

Fig. 2A-1-070. LiNbO₃. $-p_3$, Δn vs. x [91Gra]. x : Li₂O content.

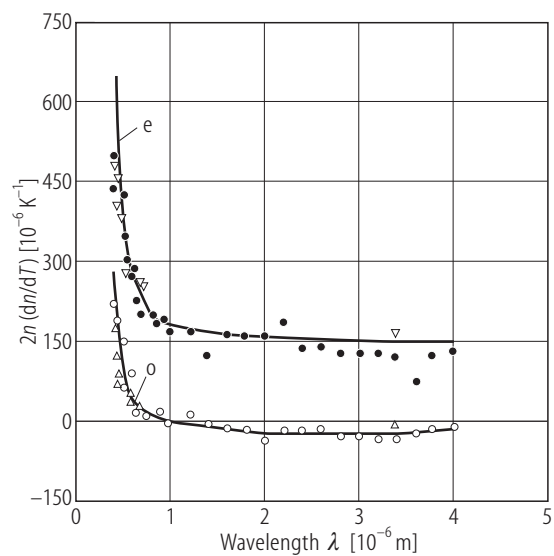


Fig. 2A-1-071. LiNbO₃. $2n(dn/dT)$ vs. λ [94Gho]. o: ordinary; e: extraordinary.

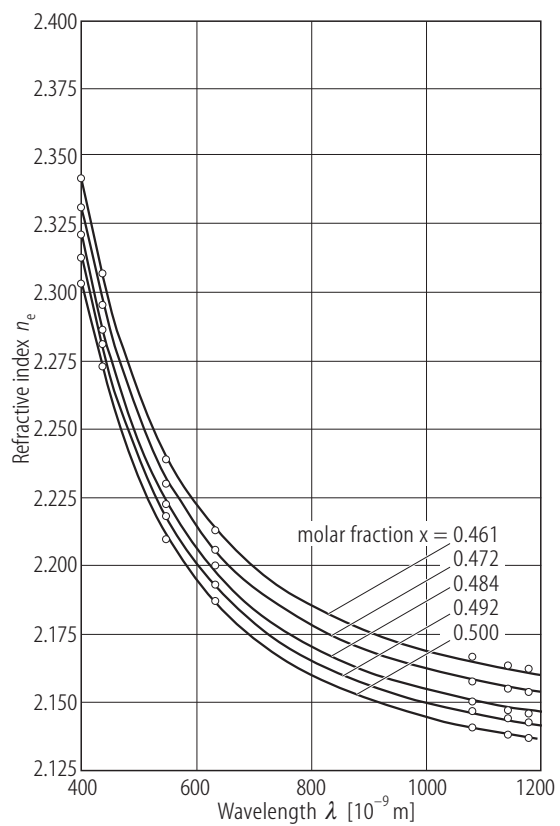


Fig. 2A-1-072. LiNbO₃. n_e vs. λ [93Sch1]. Parameter x : Li content.

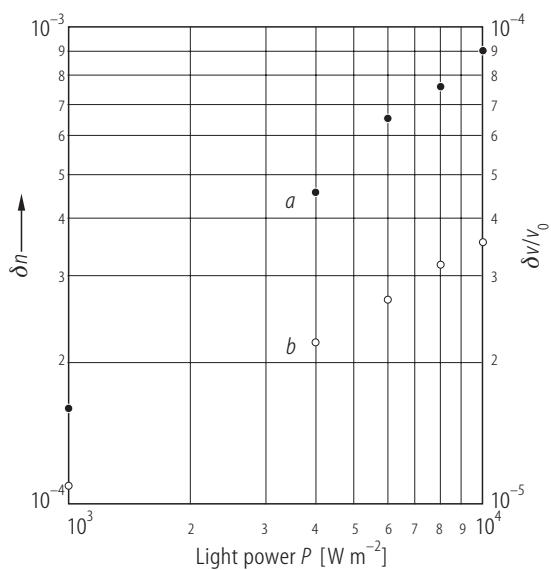


Fig. 2A-1-073. LiNbO₃. δn , $\delta v/v_0$ vs. P [92Vla]. δn : change in refractive index; $\delta v/v_0$: fractional change in ultrasound velocity; P : light power for sample illumination.

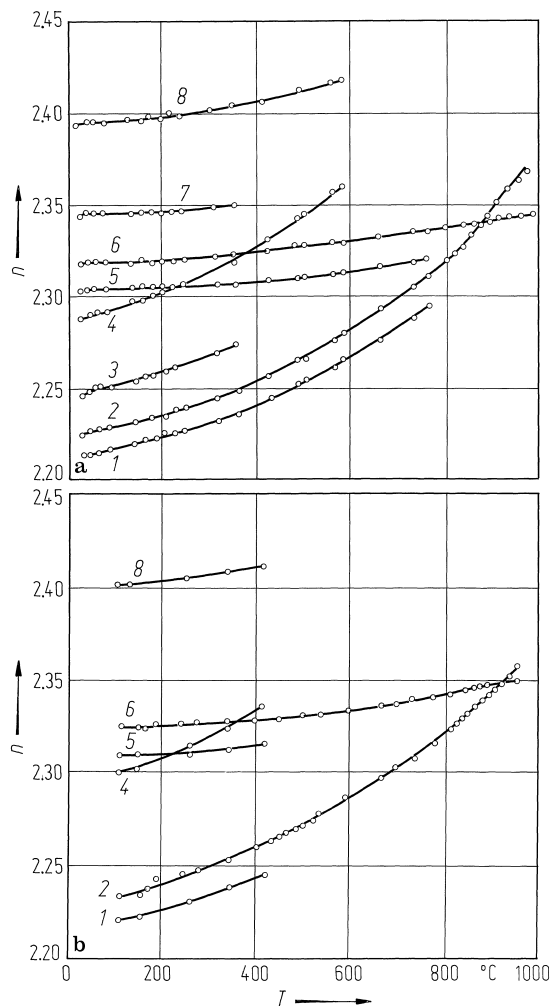


Fig. 2A-1-074. LiNbO₃, n vs. T [79Sir]. (a) pure LiNbO₃, (b) LiNbO₃ with the addition of 0.1 mol % Ti. Curves 1...4: n_e , 5...8: n_o . 1, 5: $\lambda = 585$ nm; 2, 6: $\lambda = 546$ nm; 3, 7: $\lambda = 480$ nm; 4, 8: $\lambda = 405$ nm.

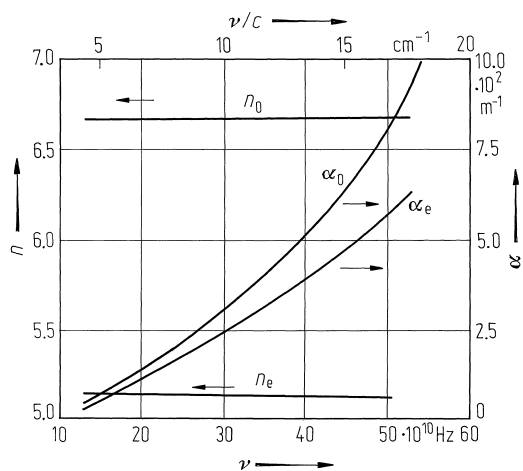


Fig. 2A-1-075. LiNbO₃, n , α vs. ν in the near millimeter wave region [83Bro]. α : power absorption coefficient.

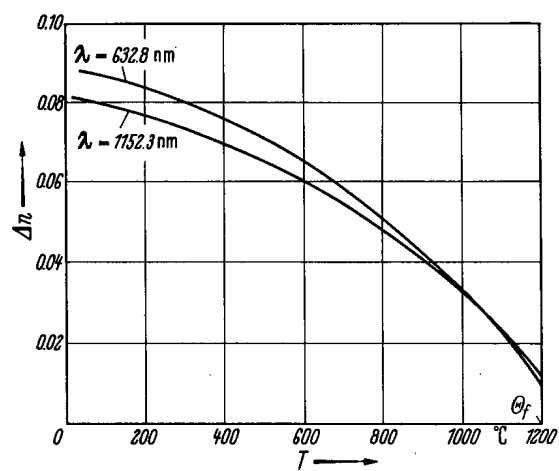


Fig. 2A-1-076. LiNbO₃, Δn vs. T [66War]. $\Delta n = n_o - n_e$.
Parameter: λ .

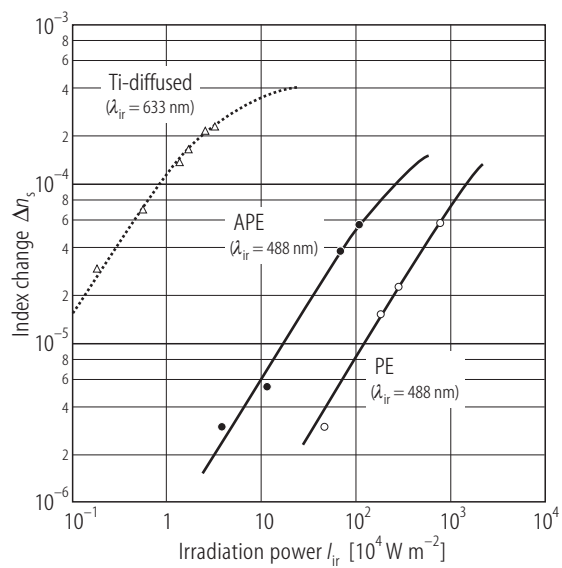


Fig. 2A-1-077. LiNbO₃. Δn_s vs. I_{ir} [93Fuj]. Δn_s : saturated refractive index change; I_{ir} : irradiation laser power. Ti-diffused: Ti-diffused sample; PE: proton-exchanged sample; APE: annealed proton-exchanged sample.

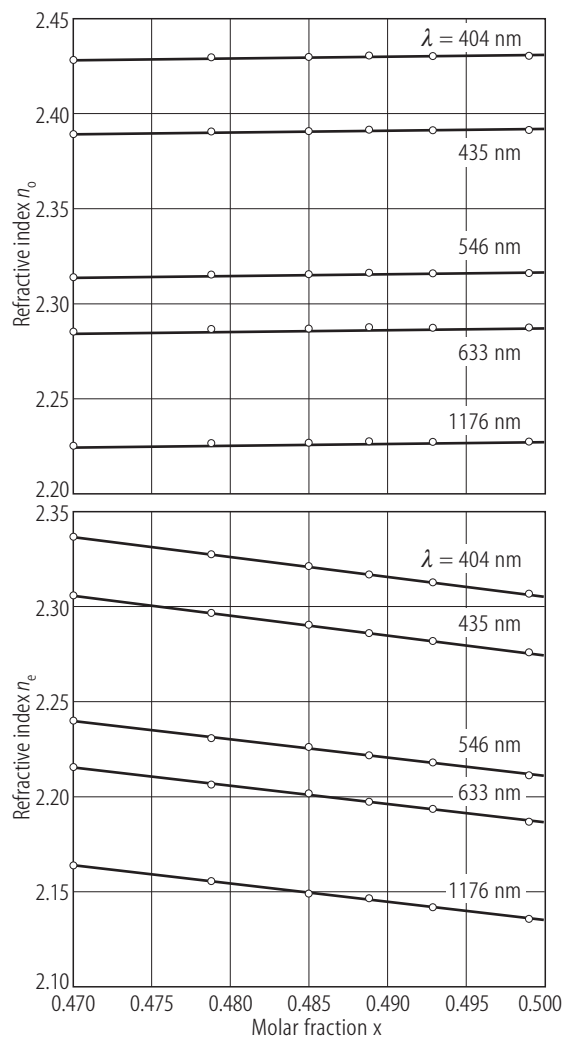


Fig. 2A-1-078. LiNbO₃. n_o , n_e vs. x [94Sch]. x : Li₂O content. Parameter: λ .

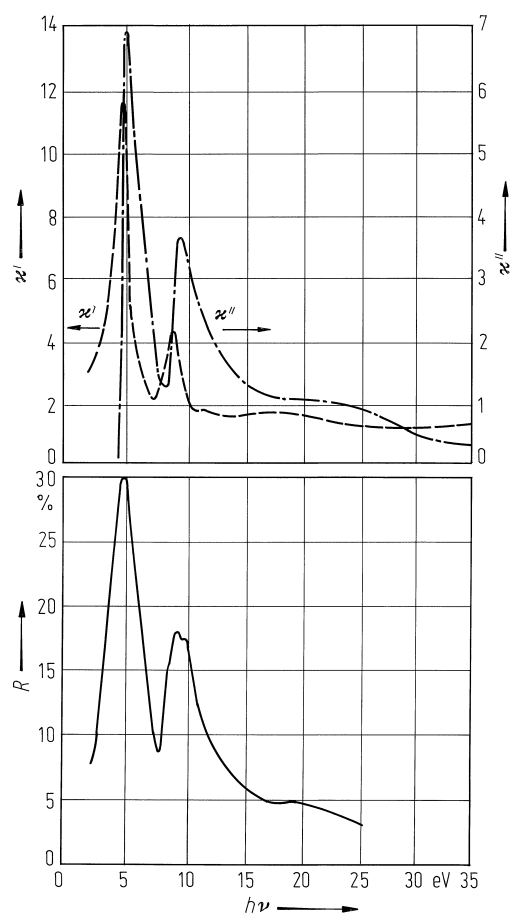


Fig. 2A-1-079. LiNbO₃. R , κ' , κ'' vs. $h\nu$ at 300 K [83Mam]. $h\nu$: photon energy. R : reflectivity.

