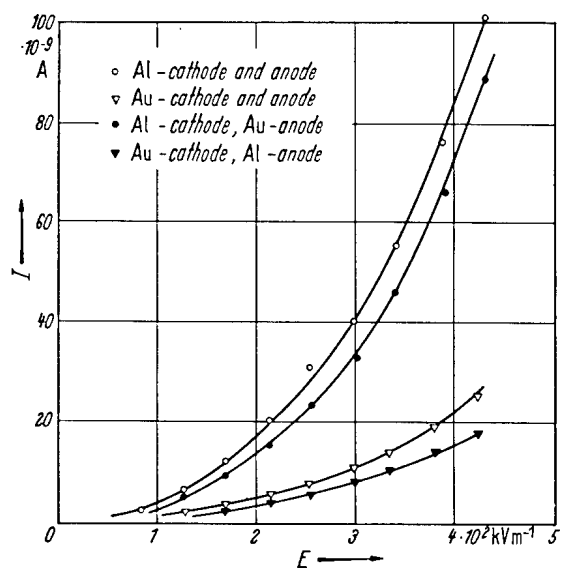


Fig. 1A-10-162. BaTiO_3 . I vs. E [62Bra]. Parameter: various cathode materials. Electrode area $7 \cdot 10^{-6}$ m 2 .

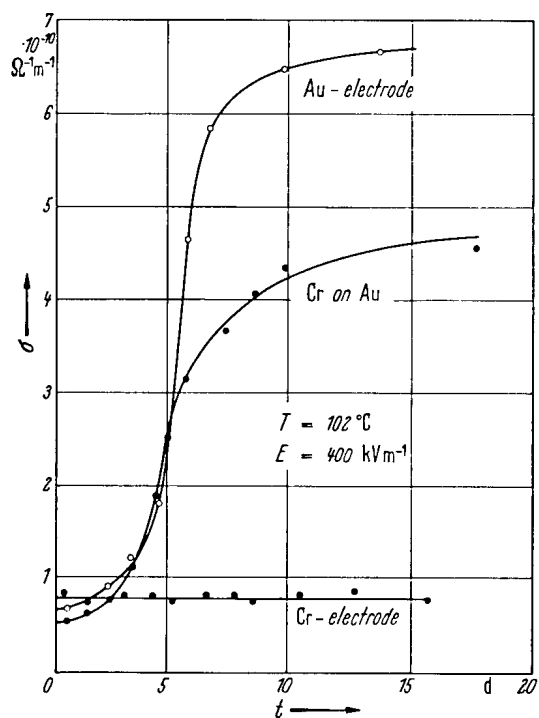


Fig. 1A-10-163. BaTiO₃. σ vs. t in days [60Bra]. Parameter: various cathode materials. Crystal thickness: $(1\ldots 3) \cdot 10^{-4}$ m. The conductivity at $t = 0$ corresponds to the value before diffusion of electrode materials into the BaTiO₃ lattice starts.

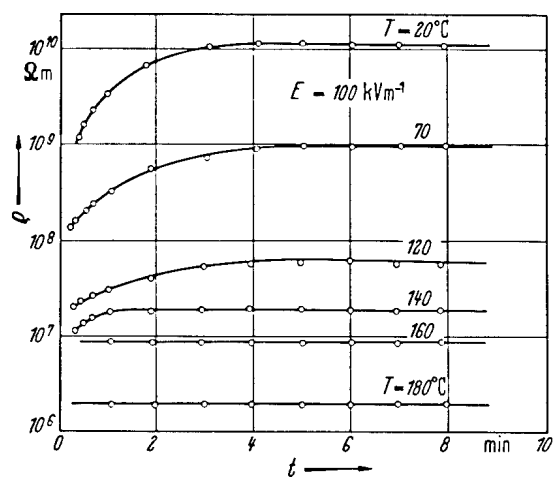


Fig. 1A-10-164. BaTiO₃ (ceramics). ρ vs. t [60Gur].
Parameter: T . ρ : electrical resistivity.

