

Fig. 39A-1-001. $(\text{NH}_4)_2\text{SO}_4$. Crystal form with the setting $b > a > c$ [08Gro].

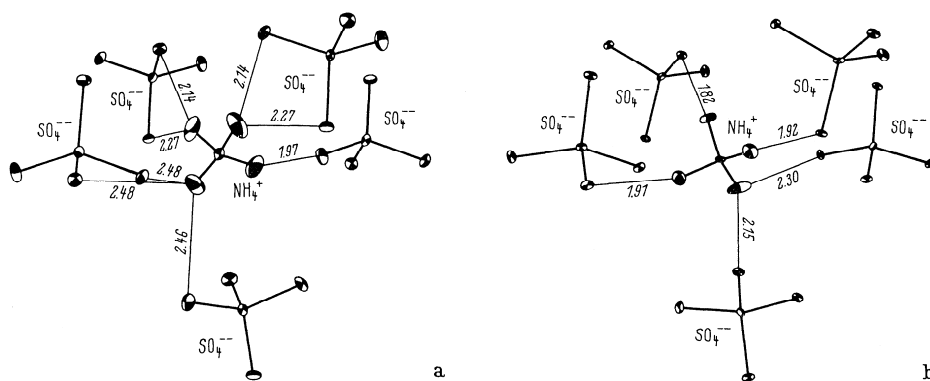


Fig. 39A-1-002. $(\text{NH}_4)_2\text{SO}_4$. Structures of phases I and II [66Sch]. Neutron diffraction. The environment of $(\text{NH}_4)^+$ (I). (a) RT, (b) about 180 K. Atomic distances in [Å].

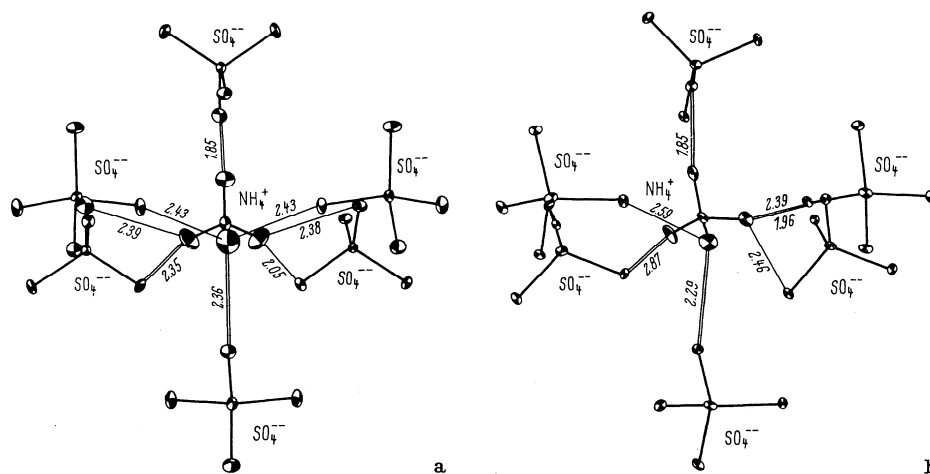


Fig. 39A-1-003. $(\text{NH}_4)_2\text{SO}_4$. Structures of phases I and II [66Sch]. Neutron diffraction. The environment of $(\text{NH}_4)^+$ (II). (a) RT, (b) about 180 K. Atomic distances in [Å].

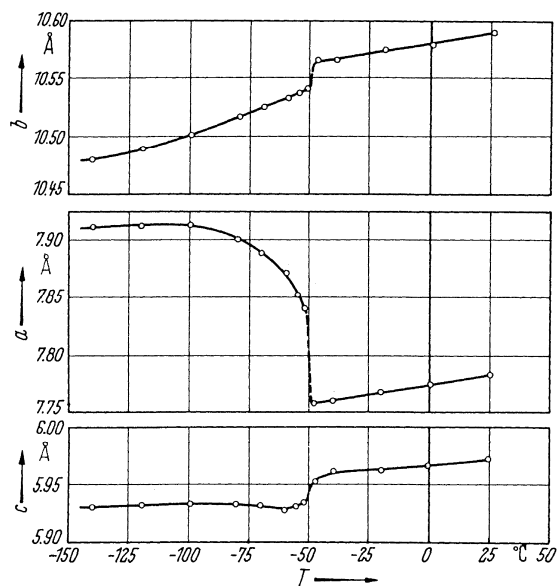


Fig. 39A-1-004. $(\text{NH}_4)_2\text{SO}_4$. a , b , c vs. T [58Hos].

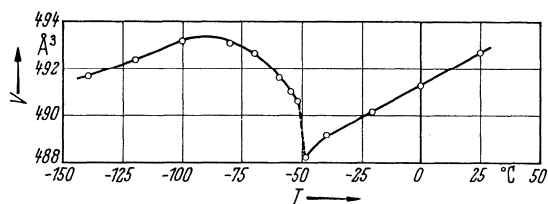


Fig. 39A-1-005. $(\text{NH}_4)_2\text{SO}_4$. V vs. T [58Hos]. V : volume of the unit cell.

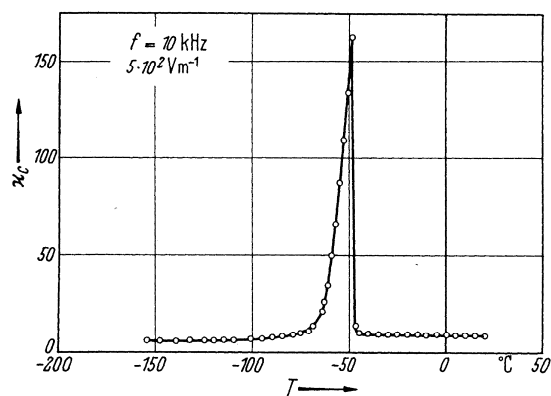


Fig. 39A-1-006. $(\text{NH}_4)_2\text{SO}_4$. κ_c vs. T [58Hos].

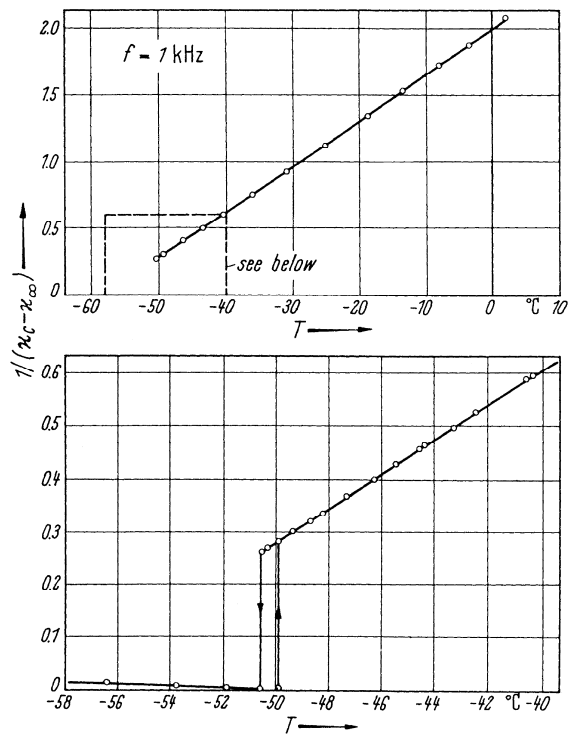


Fig. 39A-1-007. (NH₄)₂SO₄. $1/(\kappa_c - \kappa_\infty)$ vs. T [65Unr]. $\kappa_\infty = 9.4(2)$.

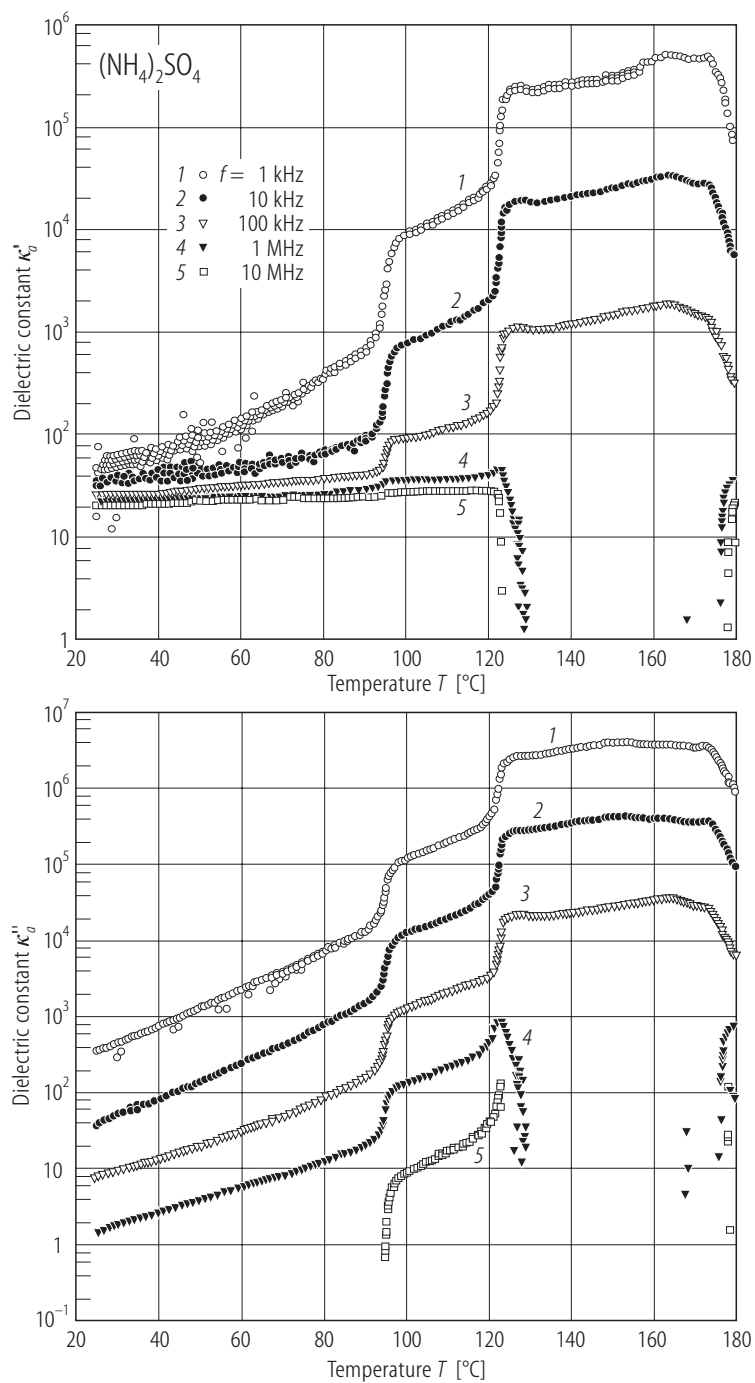


Fig. 39A-1-008. (NH₄)₂SO₄. κ'_a , κ''_a vs. T [96Kim2]. Parameter: f . On cooling.

