

**No. 1A-11 PbTiO<sub>3</sub>, Lead titanate**  
( $M = 303.1$ )

1a	Ferroelectric activity was suggested independently by Shirane et al. and Smolenskii in 1950.		50Shi, 50Smo
b	phase	II	I
	state	F	P
	crystal system	tetragonal	cubic
	space group	$P4mm - C_{4v}^1$	$Pm3m - O_h^1$
	$\Theta$ [°C]	490	
	The other phase transitions were reported to occur at about $-100$ °C, $-160$ °C or $-90$ °C in X-ray <sup>a)</sup> and optical <sup>b)</sup> measurements;		55Kob, 69Ike, 73Dos, 83Kob
	$T_{\text{melt}} = 1281(3)$ °C. $\rho = 7.96(1) \cdot 10^3 \text{ kg m}^{-3}$ . $P_s \parallel [100]$ .		52Rog 46Meg
2a	Crystal growth: flux method <sup>a)</sup> , Bridgman-Stockbarger method <sup>b)</sup> , Czochralski method <sup>c)</sup> , hydrothermal method <sup>d)</sup> , sol-gel process <sup>e)</sup> .		<sup>a)</sup> 52Nom, 58Kob, 70Rem, 93Sun <sup>b)</sup> 52Rog <sup>c)</sup> 76Gra <sup>d)</sup> 63Chr, 65Kuz, 68Sht <sup>e)</sup> 85Blu 80Tak
	Amorphous preparation: twin-roller quenching method. Phase diagram of PbO–TiO <sub>2</sub> : Fig. 1A-11-001. Non-stoichiometric PbTiO <sub>3</sub> compounds are prepared.		71Shi, 73Shi, 75Shi
3a	Positional parameters: Table 1A-11-001, Table 1A-11-002. Unit cell parameters: Table 1A-11-003; see also		61Ste 79Mab
b	Projection of crystal structure: Fig. 1A-11-002, Fig. 1A-11-003. Atomic displacements: Table 1A-11-003; Fig. 1A-11-004.		
4	Thermal expansion: Figs. 1A-11-005...1A-11-010, see also Dependence of lattice parameters on hydrostatic pressure: Fig. 1A-11-011, Fig. 1A-11-012, see also		68Bhi 62Kab
5a	Dielectric constant: Fig. 1A-11-013, Fig. 1A-11-014, see also  Small dielectric anomaly was observed at about $-100$ and $-150$ °C. Curie-Weiss constant: $C = 4.1 \cdot 10^5$ °C. Because of its high Curie temperature, the Curie-Weiss constant varies to some extent, depending on the crystal and measuring method. Effect of $p$ on $\kappa$ : Figs. 1A-11-015...1A-11-017. Phase diagram in regard to $p$ : Fig. 1A-11-018.		79Mal, 81Oku 56Kob 70Shi, 70Rem

	$[d\Theta_{II-I}/dp]_{p=0} = -8.4(3) \cdot 10^{-8} \text{ K Pa}^{-1}$ $[d\Theta_p/dp]_{p=0} = -7.1(3) \cdot 10^{-8} \text{ K Pa}^{-1}$ $[dC/dp]_{p=0} = -9.1(7) \cdot 10^{-5} \text{ K Pa}^{-1}$	71Sam
c	Spontaneous polarization: Fig. 1A-11-019. The theoretical value of $P_s$ was estimated to be $8.1 \cdot 10^{-1} \text{ Cm}^{-2}$ by Venevtsev et al. The semiempirical value to be $6.6 \cdot 10^{-1} \text{ Cm}^{-2}$ by Bhide et al. Effect of electric field on dielectric constant: see Spontaneous polarization and coercive field: spontaneous polarization and coercive field obtained from the $D$ - $E$ hysteresis are $7.5 \cdot 10^{-1} \text{ Cm}^{-2}$ and $6.75 \cdot 10^5 \text{ Vm}^{-1}$ for single crystal and $8.0 \cdot 10^{-1} \text{ Cm}^{-2}$ and $2.8 \cdot 10^7 \text{ Vm}^{-1}$ for thin film, respectively. Spontaneous polarization of $80 \pm 1 \cdot 10^{-1} \text{ Cm}^{-2}$ has been reported in single crystal having single domain; see also Fig. 1A-11-004 and	59Ven 68Bhi 95Fon 70Gav 93Tab 94Gav 70Rem 79Mal
d	Pyroelectricity: Figures of merit for infrared detector are given.  Pyroelectric coefficient: Fig. 1A-11-020.	71Yam, 76Yam
6a	Heat capacity: Fig. 1A-11-021, Fig. 1A-11-022. II-I: $\Delta Q_m = 4807 \text{ J mol}^{-1}$ , $\Delta S_m = 6.7 \text{ J K}^{-1} \text{ mol}^{-1}$ . DTA: Small exothermic and endothermic anomalies are observed near 113 K in the cooling and heating process.	51Shi 69Ike
c	Thermal conductivity: Fig. 1A-11-023.	
7a	Piezoelectric properties: Table 1A-11-004, Table 1A-11-005; Figs. 1A-11-024...1A-11-030.	
9a	Refractive indices: $n_o = 2.6676$ , $n_e = 2.6594$ for $\lambda = 0.6328 \mu\text{m}$ . $n$ was fitted to the dispersion equation $n^2 - 1 = E_d E_o / (E_o^2 - E^2)$ for $\lambda$ between 0.45 and $1.15 \mu\text{m}$ , where $E_d = 29.66 \text{ eV}$ and $E_o = 5.53 \text{ eV}$ for $n_o$ , $E_d = 30.64 \text{ eV}$ and $E_o = 5.71 \text{ eV}$ for $n_e$ , and $E$ is the photon energy in eV, $E_o$ : oscillator energy, $E_d$ : dispersion energy. Fig. 1A-11-031, Fig. 1A-11-032; see also Birefringence: Figs. 1A-11-033...1A-11-035. Infrared reflection and absorption: Figs. 1A-11-036...1A-11-040. Frequencies of infrared modes: Table 1A-11-006. Optical absorption: Fig. 1A-11-041, Fig. 1A-11-042.	71Pre 72Sin          79Mab 87Mal     92Zha
b	Electrooptic effect: $r_{13}^s = 13.8 \cdot 10^{-12} \text{ mV}^{-1}$ , $r_{33}^s = 5.9 \cdot 10^{-12} \text{ mV}^{-1}$ for $\lambda = 0.633 \mu\text{m}$ .	71Pas
d	Gyration tensor: Fig. 1A-11-034.	
e	Nonlinear optical susceptibilities: Fig. 1A-11-043. $d_{15}/ d_{11}^{\text{quartz}}  = \mp 104.1(157)$ , $d_{31}/ d_{11}^{\text{quartz}}  = \mp 117.6(176)$ , $d_{33}/ d_{11}^{\text{quartz}}  = \pm 23.4(38)$ for $\lambda = 1.064...1.319 \mu\text{m}$ . Second harmonic generation: see Fig. 1A-2-025 in No. 1A-2, KNbO <sub>3</sub> .	72Sin
10	Raman scattering: Figs. 1A-11-044...1A-11-048; see also Damping of the soft E-symmetry phonon: see Effect of pressure on the zone-center phonons: Fig. 1A-11-049, Fig. 1A-11-050. For amorphous specimen: see	70Tak 78Hei    81Nak, 84Nak
11	Conductivity: Fig. 1A-11-051. The conductivity of ceramics is $10^{-5}...10^{-6} \Omega^{-1} \text{ m}^{-1}$ at RT but very much dependent on the sintering condition of the ceramics. Also, effect of some impurities on the conductivity is seen.	69Gur 76Pro

	Seebeck effect: Fig. 1A-11-052.	
	Photoconductivity: the maximum at $\lambda \approx 410$ nm.	76Mal
	Electronic structure was calculated by P $\beta$ X $_{\alpha}$ method <sup>a)</sup> , ultra-soft-pseudopotential method <sup>b)</sup> , LCAO method <sup>c)</sup> .	<sup>a)</sup> 81Roz, 83Shv <sup>b)</sup> 94Kin <sup>c)</sup> 92Pro
	Photoconductivity: see	83Rae
13b	ESR: Fe <sup>3+</sup> , Spin-Hamiltonian parameters at $-188$ °C. For narrow line width spectrum: $D = 118.7(2) \text{ m}^{-1}$ , $a = 2.0(7) \text{ m}^{-1}$ , $F = -1.7(10) \text{ m}^{-1}$ . For wide line width spectrum: $D = 115.0(3) \text{ m}^{-1}$ , $a = 2.2(10) \text{ m}^{-1}$ , $F = -2(1) \text{ m}^{-1}$ . Table 1A-11-007. Mn <sup>2+</sup> : Table 1A-11-008. Fig. 1A-11-053; see also	76Lew      85Bal
c	Mössbauer effect: Figs. 1A-11-054...1A-11-057. Additional informations can be found in  Perturbed $\gamma$ - $\gamma$ directional correlation for the first excited state of <sup>44</sup> Sc in PbTiO <sub>3</sub> at $-189$ °C: $ QV_{zz}  = 4.09(9) \cdot 10^{-8} \text{ V}$ .	69Yam, 71Yag  73Dos
14b	Inelastic neutron scattering: Figs. 1A-11-058...1A-11-060.	
c	EXAFS: Fig. 1A-11-061; see also	81Ter, 94Sic
15a	Domain structure: see	52Nom, 58Kob, 59Kob, 68Bhi, 72Fes, 73Fes, 75Tan
b	Domain switching: the mobility of side motion ranges from $10^{-8}$ to $10^{-7} \text{ m}^2\text{V}^{-1}\text{s}^{-1}$ .	76Fes
16	Surface layer: X-ray diffraction measurements under d.c. electric field show the existence of a surface layer with thickness of about $1.04 \cdot 10^{-6} \text{ m}$ . Radiation damage: Fig. 1A-11-062. Thin film growth: RF sputtering, chemical vapor deposition, laser annealing.	74Dud, 78Dud  79Oku, 83Koj, 81Mat
	Thin film dielectric property: see Fig. 1A-11-014 in 5a and Fig. 1A-11-020 in 5d; see also	81Oku, 85Shi, 86Iij

**Table 1A-11-001.** PbTiO<sub>3</sub>. Fractional coordinates of atoms in the unit cell of phase II at RT [56Shi].

Atom	<i>x</i>	<i>y</i>	<i>z</i>
Pb	0	0	0
Ti	1/2	1/2	0.540
O(1)	1/2	1/2	0.112
O(2)	1/2	0	0.612
	0	1/2	0.612

**Table 1A-11-002.**  $\text{PbTiO}_3$ . Bond lengths [in Å] in phase II at RT and in phase I at 490 °C [56Shi].  $\text{O}(1)_+$  and  $\text{O}(1)_-$  represent the O(1) atoms closer to and further away from Ti atom, respectively. Similarly  $\text{O}(2)_+$  is closer to Pb atom.

Phase	II (at RT)	I (at 490°C)
Ti–O(1) <sub>+</sub>	1.78	1.89
Ti–O(1) <sub>–</sub>	2.38	
Ti–O(2)	1.98	
Pb–O(1)	2.80	2.80
Pb–O(2) <sub>+</sub>	2.53	
Pb–O(2) <sub>–</sub>	3.20	

**Table 1A-11-003.** PbTiO<sub>3</sub>. Refined structural parameters for tetragonal PbTiO<sub>3</sub> at 295 K and 700 K [85Nad].  $\delta_z$ : displacements of atoms from the cubic phase sites (in fraction).  $U_{ij}$ : the mean-square thermal amplitudes (in Å<sup>2</sup> units) in the unit cell of phase II.  $U_{ij}$  are defined by Eq. (d) in Introduction.  $a$ ,  $c$ : lattice constants.  $R_w$ : final weighted  $R$ -factor.