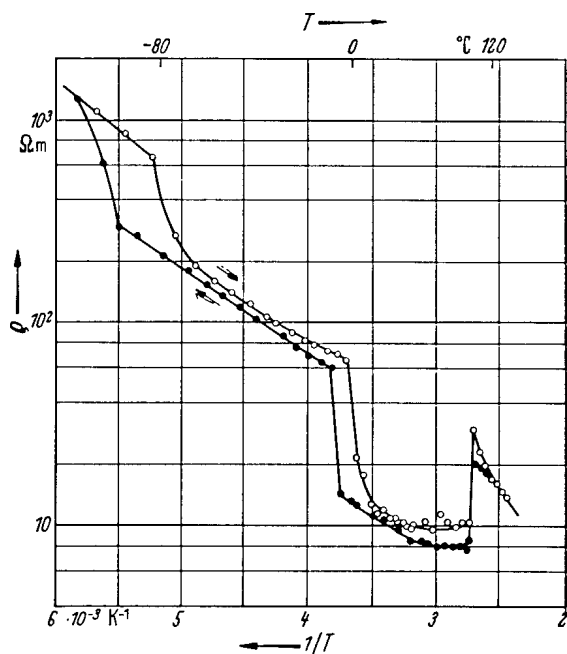


Fig. 1A-10-165. BaTiO₃ (reduced at 800 $^{\circ}\text{C}$ in H₂). ρ vs. T^{-1} [64Ike].

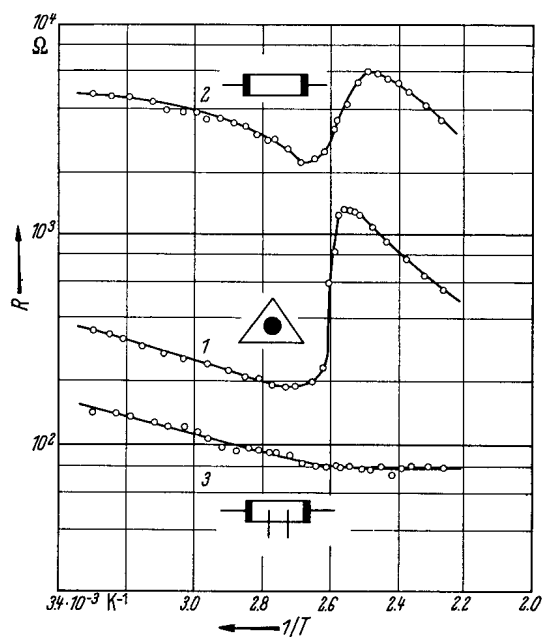


Fig. 1A-10-166. BaTiO_3 (reduced). R vs. T^{-1} [65Ued].
 Curve 1: Resistance along thickness of the crystal.
 2: Resistance along direction of length of a rectangular plate cut from the triangular crystal. 3: Resistance obtained by the four terminal method along the same direction as that of curve 2.

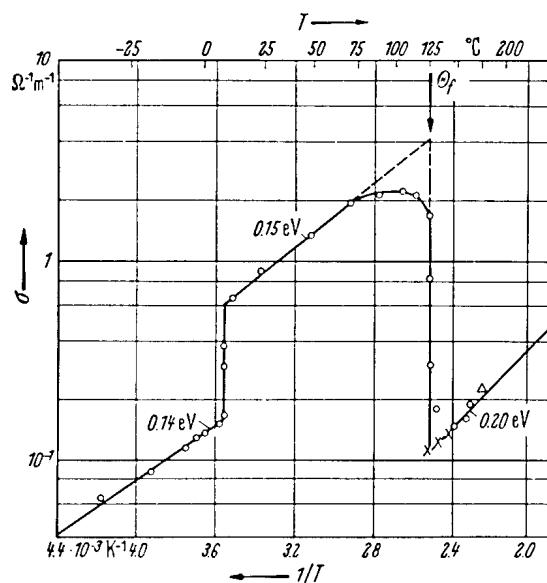


Fig. 1A-10-167. BaTiO₃ (Nb doped). σ vs. T^{-1} [64Bro].
The activation energies are given at the respective parts of the curve.

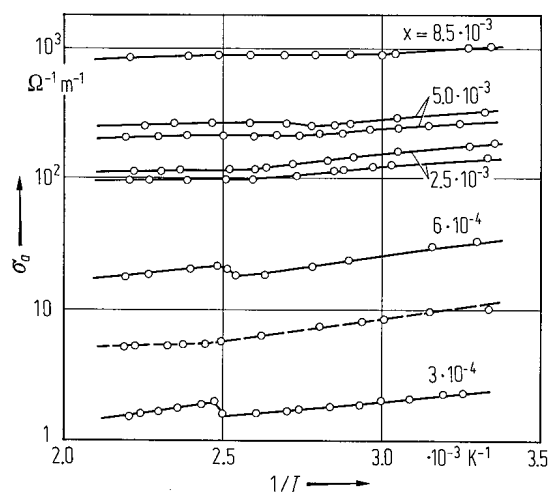


Fig. 1A-10-168. BaTiO_3 (Nb doped, single crystal). σ_a vs. T^{-1} for $\text{BaNb}_x^{\text{V}}\text{Ti}_x^{\text{III}}\text{Ti}_{1-2x}^{\text{IV}}\text{O}_3$ [77Fel]. Parameter: x , Nb concentration. σ_a : conductivity along the a -axis. Dashed curve: for single crystal reduced in H_2 .

