

Table 43A-12-001. K₂Cd₂(SO₄)₃ (2% Co doped). Fractional atomic coordinates [89Per]. Parameter: *T*.

Atom	Parameter	440 K	540 K	640 K
K(1)	<i>x</i>	0.8146(2)	0.8146(2)	0.8148(3)
K(2)	<i>x</i>	0.0520(2)	0.0518(2)	0.0516(4)
Cd(1)	<i>x</i>	0.32984(5)	0.3303(1)	0.3305(1)
Cd(2)	<i>x</i>	0.58701(5)	0.5872(1)	0.5876(1)
S(1)	<i>x</i>	0.3752(2)	0.3752(2)	0.3749(2)
	<i>y</i>	0.2266(2)	0.2264(2)	0.2263(2)
	<i>z</i>	0.0117(1)	0.0119(2)	0.0121(2)
O(1)	<i>x</i>	0.0962(7)	0.0962(7)	0.0978(10)
	<i>y</i>	0.0064(11)	0.0088(11)	0.0080(15)
	<i>z</i>	0.3244(8)	0.3242(9)	0.3260(11)
O(2)	<i>x</i>	0.2811(8)	0.2816(8)	0.2865(12)
	<i>y</i>	0.3180(8)	0.3176(7)	0.3171(11)
	<i>z</i>	0.9602(10)	0.9608(10)	0.9583(15)
O(3)	<i>x</i>	0.4175(10)	0.4160(10)	0.4170(14)
	<i>y</i>	0.2602(11)	0.2600(11)	0.2632(14)
	<i>z</i>	0.1431(7)	0.1437(8)	0.1412(11)
O(4)	<i>x</i>	0.9311(9)	0.9322(9)	0.9335(12)
	<i>y</i>	0.4891(10)	0.4890(10)	0.4878(13)
	<i>z</i>	0.2394(12)	0.2394(11)	0.2434(14)

Table 43A-12-002. K₂Cd₂(SO₄)₃ (2% Co doped). Temperature parameters [89Per]. Parameter: T . U_{ij} [Å²] is defined by Eq. (d) in Introduction.

Atom	Parameter	440 K	540 K	640 K
K(1)	U_{11}	0.0405(7)	0.0480(8)	0.0528(13)
	U_{23}	−0.0022(7)	−0.0017(9)	−0.0014(13)
K(2)	U_{11}	0.0432(8)	0.0514(10)	0.0576(15)
	U_{23}	0.0008(8)	0.0014(10)	0.0020(15)
Cd(1)	U_{11}	0.0277(2)	0.0299(2)	0.0322(3)
	U_{23}	−0.0040(2)	−0.0031(2)	−0.0027(3)
Cd(2)	U_{11}	0.0236(2)	0.0278(2)	0.0311(3)
	U_{23}	−0.0010(2)	−0.0012(2)	−0.0016(3)
S(1)	U_{11}	0.0185(6)	0.0216(7)	0.0239(10)
	U_{22}	0.0189(7)	0.0218(8)	0.0242(12)
	U_{33}	0.0183(7)	0.0207(7)	0.0229(11)
	U_{23}	0.0008(5)	0.0003(6)	0.0000(9)
	U_{13}	0.0018(5)	0.0019(5)	0.0018(8)
	U_{12}	0.0040(5)	0.0042(5)	0.0047(8)
O(1)	U_{11}	0.0263(31)	0.0257(34)	0.0301(49)
	U_{22}	0.1168(76)	0.1202(79)	0.1184(108)
	U_{33}	0.0647(46)	0.0763(54)	0.0715(72)
	U_{23}	−0.0384(50)	−0.0401(54)	−0.0377(73)
	U_{13}	−0.0119(32)	−0.0128(34)	−0.0188(47)
	U_{12}	0.0130(39)	0.0127(42)	0.0153(60)
O(2)	U_{11}	0.0572(52)	0.0589(54)	0.0625(79)
	U_{22}	0.0543(49)	0.0563(51)	0.0592(75)
	U_{33}	0.1137(80)	0.1148(79)	0.1249(118)
	U_{23}	0.0061(47)	0.0041(48)	−0.0010(70)
	U_{13}	−0.0339(48)	−0.0387(49)	−0.0465(72)
	U_{12}	0.0331(40)	0.0304(41)	0.0340(61)
O(3)	U_{11}	0.0790(61)	0.0880(68)	0.0819(88)
	U_{22}	0.1078(72)	0.1040(72)	0.0909(90)
	U_{33}	0.0343(36)	0.0405(41)	0.0460(59)
	U_{23}	−0.0309(41)	−0.0309(41)	−0.0309(55)
	U_{13}	−0.0160(39)	−0.0187(44)	−0.0091(60)
	U_{12}	0.0079(59)	0.0116(62)	0.0052(78)
O(4)	U_{11}	0.0826(70)	0.0773(66)	0.0749(87)
	U_{22}	0.0638(58)	0.0663(61)	0.0638(80)
	U_{33}	0.1433(91)	0.1387(87)	0.1122(100)
	U_{23}	0.0253(60)	0.0283(60)	0.0191(72)
	U_{13}	0.0258(71)	0.0184(68)	0.0061(82)
	U_{12}	0.0615(58)	0.0535(57)	0.0483(74)