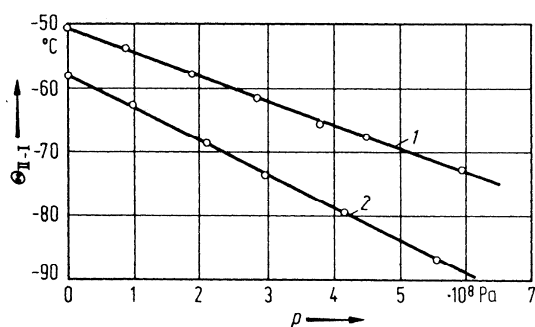


Fig. 39A-1-009. (NH₄)₂SO₄, [(NH₄)₂SO₄]_{0.9}[NH₄BeF₃]_{0.1}. $\Theta_{\text{II-I}}$ vs. p [72Pol]. Curve 1: for (NH₄)₂SO₄, 2: for the solid solution.

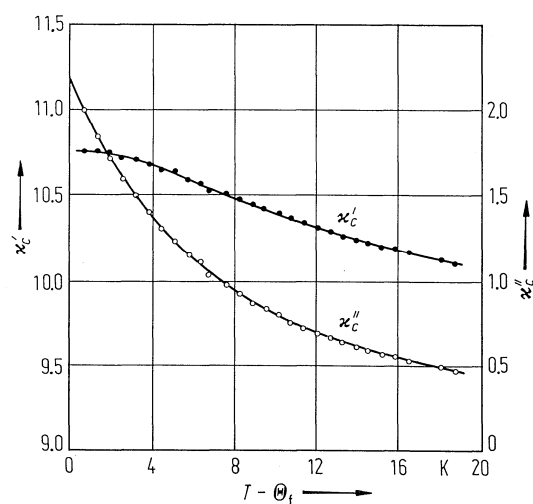


Fig. 39A-1-010. (NH₄)₂SO₄. κ'_c , κ''_c vs. $T - \Theta_f$ [80Lut]. $f = 2.3 \cdot 10^{10}$ Hz.

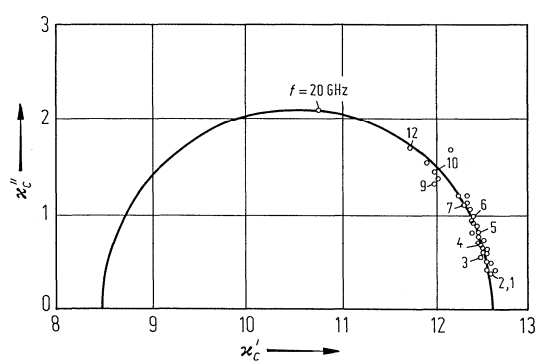


Fig. 39A-1-011. (NH₄)₂SO₄. Cole-Cole plot of the complex dielectric constant in the paraelectric phase [79Lut]. $T = -50.482$ °C, 0.42 °C above Θ_f .

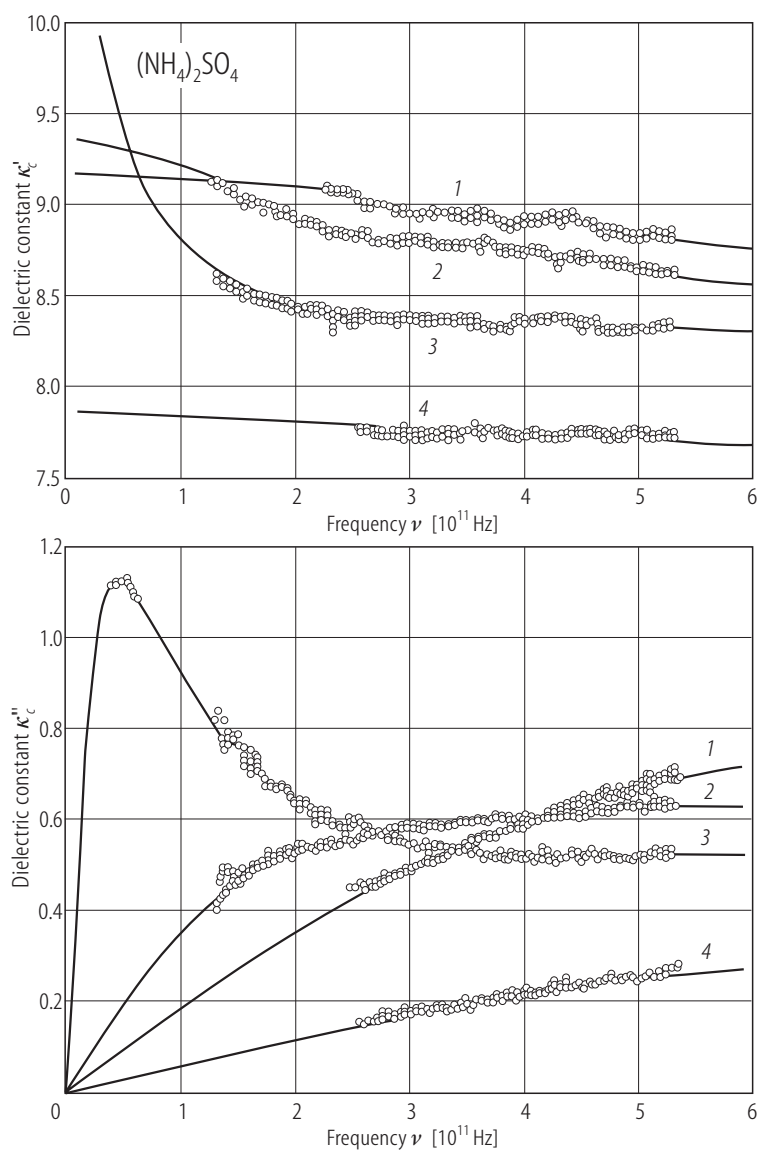


Fig. 39A-1-012. (NH₄)₂SO₄. κ'_c , κ''_c vs. ν [88Vol]. Parameter: T . Curve 1: $T = 376$ K, 2: 297 K, 3: 225 K, 4: 196 K.

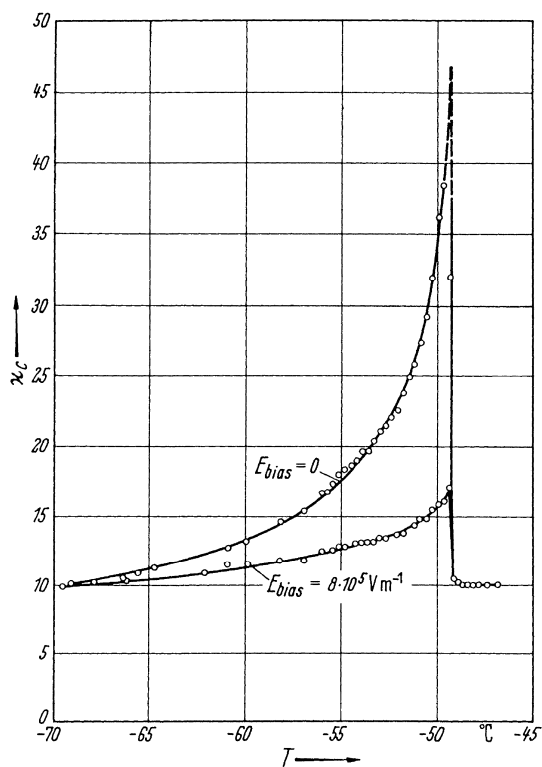


Fig. 39A-1-013. (NH₄)₂SO₄. κ_c vs. T [58Hos]. $f = 10$ kHz. Parameter: E_{bias} .

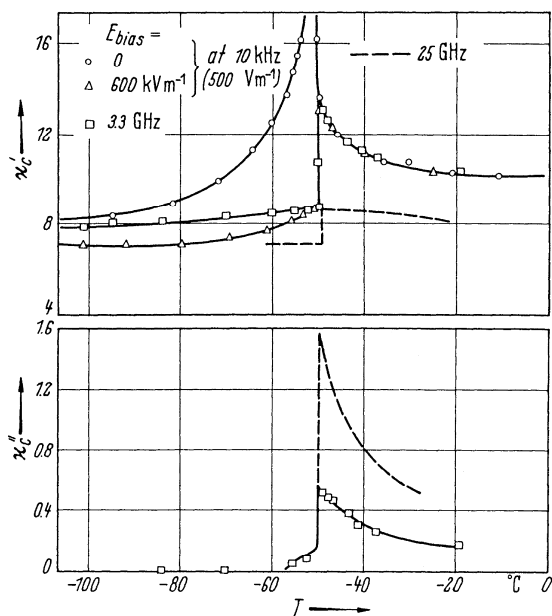


Fig. 39A-1-014. (NH₄)₂SO₄. κ'_c , κ''_c vs. T [66Ohs]. Dashed curve: [56Cou].

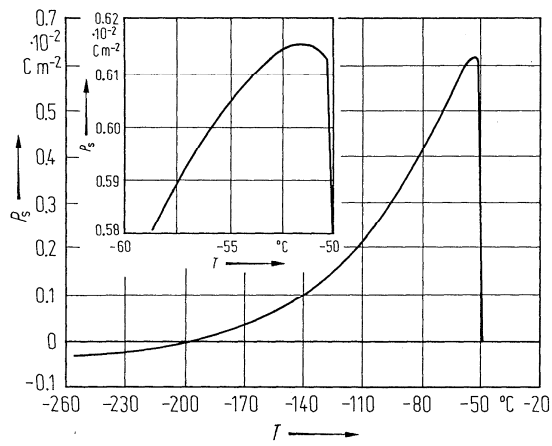


Fig. 39A-1-015. (NH₄)₂SO₄. P_s vs. T [70Unr].

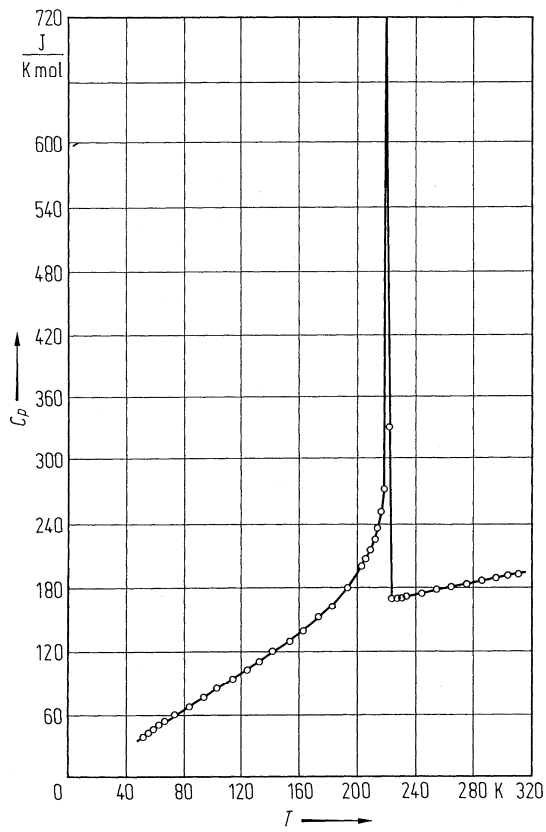


Fig. 39A-1-016. (NH₄)₂SO₄. C_p vs. T [45Sho]. C_p : molar heat capacity at constant pressure.

