

3.2.4 Larger non-linear free radicals

3.2.4.1 Preliminary remarks

1 Introduction

In the present edition, there are twenty two free radicals in this category (larger non-linear free radicals) compared with seven in the previous supplement. Once again, this is a significant increase, driven to a large extent by the need to obtain information for the identification of species in astrophysical sources.

All the molecules in this section are asymmetric rotors in doublet open-shell states. The effective Hamiltonian is the same as that used for the triatomic asymmetric rotors in section 3.2.2 with the sole exception that the centrifugal distortion of the rotational kinetic energy have been extended for some molecules to higher degree terms [00Bre]. For example, in the Watson S-reduction [77Wat], the extended centrifugal distortion Hamiltonian has the form

$$\begin{aligned} H_{\text{cd}} = & -D_N(N^2)^2 - D_{NK}N^2N_z^2 - D_KN_z^4 + d_1N^2(N_+^2 + N_-^2) + d_2(N_+^4 + N_-^4) \\ & + H_N(N^2)^3 + H_{NK}(N^2)^2N_z^2 + H_{KN}N^2N_z^4 + H_KN_z^6 \\ & + h_1(N^2)^2(N_+^2 + N_-^2) + h_2N^2(N_+^4 + N_-^4) + h_3(N_+^6 + N_-^6) \\ & + L_{NNK}(N^2)^3N_z^2 + L_{NK}(N^2)^2N_z^4 + L_{KKN}N^2N_z^6 + \dots \\ & + P_{NNK}(N^2)^3N_z^4 + P_{NKK}(N^2)^2N_z^6 + P_{KN}N^2N_z^8 + \dots \end{aligned}$$

The operators involved are defined in section 3.2.3.1.

2 List of tabulated parameters

These are the same as those listed in section 3.2.3.2. In addition:

P_{NNK}, P_{NKK}, P_{KN} decadic centrifugal distortion parameters
 $v_i, i=1, 2, 3, \dots, 3N-6$ vibrational quantum number of the i -th normal mode of a radical consisting of N atoms

3 List of symbols used

See section 3.2.3.1, non-linear triatomic radicals

4 Arrangement of molecules

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|------------------------|-------------------------------------|-----------------------|
| 1. c-C ₃ H | 9. CH ₂ C ₄ H | 17. HC ₄ O |
| 2. CH ₂ F | 10. CH ₂ CN | 18. HOCO |
| 3. CHF ₂ | 11. CH ₂ CCCN | 19. HNCN |
| 4. CH ₂ Cl | 12. CH ₂ CP | 20. MgNH ₂ |
| 5. CH ₂ Br | 13. H ₂ NS | 21. CaNH ₂ |
| 6. CH ₂ N | 14. H ₂ PO | 22. SrNH ₂ |
| 7. CH ₂ P | 15. HCCO | |
| 8. CH ₂ CCH | 16. HCCCO | |

5 References

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 00Bre Brewster, M.A., Ziurys, L.M.: J. Chem Phys. **113**(2000)3141.