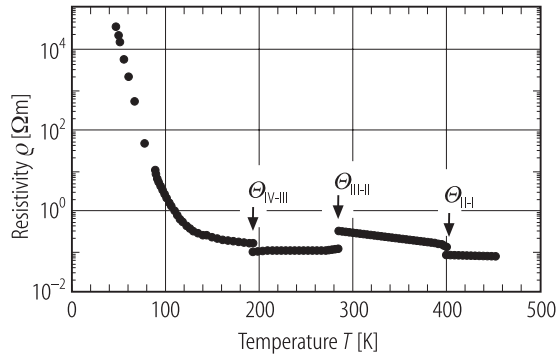


Fig. 1A-10-169. BaTiO₃ (0.2 at% Nb doped). ρ vs. T [94Gil2]. The crystal was grown by the top-seeded solution method. $\Theta_{\text{II-I}} = 403.5$ K, $\Theta_{\text{III-II}} = 285$ K, $\Theta_{\text{IV-III}} = 194$ K. See also [92Gil].

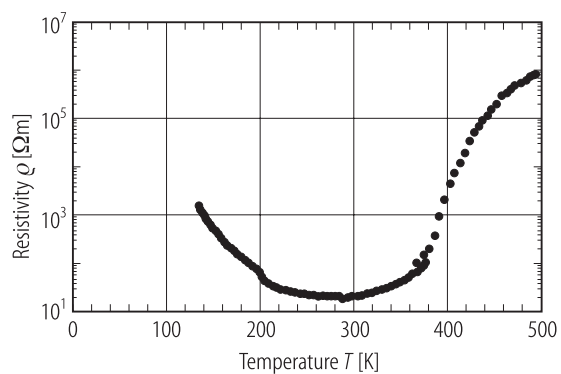


Fig. 1A-10-170. BaTiO₃ (0.2% Nb doped, ceramics). ρ vs. T [94Gil2].

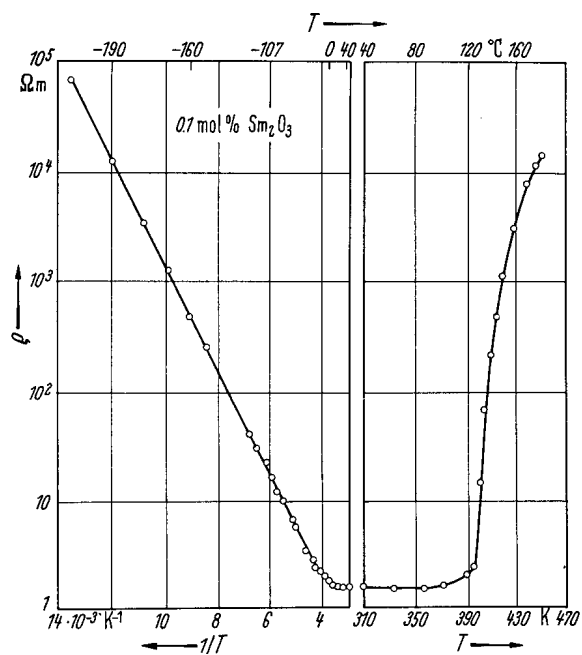


Fig. 1A-10-171. BaTiO₃ (Sm doped, ceramics). ρ vs. T [57Har]. Note change of abscissa scale.

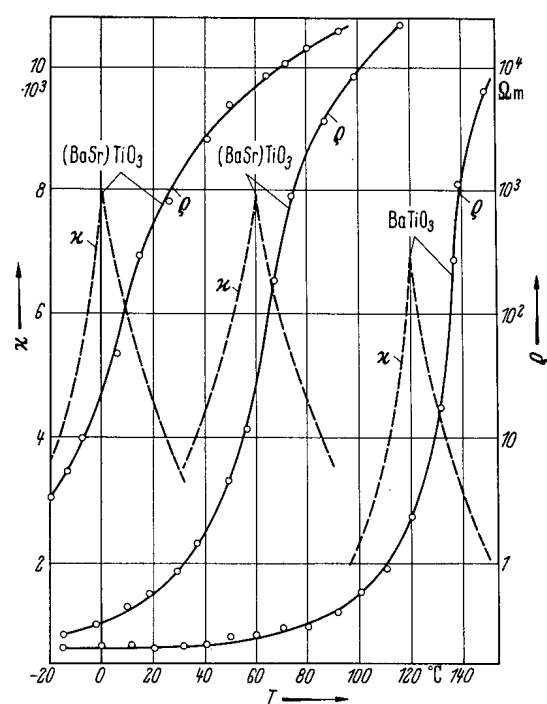


Fig. 1A-10-172. BaTiO₃ (semiconducting ceramics). ρ , κ vs. T [61Hey].