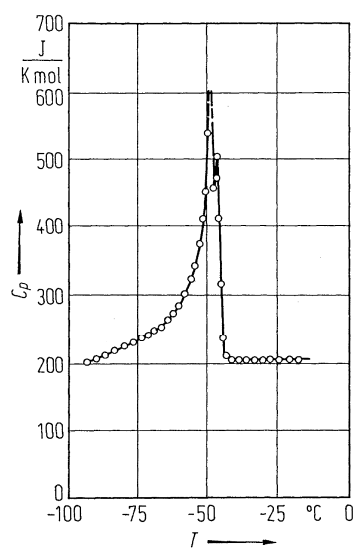


Fig. 39A-1-017. $(\text{NH}_4)_2\text{SO}_4$. C_p vs. T [58Hos]. C_p : molar heat capacity at constant pressure.

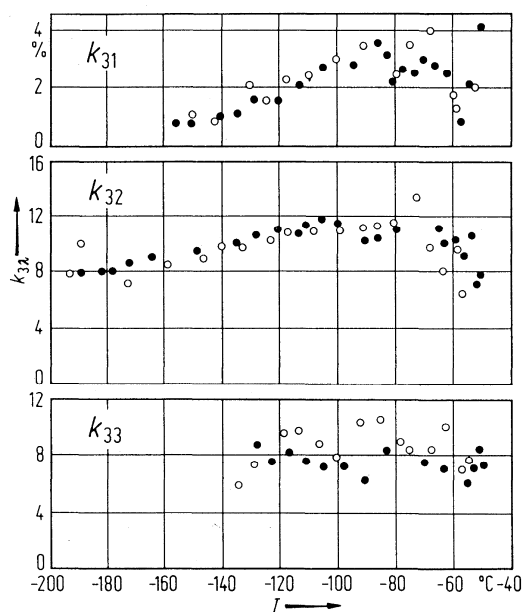


Fig. 39A-1-018. $(\text{NH}_4)_2\text{SO}_4$. $k_{3\lambda}$ vs. T [73Ike]. $k_{3\lambda}$: electromechanical coupling coefficient. Open circles: on cooling, full circles: on heating.

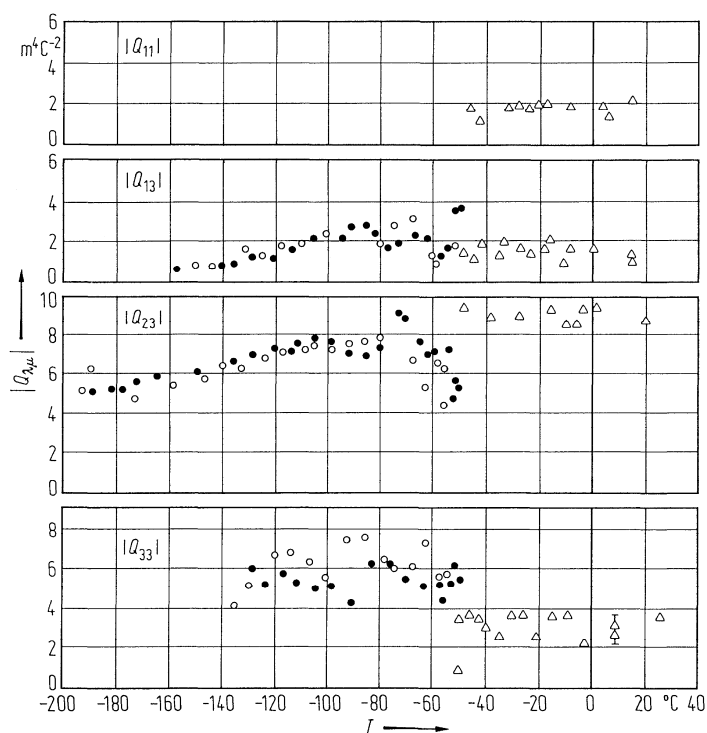


Fig. 39A-1-019. (NH₄)₂SO₄. $|Q_{\lambda\mu}|$ vs. T [73Ike, 75Zai]. $Q_{\lambda\mu}$: electrostrictive constant. Values below $\Theta_{\text{II-I}}$ are obtained from piezoelectric constants assuming that P_s is independent of temperature and should be corrected by taking account of the decrease of P_s at lower temperatures [73Ike]. Open circles: on cooling, full circles: on heating. The triangles above $\Theta_{\text{II-I}}$ are determined by the measurement of electrostrictive strains at low frequencies [75Zai].

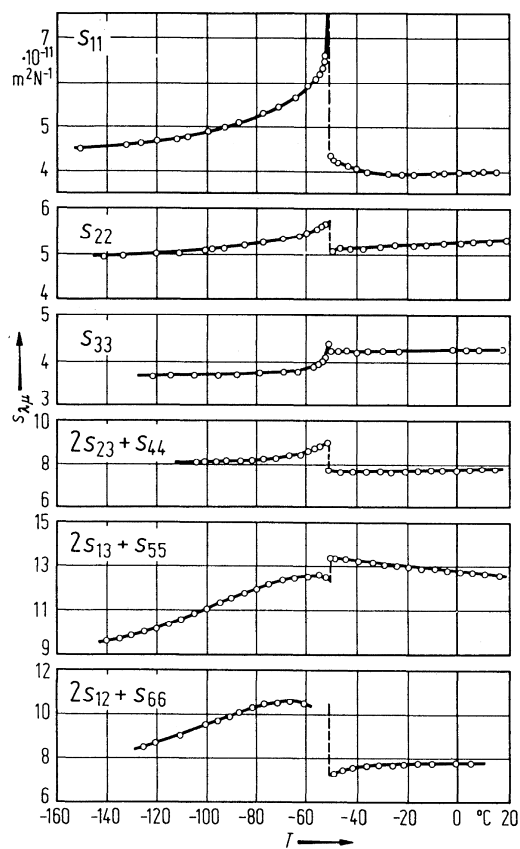


Fig. 39A-1-020. (NH₄)₂SO₄. $s_{\lambda\mu}$ vs. T [73Ike]. $s_{\lambda\mu}$: elastic compliance.

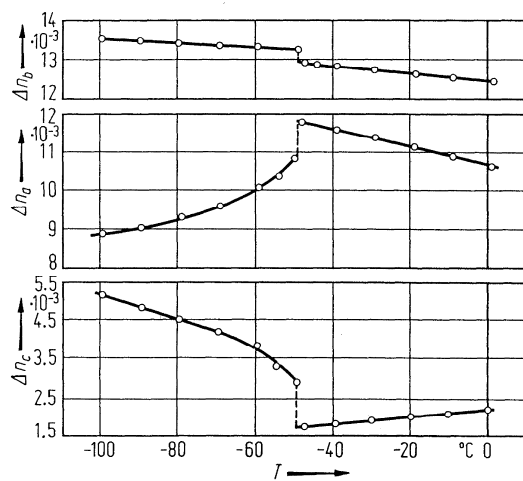


Fig. 39A-1-021. (NH₄)₂SO₄. Δn_a , Δn_b , Δn_c vs. T [70Ani]. $\Delta n_a = |n_b - n_c|$, $\Delta n_b = |n_c - n_a|$, $\Delta n_c = |n_a - n_b|$.

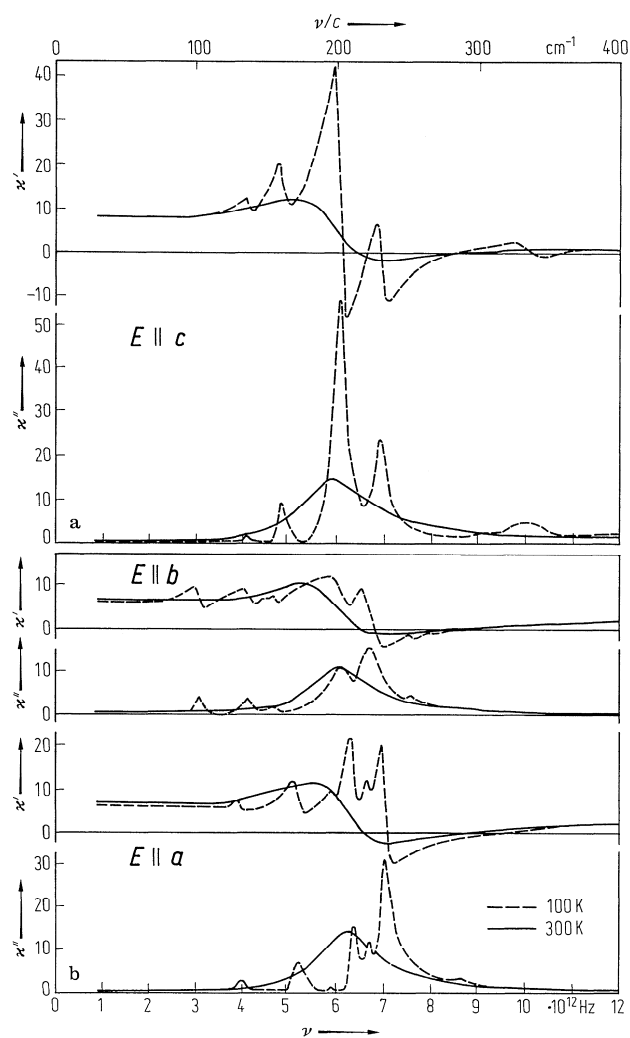


Fig. 39A-1-022. (NH₄)₂SO₄. κ' , κ'' vs. ν [74Pet]. ν : frequency of incident far-infrared radiation. Polarization of the radiation is parallel to the c axis in (a), and parallel to the b and the c axis in (b). κ' , κ'' are obtained from the reflectivity spectra using Kramers-Kronig analysis.