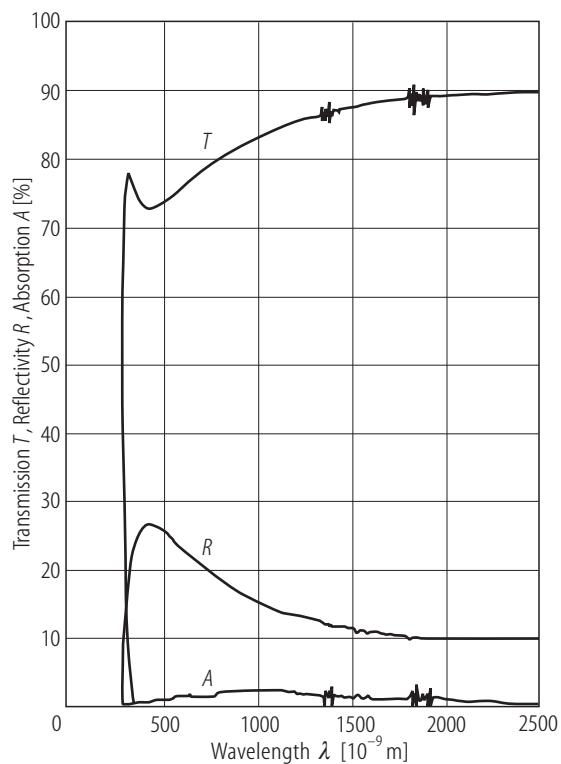


Fig. 2A-1-080. LiNbO₃ (thin film). *T*, *R*, *A* vs. λ [92Ben].
Sample: thin film prepared by RF sputtering. *T*, *R*, *A*: transmission, reflectivity and absorption of the light. $d = 85$ nm.

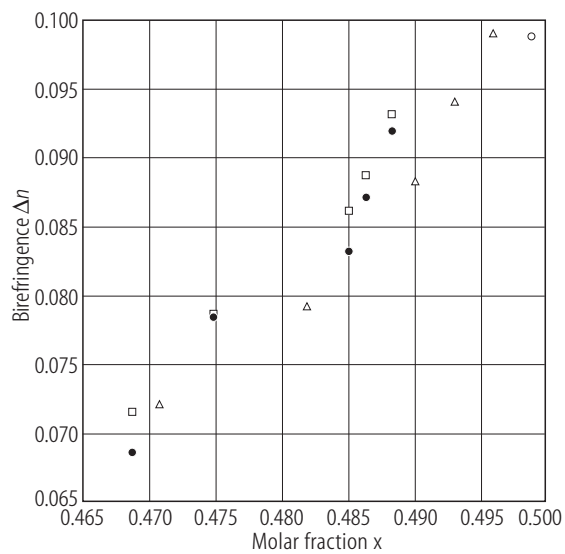


Fig. 2A-1-081. LiNbO₃. Δn vs. x [92Sch]. Δn : $n_o - n_e$; x : Li₂O content. Open circles, squares, triangles, full circles: samples of different origins.

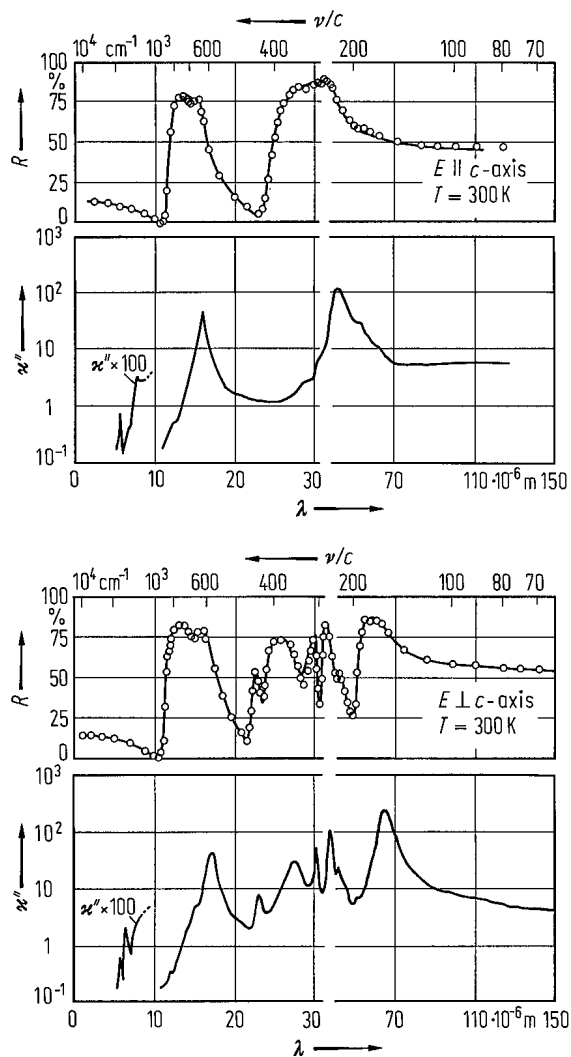


Fig. 2A-1-082. LiNbO₃. R , κ'' vs. λ [67Bar]. R : reflectivity. κ'' for $\lambda > 10 \cdot 10^{-6}$ m results from Kramers-Kronig relations; for $10 \cdot 10^{-6} \text{ m} > \lambda > 5 \cdot 10^{-6} \text{ m}$, κ'' is deduced from transmission measurement.

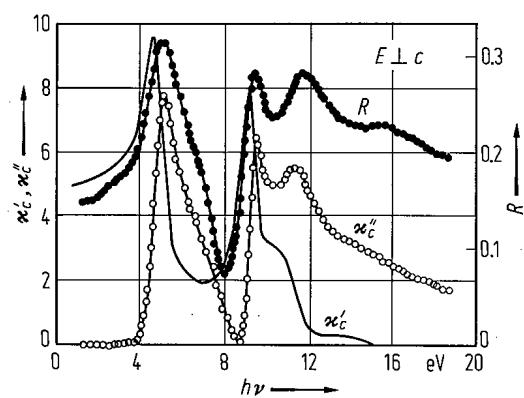


Fig. 2A-1-083. LiNbO₃. R , κ'_c , κ''_c vs. $h\nu$ [75Vis]. R : reflectivity.

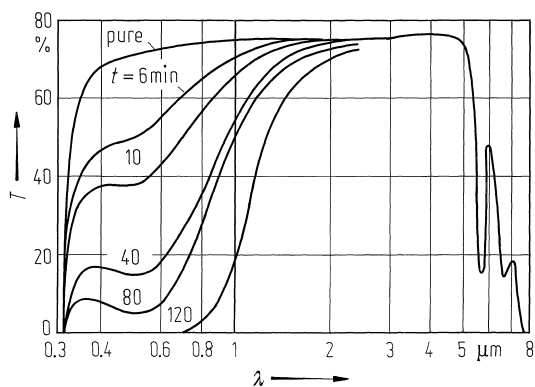


Fig. 2A-1-084. LiNbO₃ (pure and reduced). T vs. λ [86Jha].
 T : transmission. Parameter: heat treatment time in H_2 at 830 K. Sample thickness is 50 μm .

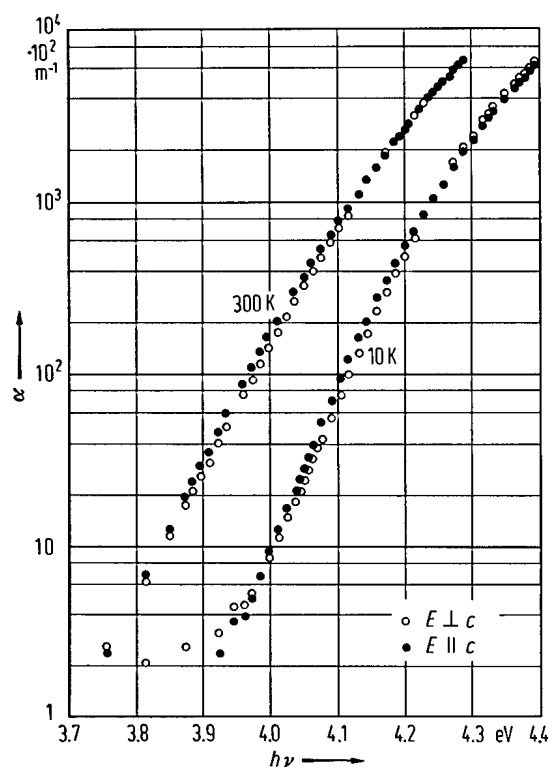


Fig. 2A-1-085. LiNbO₃. α vs. $h\nu$ [75Wea]. α : optical absorption coefficient. The crystal was grown from a congruent melt. See also [74Red].

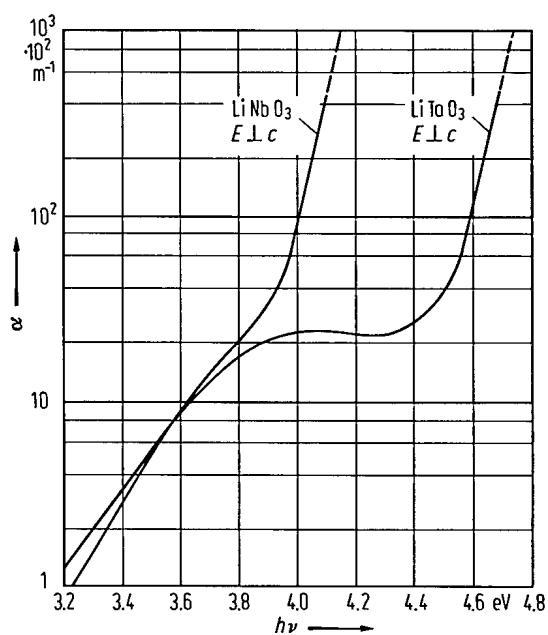


Fig. 2A-1-086. LiNbO₃, LiTaO₃. α vs. $h\nu$ [74Kas]. α : optical absorption coefficient.

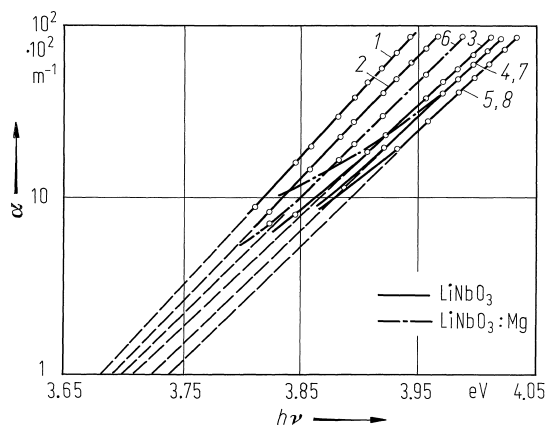


Fig. 2A-1-087. LiNbO₃ (undoped and MgO-doped). α vs. $h\nu$ [86Fen]. α : optical absorption coefficient. $h\nu$: photon energy. Parameter: composition. Curve 1: [Li] / [Nb] = 0.90. 2: 0.945. 3: 1.00. 4: 1.04. 5: 1.083. 6: 0.945. 7: 1.00. 8: 1.04. Curves 6, 7 and 8: additionally 1.0 mol % MgO.

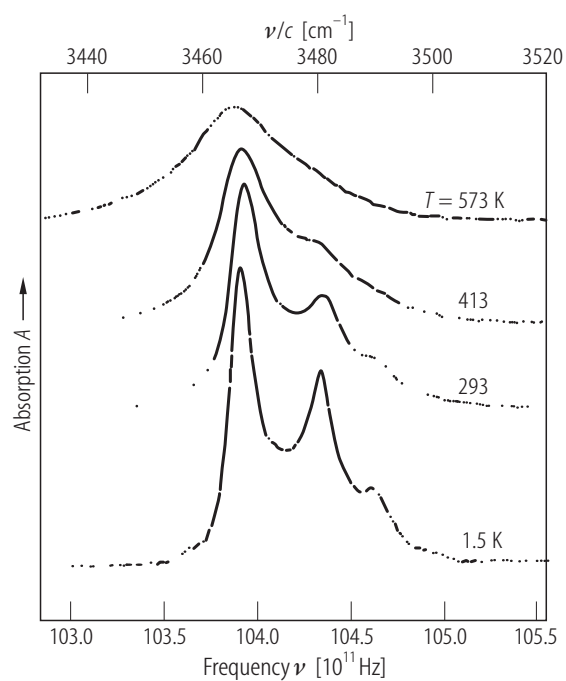


Fig. 2A-1-088. LiNbO₃. A vs. ν [91Kov]. A : absorption. OH absorption bands. Parameter: T .

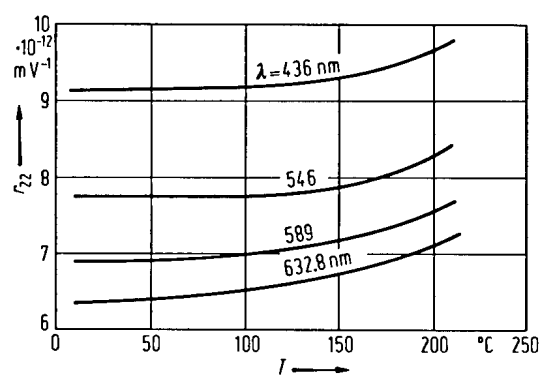


Fig. 2A-1-089. LiNbO₃. r_{22} vs. T [68Iwa]. Parameter: λ .

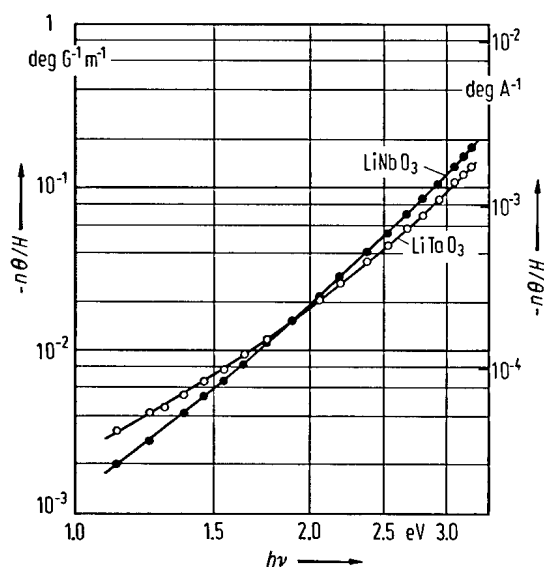


Fig. 2A-1-090. LiNbO₃, LiTaO₃. $-n\theta/H$ vs. $h\nu$ [74Kas]. n : refractive index, θ : Faraday rotation per unit length, H : magnetic field strength.