		$T = 295 \text{ K}$	$T = 700 \text{ K}$
Pb	$U_{11} [\text{Å}^2]$	0.0089(2)	0.0231(3)
	$U_{22} [\text{Å}^2]$	0.0083(2)	0.0226(3)
Ti	$\delta_z$	0.0377(4)	0.0250(4)
	$U_{11} [\text{Å}^2]$	0.0037(3)	0.0090(3)
	$U_{33} [\text{Å}^2]$	0.0054(5)	0.0118(7)
OI	$\delta_z$	0.1118(3)	0.0799(4)
	$U_{11} [\text{Å}^2]$	0.0095(2)	0.0207(3)
	$U_{33} [\text{Å}^2]$	0.0055(3)	0.0098(3)
OII	$\delta_z$	0.1174(3)	0.0845(4)
	$U_{11} [\text{Å}^2]$	0.0081(2)	0.0197(3)
	$U_{22} [\text{Å}^2]$	0.0039(2)	0.0081(2)
	$U_{33} [\text{Å}^2]$	0.0095(2)	0.0203(3)
	$a [\text{Å}]$	3.902(3)	3.940(6)
	$c [\text{Å}]$	4.156(3)	4.063(6)
	$R_w$	0.025	0.036

**Table 1A-11-004.** PbTiO<sub>3</sub>. Physical constants at RT.

	Single crystal [71Gav]	Ceramics <sup>a)</sup> [71Ike] <sup>b)</sup>		Single crystal [71Gav]	Ceramics <sup>a)</sup> [71Ike] <sup>b)</sup>
$a$ [Å]		3.9072	Elastic constants		
$c$ [Å]		4.1187	$s_{11}^E$ [ $\cdot 10^{-12} \text{ m}^2 \text{ N}^{-1}$ ]	7.2	7.5
$\Theta_f$ [°C]		470	$s_{12}^E$	−2.1	−1.5
$\rho_{\text{exp}}$ [ $\cdot 10^3 \text{ kg m}^{-3}$ ]	7.9	7.87	$s_{13}^E$		−1.1
Porosity [%]		1.5	$s_{33}^E$	32.5	8.0
$\rho$ [ $\Omega \text{ m}$ ]		$2 \cdot 10^{10}$	$s_{44}^E$	12.2	17.9
Dielectric constants			$s_{11}^D$		7.5
$\kappa_{11}^T$	210	1 kHz 230	$s_{12}^D$		−1.5
$\kappa_{11}^S$	115	210	$s_{13}^D$		−1.1
$\kappa_{33}^T$	126	1 kHz 170	$s_{33}^D$		6.4
$\kappa_{33}^S$	51	140	$s_{44}^D$		16.5
Dissipation factor $D$		0.008	$s_{66}$	7.9	18.0
Frequency constants			$c_{33}^E$ [ $\cdot 10^{10} \text{ Nm}^{-2}$ ]		13.1
$N_p$ [Hz m]		2640	$c_{33}^D$		16.6
$N_{31}$		2050	Poisson's ratio $\sigma^E$		
$N_{33}$		2020	Mechanical		
$N_t$		2070	quality factor $Q$		
$N_{15}$		1340			

<sup>a)</sup> 1.25 mol% La<sub>2</sub>O<sub>3</sub>, 1.0 mol% MnO<sub>2</sub>. <sup>b)</sup> See also [72Ued].

**Table 1A-11-005.** PbTiO<sub>3</sub>. Electromechanical constants at RT.

	Single crystal [71Gav]	Ceramics (1.25 mol% La <sub>2</sub> O <sub>3</sub> , 1.0 mol% MnO <sub>2</sub> ) [71Ike] <sup>a)</sup>
Coupling factors		
$k_p$	0.40	0.07
$k_{31}$	0.24	0.04
$k_{33}$	0.64	0.46
$k_t$		0.46
$k_{15}$	0.43	0.28
Piezoelectric constants		
$d_{31}$ [ $\cdot 10^{-12}$ C N <sup>-1</sup> ]	-2.5	-4.4
$d_{33}$	11.7	51
$d_{33}^b$	19.3	
$d_{15}$	6.5	53
$g_{31}$ [ $\cdot 10^{-3}$ m <sup>2</sup> C <sup>-1</sup> ]		-2.9
$g_{33}$		33
$g_{15}$		27
Electrostrictive constants for $P$		
$Q_{11}$ [m <sup>4</sup> C <sup>-2</sup> ]	0.080	
$Q_{12}$	-0.029	

<sup>a)</sup> See also [72Ued]. <sup>b)</sup> Static method.



**Table 1A-11-006.**  $\text{PbTiO}_3$ . Wave numbers in  $10^2 \text{ m}^{-1}$  and symmetries of infrared modes of lattice vibration obtained from Kramers-Kronig analysis of the reflectance data at RT [65Per].

$\tilde{\nu}_1$ (Ti–O stretch)	$\tilde{\nu}_2$ (Ti–O <sub>3</sub> torsion)	$\tilde{\nu}_3$ (O–Ti–O bend)	$\tilde{\nu}_4$ (cation–TiO <sub>3</sub> lattice mode)
530 (E <sub>u</sub> , A <sub>1</sub> )	400 (B <sub>1</sub> , E <sub>u</sub> )	220 (E <sub>u</sub> , A <sub>1</sub> ) 172	83 (E <sub>u</sub> , A <sub>1</sub> )

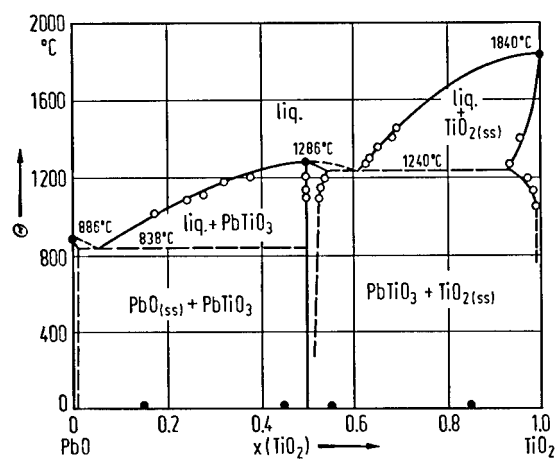
**Table 1A-11-007.** PbTiO<sub>3</sub>. *g*-factor at RT [64Gai].

Paramagnetic center	Site	<i>S</i>	<b>H</b>	<i>ν</i> [GHz]	<i>T</i> [K]	<i>g</i> <sub>  </sub>	<i>g</i> <sub>⊥</sub>
Fe <sup>3+</sup>	Ti <sup>4+</sup>	1/2 <sup>*)</sup>	(2)	8.8	RT	2.009(5)	5.97(2)

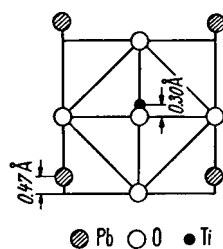
<sup>\*)</sup> Observed spectrum could be described by effective spin of 1/2, which was explained in terms of a strong tetragonal component of the crystalline field.

**Table 1A-11-008.** PbTiO<sub>3</sub>. ESR data [69Ike].

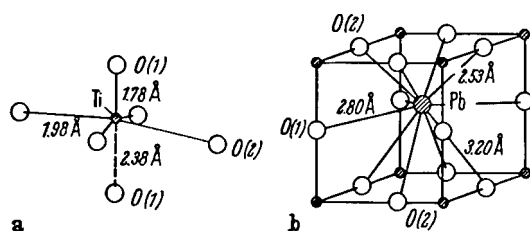
Paramagnetic center	Mn <sup>2+</sup>		Cubic phase	Tetragonal phase
Site	Ti <sup>4+</sup>	<i>g</i> -factor	1.996(2)	2.084(2)
<i>S</i>	5/2	FS [ $\cdot 10^{-2} \text{ m}^{-1}$ ]	$D = 0$	$D = + 502(4)$
<b>H</b>	(7)		–	$a = + 43(2)$
$\nu$ [GHz]	$\approx 9$	HFS [ $\cdot 10^{-2} \text{ m}^{-1}$ ]	–79.8(5)	–82.0(15)



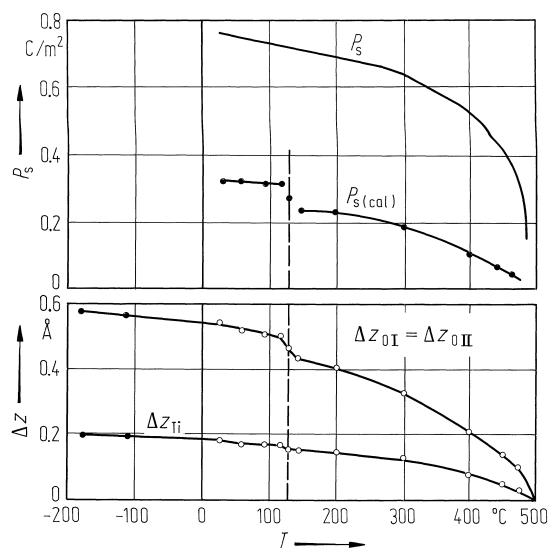
**Fig. 1A-11-001.**  $\text{PbTiO}_3$ . Phase diagram of system  $(\text{PbO})_{1-x}(\text{TiO}_2)_x$  [76Hol]. Refined by mass-loss Knudsen effusion. Full circles: DTA, open circles: mass-loss Knudsen effusion.



**Fig. 1A-11-002.**  $\text{PbTiO}_3$ . Projection of the structure of phase II along the  $a$ -axis [56Shi].



**Fig. 1A-11-003.** PbTiO<sub>3</sub>. A sketch of the structure of phase II at RT [56Shi]. (a) environment of Ti atom, (b) environment of Pb atom.



**Fig. 1A-11-004.**  $\text{PbTiO}_3$ .  $\Delta z$  and  $P_s$  vs.  $T$  [83Kup].  $\Delta z$ : atomic displacement.  $P_s(\text{cal})$ : spontaneous polarization due to the displacement of cation from the center of oxygen octahedron. Full circles: data from [78Gla].

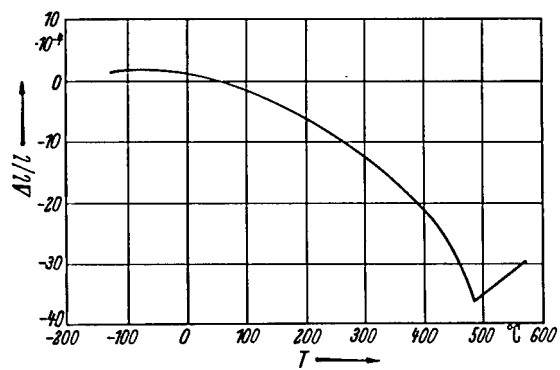


Fig. 1A-11-005. PbTiO<sub>3</sub> (ceramics).  $\Delta l/l$  vs.  $T$  [51Shi1].