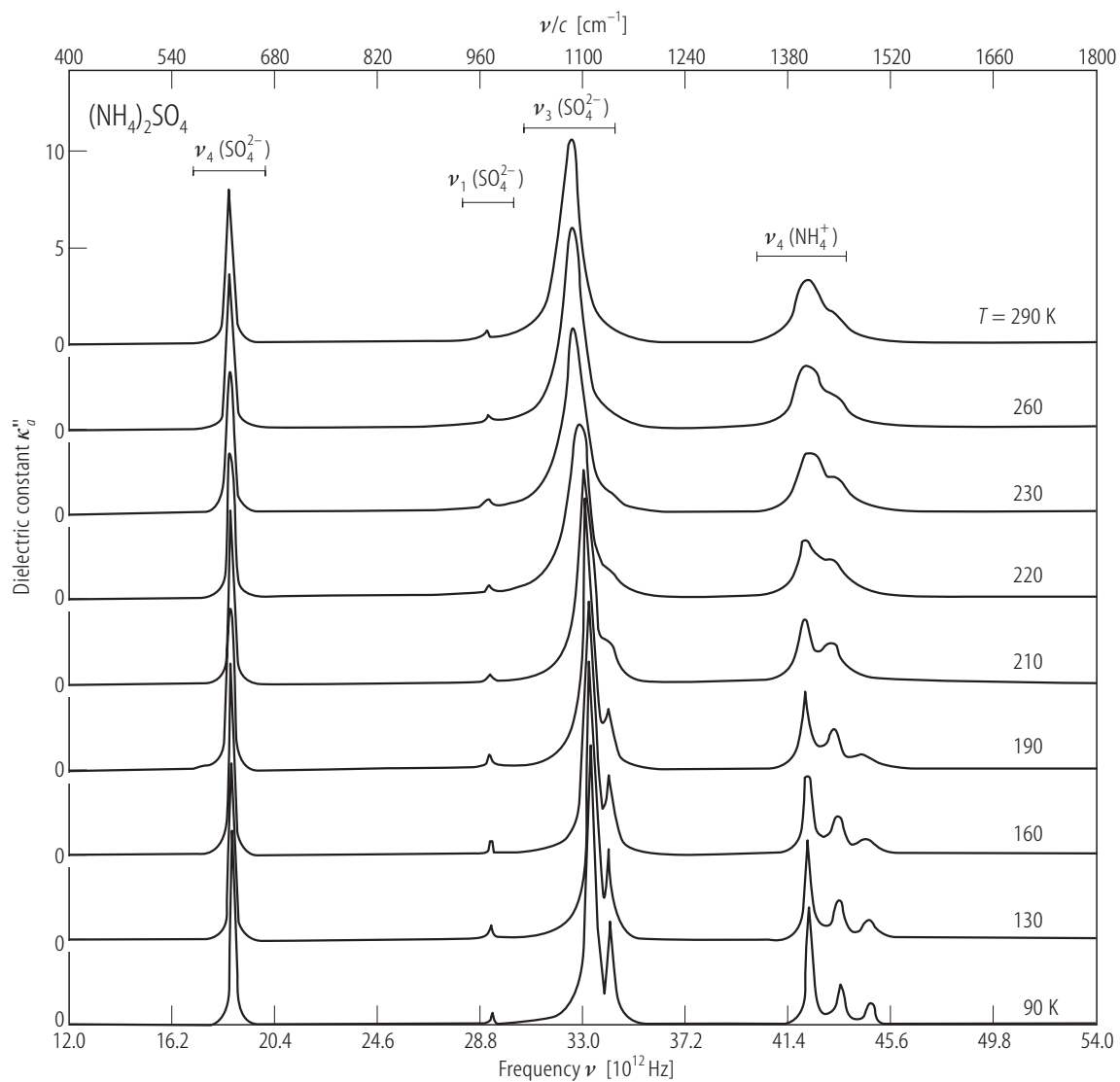


Fig. 39A-1-023. (NH₄)₂SO₄. κ''_a vs. ν [95DeS]. ν : frequency of incident far-infrared radiation. Parameter: T . κ''_a is obtained from the reflectivity data by the fit of the factorized form of κ .

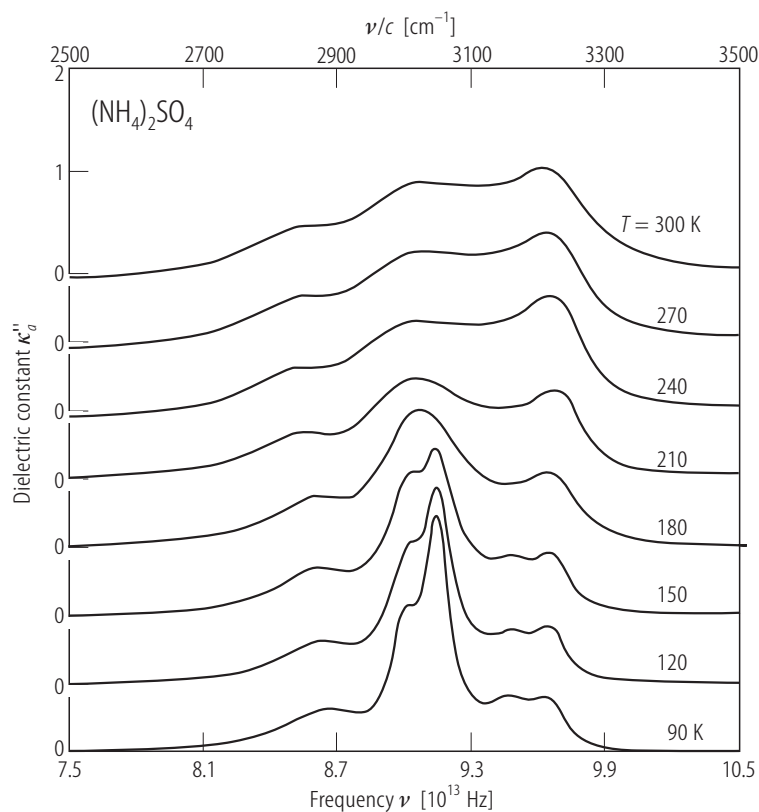


Fig. 39A-1-024. (NH₄)₂SO₄. κ''_a vs. ν [95DeS]. ν : frequency of incident far-infrared radiation. Parameter: T . κ''_a is obtained from the reflectivity data by the fit of the factorized form of κ .

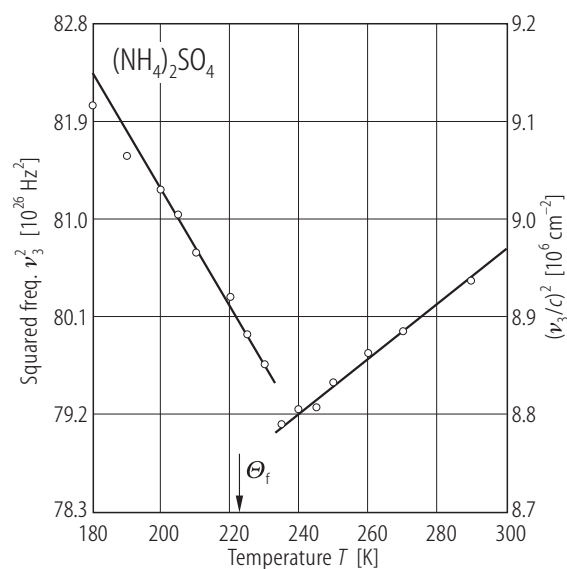


Fig. 39A-1-025. (NH₄)₂SO₄. ν_3^2 vs. T [95DeS]. ν_3 : frequency of the internal ν_3 mode of NH₄⁺ near 3000 cm⁻¹.

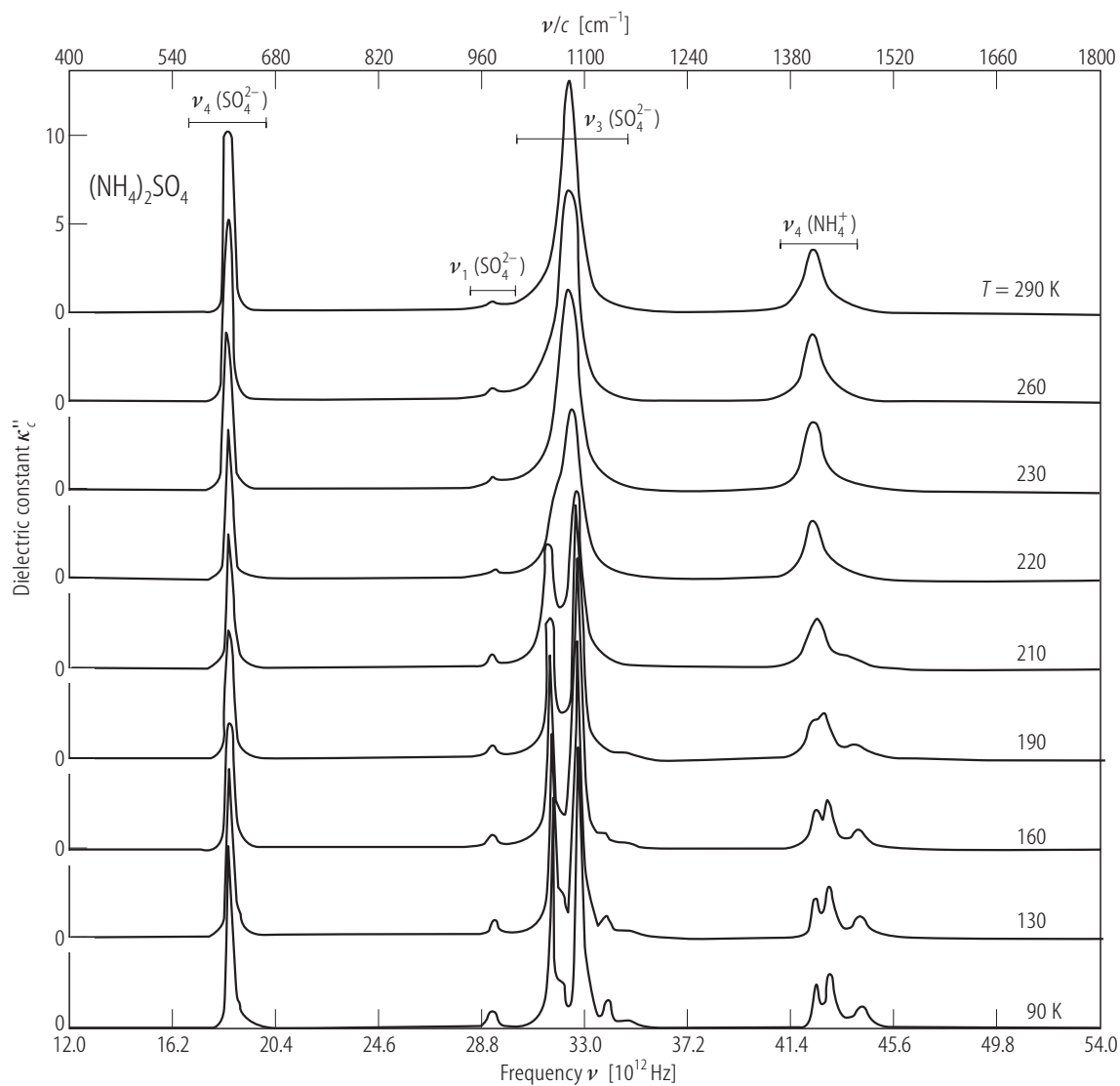


Fig. 39A-1-026. (NH₄)₂SO₄. κ''_c vs. ν [95DeS]. ν : frequency of incident far-infrared radiation. κ''_c is obtained from the reflectivity data by the fit of the factorized form of κ .

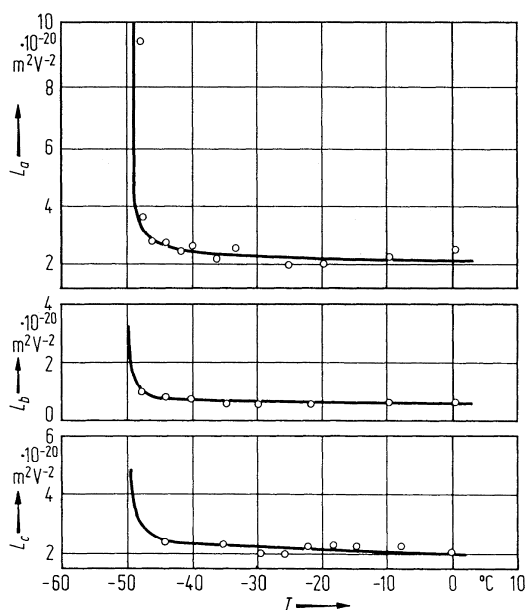


Fig. 39A-1-027. (NH₄)₂SO₄. L_a , L_b , L_c vs. T [70Ani]. $L_a = n_c^3 L_{33} - n_b^3 L_{23}$, $L_b = n_a^3 L_{13} - n_c^3 L_{33}$, $L_c = n_b^3 L_{23} - n_a^3 L_{13}$. $L_{\lambda\mu}$: quadratic electrooptic constant for E .