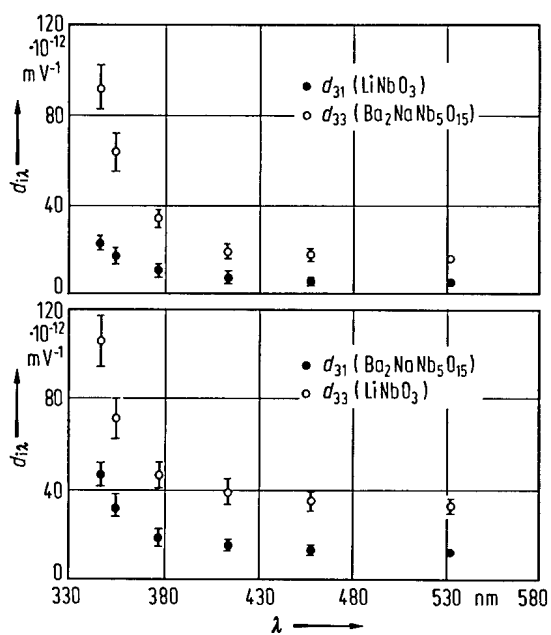


Fig. 2A-1-091. LiNbO₃, Ba₂NaNb₅O₁₅. $d_{i\lambda}$ vs. λ [76Dor].
 $d_{i\lambda}$: nonlinear optical susceptibility.

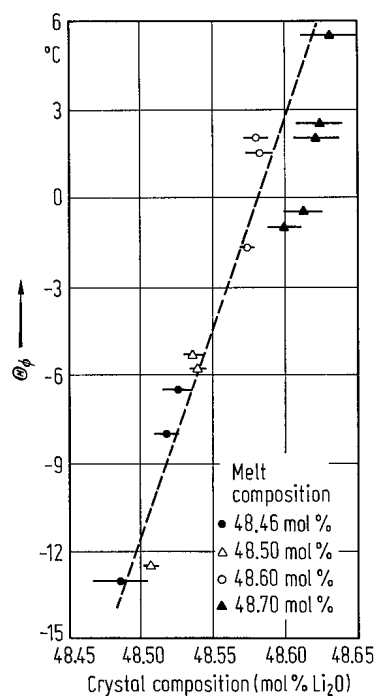


Fig. 2A-1-092. LiNbO₃. Θ_ϕ vs. crystal composition [74Cho]. Θ_ϕ : phase matching temperature in optical second harmonic generation.

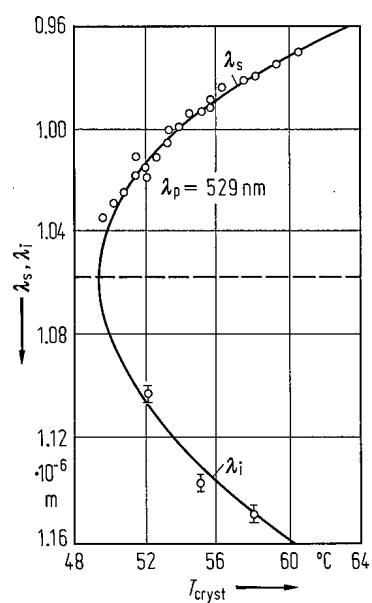


Fig. 2A-1-093. LiNbO₃. λ_s , λ_i vs. T_{cryst} [65Gio]. λ_s , λ_i : wavelengths of signal and idler light in parametric oscillation. T_{cryst} : phase matching crystal temperature, λ_p : wavelength of pump light obtained as the second harmonic of a Nd-doped CaWO₄ laser.

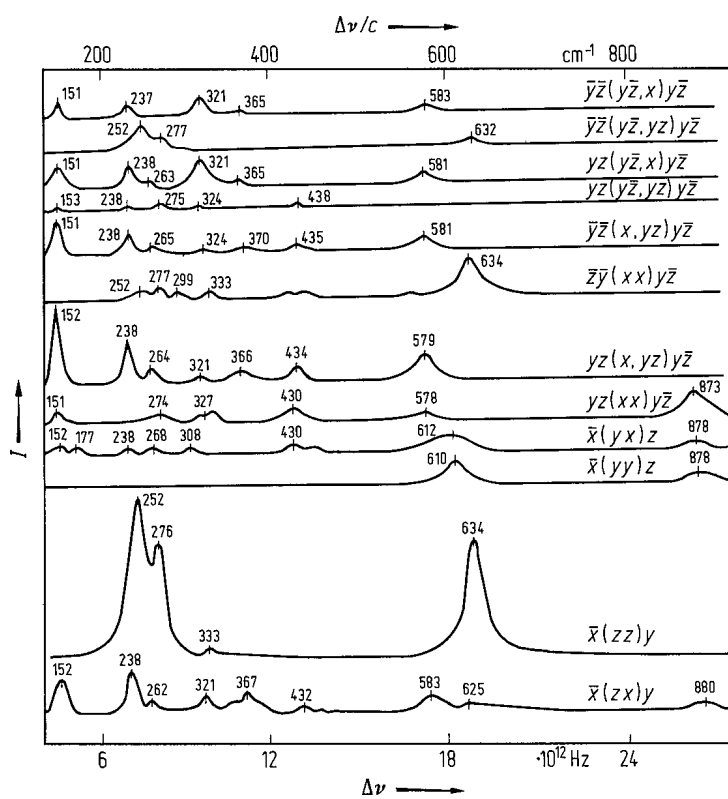


Fig. 2A-1-094. LiNbO₃. I vs. $\Delta\nu$ at RT [67Bar]. I : Raman scattering intensity, $\Delta\nu$: frequency shift. The numbers in the figure indicate the peak frequencies in the units of cm^{-1} . See also [66Sch].

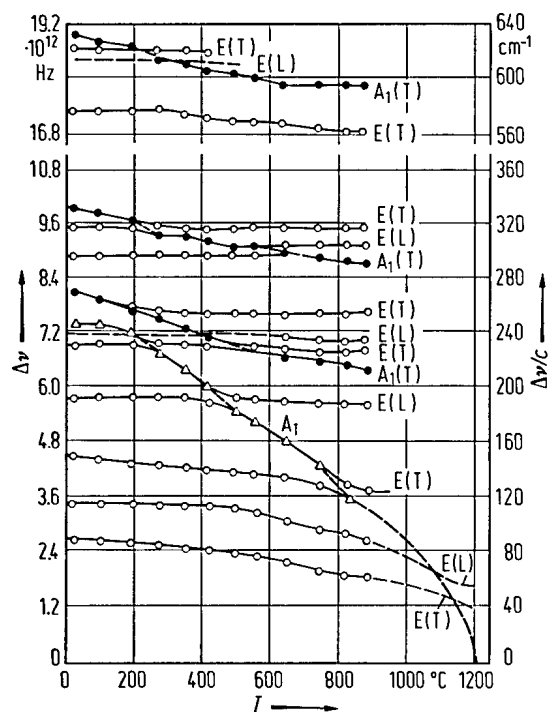


Fig. 2A-1-095. LiNbO₃. $\Delta\nu$ vs. T [68Joh]. $\Delta\nu$: Raman frequency shift. Triangles indicate the soft-mode peak. Broken lines are mere extrapolation of the data. For the assignment of the modes, see Table 2A-1-023.

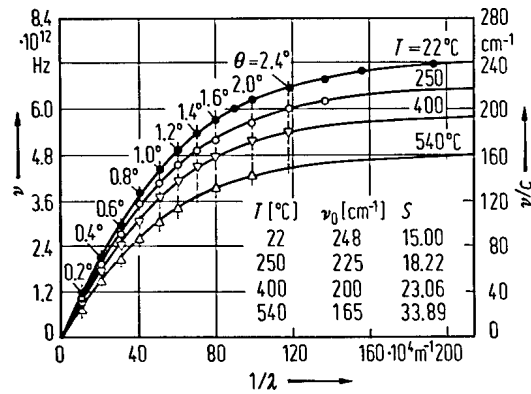


Fig. 2A-1-096. LiNbO₃. ν vs. $1/\lambda$ [72Rok]. ν : frequency of polariton associated with the 248 cm^{-1} soft mode, $1/\lambda$: wave number of polariton. θ : scattering angle. The mode frequency ν_0 and the mode strength S are also indicated in the figure.

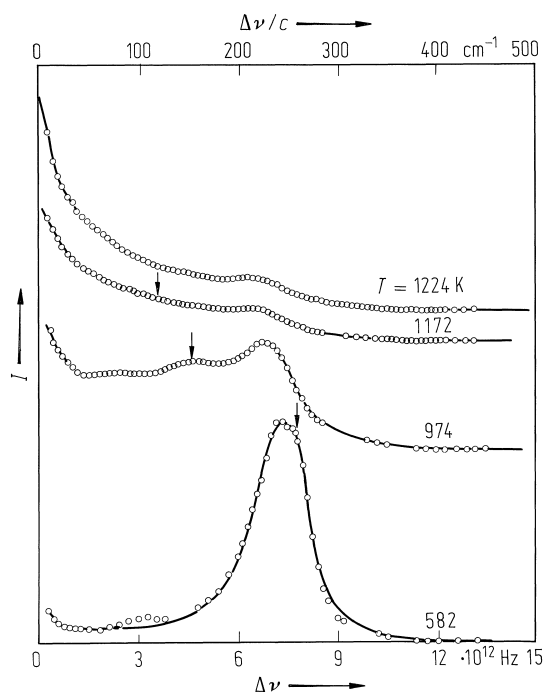


Fig. 2A-1-097. LiNbO₃. I vs. $\Delta\nu$ [85Oka]. I : Raman scattering intensity for cc polarizability, $\Delta\nu$: Raman scattering frequency shift. Parameter: T . Arrow on each trace marks the position of soft mode frequency.

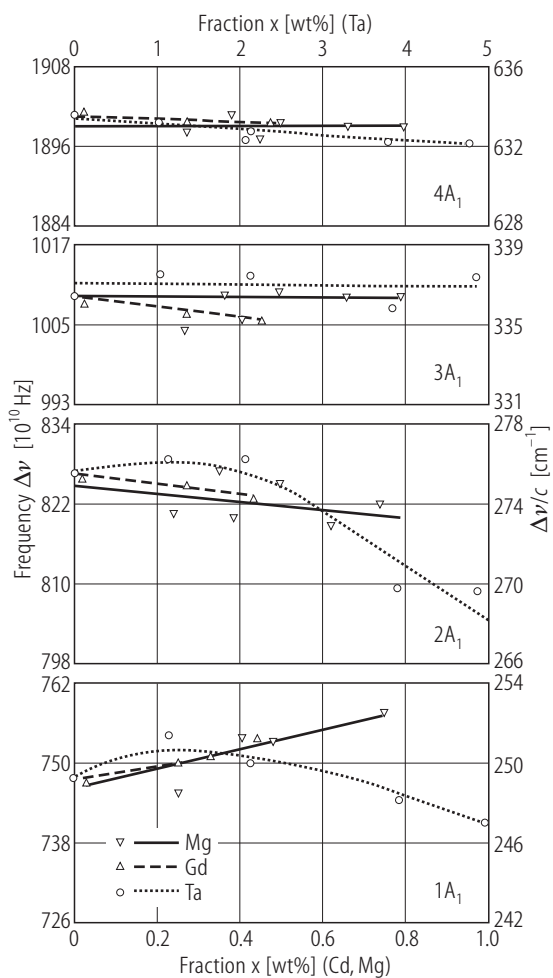


Fig. 2A-1-098. LiNbO₃ (Mg, Gd, Ta-doped). $\Delta\nu$ vs. x [94Sid]. $\Delta\nu$: Raman shift. x : wt% concentration.

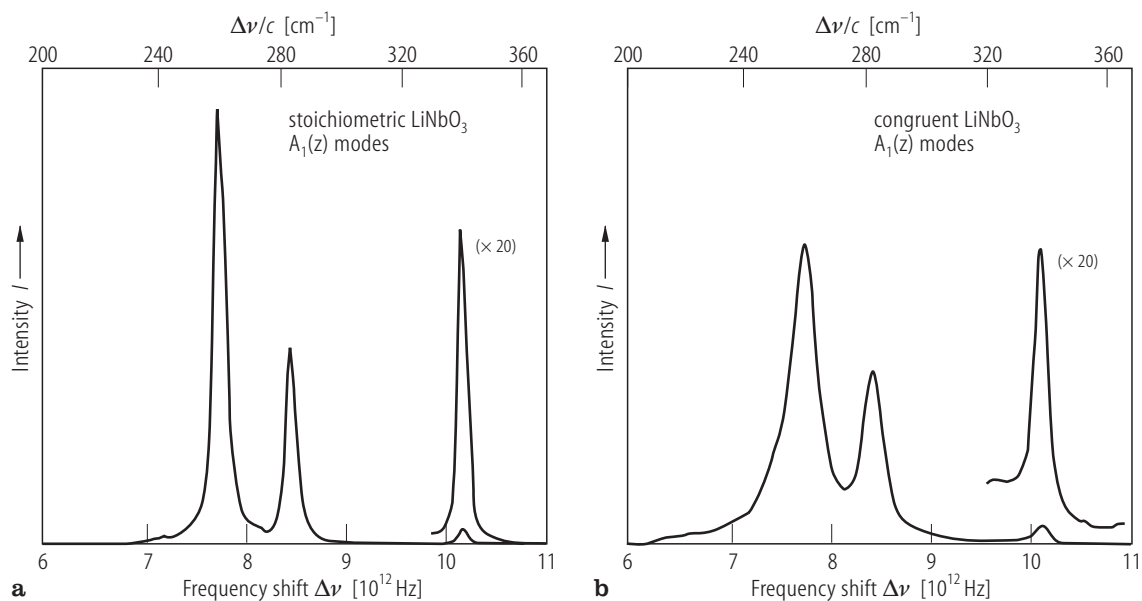


Fig. 2A-1-099. LiNbO₃. Raman spectra of A₁(z) modes [94Koj]. (a): stoichiometric; (b): congruent crystal. *T* = 95 K.