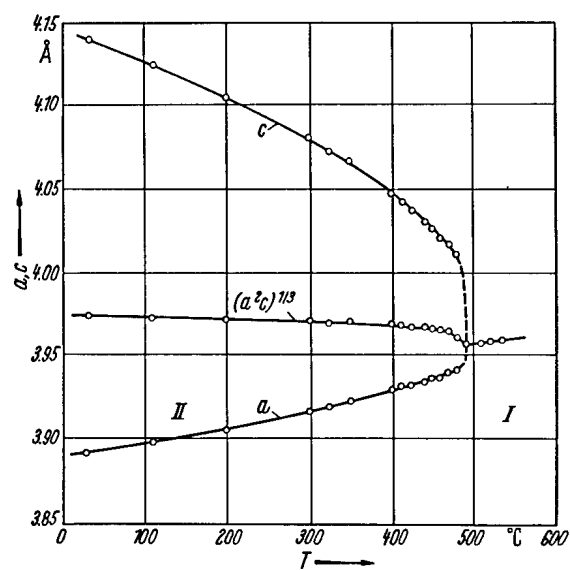


Fig. 1A-11-006.  $\text{PbTiO}_3$ . Lattice constants  $a$ ,  $c$  vs.  $T$  [51Shi1].

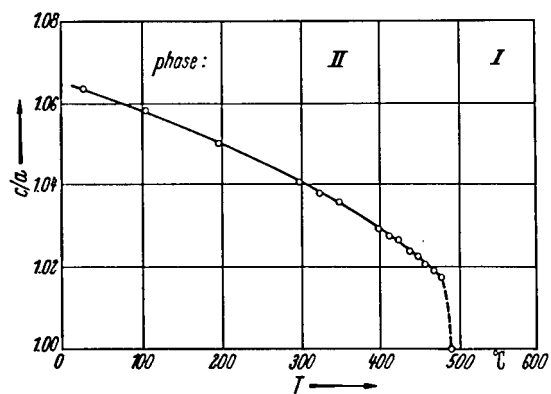
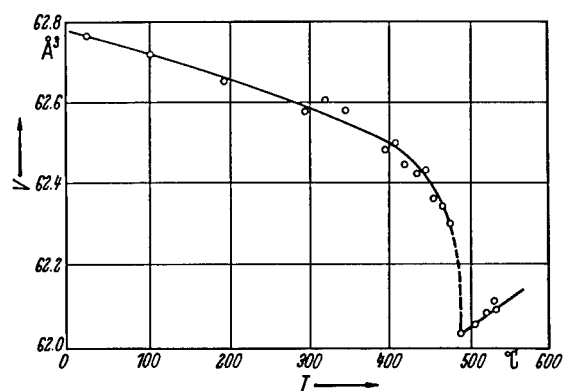
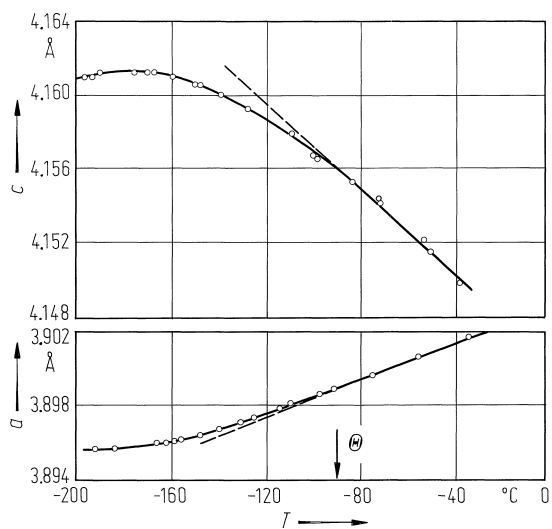


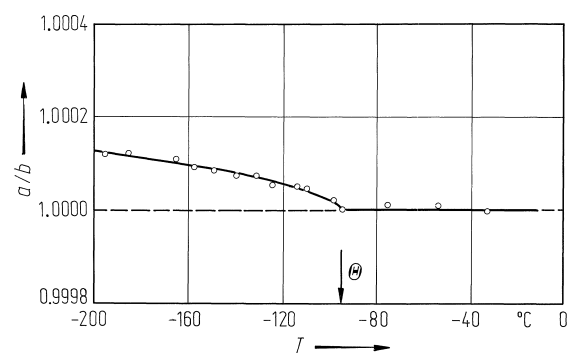
Fig. 1A-11-007.  $\text{PbTiO}_3$ ,  $c/a$  vs.  $T$  [51Shi1].  $c$ ,  $a$ : lattice constants.



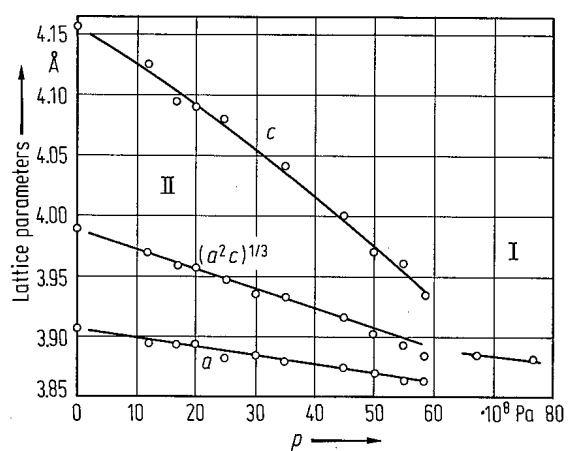
**Fig. 1A-11-008.**  $\text{PbTiO}_3$ .  $V$  vs.  $T$  [51Shi1].  $V$ : unit cell volume.



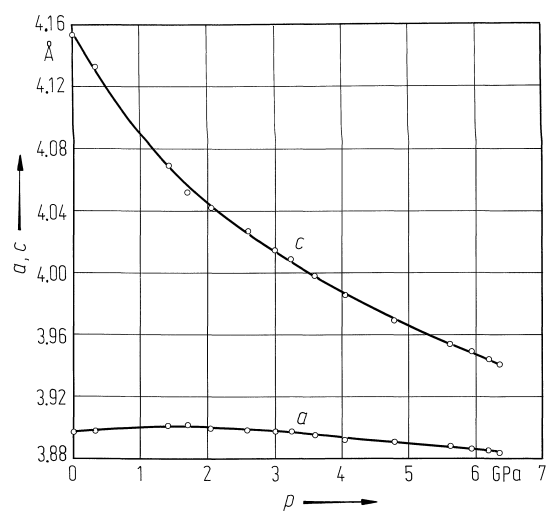
**Fig. 1A-11-009.**  $\text{PbTiO}_3$ . Lattice constants  $a$ ,  $c$  vs.  $T$  [83Kac].



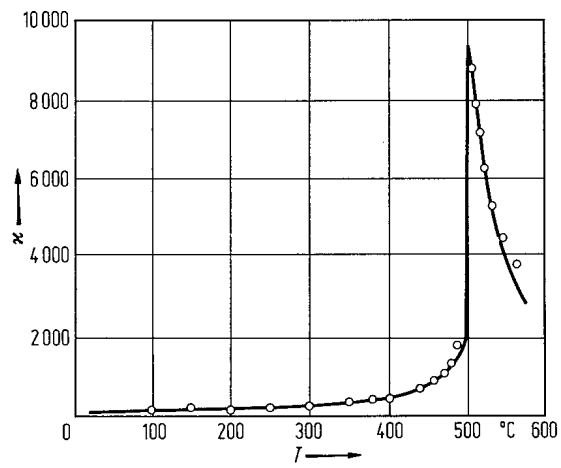
**Fig. 1A-11-010.**  $\text{PbTiO}_3$ .  $a/b$  vs.  $T$  [83Kac].



**Fig. 1A-11-011.** PbTiO<sub>3</sub>. Lattice constants  $a$ ,  $c$  vs.  $p$  [75Ike].

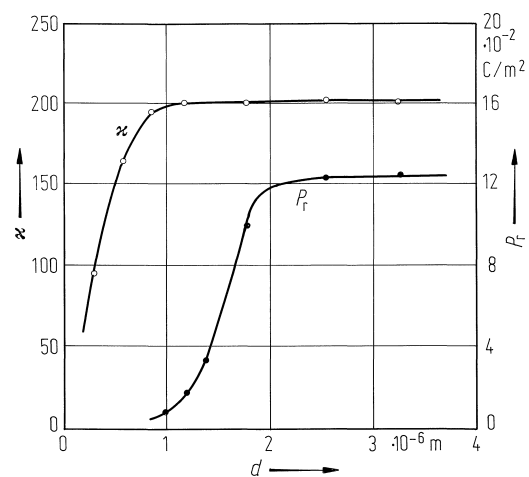


**Fig. 1A-11-012.**  $\text{PbTiO}_3$ .  $a$ ,  $c$  vs.  $p$  at  $T = 292 \text{ K}$  [86Nel].  
 $p$ : hydrostatic pressure.