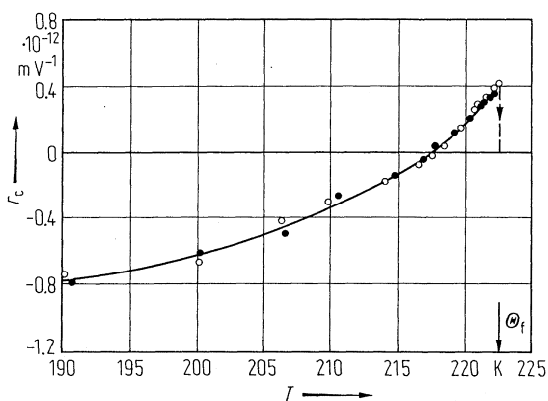


Fig. 39A-1-028. (NH₄)₂SO₄. r_c vs. T [76Fou]. $r_c = r_{13} - r_{23}$.

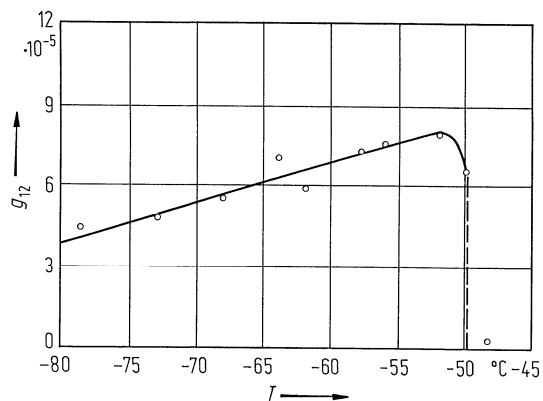


Fig. 39A-1-029. (NH₄)₂SO₄. g_{12} vs. T [80Kob]. g_{12} : gyration tensor component. $\lambda = 456.5$ nm.

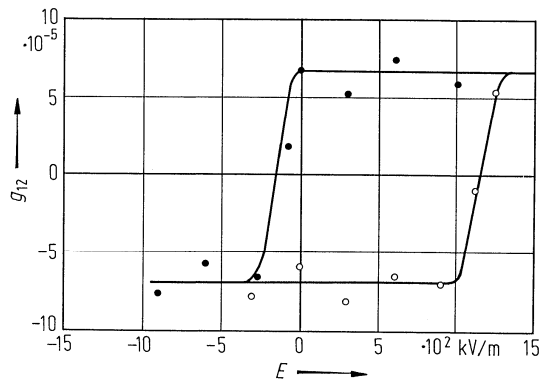


Fig. 39A-1-030. (NH₄)₂SO₄. g_{12} vs. E [80Kob]. g_{12} : gyration tensor component. $T = -57$ °C. $\lambda = 456.5$ nm.

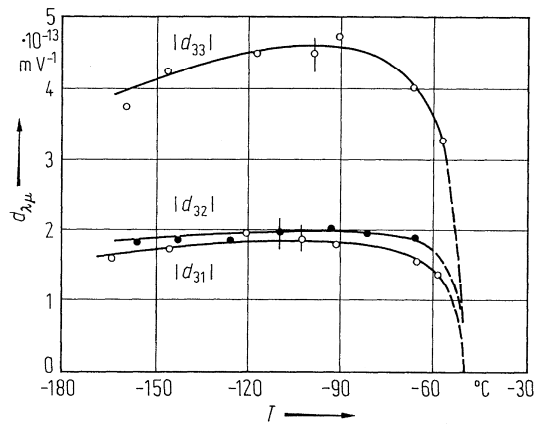


Fig. 39A-1-031. (NH₄)₂SO₄. $d_{\lambda\mu}$ vs. T [77Suz]. $d_{\lambda\mu}$: nonlinear optical susceptibility.

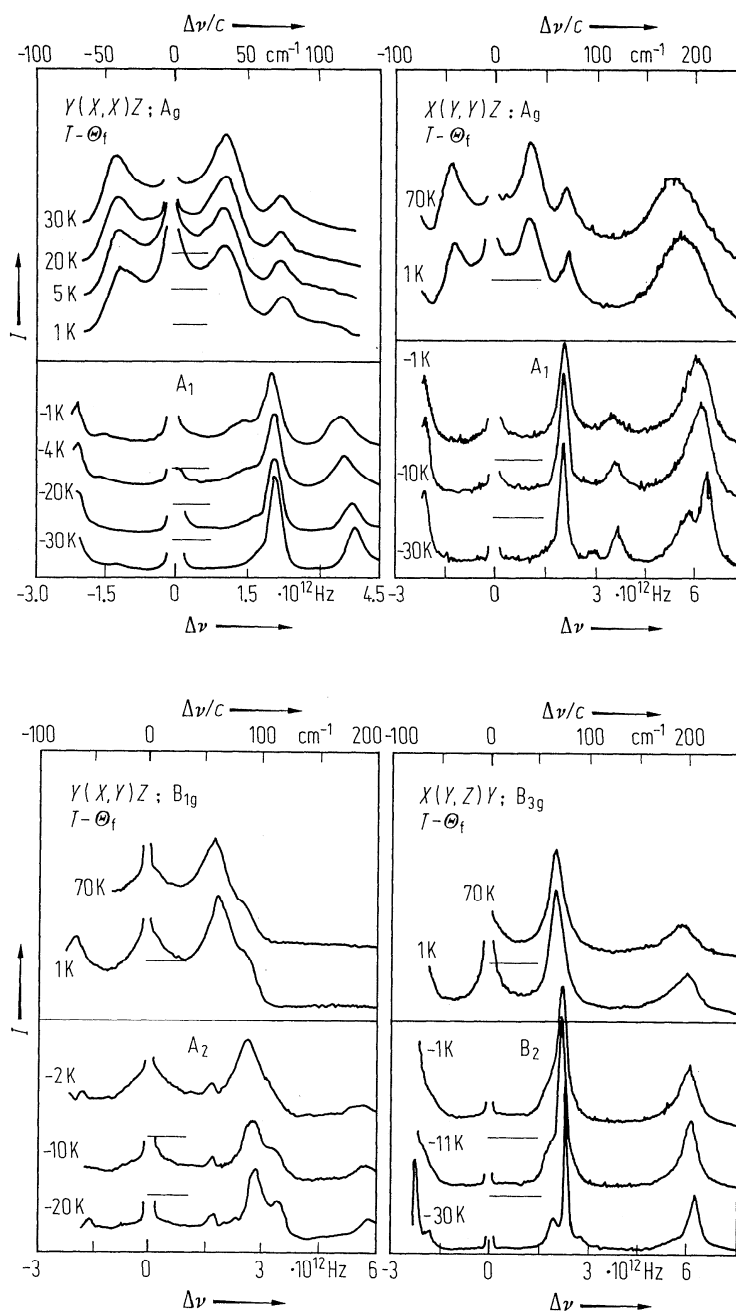


Fig. 39A-1-032. (NH₄)₂SO₄. *I* vs. $\Delta\nu$ above and below Θ_f in the low frequency region [78Unr]. *I*: Raman scattering intensity, $\Delta\nu$: frequency shift of Raman scattering. Scattering geometries and the corresponding symmetries of phonons are given in the figure.

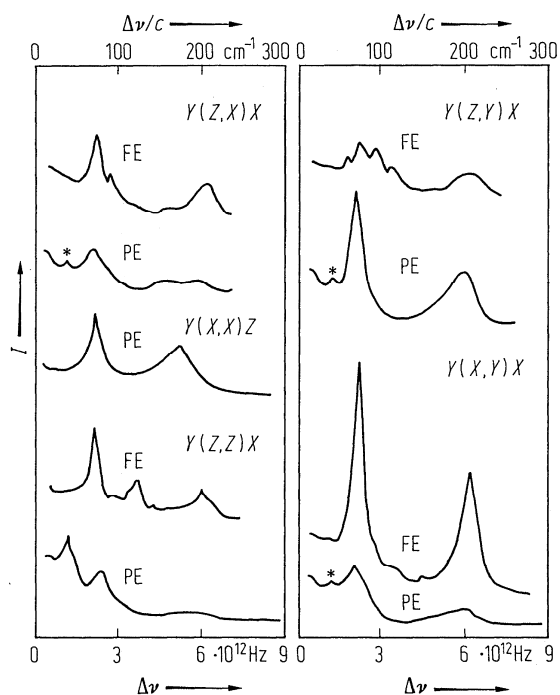


Fig. 39A-1-033. (NH₄)₂SO₄. *I* vs. $\Delta\nu$ [76Iqb]. *I*: Raman scattering intensity of the low frequency region in the paraelectric phase (PE) at 240 K and ferroelectric phase (FE) at 214 K. $\Delta\nu$: frequency shift of Raman scattering. Asterisks indicate spill-over of peaks of other symmetry.

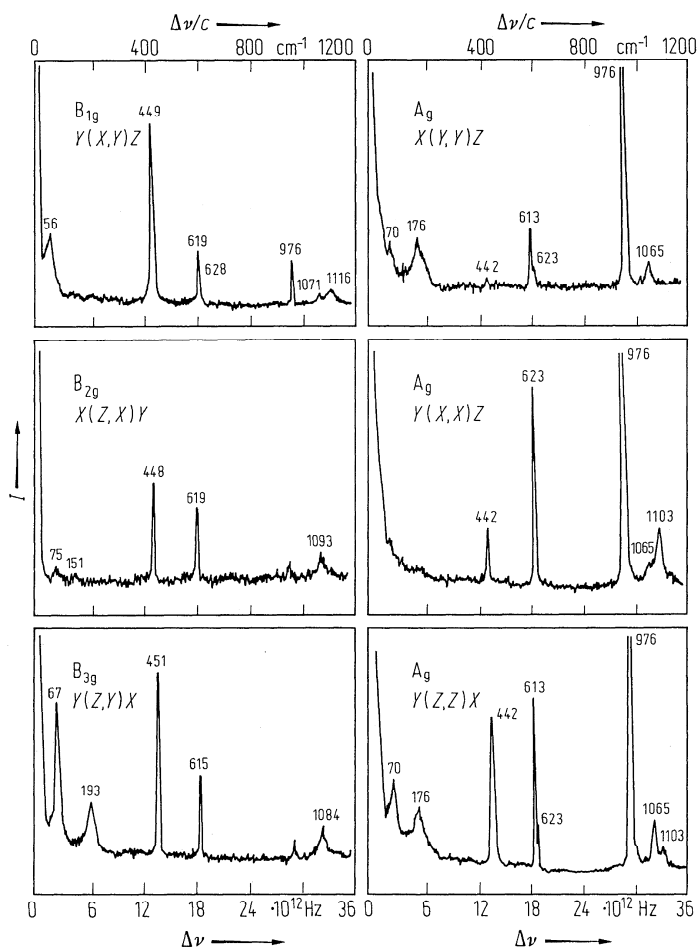


Fig. 39A-1-034. (NH₄)₂SO₄. I vs. $\Delta\nu$ [74Uch]. I : Raman scattering intensity at room temperature for various scattering geometries. $\Delta\nu$: frequency shift of Raman scattering. Peak frequencies are given in the figure in units of cm^{-1} .