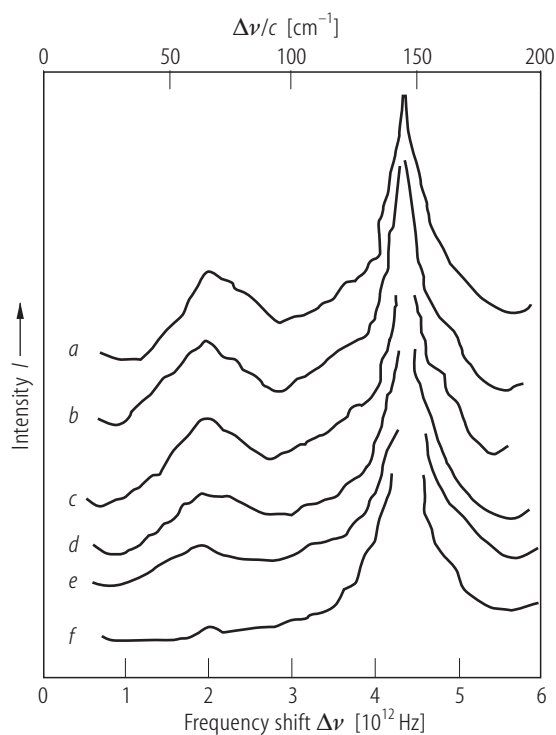


Fig. 2A-1-100. LiNbO₃. Raman spectra from a proton-exchanged waveguide [92Sav]. Curve *a*: after the proton-exchange, *b*, *c*, *d*, *e*: after consecutive polishings of very thin layers from the surface, *f*: bulk spectrum.

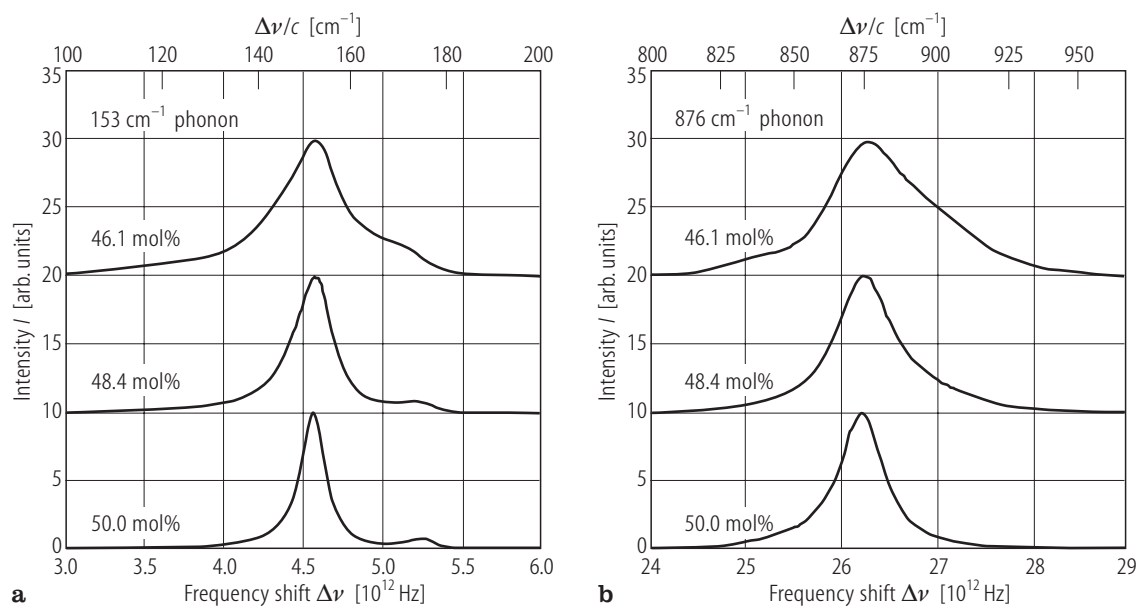


Fig. 2A-1-101. LiNbO₃. I vs. $\Delta\nu$ [93Sch2]. I : Raman intensity. $\Delta\nu$: frequency shift. Parameter: Li content [mol%].

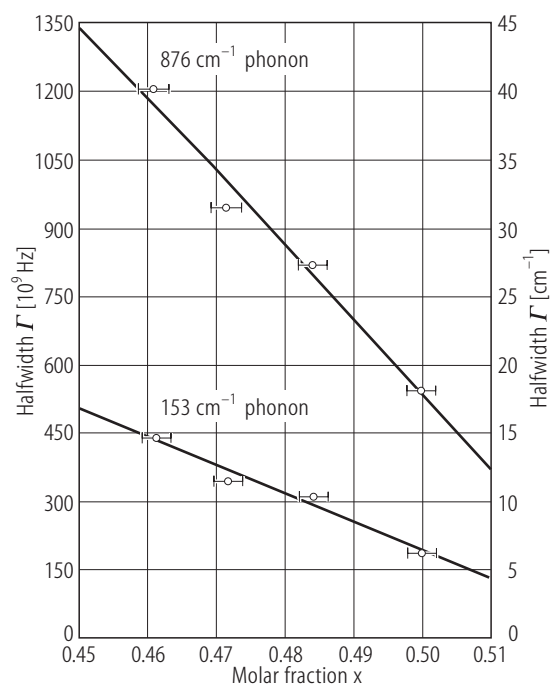


Fig. 2A-1-102. LiNbO₃. Γ vs. x [93Sch2]. Γ : halfwidths of the Raman lines at 153 cm^{-1} and 876 cm^{-1} . x : Li content.

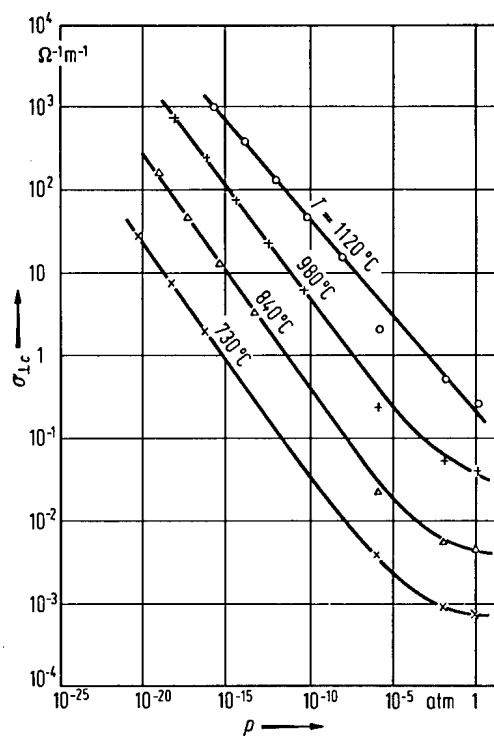


Fig. 2A-1-103. LiNbO₃. σ_{Lc} vs. p [68Ber]. p : oxygen partial pressure of ambient gas. 1 atm = 101325 Pa. Parameter: T .

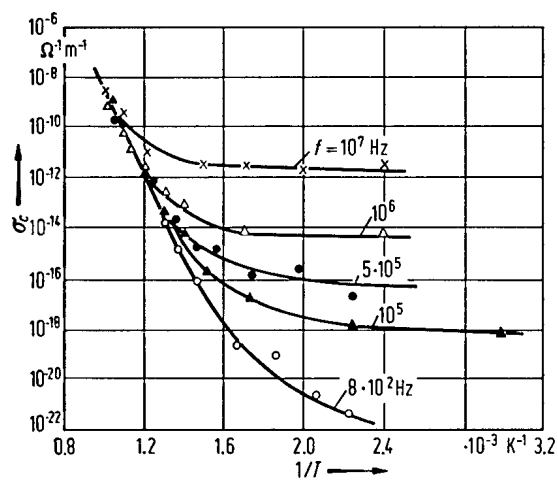


Fig. 2A-1-104. LiNbO_3 . σ_c vs. $1/T$ [71Glu]. Parameter: f .

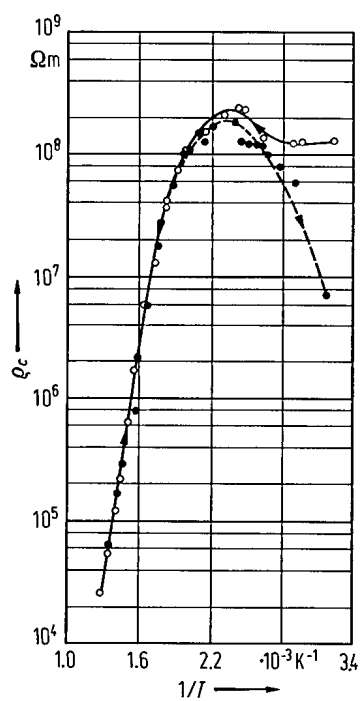


Fig. 2A-1-105. LiNbO₃. ρ_c vs. $1/T$ [78Kau].

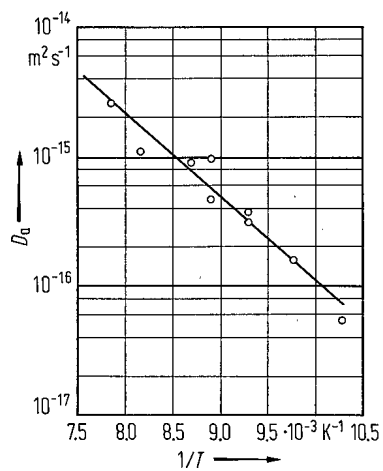


Fig. 2A-1-106. LiNbO₃. D_a vs. $1/T$ [69Jor]. D_a : oxygen diffusion coefficient along the a -axis. $D_c \approx D_a$.

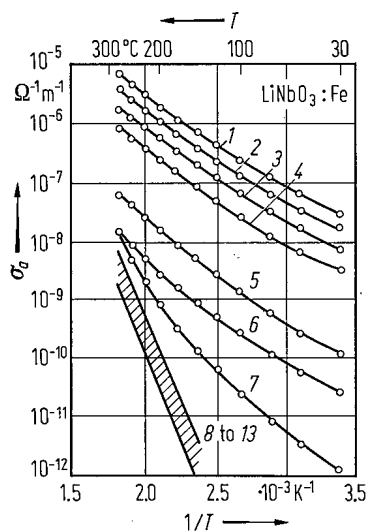


Fig. 2A-1-107. LiNbO₃. σ_a vs. $1/T$ [77Bar]. σ_a : conductivity of Fe doped single-domain crystals along the a -axis. Parameter: concentration of Fe dopants in wt %. 1: 0.75, 2: 0.65, 3: 0.5, 4: 0.44, 5: 0.3, 6: 0.25, 7: 0.18, 8: 0.1, 9: 0.08, 10: 0.057, 11: 0.018, 12: 0.014, 13: 0.004.

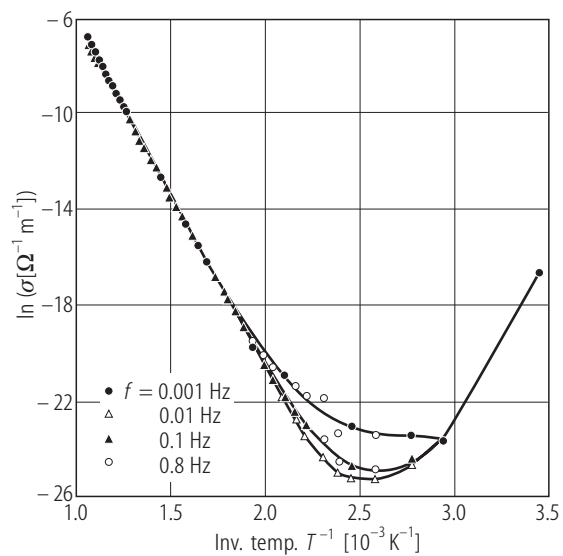


Fig. 2A-1-108. LiNbO₃. $\ln \sigma$ vs. $1/T$ [92Bak1]. σ : conductivity of a cleaved crystal. Parameter: f .

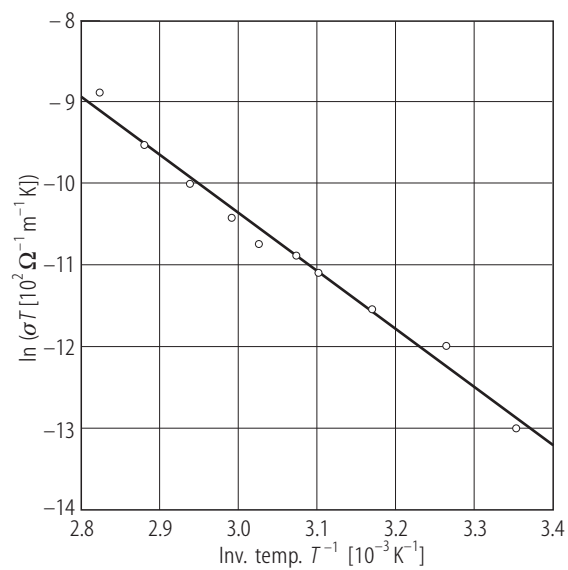


Fig. 2A-1-109. LiNbO₃ (thin film). $\ln(\sigma T)$ vs. $1/T$ [92Ben].
Sample: thin film prepared by RF sputtering.
 σ : electric conductivity.