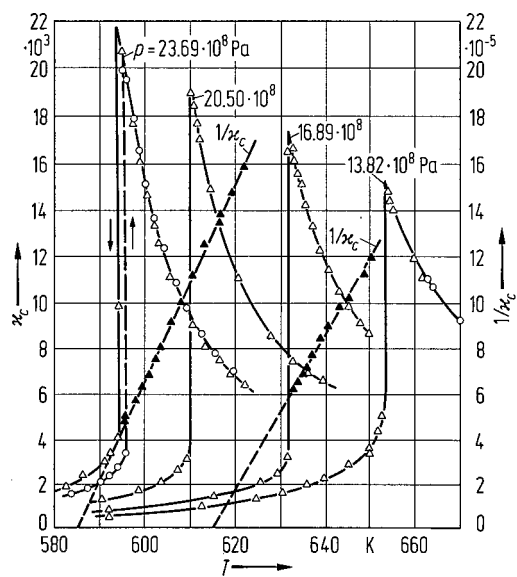


**Fig. 1A-11-013.**  $\text{PbTiO}_3$ .  $\kappa$  vs.  $T$  [70Rem].

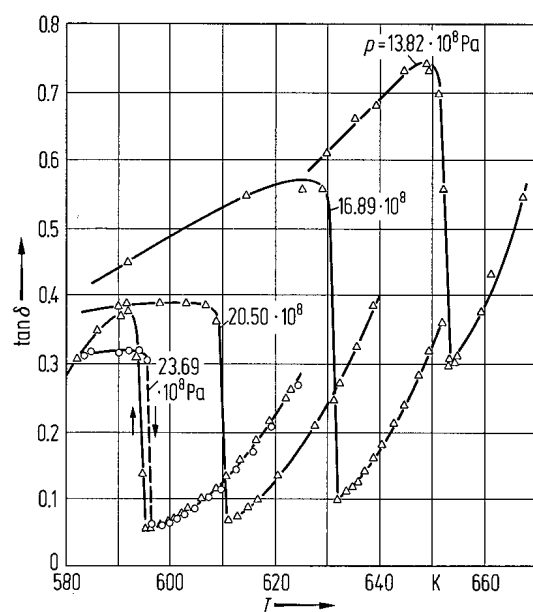




**Fig. 1A-11-014.** PbTiO<sub>3</sub> (thin film).  $\kappa$ ,  $P_r$  vs.  $d$  [81Oku].  
 $d$ : film thickness.



**Fig. 1A-11-015.** PbTiO<sub>3</sub>.  $\kappa_c$ ,  $1/\kappa_c$  vs.  $T$  [71Sam].  
Parameter:  $p$ .



**Fig. 1A-11-016.**  $\text{PbTiO}_3$ .  $\tan \delta$  vs.  $T$  [71Sam]. Parameter:  $p$ .

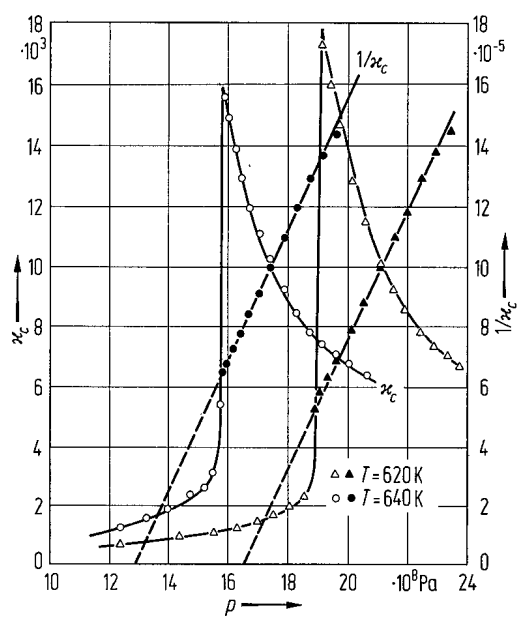
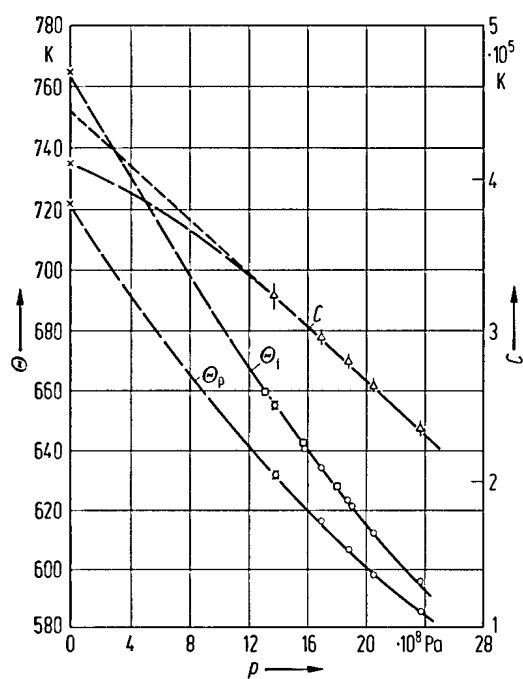


Fig. 1A-11-017. PbTiO<sub>3</sub>.  $\kappa_c$ ,  $1/\kappa_c$  vs.  $p$  [71Sam].



**Fig. 1A-11-018.**  $\text{PbTiO}_3$ .  $\Theta_p$ ,  $\Theta_f$ ,  $C$  vs.  $p$  [71Sam].  
 $C$ : Curie-Weiss constant.

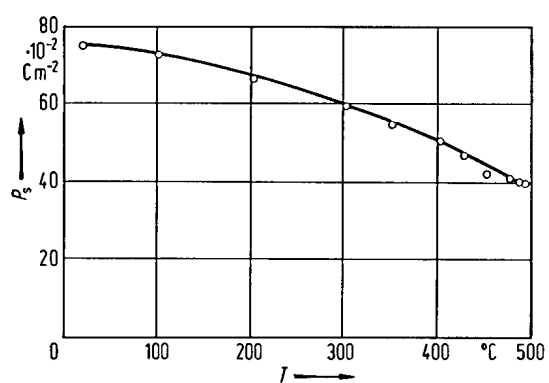
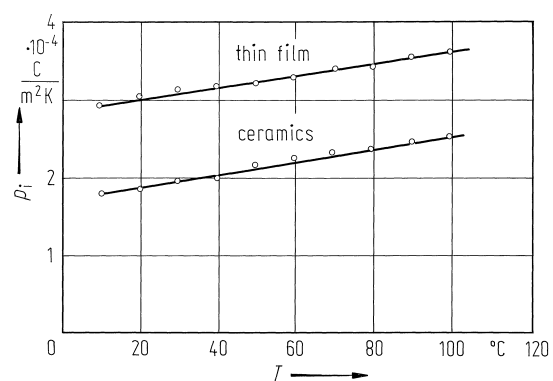
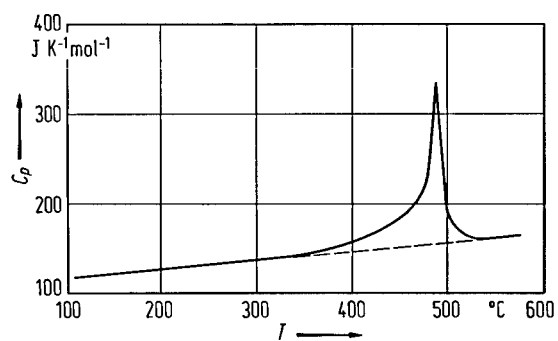


Fig. 1A-11-019.  $\text{PbTiO}_3$ .  $P_s$  vs.  $T$  [70Gav].

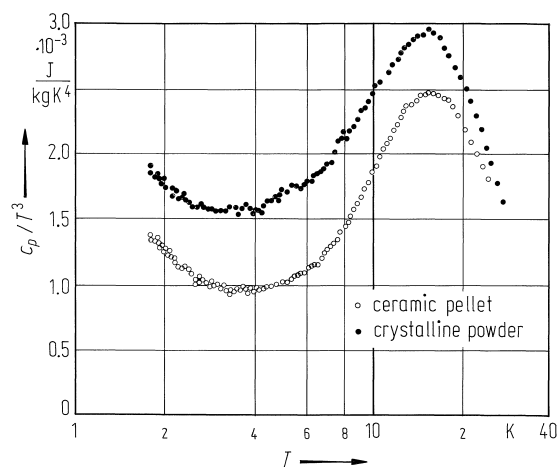


**Fig. 1A-11-020.**  $\text{PbTiO}_3$  (thin film and ceramics).  $p_i$  vs.  $T$   
[83Iij].  $p_i$ : pyroelectric coefficient.

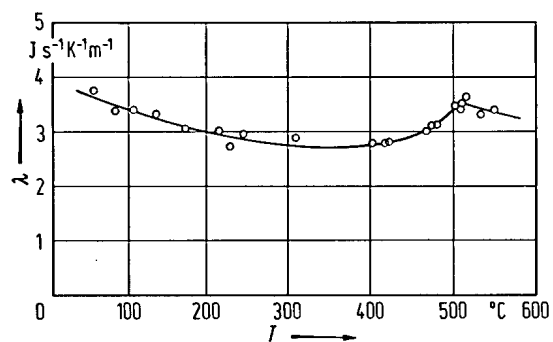


**Fig. 1A-11-021.**  $\text{PbTiO}_3$ .  $C_p$  vs.  $T$  [51Shi2].  $C_p$ : molar heat capacity at constant pressure.





**Fig. 1A-11-022.**  $\text{PbTiO}_3$ .  $c_p/T^3$  vs.  $T$  [83Law].  $c_p$ : specific heat capacity at constant pressure.



**Fig. 1A-11-023.** PbTiO<sub>3</sub>.  $\lambda$  vs.  $T$  [60Yos].  $\lambda$ : thermal conductivity.

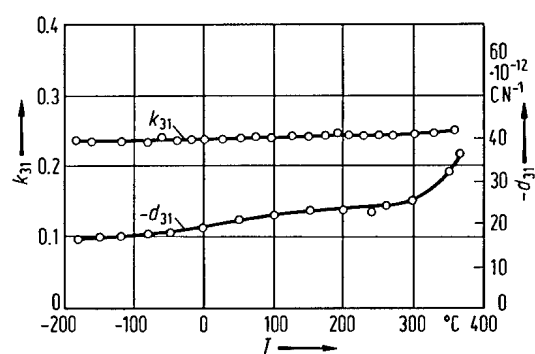


Fig. 1A-11-024.  $\text{PbTiO}_3$ .  $k_{31}$ ,  $-d_{31}$  vs.  $T$  [71Gav].

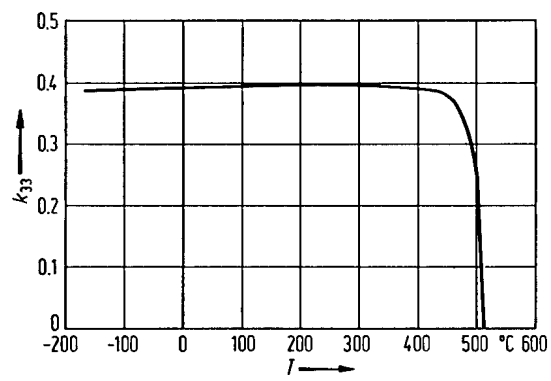
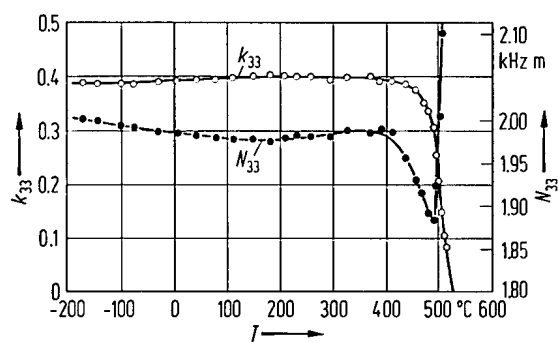
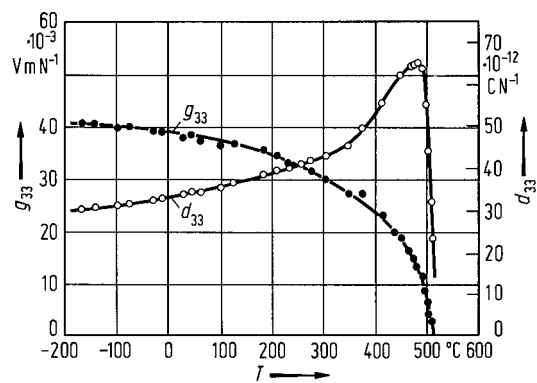


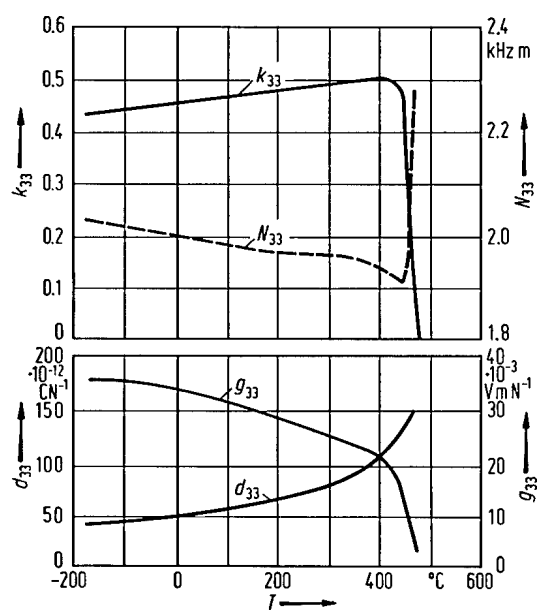
Fig. 1A-11-025.  $\text{PbTiO}_3$  (ceramics).  $k_{33}$  vs.  $T$  [70Ike].