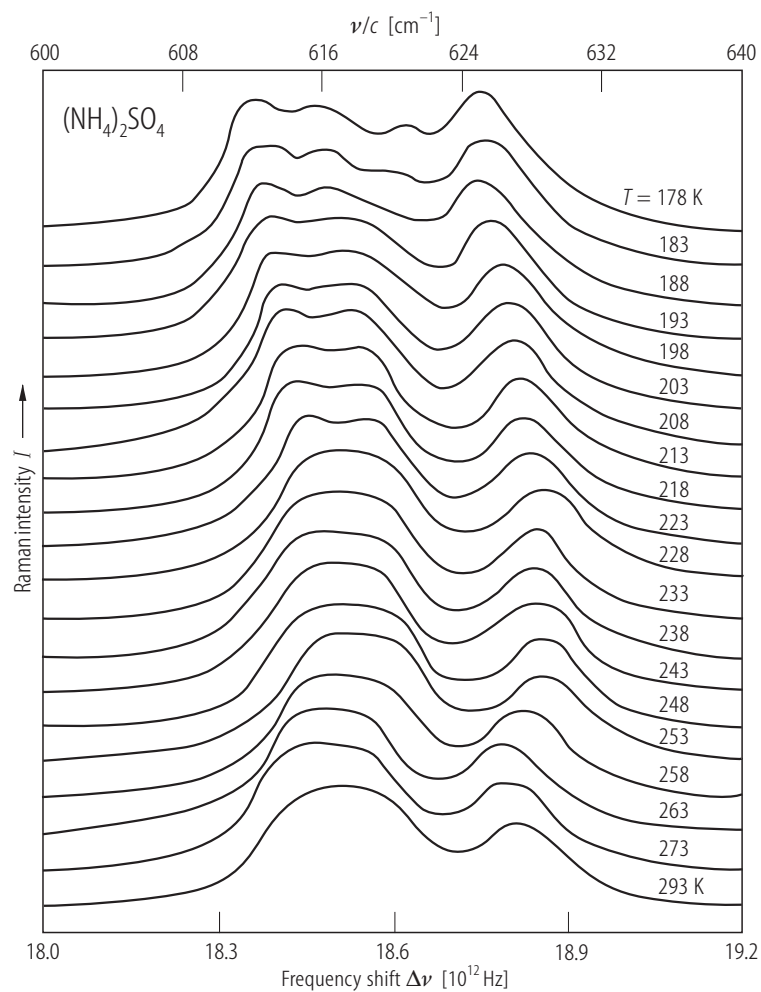


Fig. 39A-1-035. (NH₄)₂SO₄. I vs. $\Delta\nu$ [90Baj]. I : Raman scattering intensity of the ν_4 mode of the SO_4^{2-} ion. $\Delta\nu$: frequency shift of Raman scattering. Parameter: T .

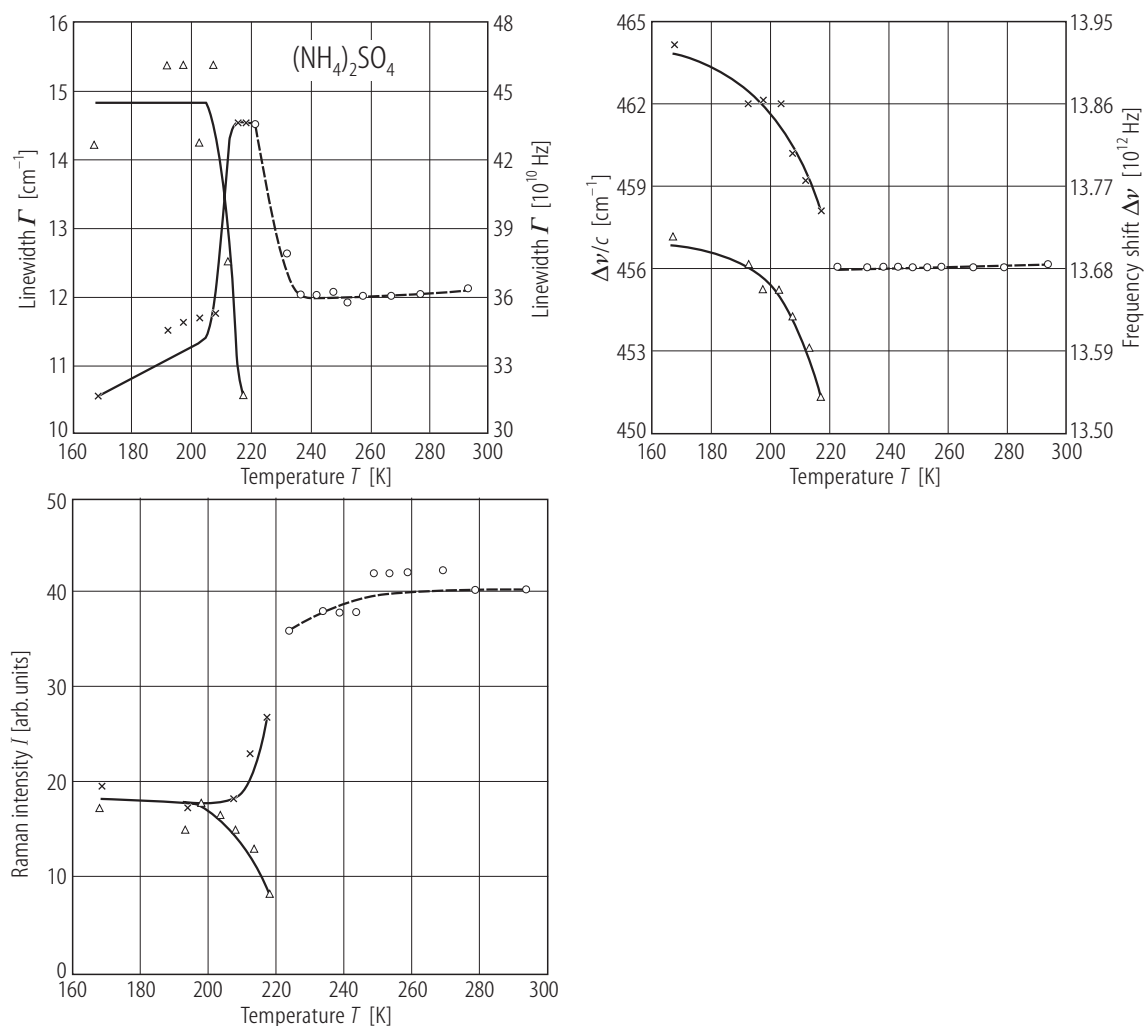


Fig. 39A-1-036. (NH₄)₂SO₄. Γ , $\Delta\nu$, I vs. T [90Ba]. The ν_2 mode of the SO₄²⁻ ion. Γ : Raman scattering linewidth, $\Delta\nu$: frequency shift of Raman scattering, I : Raman scattering intensity. The ν_2 mode splits into two branches (triangle and cross) below Θ .

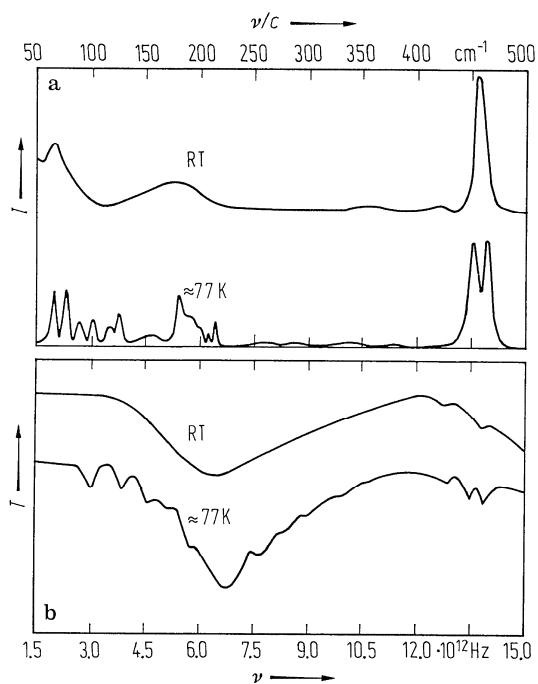


Fig. 39A-1-037. (NH₄)₂SO₄. I , T vs. ν [72Tor]. I : Raman scattering intensity (a), T : transmittance for the infrared radiation (b). Sample: polycrystal. ν : frequency shift in (a), and frequency of the incident radiation in (b).

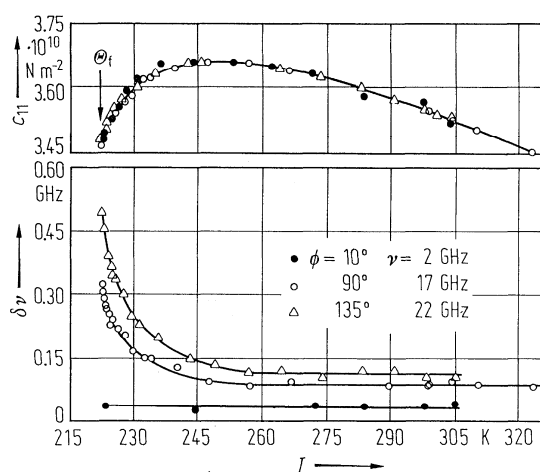


Fig. 39A-1-038. (NH₄)₂SO₄. c_{11} , $\delta\nu$ vs. T [78Unr]. c_{11} : elastic stiffness. $\delta\nu$: linewidth of phonon spectrum at different frequencies ν , i.e. different scattering angles ϕ .

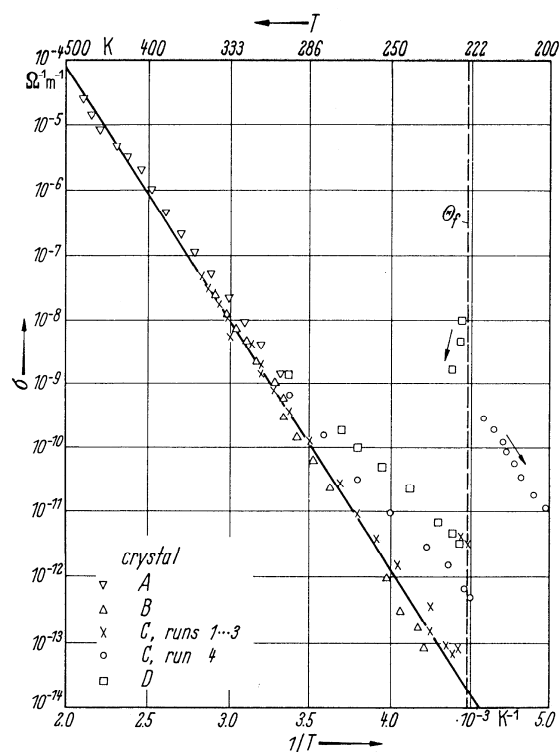


Fig. 39A-1-039. (NH₄)₂SO₄. σ vs. T^{-1} [63Sch]. All points near and below Θ_f were obtained from cooling runs, except for the three isolated high-conductivity points for crystal D which were obtained just above Θ_f while warming the crystal immediately after having cooled through the transition point. Conducting silver paint was applied on the (010) faces as electrodes: dc field of about 400 V was applied across crystals of 1...2 mm thickness. Probably the charge carriers are protons.

