

**Table 27A-2-001.** HBr, DBr. Unit cell parameters.

Compound	Phase	$T$ [K]	$a, b, c$ [Å]	Ref.
HBr	I	130	$a = 5.815(2)$	71Sim
	II	115	$a = 5.785(2)$	
	III	100	$a = 5.555(3)$ $b = 5.648(2)$ $c = 6.086(2)$	
	IV	85	$a = 5.462(3)$ $b = 5.643(3)$ $c = 6.111(3)$	
DBr	III	107	$a = 5.559(5)$ $b = 5.649(5)$ $c = 6.095(5)$	68San
	IV	84	$a = 5.44(2)$ $b = 5.614(5)$ $c = 6.120(5)$	

**Table 27A-2-002.** DBr. Fractional coordinates and temperature parameters [68San].  $B$  is defined by Eq. (e) in Introduction.

Atom	$x$	$y$	$z$	$B$ [Å <sup>2</sup> ]	$x$	$y$	$z$	$B$ [Å <sup>2</sup> ]
	$T = 84$ K (phase IV)				$T = 107$ K (phase III)			
D	0.075(2)	0.179(3)	0	5.0(5)	−0.172(6)	0.177(4)	0	8.4(8)
Br	0.25	0	0	2.1(4)	0	0	0	4.4(8)

**Table 27A-2-003.** HBr. Linear thermal expansion coefficients [71Sim].

Phase	$T$ [K]	$\alpha_{ij}$ [ $\cdot 10^{-6}$ K <sup>−1</sup> ]
I	130	$\alpha_{11} = 225(15)$
II	115	$\alpha_{11} = 225(40)$
III	100	$\alpha_{11} = 430(20)$
		$\alpha_{22} = 370(20)$
		$\alpha_{33} = -60(30)$

**Table 27A-2-004.** HBr, DBr. Transition heats and transition entropies.

Compound	Transition	$\Delta Q_m$ [J mol <sup>−1</sup> ]	$\Delta S_m$ [J K <sup>−1</sup> mol <sup>−1</sup> ]	Ref.
HBr	I–II–III	691	5.98	24Euc
	III–IV	472	5.27	
DBr	I–III	1267	10.5	47Clu
	III–IV	709	7.6	