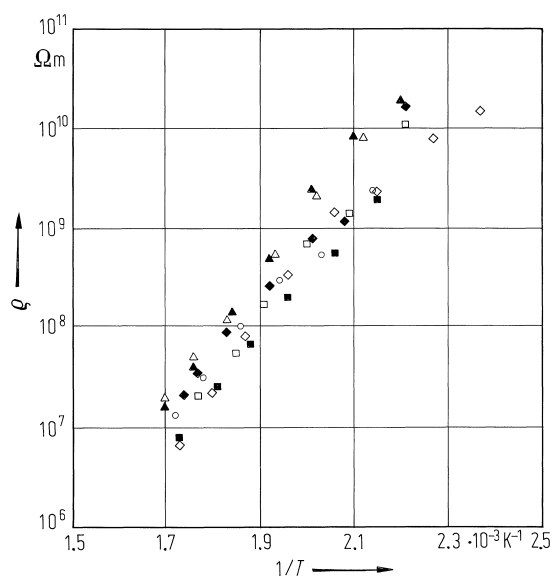


Fig. 2A-1-110. LiNbO₃ (undoped and Mg-, Fe-doped). Q vs. $1/T$ [86Ger]. Open circles: commercial optical quality; open and full squares: floating zone purified; open and full triangles: 5 mol% MgO-doped; open diamonds: 5 mol% MgO-doped and 100 ppm Fe; full diamonds: 5 mol% MgO-doped and 200 ppm Fe.

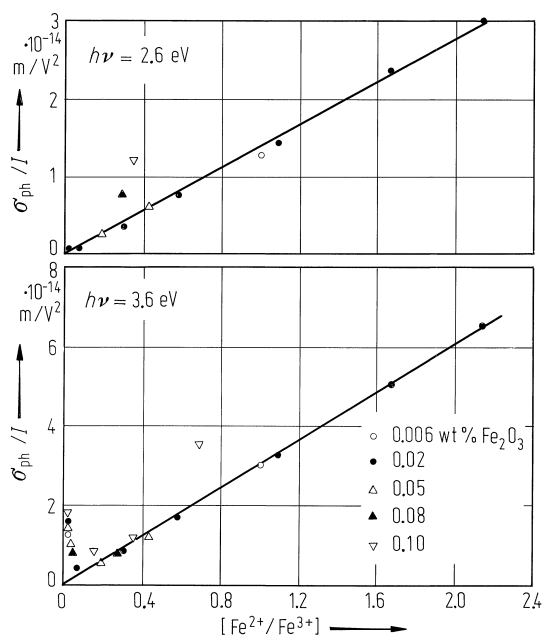


Fig. 2A-1-111. LiNbO₃ (Fe-doped). σ_{ph}/I vs. $[Fe^{2+}/Fe^{3+}]$ [78Kra]. σ_{ph} : photoconductivity, I : irradiated light intensity, $[Fe^{2+}/Fe^{3+}]$: Fe ions concentration ratio. Parameter: Fe doping level.

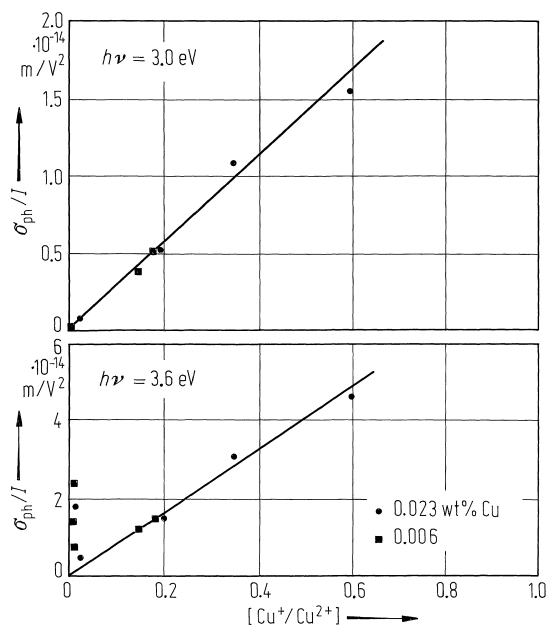


Fig. 2A-1-112. LiNbO₃ (Cu-doped). σ_{ph}/I vs. $[Cu^+/Cu^{2+}]$ [80Kra]. σ_{ph} : photoconductivity, I : irradiated light intensity, $[Cu^+/Cu^{2+}]$: Cu ions concentration ratio. Parameter: Cu doping level.

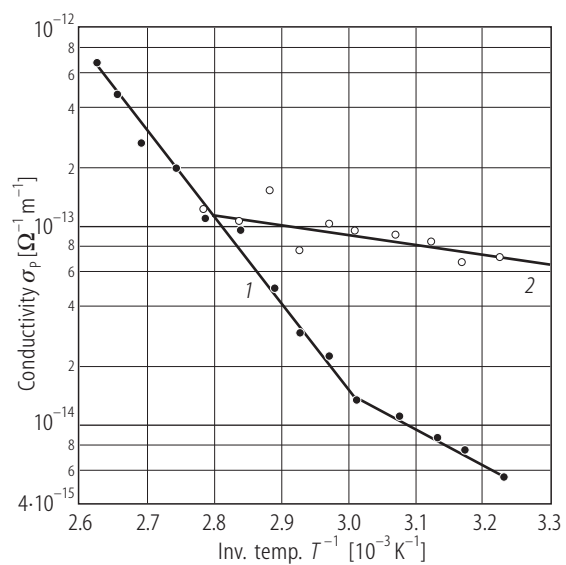


Fig. 2A-1-113. LiNbO₃. $\log \sigma_p$ vs. $1/T$ [86Bli]. σ_p : photoconductivity. Curve 1: in the dark; 2: with light illumination.

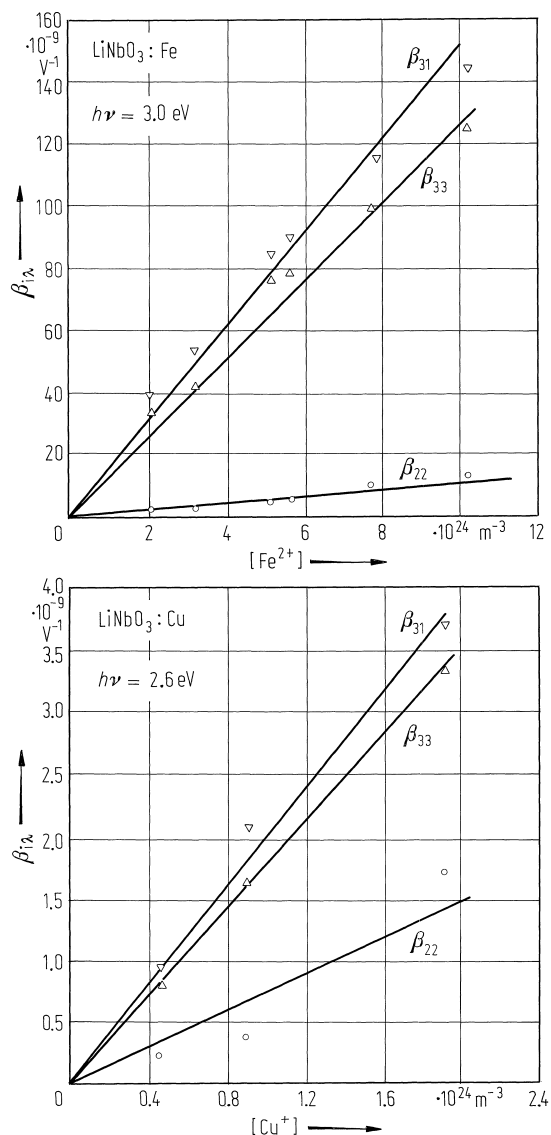


Fig. 2A-1-114. LiNbO₃ (Fe- and Cu-doped). $\beta_{i\lambda}$ vs. $[M]$ [82Fes]. $\beta_{i\lambda}$: photovoltaic tensor component, $[M]$: doped ion concentration. $\beta_{i\lambda} = \beta_{ijk}$. β_{ijk} is defined by the equation $j_i/I = \beta_{ijk}e_j e_k$, where j_i is a component of photovoltaic current density, I is the light intensity and e_j is a polarization vector component of the light.

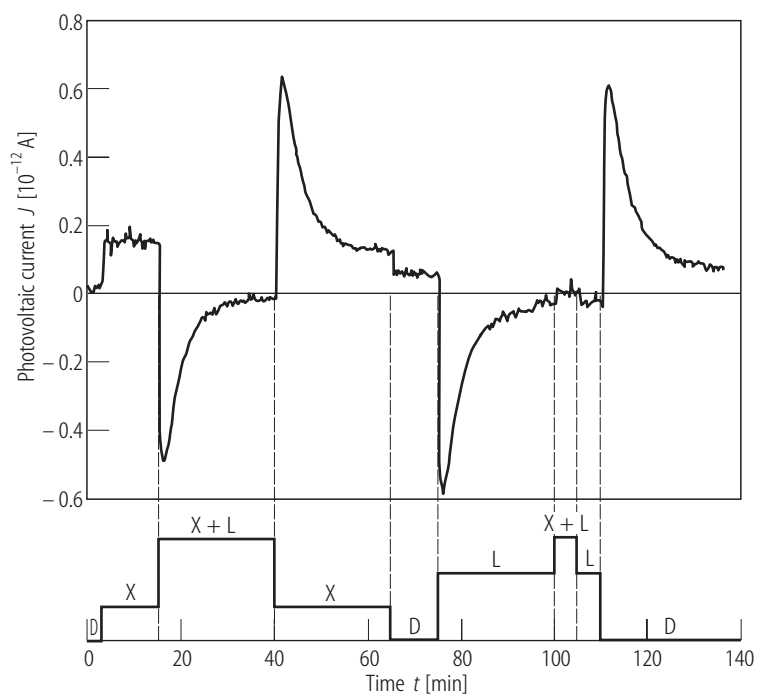


Fig. 2A-1-115. LiNbO₃. J vs. t [93Fri]. J : photovoltaic current. Irradiation: X-rays at 4 keV (X); visible light (L); combined excitation (X+L); darkness (D).

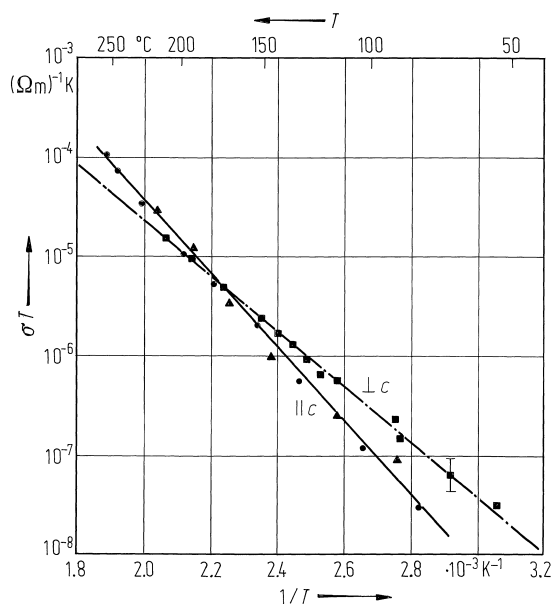


Fig. 2A-1-116. LiNbO₃. σT vs. $1/T$ [84Fra]. σ : dc Li⁺ ionic conductivity. Circles: $\parallel c$ in vacuum, triangles: $\parallel c$ in argon, squares: $\perp c$. Li/Nb = 48.6/51.4.

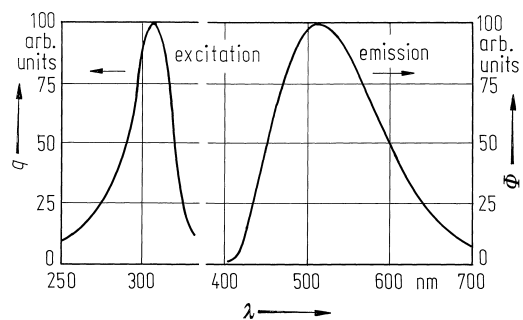


Fig. 2A-1-117. LiNbO₃. Photoluminescence spectrum at 5 K [80Kro]. q : relative quantum output, Φ : radiation power per constant wavelength interval. For the excitation spectrum $\lambda_{\text{exc}} = 275$ nm, for the emission spectrum $\lambda_{\text{em}} = 500$ nm.

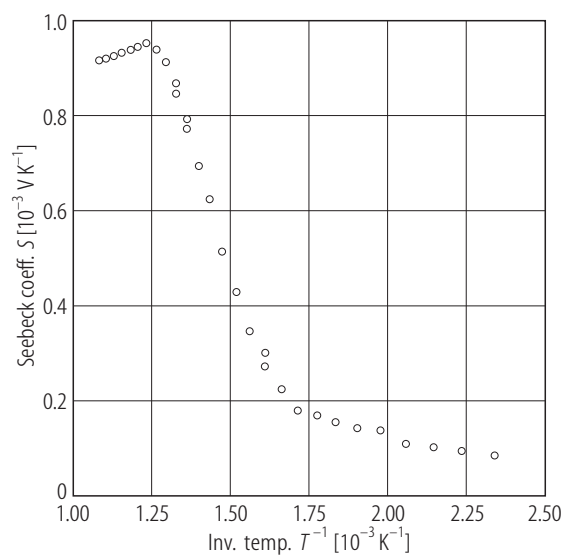


Fig. 2A-1-118. LiNbO₃. S vs. $1/T$ [92Bak2]. S : Seebeck coefficient.