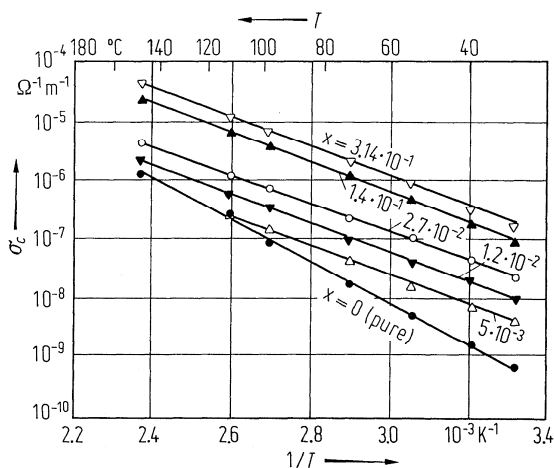


Fig. 39A-1-040. (NH₄)₂SO₄. σ_c vs. T^{-1} for (HSO₄)⁻¹ doped crystals [74Kha]. x: contents of (HSO₄)⁻¹ ions in (NH₄)₂SO₄ in mol%.

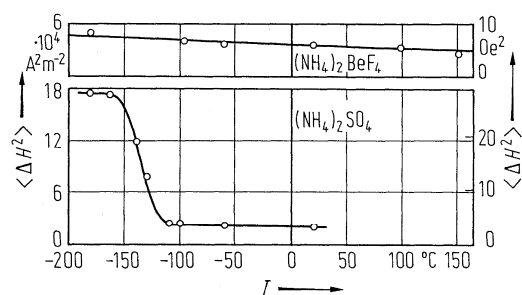


Fig. 39A-1-041. $(\text{NH}_4)_2\text{SO}_4$, $(\text{NH}_4)_2\text{BeF}_4$. $\langle \Delta H^2 \rangle$ vs. T [60Bli]. $\langle \Delta H^2 \rangle$: second moment of magnetic resonance curve of ^1H .

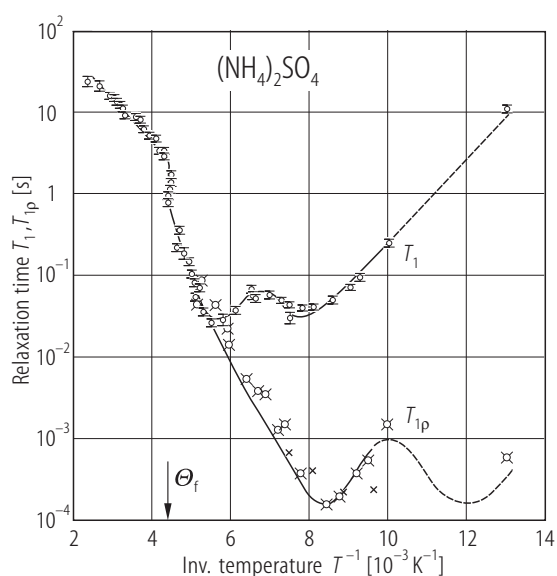


Fig. 39A-1-042. $(\text{NH}_4)_2\text{SO}_4$. T_1 , T_{1p} vs. T^{-1} [67ORE]. T_1 : proton spin-lattice relaxation time; T_{1p} : proton spin-lattice relaxation time along the radio frequency field. Open circle: T_1 at 42 MHz; crossed circle: T_{1p} by the method of Look et al. [66Loo]; cross: T_{1p} by the method of Ostroff and Waugh [66Ost].

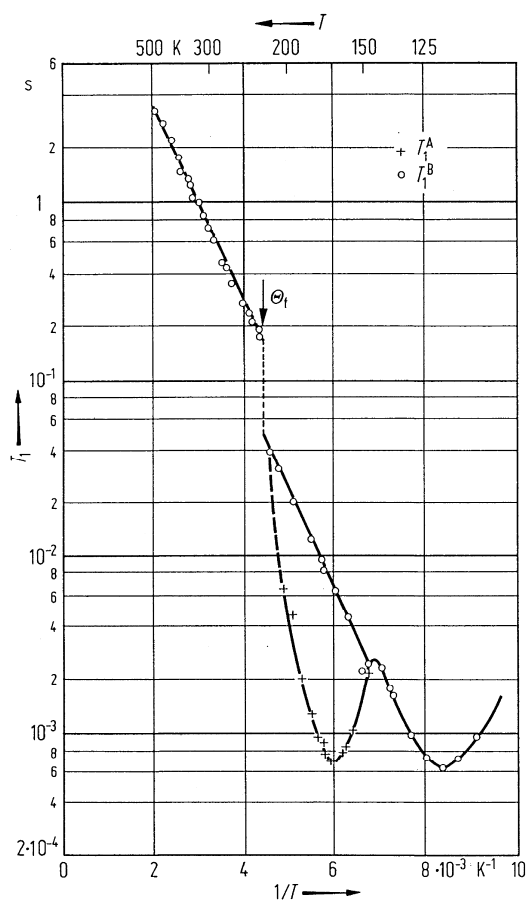


Fig. 39A-1-043. (ND₄)₂SO₄. T_1 vs. T^{-1} [69Kyd]. T_1 : deuteron spin-lattice relaxation time. There are two well-separated sets of T_1 around Θ_f , denoted as T_1^A and T_1^B . $\nu_L = 11.5$ MHz.

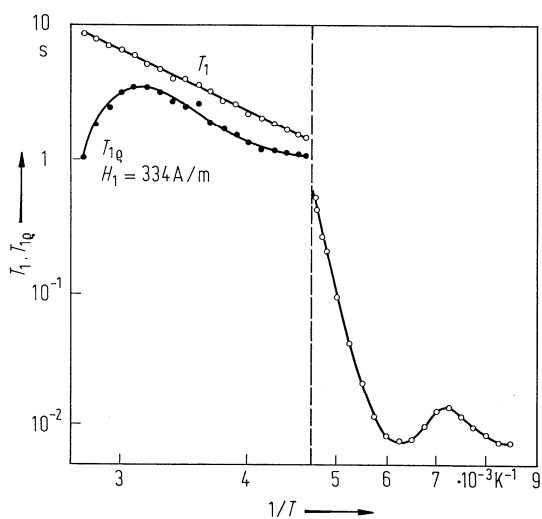


Fig. 39A-1-044. (NH₄)₂SO₄. T_1 , $T_{1\rho}$ vs. $1/T$ [84Mor]. T_1 and $T_{1\rho}$ are spin-lattice relaxation times of proton in laboratory frame and rotating frame, respectively. $\nu_L = 16$ MHz. The scale of abscissa is changed at $1/T = 4.5 \cdot 10^{-3} \text{ K}^{-1}$.

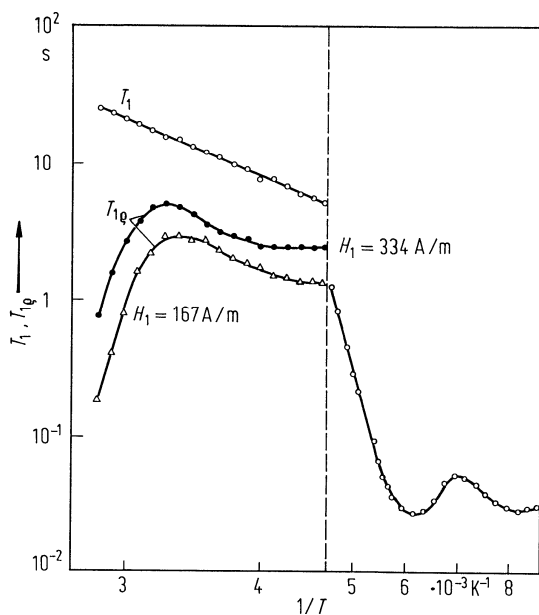


Fig. 39A-1-045. (NH₄(1-x)D_{4x})₂SO₄ (x = 0.75). T_1 , $T_{1\rho}$ vs. $1/T$ [84Mor]. Parameter: H_1 . T_1 and $T_{1\rho}$ are spin-lattice relaxation times of proton in laboratory frame and rotating frame, respectively. $\nu_L = 16$ MHz. H_1 : strength of rf field. The scale of abscissa is changed at $1/T = 4.5 \cdot 10^{-3} \text{ K}^{-1}$.

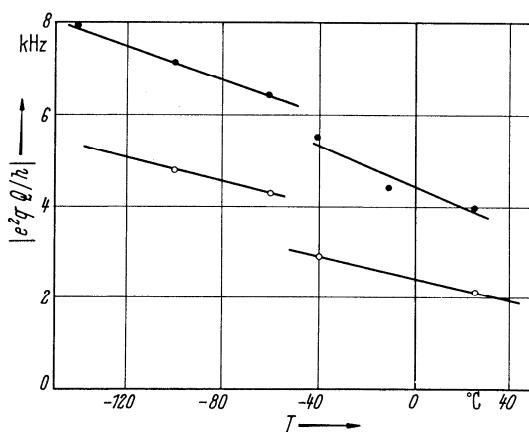


Fig. 39A-1-046. (ND₄)₂SO₄. $|e^2qQ/h|$ vs. T [67ORE]. $|e^2qQ/h|$: quadrupole coupling constant of ²D. The different marks of data points correspond to the two inequivalent (ND₄)⁺ ions.

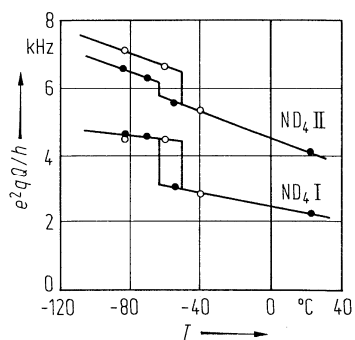


Fig. 39A-1-047. (ND₄)₂SO₄. e^2qQ/h vs. T [75Vin]. e^2qQ/h : quadrupole coupling constant of ²D. ND₄ I and ND₄ II correspond to the two non-equivalent (ND₄)⁺ ions. Open circles: at atmospheric pressure; full circles: under hydrostatic pressure of $3.5 \cdot 10^8$ Pa.

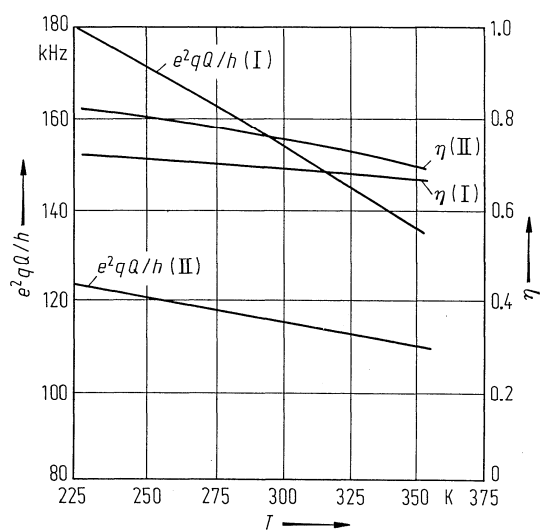


Fig. 39A-1-048. (NH₄)₂SO₄. e^2qQ/h , η vs. T for ¹⁴N [80Bat]. I and II refer to different ¹⁴N sites.