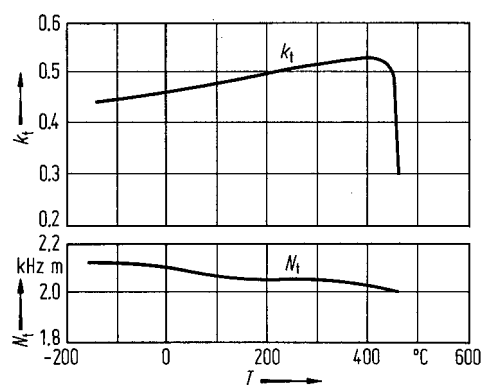
**Fig. 1A-11-026.**  $\text{PbTiO}_3$  (ceramic with additive 1 mol%  $\text{MnO}_2$ ).  $k_{33}$ ,  $N_{33}$  vs.  $T$  [72Vie].  $N_{33}$ : frequency constant.



**Fig. 1A-11-027.** PbTiO<sub>3</sub> (ceramic with additive 1 mol% MnO<sub>2</sub>).  $d_{33}$ ,  $g_{33}$  vs.  $T$  [72Vie].  $1 \text{ V m N}^{-1} = 1 \text{ m}^2 \text{C}^{-1}$ .

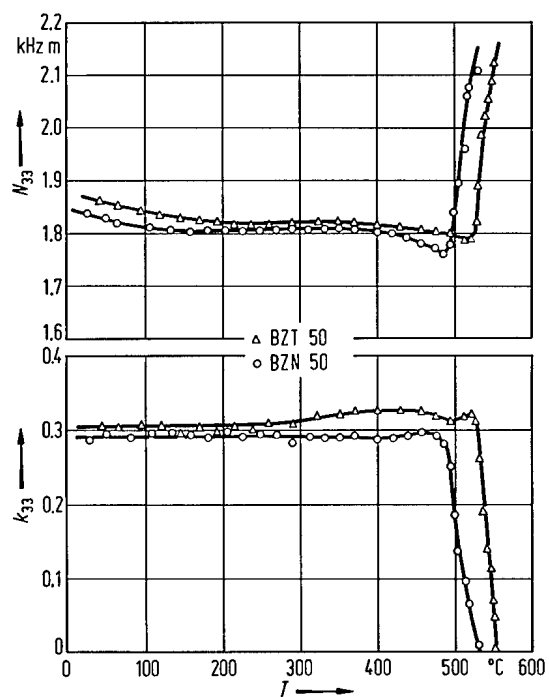


**Fig. 1A-11-028.**  $\text{PbTiO}_3$  (ceramics containing  $\text{La}_2\text{O}_3$  and  $\text{MnO}_2$ ).  $k_{33}$ ,  $N_{33}$ ,  $d_{33}$ ,  $g_{33}$  vs.  $T$  [71Ike].  $N_{33}$ : frequency constant.

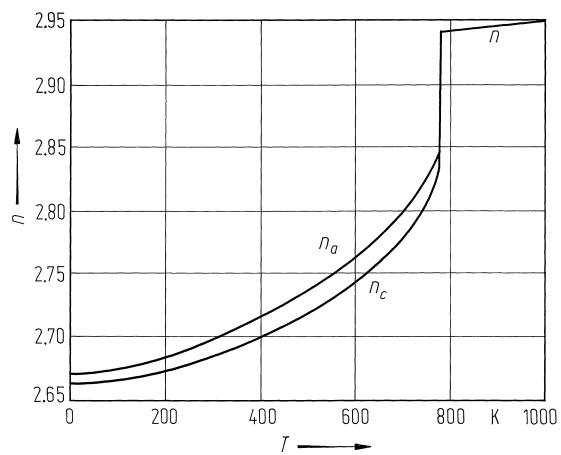


**Fig. 1A-11-029.** PbTiO<sub>3</sub> (ceramics containing La<sub>2</sub>O<sub>3</sub> and MnO<sub>2</sub>).  $k_t$ ,  $N_t$  vs.  $T$  [71Ike].  $k_t$ : thickness coupling factor,  $N_t$ : frequency constant (thickness vibration).

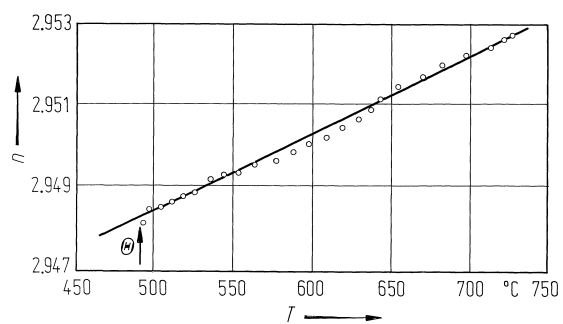




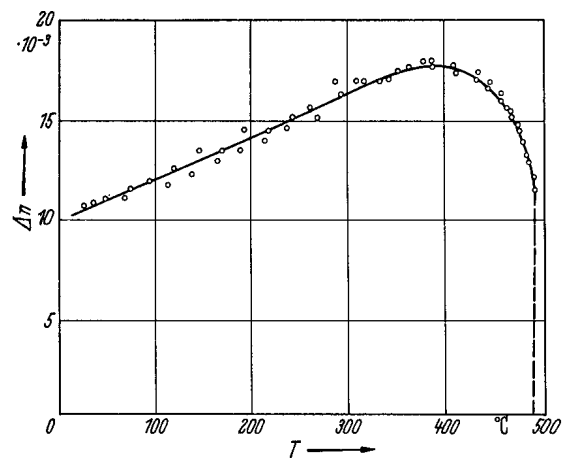
**Fig. 1A-11-030.**  $\text{PbTiO}_3$  (ceramics).  $N_{33}$ ,  $k_{33}$ , vs.  $T$  [68Ued].  $N_{33}$ : frequency constant of the length extensional mode. BZT50: with additive  $\text{BiZn}_{1/2}\text{Ti}_{1/2}\text{O}_3$ , BZN50: with additive  $\text{Bi}_{2/3}\text{Zn}_{1/3}\text{Nb}_{2/3}\text{O}_3$ .



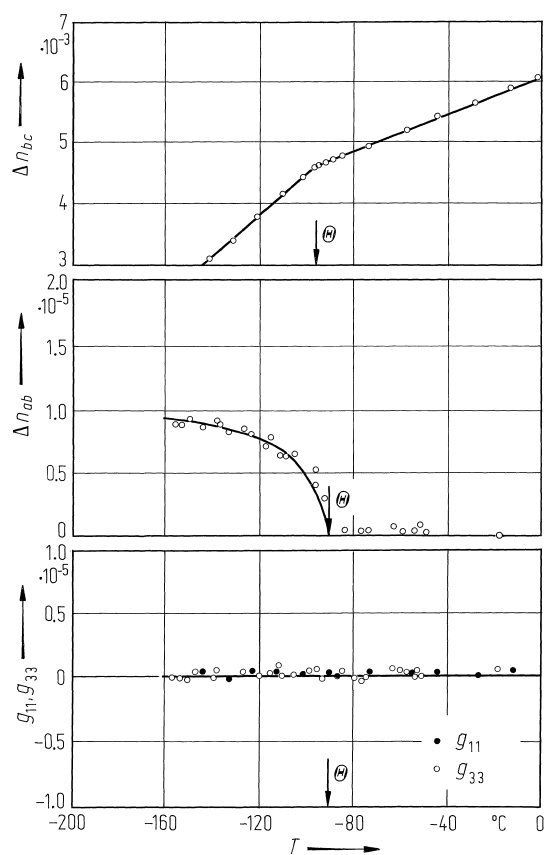
**Fig. 1A-11-031.** PbTiO<sub>3</sub>.  $n_a$ ,  $n_c$ ,  $n$  vs.  $T$  [86Kle].  
 $\lambda = 589.3$  nm.  $n_a$ ,  $n_c$ : refractive indices for  $E$  of light parallel to  $a$  and  $c$ .



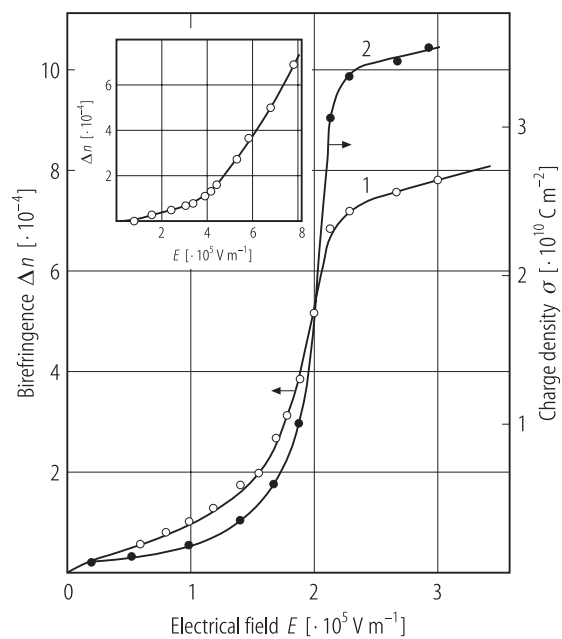
**Fig. 1A-11-032.**  $\text{PbTiO}_3$ .  $n$  vs.  $T$  [82Bur].  $\lambda = 632.8 \text{ nm}$ .



**Fig. 1A-11-033.** PbTiO<sub>3</sub>.  $\Delta n$  vs.  $T$  [56Shi].  $\Delta n = n_a - n_c$ . Na light was used.

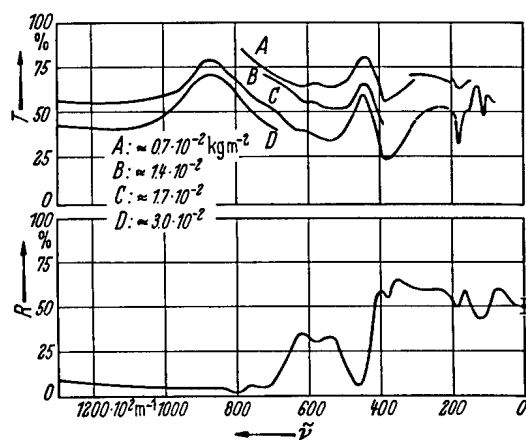


**Fig. 1A-11-034.**  $\text{PbTiO}_3$ .  $\Delta n_{bc}$ ,  $\Delta n_{ab}$ ,  $g_{11}$ ,  $g_{33}$  vs.  $T$  [83Kob].  $\Delta n_{bc} = n_b - n_c$ .  $\Delta n_{ab} = n_a - n_b$ .  $g_{11}$ ,  $g_{33}$ : component of gyration tensor for optical activity.

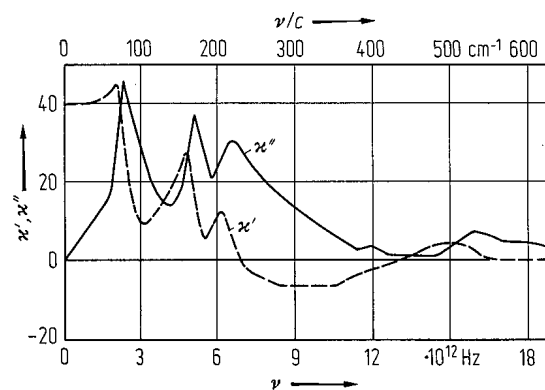


**Fig. 1A-11-035.** PbTiO<sub>3</sub>.  $\Delta n$ ,  $\sigma$  vs.  $E$  [90Che].

$\Delta n$ : birefringence,  $\sigma$ : charge density,  $E$ : electric field.