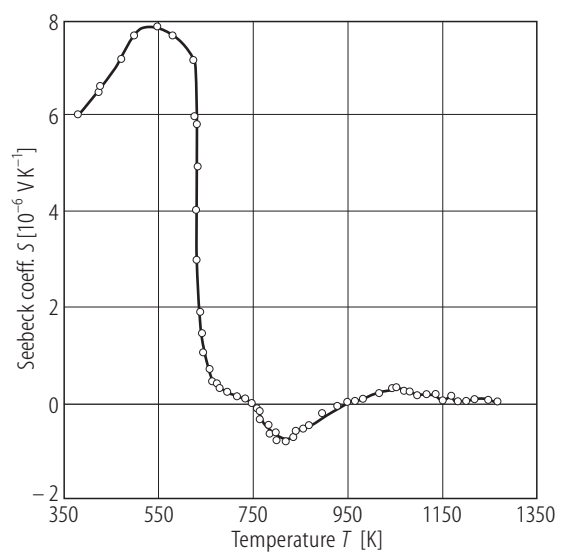


Fig. 2A-1-119. LiNbO₃. S vs. T [88Kha]. S : Seebeck coefficient.

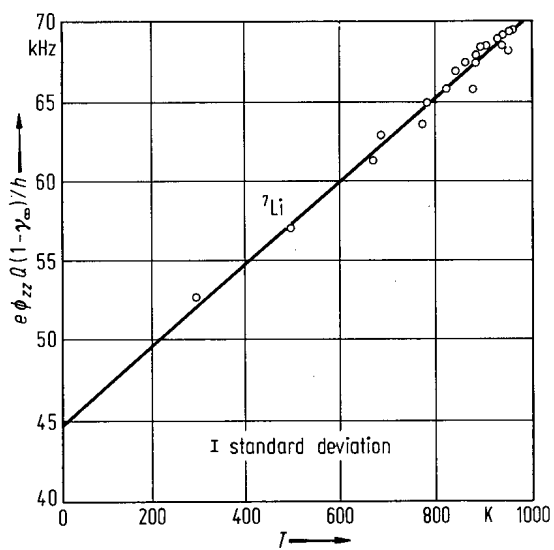


Fig. 2A-1-120. LiNbO₃. $e\phi_{zz}Q(1-\gamma_\infty)/h$ vs. T for ${}^7\text{Li}$ NMR [70Hal]. γ_∞ : Sternheimer antishielding factor. $1-\gamma_\infty = 0.7448$.

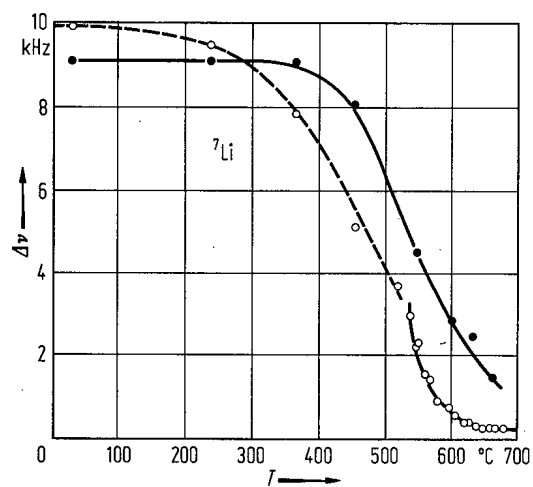


Fig. 2A-1-121. LiNbO_3 . $\Delta\nu$ vs. T for ${}^7\text{Li}$ NMR [70Hal].
 $\Delta\nu$: linewidth. Open circles: central line, full circles: satellite line.

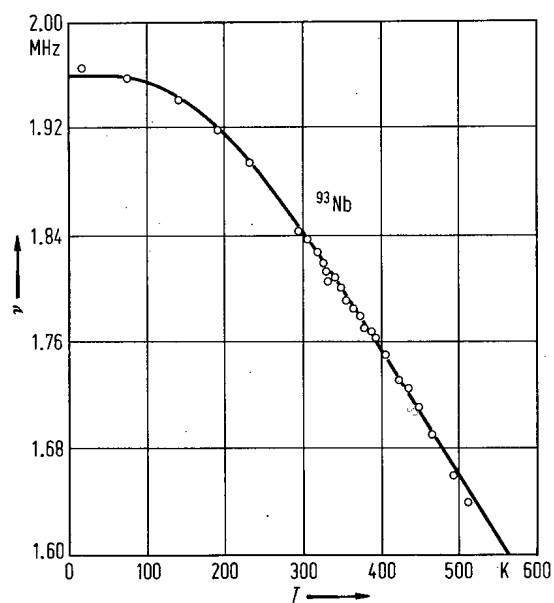


Fig. 2A-1-122. LiNbO₃. ν vs. T for ⁹³Nb NMR [70Sch]. ν : transition frequency for $m = 5/2 \leftrightarrow m = \pm 3/2$ transition.

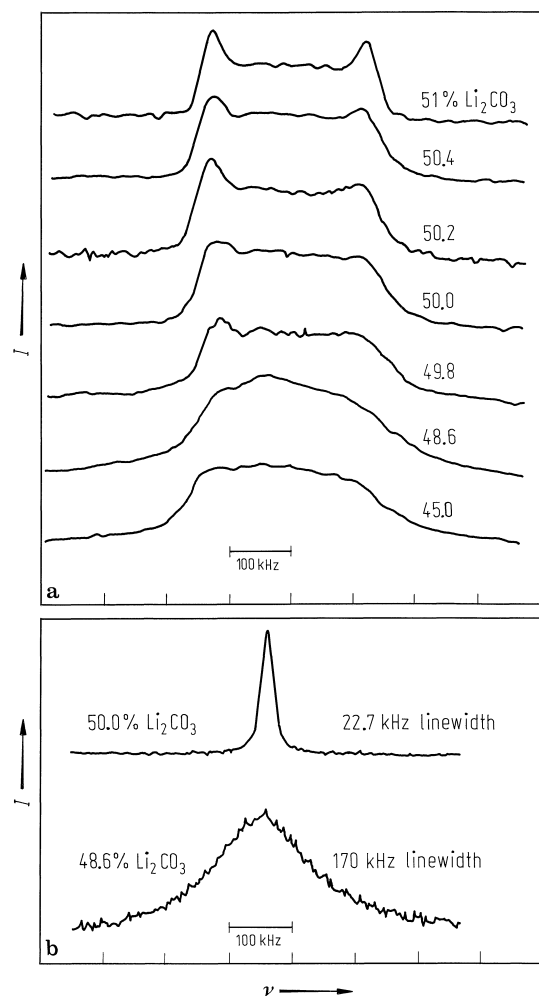


Fig. 2A-1-123. LiNbO₃. ^{93}Nb NMR spectra [86Dou].
 (a) $1/2 \leftrightarrow -1/2$ transition of ceramic samples, $f = 48.5$ MHz;
 (b) $1/2 \leftrightarrow 3/2$ transition in single crystals, $f = 49.42$ MHz.
 $H \perp c$. Parameter: Li_2CO_3 mol% in the starting materials.

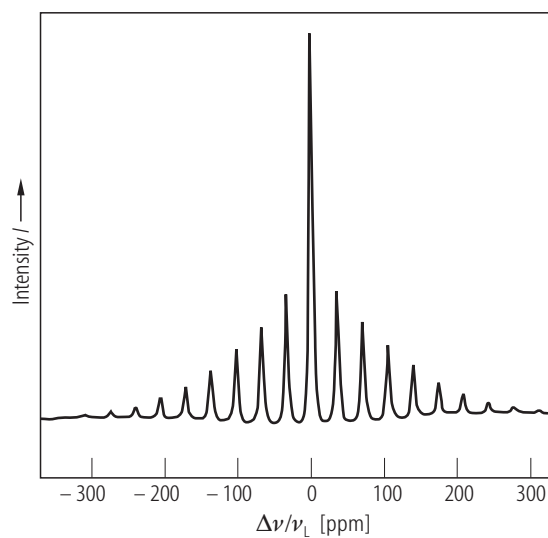


Fig. 2A-1-124. LiNbO₃. ⁷Li MAS-NMR spectrum [94Blu].
I: signal intensity. Powder sample. ν_L = 116.6 MHz. MAS:
magic angle spinning.

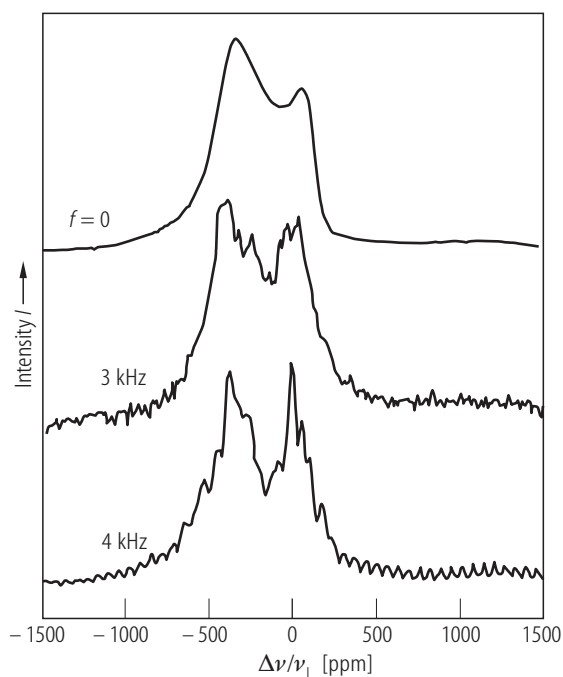


Fig. 2A-1-125. LiNbO₃. ⁹³Nb MAS-NMR spectrum [94Blu]. *I*: signal intensity. Powder sample. $\nu_L = 73.5$ MHz. Parameter: spinning frequency of the sample. MAS: magic angle spinning.

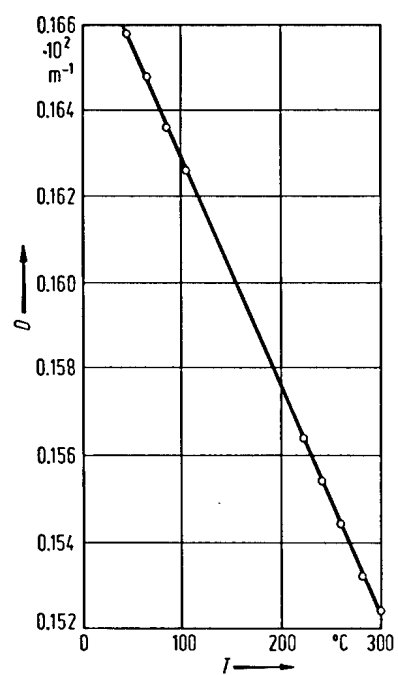


Fig. 2A-1-126. $\text{LiNbO}_3:\text{Fe}^{3+}$. D vs. T [72Meh2]. D : second order spin Hamiltonian parameter determined by ESR.

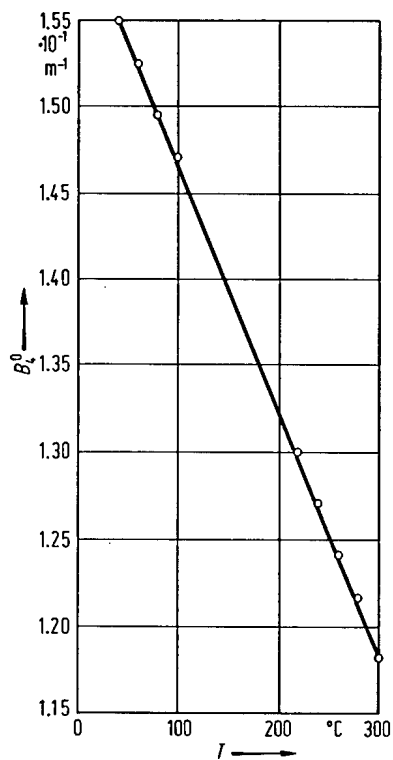


Fig. 2A-1-127. LiNbO₃:Fe³⁺. B_4^0 vs. T [72Meh]. B_4^0 : fourth order spin Hamiltonian parameter determined by ESR.

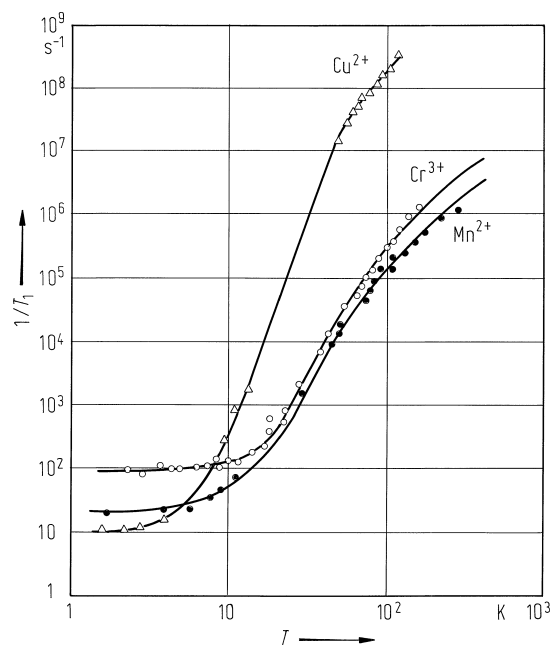


Fig. 2A-1-128. LiNbO_3 . $1/T_1$ vs. T [84Kor]. T_1 : spin-lattice relaxation time of Cu^{2+} , Cr^{3+} , Mn^{2+} ions. Transition: $-1/2 \leftrightarrow 1/2$. $H \parallel c$. $f \approx 9$ GHz.