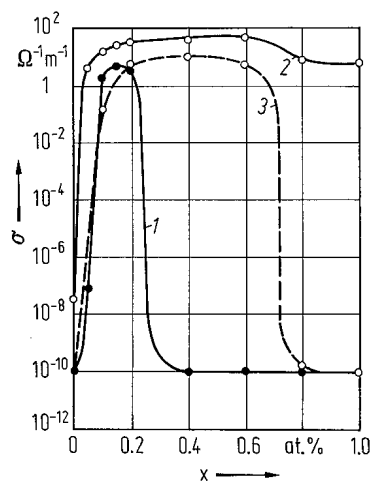


Fig. 1A-10-173. BaTiO₃ (Gd doped ceramics). σ vs. x at RT [73Mur]. x : Gd concentration [%]. Curve 1: as-fired in air, 2: as-fired in N₂, 3: reheated in air at 1200 °C after N₂ firing.

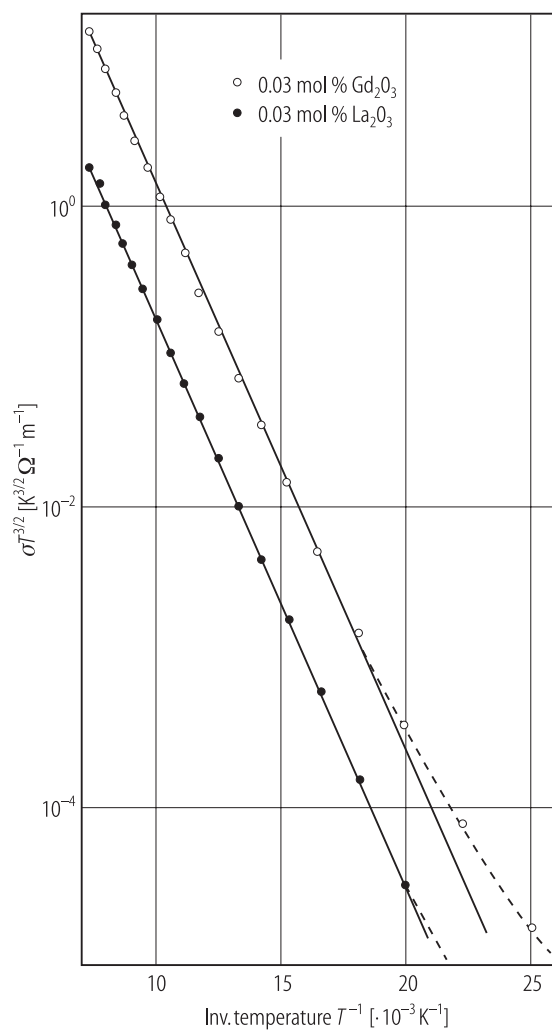


Fig. 1A-10-174. BaTiO₃ (La, Gd doped ceramics). $\sigma \cdot T^{3/2}$ vs. $1/T$ [911gu]. Full circles: 0.03 mol% La₂O₃ doped, open circles: 0.03 mol% Gd₂O₃ doped.

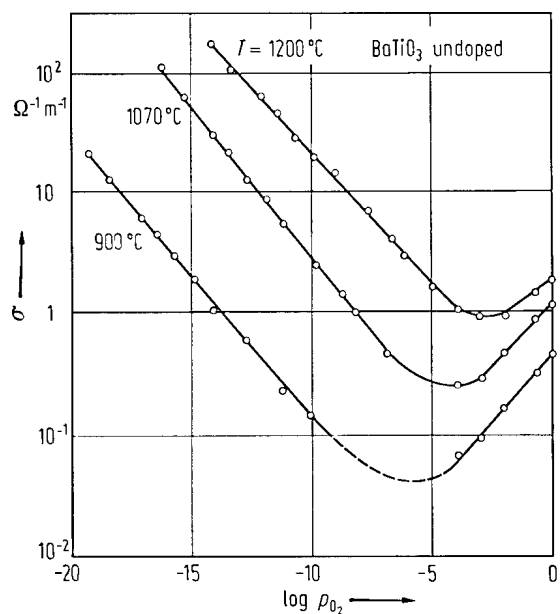


Fig. 1A-10-175. BaTiO₃ (ceramics). σ vs. $\log p_{\text{O}_2}$ [76Dan2]. Parameter: T where the measurements were made. p_{O_2} : oxygen partial pressure in units of atm. 1 atm = 101325 Pa.