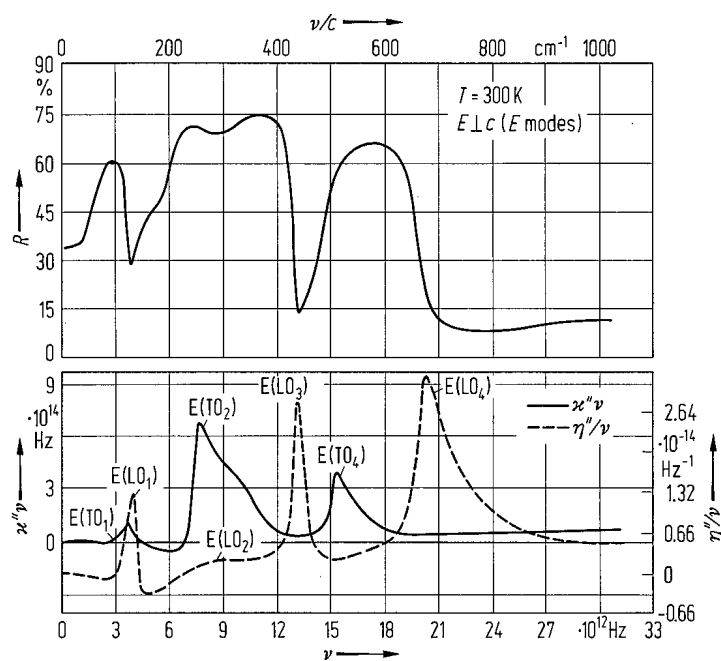


**Fig. 1A-11-036.**  $\text{PbTiO}_3$ .  $T$ ,  $R$  vs.  $\tilde{\nu}$  [64Per].  $T$ : transmittance,  $R$ : reflectance,  $\tilde{\nu}$ : wave number. The  $\text{PbTiO}_3$  concentrations of the samples are given in the figure.

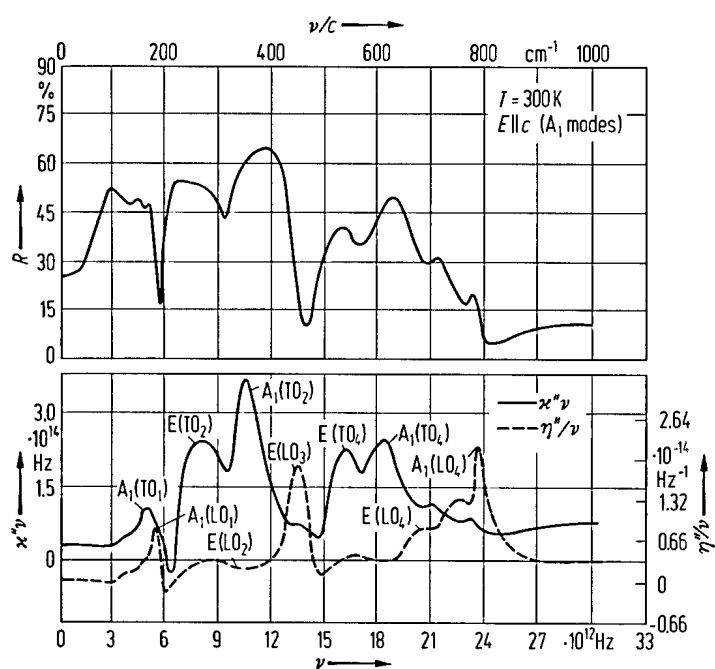


**Fig. 1A-11-037.**  $\text{PbTiO}_3$ .  $\kappa'$ ,  $\kappa''$  vs.  $\nu$  [64Per]. The curves were obtained from the reflectivity data using Kramers-Kronig relation.

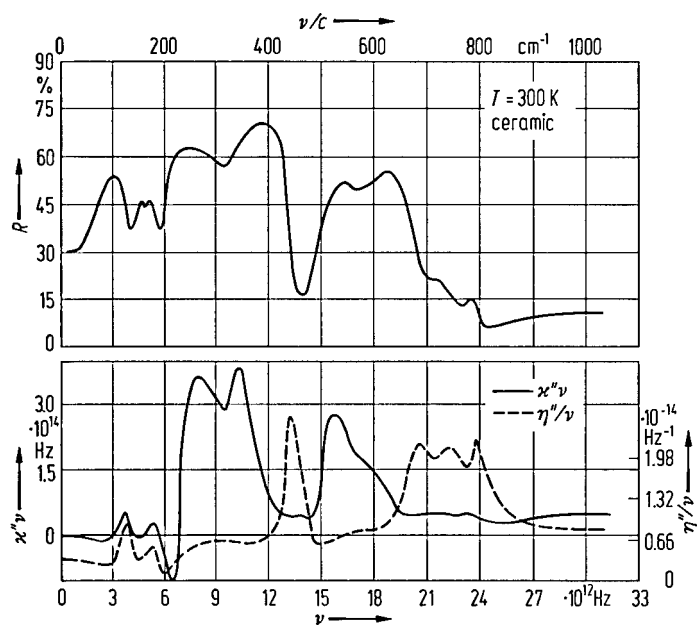




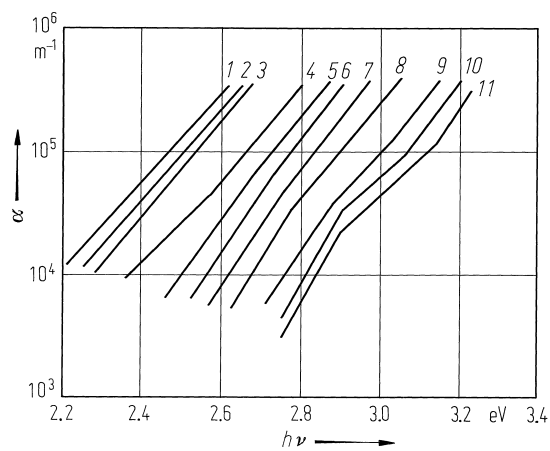
**Fig. 1A-11-038.**  $\text{PbTiO}_3$ . Infrared reflectance, conductivity, resistivity vs.  $\nu$  [70Tor].  $\eta''$ : imaginary part of  $1/\kappa$ . Polarization of the incident light is normal to the  $c$ -axis (E-modes).



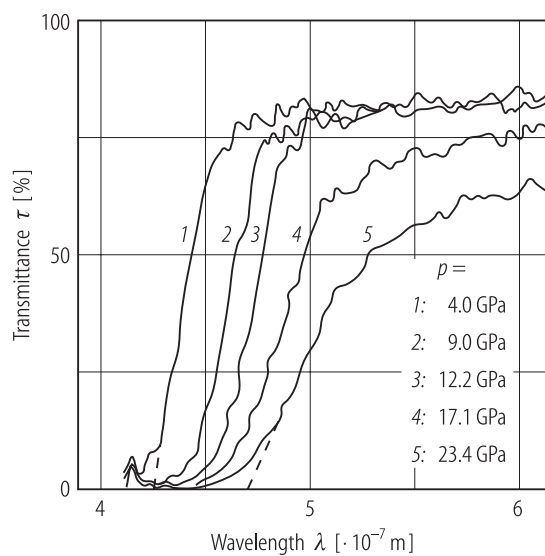
**Fig. 1A-11-039.** PbTiO<sub>3</sub>. Infrared reflectance, conductivity, resistivity vs.  $\nu$  [70Tor].  $\eta''$ : imaginary part of  $1/\kappa$ . Polarization of the incident light is parallel to the  $c$ -axis ( $A_1$ -modes).



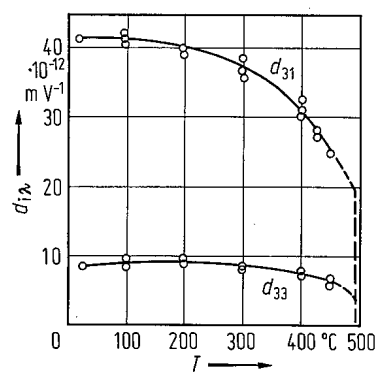
**Fig. 1A-11-040.**  $\text{PbTiO}_3$  (ceramics). Infrared reflectance, conductivity, resistivity vs.  $\nu$  [70Tor].  $\eta''$ : imaginary part of  $1/\kappa$ .



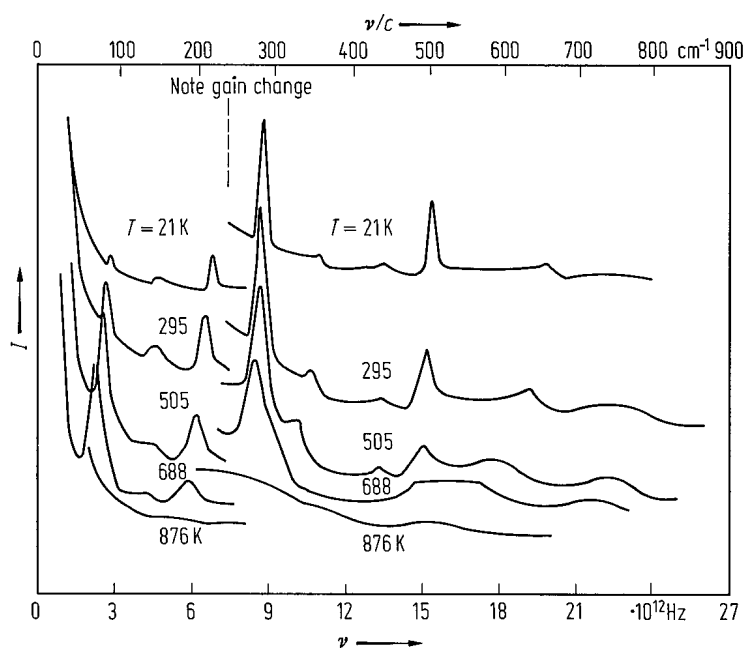
**Fig. 1A-11-041.** PbTiO<sub>3</sub>.  $\alpha$  vs.  $h\nu$  [87Zam].  $\alpha$ : optical absorption coefficient. Parameter:  $T$ . Curve 1: 856 °C, 2: 802 °C, 3: 760 °C, 4: 758 °C, 5: 680 °C, 6: 645 °C, 7: 589 °C, 8: 526 °C, 9: 425 °C, 10: 343 °C, 11: 293 °C.



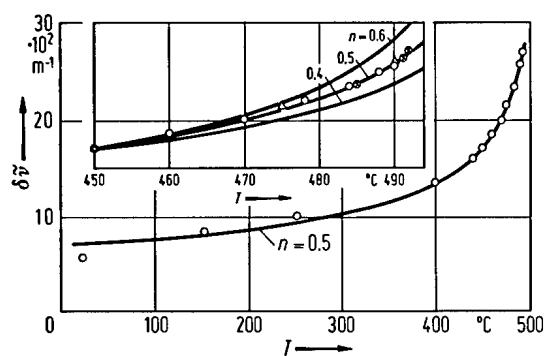
**Fig. 1A-11-042.**  $\text{PbTiO}_3$ . Transmittance  $\tau$  vs.  $\lambda$  [92Zha].  
 $\lambda$ : wavelength. Parameter:  $p$ .



**Fig. 1A-11-043.** PbTiO<sub>3</sub>.  $d_{i\lambda}$  vs.  $T$  [77Ber].  $d_{i\lambda}$ : nonlinear optical coefficient.