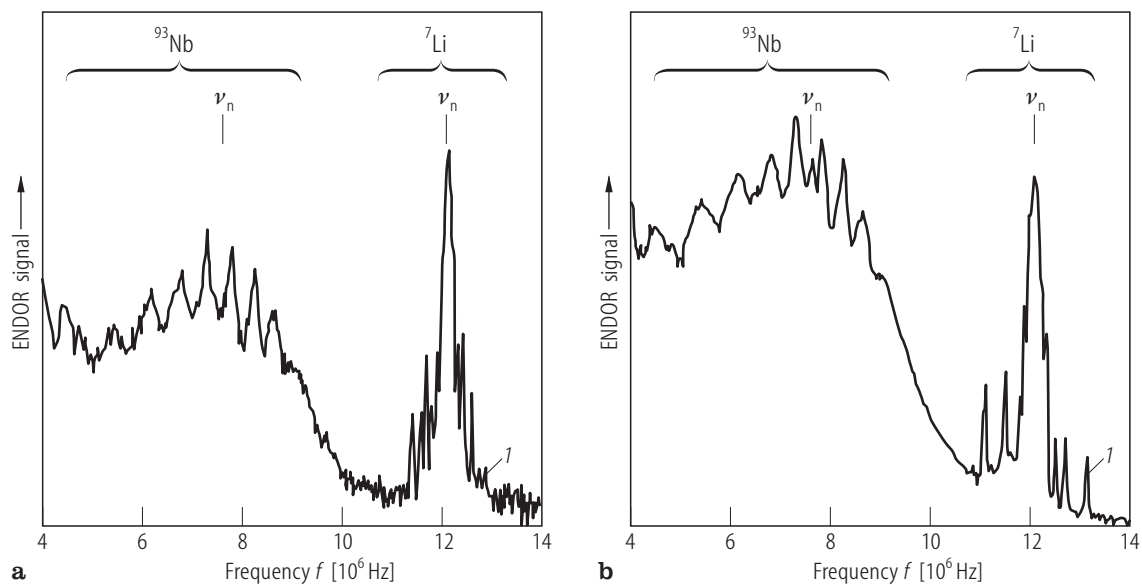


Fig. 2A-1-129. LiNbO₃:Fe³⁺. ENDOR spectrum [92Sot]. $H_0 = 7300$ Oe. $T = 4.5$ K. (a) $H_0 \parallel y$, (b) $H_0 \parallel x$. ν_n : central frequency for nucleus.

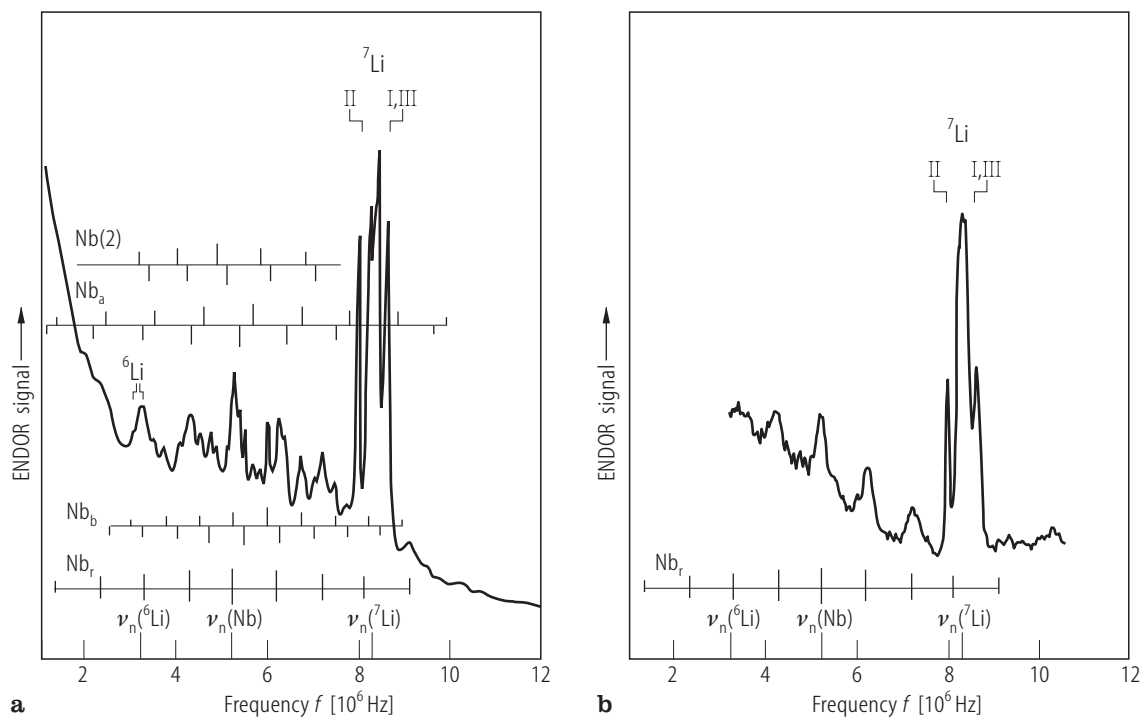


Fig. 2A-1-130. LiNbO₃:Mn²⁺. ENDOR spectrum of Mn²⁺ centers [90Cor]. $H \parallel c$. $H = 500.4$ mT. $T = 8$ K. ν_n : central frequency for nucleus. **(a)** Stoichiometric crystal. Nb_a: on axis; Nb_b: neighbors; Nb_r: remote; Nb(2): off axis. **(b)** Congruent crystal. Nb_r: remote nuclei.

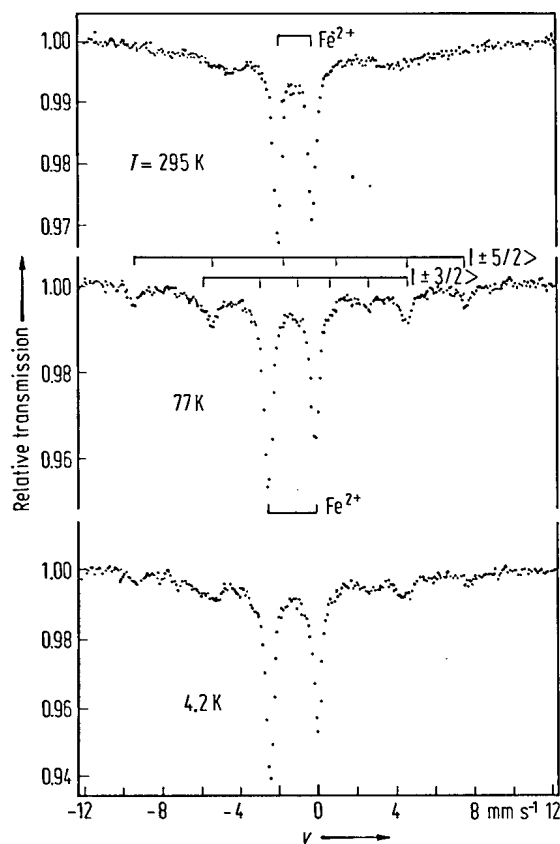


Fig. 2A-1-131. LiNbO₃:⁵⁷Co. Mössbauer spectra [75Keu].
 v : source velocity. As absorbers, enriched Na₄Fe(CN)₆ · 10 H₂O is employed at 295 K, and ⁵⁷Fe in Nb foil at 77 K and 4.2 K. The c -axis is perpendicular to the γ -ray direction.

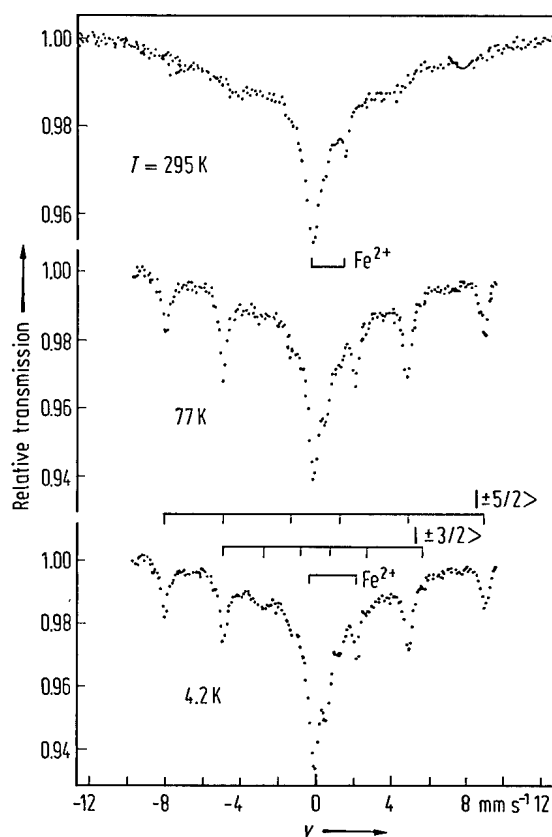


Fig. 2A-1-132. LiNbO₃:⁵⁷Fe. Mössbauer spectra [75Keu]. v : absorber velocity in original preparation state. ⁵⁷Co in Pd matrix is served as a source. Estimated Fe concentration is about 0.1 at %. The c -axis is perpendicular to the γ -ray direction.

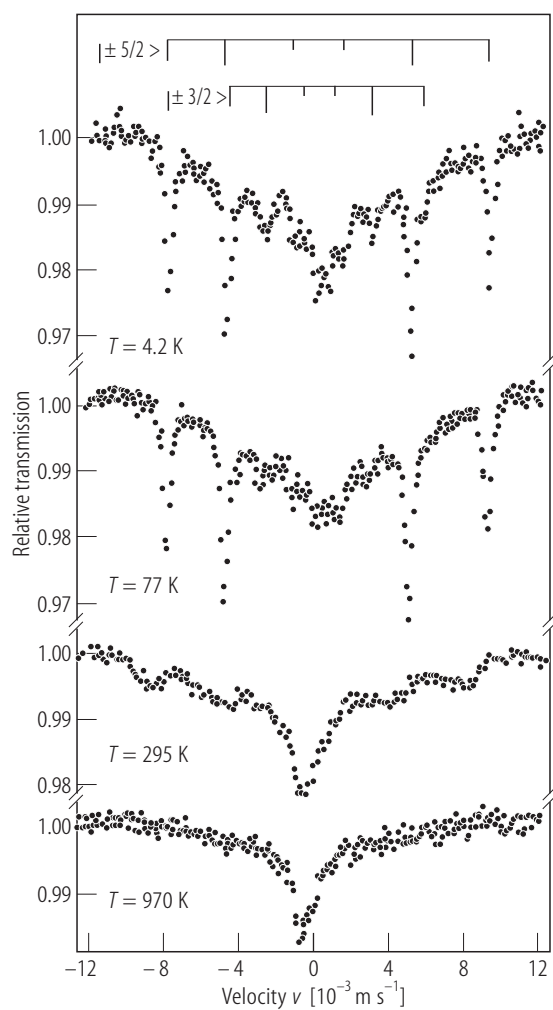


Fig. 2A-1-133. LiNbO₃:Fe. ⁵⁷Fe Mössbauer spectra [88Eng]. Sample: single crystal oxidized in air for 24 h at 300 °C. Parameter: *T*.

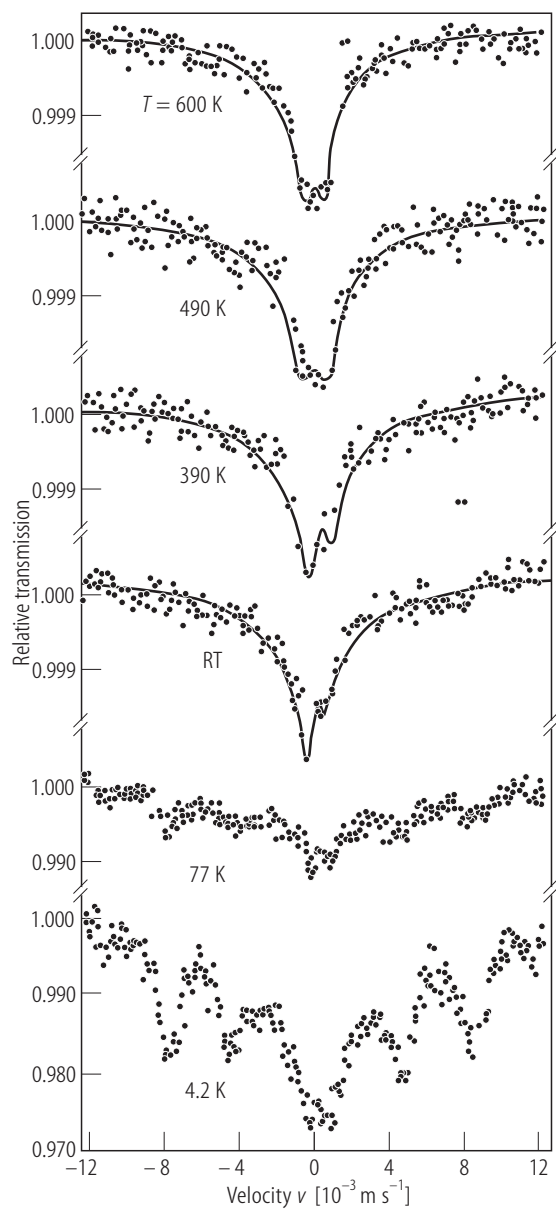


Fig. 2A-1-134. LiNbO₃:Fe. ⁵⁷Fe Mössbauer spectra [88Eng]. Sample: amorphous LiNbO₃ oxidized for 24 h in air at 300 °C. Parameter: *T*.