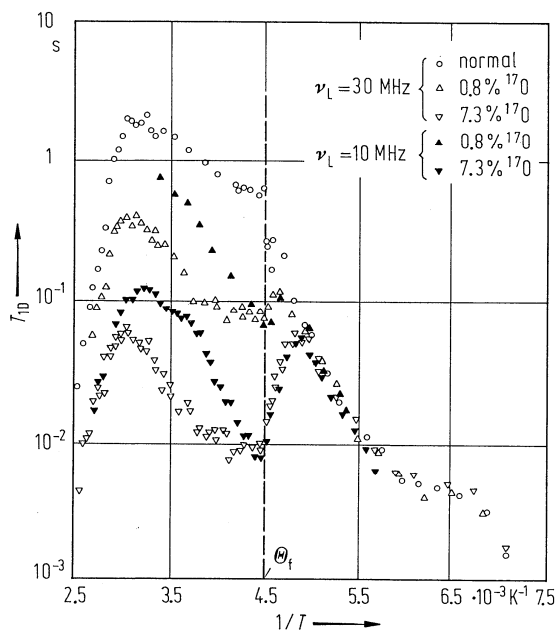


Fig. 39A-1-049. (NH₄)₂SO₄ (¹⁷O enriched). T_{1D} vs. $1/T$ [85Chi]. T_{1D} : ¹⁷O spin-lattice relaxation time in dipolar field.

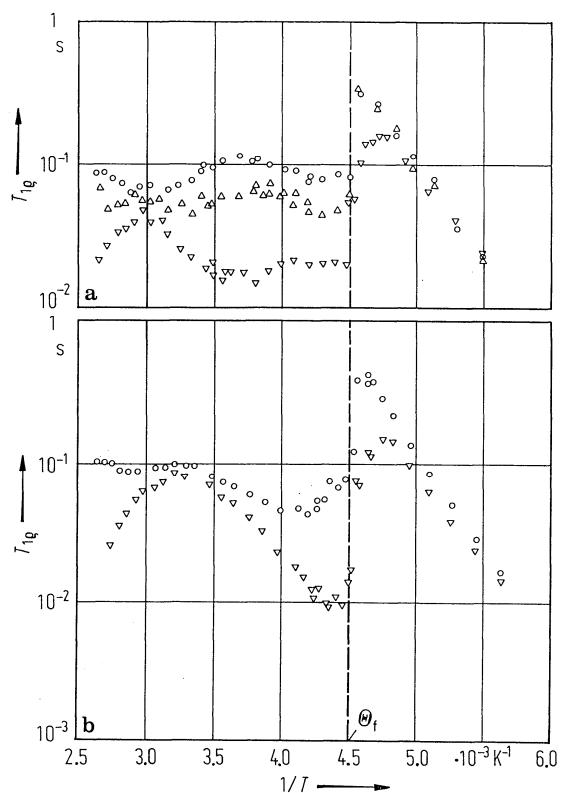


Fig. 39A-1-050. (NH₄)₂SO₄ (7.3% ¹⁷O enriched). $T_{1\rho}$ vs. $1/T$ [85Chi]. Parameter: H_1 . $T_{1\rho}$: ¹⁷O spin-lattice relaxation time in rotating frame. (a) $\nu_L = 30$ MHz; $H_1 = 1.4 \cdot 10^3$ A m⁻¹ (circle), $1.0 \cdot 10^3$ A m⁻¹ (upside triangle), $0.32 \cdot 10^3$ A m⁻¹ (downside triangle). (b) $\nu_L = 10$ MHz; $H_1 = 1.6 \cdot 10^3$ A m⁻¹ (circle), $0.32 \cdot 10^3$ A m⁻¹ (downside triangle).

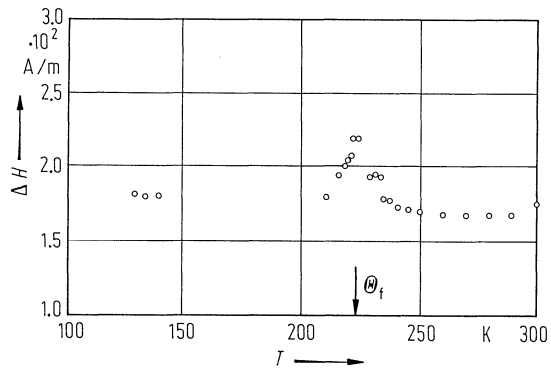


Fig. 39A-1-051. (NH₄)₂SO₄. ΔH vs. T [86Rav]. ESR. ΔH : linewidth for the high field singlet of the (NH₃)⁺ radicals. $f = 9.4$ GHz.

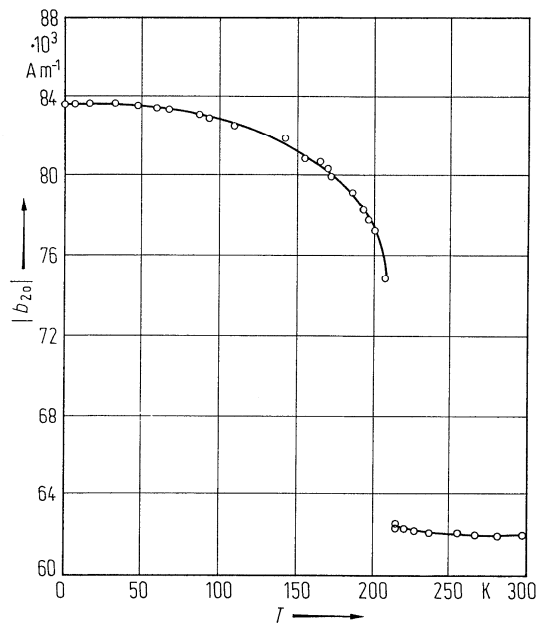


Fig. 39A-1-052. (NH₄)₂SO₄:Mn²⁺. $|b_{20}|$ vs. T [72Adb]. ESR for Mn²⁺ doped crystal. b_{20} : FS parameter.

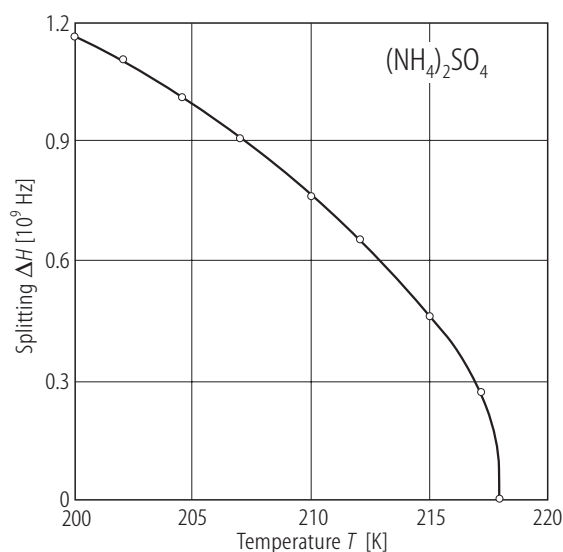


Fig. 39A-1-053. (NH₄)₂SO₄:Mn²⁺. ΔH vs. T [88Mis]. ESR. ΔH : splitting of the $(-3/2, 5/2 \leftrightarrow -5/2, 5/2)$ transition of Mn²⁺. The continuous line indicates calculated values of $\Delta H = H_0[(\Theta - T)/\Theta]^\beta$ with $H_0 = 4.053$ GHz, $\Theta = 218$ K, $\beta = 0.5$.

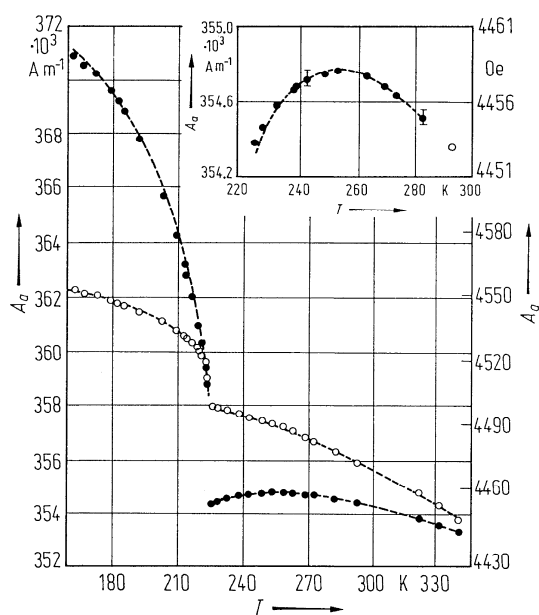


Fig. 39A-1-054. (NH₄)₂SO₄:Cd⁺. A_a vs. T [78Bar]. ESR of Cd⁺ impurity. A_a : hyperfine structure parameter of Cd⁺. External magnetic field is parallel to the a axis, one of the principal axes of the A tensor. Full circles: Cd⁺ in α -position; open circles: Cd⁺ in β -position.

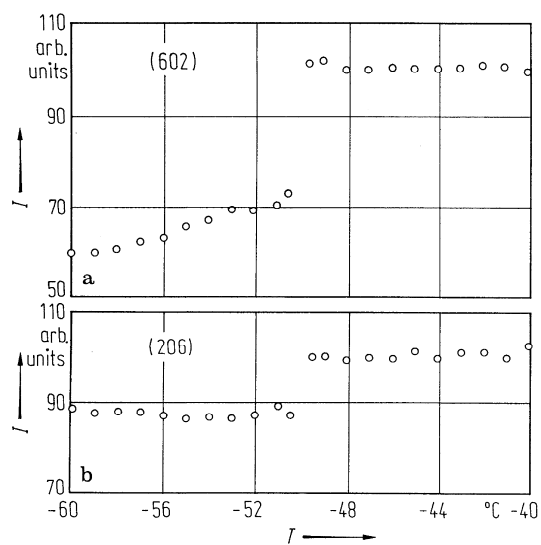


Fig. 39A-1-055. $(\text{NH}_4)_2\text{SO}_4$. (a) $I_{(6,0,2)}$, (b) $I_{(2,0,6)}$ vs. T [77Has]. I : integrated intensity of X-ray Bragg reflection.

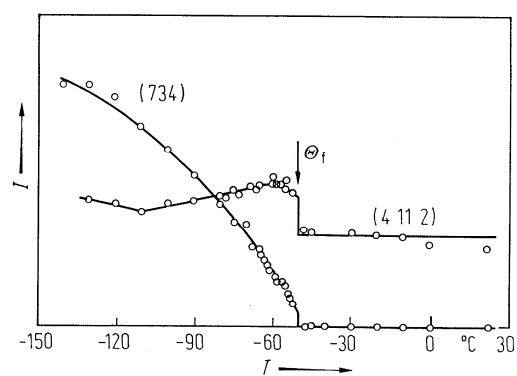


Fig. 39A-1-056. $(\text{NH}_4)_2\text{SO}_4$. $I_{(7,3,4)}$, $I_{(4,11,2)}$ vs. T [78Ono]. I : integrated intensity of X-ray Bragg reflection.