**Fig. 1A-11-044.**  $\text{PbTiO}_3$ . Raman spectra. Intensity vs.  $\nu$  [70Bur]. Parameter:  $T$ .

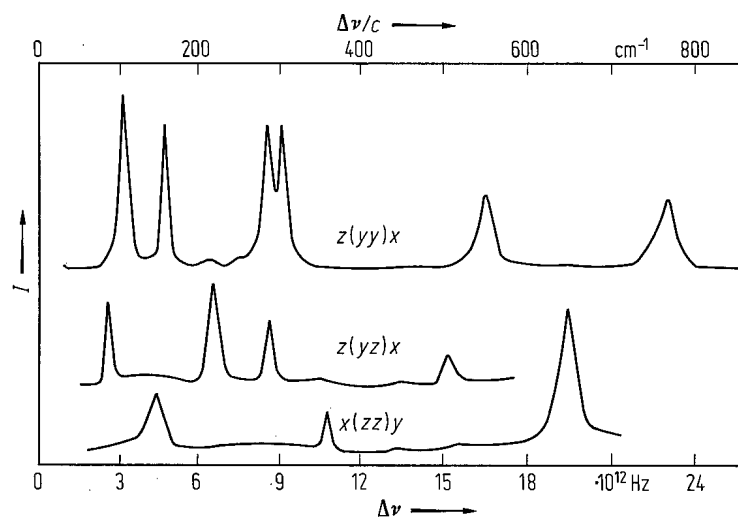


**Fig. 1A-11-045.** PbTiO<sub>3</sub>.  $\delta\tilde{\nu}$  vs.  $T$  [70Bur].  $\delta\tilde{\nu}$ : Raman line width. The solid lines show the theoretical curves:

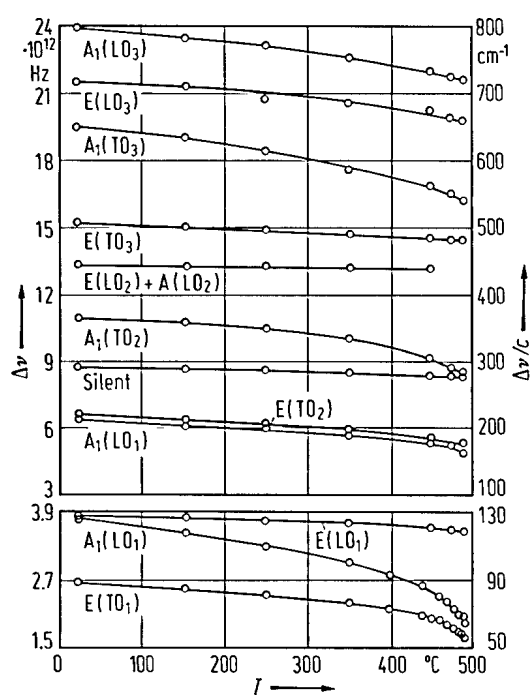
$$\delta\tilde{\nu} \propto \chi_{33}^n \propto \left[ \left( 1 - \frac{3}{4} \frac{T - \Theta_p}{\Theta_f - \Theta_p} \right) + \left( 1 - \frac{3}{4} \frac{T - \Theta_p}{\Theta_f - \Theta_p} \right)^{\frac{1}{2}} \right]^{-n}$$

the best agreement with the experimental points occurs at  $n = 0.5$ .

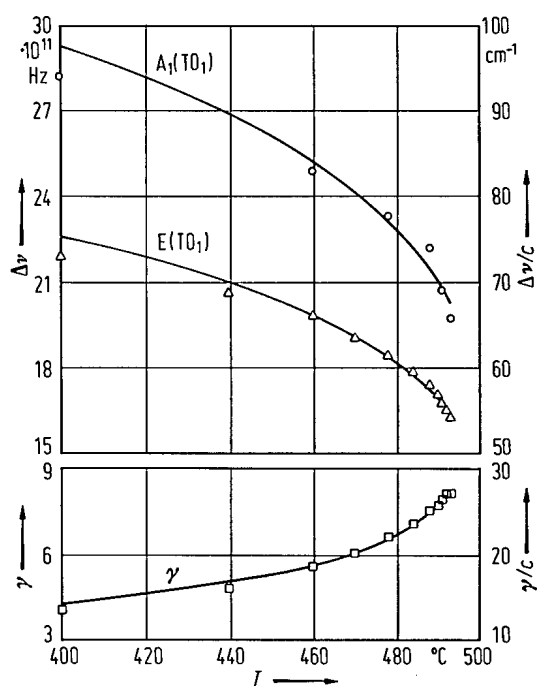




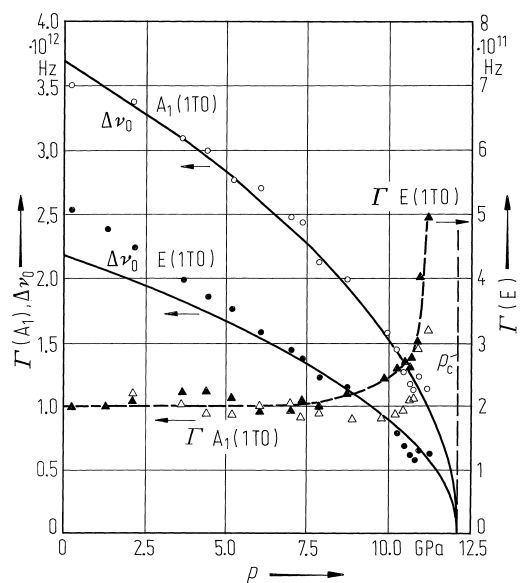
**Fig. 1A-11-046.**  $\text{PbTiO}_3$ .  $I$  vs.  $\Delta\nu$  at RT [70Bur].  $I$ : Raman scattering intensity. See also [73Fre].



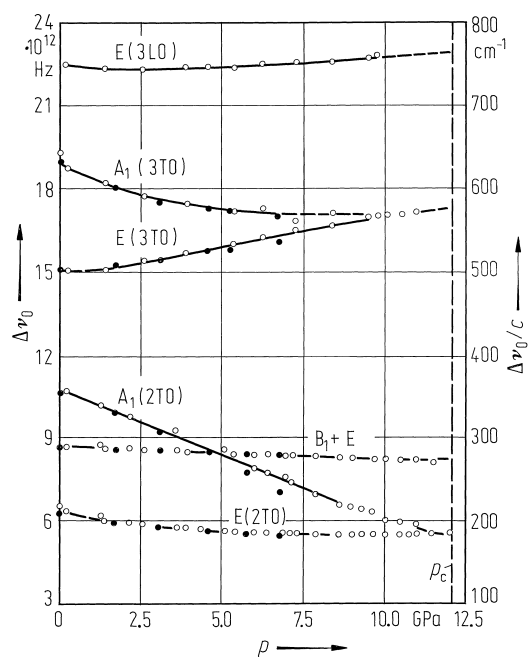
**Fig. 1A-11-047.** PbTiO<sub>3</sub>,  $\Delta\nu$  vs.  $T$  [73Bur].  $\Delta\nu$ : Raman frequency shift.



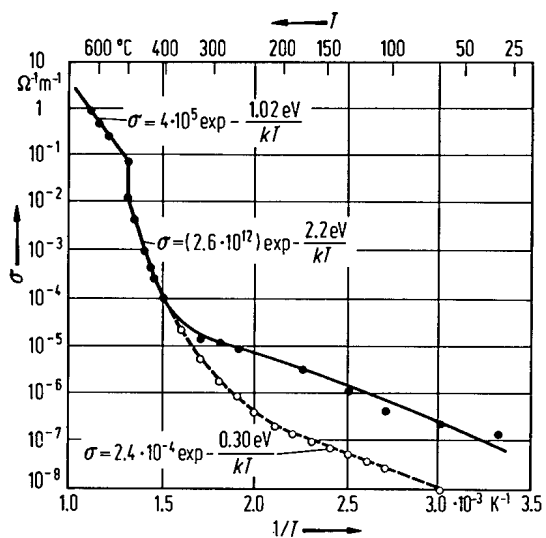
**Fig. 1A-11-048.**  $\text{PbTiO}_3$ .  $\Delta\nu$ ,  $\gamma$  vs.  $T$  [73Bur].  $\Delta\nu$ : Raman frequency shift of the lowest  $A_1$  mode (open circles) and E mode (open triangles).  $\gamma$ : damping constant of the the lowest E mode. Solid lines are calculated results on an empirical model. See also [70Bur].



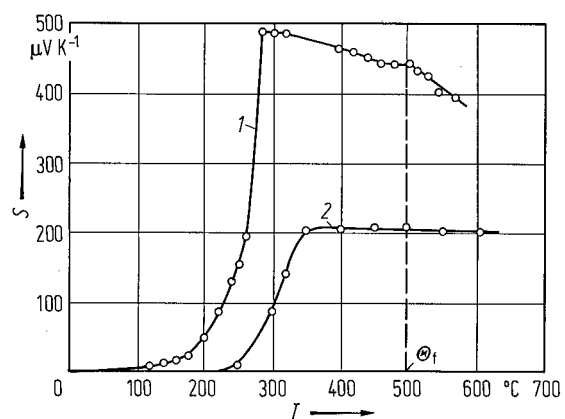
**Fig. 1A-11-049.** PbTiO<sub>3</sub>.  $\Delta\nu_0$ ,  $\Gamma$  vs.  $p$  [83San].  $\Delta\nu_0$ : Raman shift,  $\Gamma$ : damping constant for the A<sub>1</sub> mode. Open circles and triangles: Raman shift and damping constant for A<sub>1</sub> mode. Full circles and triangles: Raman shift and damping constant for the E mode.



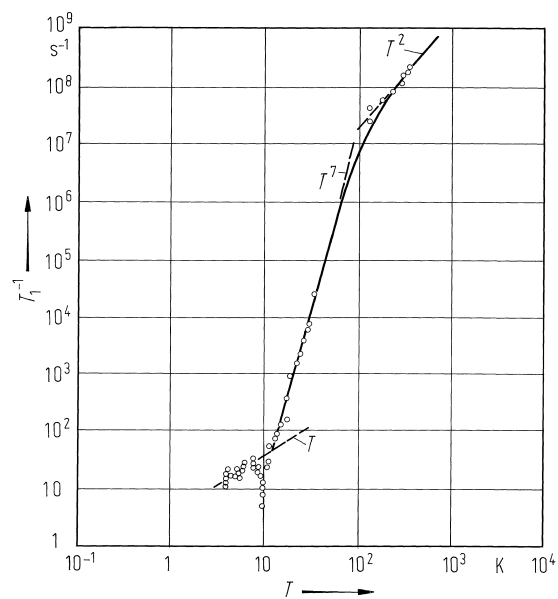
**Fig. 1A-11-050.**  $\text{PbTiO}_3$ ,  $\Delta\nu_0$  vs.  $p$  [83San].  $\Delta\nu_0$ : Raman shift. Full circles: data from [75Cer]. nTO and nLO: n-th TO and LO modes.



**Fig. 1A-11-051.** PbTiO<sub>3</sub>.  $\sigma$  vs.  $T^{-1}$  [70Rem]. Open circles: dc conductivity, full circles: conductivity measured at 1 kHz, which includes the dielectric loss.