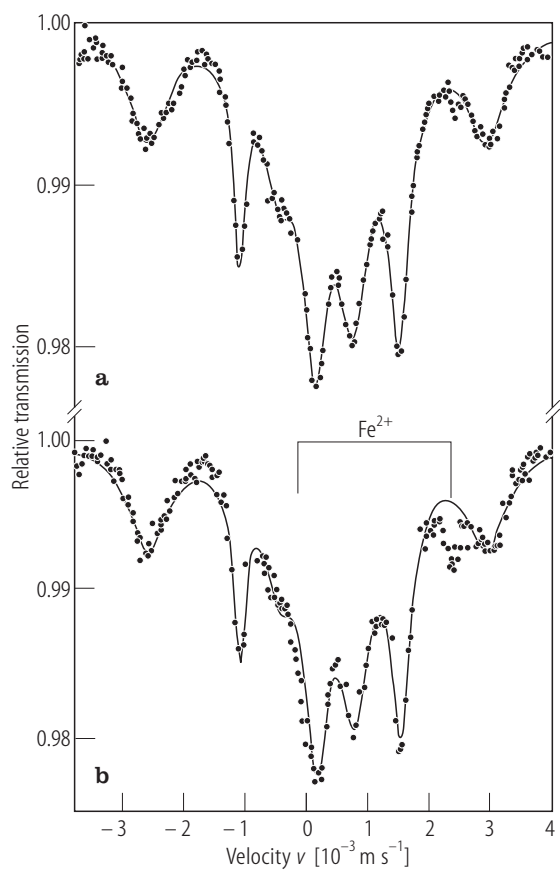


Fig. 2A-1-135. LiNbO₃:Fe. ⁵⁷Fe Mössbauer spectra at 4.2 K [88Eng]. Sample: a fully oxidized LiNbO₃. (a): before the light irradiation; (b): after the light irradiation ($\lambda = 366 \text{ nm}$).

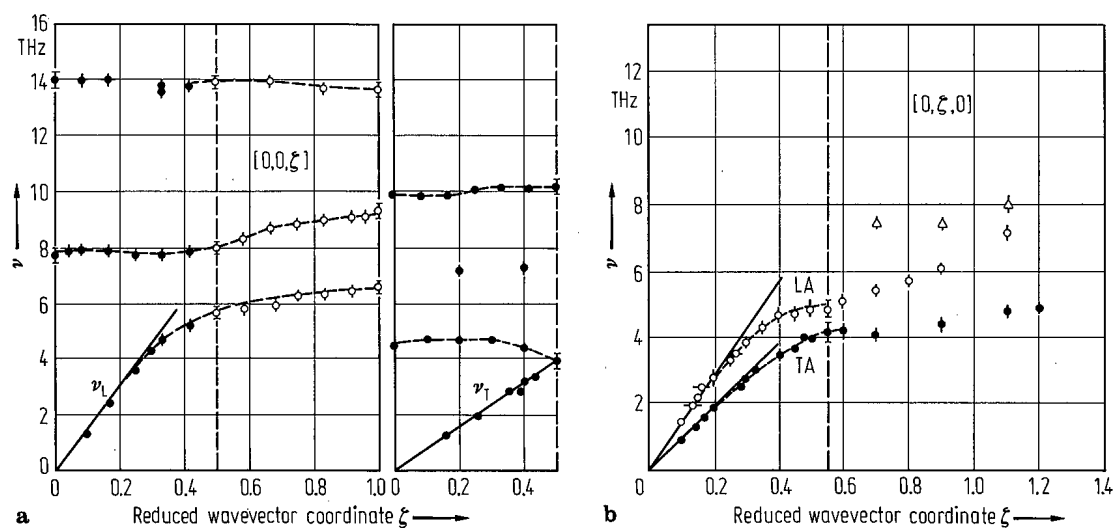


Fig. 2A-1-136. LiNbO₃. Phonon dispersion relations, (a) along the [001] directions ($q = [0,0,\xi] \cdot 6\pi/c$), and (b): along the [010] directions ($q = [0,\xi,0] \cdot 4\pi/\sqrt{3}a$) at 295 K [78Cho]. ν : frequency. The broken curves drawn through the experimental points are only for a guide. The solid lines represent the sound velocities.

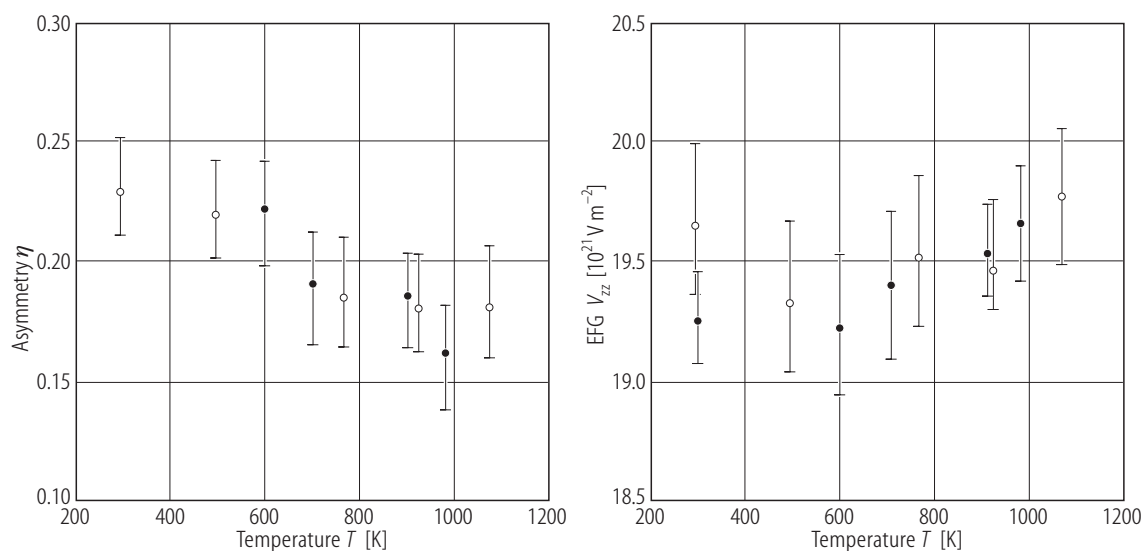


Fig. 2A-1-137. LiNbO₃. V_{zz} , η vs. T [91Cat2]. V_{zz} , η : electric field gradient and asymmetry parameter at Li sites determined by $^{181}\text{Hf} \rightarrow ^{181}\text{Ta}$ perturbed-angular correlation spectroscopy.

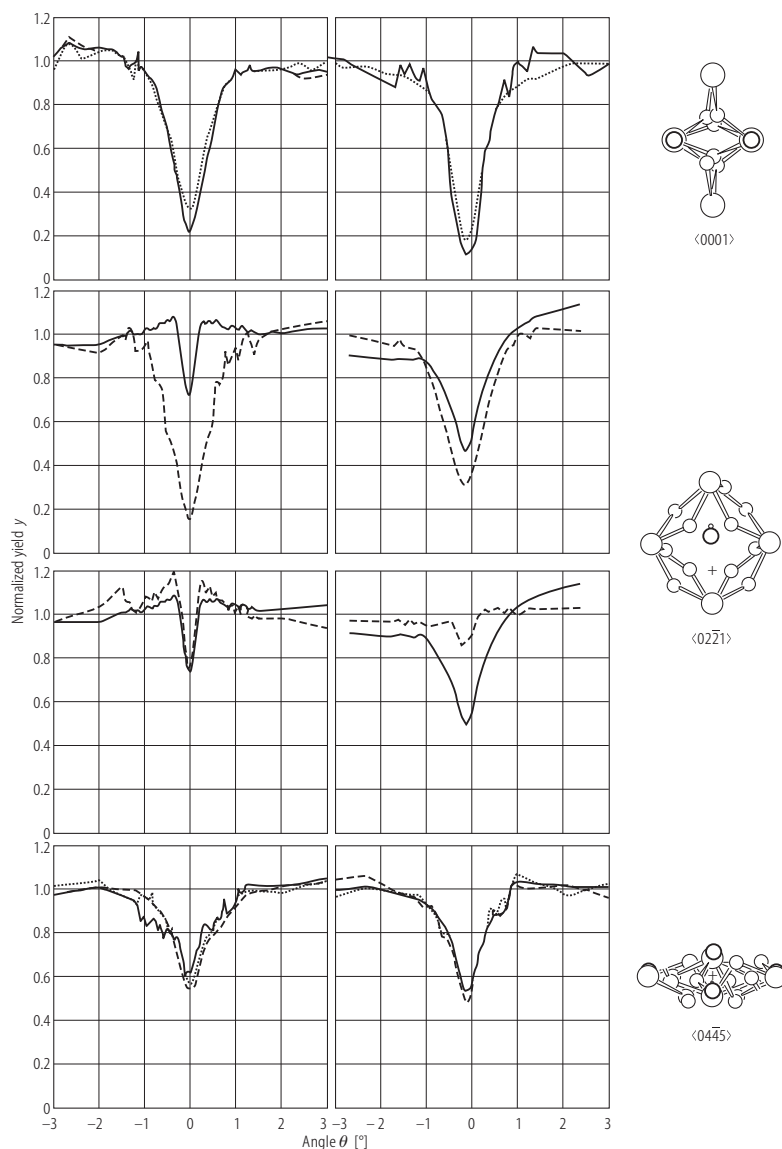


Fig. 2A-1-138. LiNbO₃:Ti. Comparison of experimental angular PIXE, RBS and NRA yield profiles [92Kol]. PIXE: particle induced X-ray emission. RBS: Rutherford backscattering. NRA: nuclear reaction analysis. y : normalized yield; θ : angle relative to the axis. Upper row: $\langle 0001 \rangle$ axis (c -axis) for the Ti-diffused LiNbO₃ crystal. Left: PIXE with 3.1 MeV He ions, (full line) Ti(K), (dotted line) Nb(K), (dashed line) Nb(L), the Nb(L) yield curve coincides for most data points with the Ti(K) curve. Right: RBS and NRA with 1.77 MeV protons, RBS (dotted line) Nb, NRA (full line) ${}^7\text{Li}(p, \alpha){}^4\text{He}$. Second and third row from above: $\langle 02\bar{2}1 \rangle$ axis for 1.77 MeV protons. Left: 3 mol% Ti bulk doped LiNbO₃ crystal. Right: Ti-diffused LiNbO₃ crystal. Second row: PIXE (full lines) Ti(K), (dashed lines) Nb(L). Third row: PIXE (full lines) Ti(K), NRA (dashed lines) ${}^7\text{Li}(p, \alpha){}^4\text{He}$. Lowest row: $\langle 04\bar{4}5 \rangle$ axis, PIXE with He ions (full lines) Ti(K), (dotted lines) Nb(K), (dashed lines) Nb(L). Left: 3 mol% Ti bulk doped LiNbO₃ crystal. Right: Ti-diffused LiNbO₃ crystal. At the right side projections on the corresponding axes are depicted, where the large circles symbolize Nb atoms, the smallest circles represent O and the medium size circles stand for Li. The cross represents the interstitial intrinsic octahedral site. Bonds between O and Nb atoms are represented by rods.

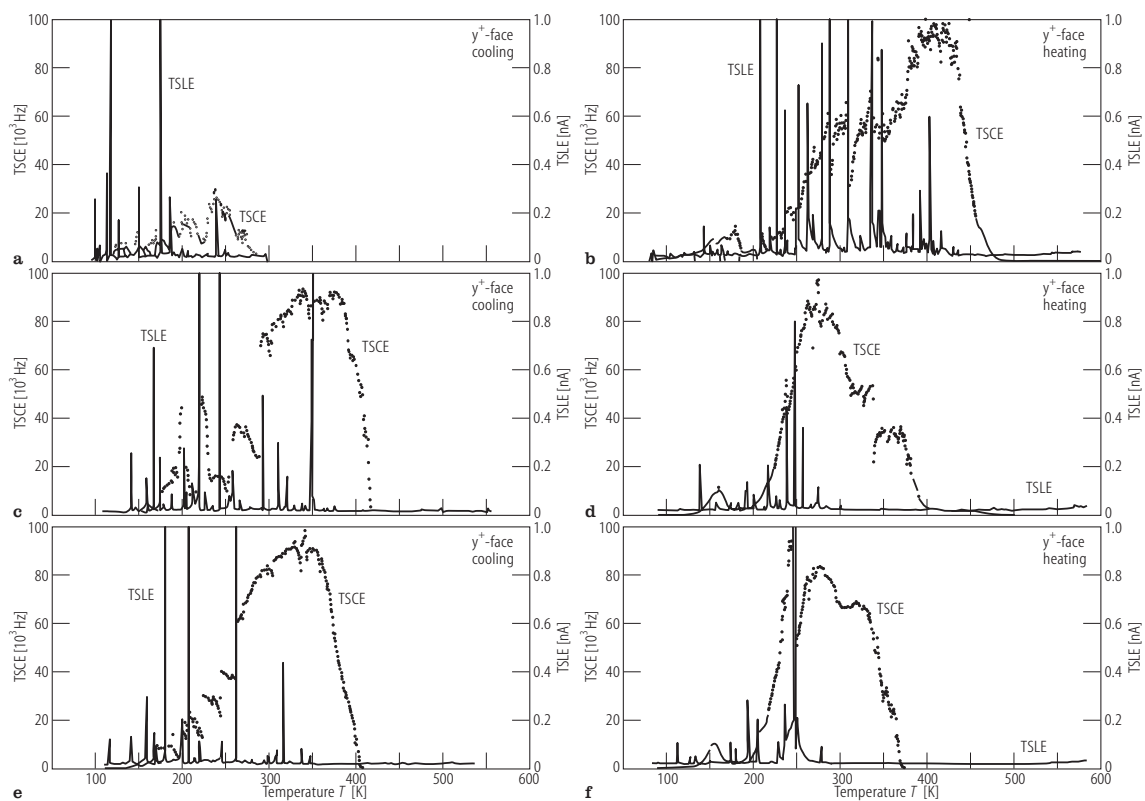


Fig. 2A-1-139. LiNbO₃. TSCE and TSLE vs. T [92Tom]. TSCE: thermally stimulated charge emission; TSLE: thermally stimulated light emission. Y^+ -face. Measurements were done in alphabetical order indicated in the figure.