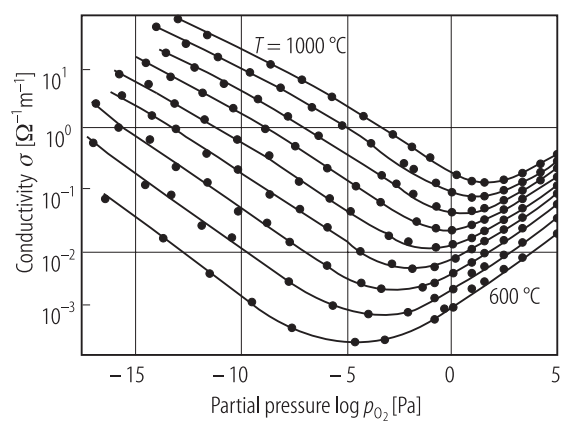


Fig. 1A-10-176. BaTiO_3 (ceramics). σ vs. $\log p_{\text{O}_2}$ [94Smy]. Parameter: T from 600 °C to 1000 °C, 50 °C step. p_{O_2} : oxygen partial pressure in units of Pa.

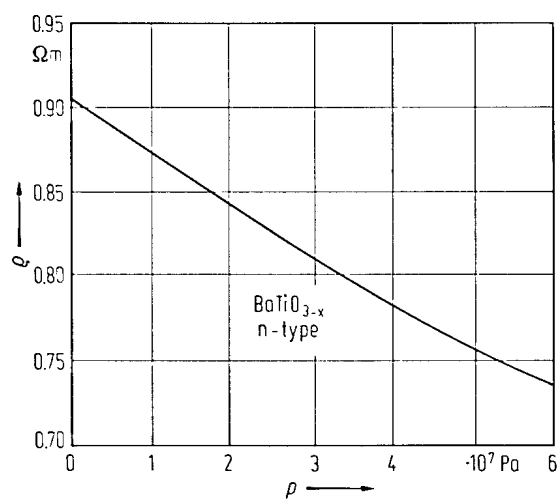


Fig. 1A-10-177. BaTiO₃ (reduced, ceramics). ρ vs. p [59Sau].

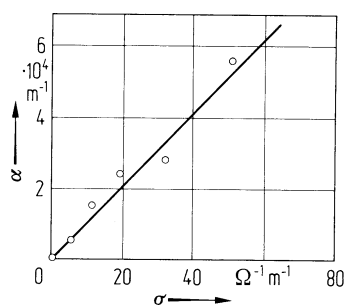


Fig. 1A-10-178. BaTiO_3 (semiconducting). α at $\lambda = 2 \mu\text{m}$ vs. σ for different semiconducting crystals containing small amount of rare earth (La, Gd, Dy) [71Bur]. α : absorption coefficient. $T = 140^\circ\text{C}$.

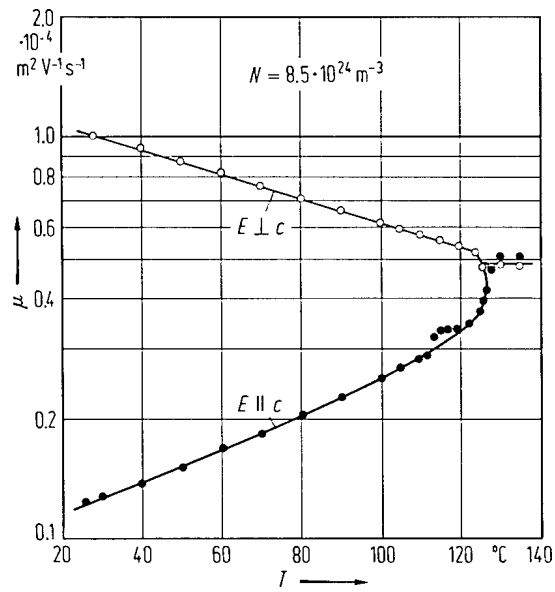


Fig. 1A-10-179. BaTiO₃ (reduced, n-type). μ vs. T [67Ber]. μ_{H} : Hall mobility. N : carrier concentration estimated from the Hall coefficient.