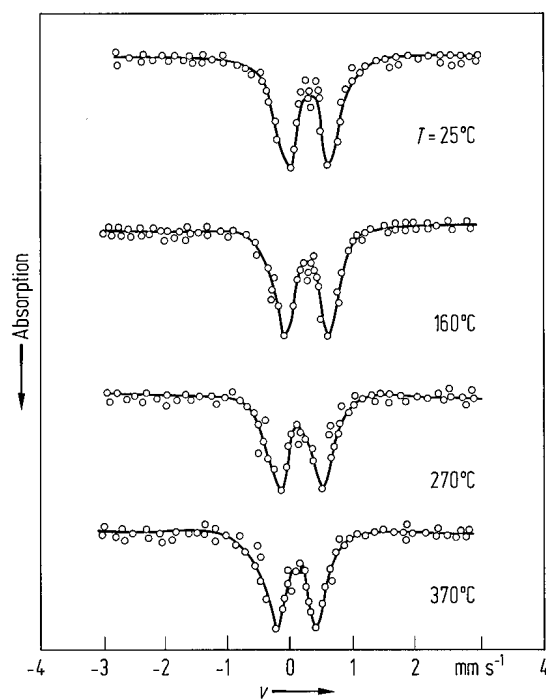


**Fig. 1A-11-052.**  $\text{PbTiO}_3$ .  $S$  vs.  $T$  [78Han].  $S$ : Seebeck coefficient. Curve 1: for single crystals highly reduced during the growing process, curve 2: for single crystals unreduced.

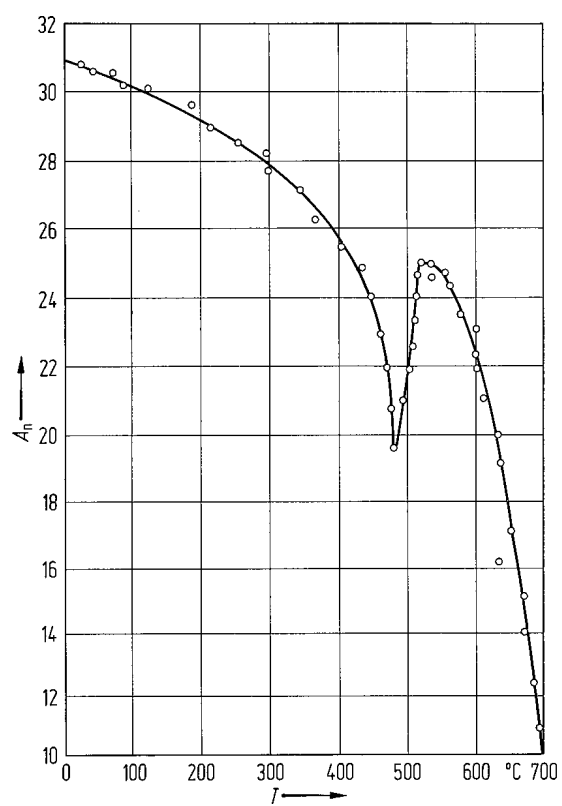


**Fig. 1A-11-053.** PbTiO<sub>3</sub> (ceramics).  $T_1^{-1}$  vs.  $T$  [86Bot].  $T_1$ : spin-lattice relaxation time.

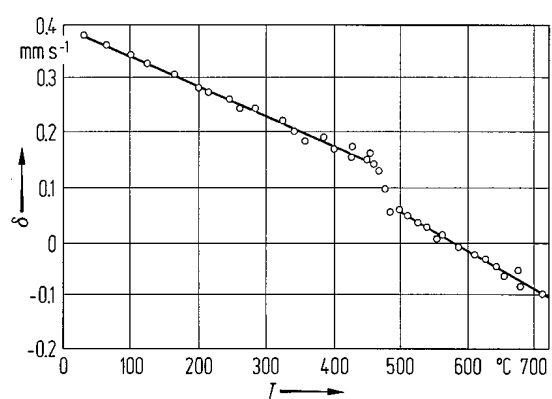




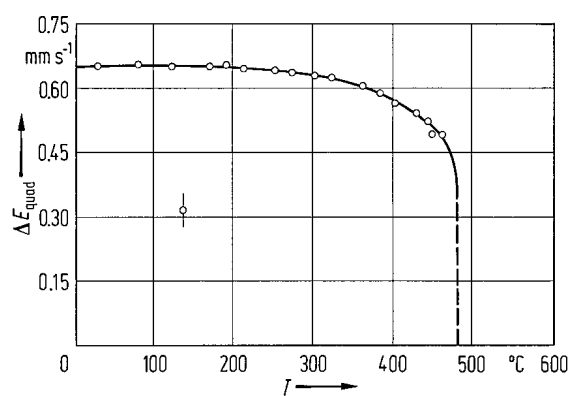
**Fig. 1A-11-054.**  $\text{PbTiO}_3 : {}^{57}\text{Co}$ . Mössbauer spectra [72Bhi]. Parameter:  $T$ .  $v$ : source velocity. The spectra matched against  $\text{K}_4\text{Fe}(\text{CN})_6 \cdot 3\text{H}_2\text{O}$  single crystal absorber.



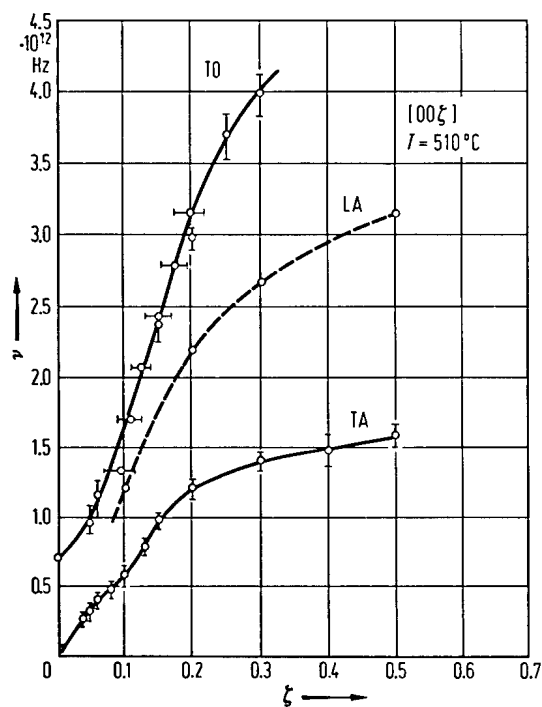
**Fig. 1A-11-055.**  $\text{PbTiO}_3$  :  $^{57}\text{Co}$ .  $A_n$  vs.  $T$  [72Bhi].  
 $A_n$ : normalized area of Mössbauer absorption spectrum.



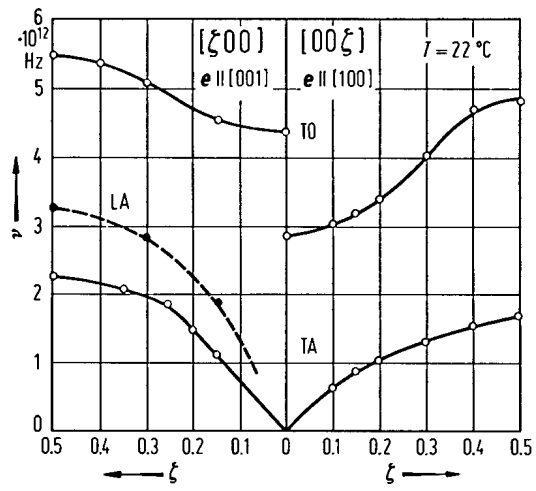
**Fig. 1A-11-056.**  $\text{PbTiO}_3 : ^{57}\text{Co}$ .  $\delta$  vs.  $T$  [72Bhi].  $\delta$ : isomer shift. The values are relative to 310 ESS absorber.



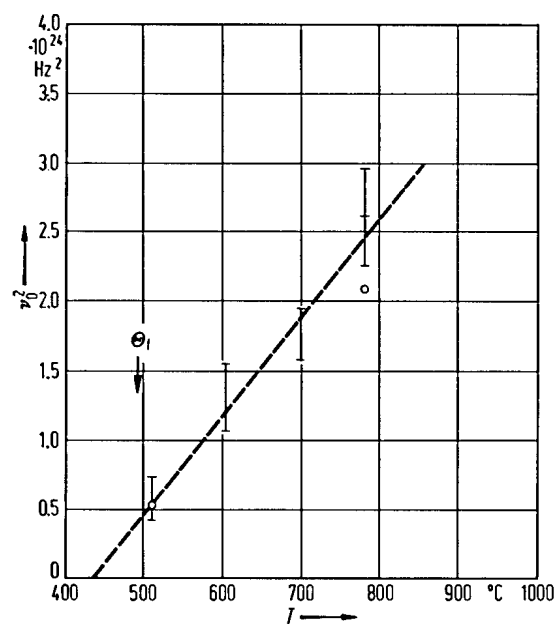
**Fig. 1A-11-057.**  $\text{PbTiO}_3 : ^{57}\text{Co}$ .  $\Delta E_{\text{quad}}$  vs.  $T$  [72Bhi].  
 $\Delta E_{\text{quad}}$ : quadrupole splitting.



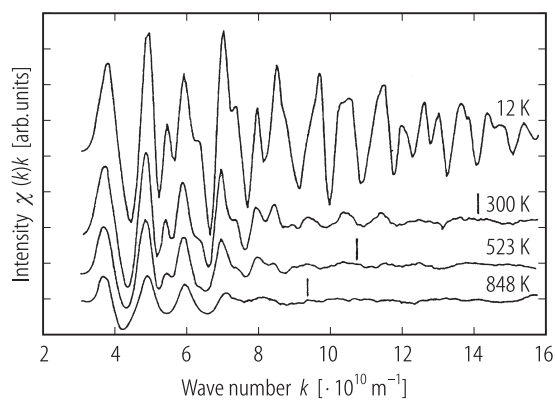
**Fig. 1A-11-058.**  $\text{PbTiO}_3$ .  $\nu$  vs.  $\zeta$  at 22 °C [70Shi].  $\nu$ : Phonon frequency,  $\zeta$ : reduced wave vector coordinate. A single domain crystal was used, which was produced by the topseeded solution technique.



**Fig. 1A-11-059.** PbTiO<sub>3</sub>.  $\nu$  vs.  $\zeta$  at 22 °C [70Shi].  $e$ : polarization vector.  $\nu$ : Phonon frequency,  $\zeta$ : reduced wave vector coordinate.

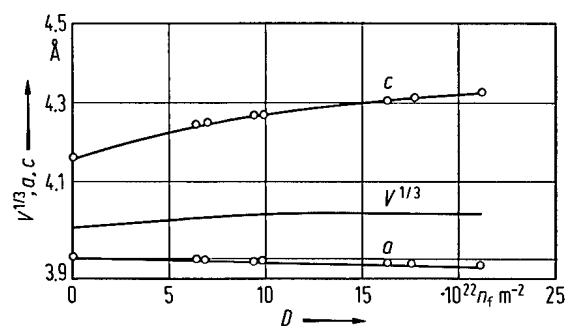


**Fig. 1A-11-060.**  $\text{PbTiO}_3$ ,  $\nu_0^2$  vs.  $T$  [70Shi].  $\nu_0$ : frequency of lowest energy optic phonon at  $q = 0$ .



**Fig. 1A-11-061.** PbTiO<sub>3</sub>.  $\chi(k)$   $k$  vs.  $k$  [94Sic].  $\chi(k)$   $k$ : intensity (Pb edge),  $k$ : wave number.





**Fig. 1A-11-062.**  $\text{PbTiO}_3$ . Lattice parameters vs. neutron dose,  $D$  [66Hau].  $n_f$ : number of fast neutrons.

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