Table 43A-12-003. K₂Cd₂(SO₄)₃. Atomic coordinates [78Abr]. Five rows given for each atom correspond to the coordinates at 298, 351, 390.5, 417.5 K (phase II) and 443.5 K (phase I).

Atom	x	y	z	Atom	x	y	z
K (1)	0.17012 (9)	0.31356 (10)	0.69223 (10)	O (4)	0.2110 (3)	0.4345 (3)	0.1572 (3)
	0.16941 (20)	0.31319 (21)	0.69278 (24)		0.2104 (7)	0.4344 (7)	0.1567 (7)
	0.17104 (12)	0.31240 (13)	0.69221 (12)		0.2158 (4)	0.4333 (4)	0.1548 (4)
	0.17315 (21)	0.31260 (19)	0.69123 (23)		0.2206 (9)	0.4318 (7)	0.1542 (8)
	0.18522 (37)	0.31478	0.68522		0.2594	0.4178	0.1431
K (2)	0.03815 (10)	0.05562 (12)	0.06101 (10)	O (5)	0.3054 (3)	0.9308 (3)	0.3384 (3)
	0.03830 (24)	0.05571 (27)	0.06070 (26)		0.3042 (8)	0.9315 (7)	0.3391 (8)
	0.03993 (15)	0.05457 (16)	0.06013 (12)		0.3028 (4)	0.9343 (5)	0.3382 (4)
	0.04203 (25)	0.05334 (24)	0.05926 (23)		0.2981 (7)	0.9418 (8)	0.3349 (6)
	0.05282 (39)	0.05282	0.05282		0.2786 (4)	0.9616 (7)	0.3165 (4)
Cd (1)	0.30681 (2)	0.34105 (3)	0.33136 (3)	O (6)	0.3909 (3)	0.1427 (3)	0.2881 (3)
	0.30791 (5)	0.34086 (5)	0.33162 (5)		0.3912 (9)	0.1435 (7)	0.2879 (9)
	0.30957 (3)	0.34058 (3)	0.33177 (3)		0.3937 (6)	0.1445 (5)	0.2843 (5)
	0.31304 (6)	0.33944 (4)	0.33198 (6)		0.3968 (9)	0.1460 (6)	0.2796 (7)
	0.32987 (9)	0.32987	0.32987		0.4178 (6)	0.1431 (4)	0.2594 (7)
Cd (2)	0.39831 (2)	0.09039 (3)	0.92580 (3)	O (7)	0.5176 (3)	0.9550 (3)	0.2517 (4)
	0.39920 (5)	0.08996 (5)	0.92526 (5)		0.5214 (8)	0.9532 (8)	0.2526 (11)
	0.40030 (3)	0.08974 (3)	0.92451 (3)		0.5168 (4)	0.9531 (5)	0.2529 (6)
	0.40262 (6)	0.08920 (5)	0.92298 (5)		0.5129 (8)	0.9499 (8)	0.2476 (9)
	0.41305 (9)	0.08695	0.91305		0.4904	0.9303	0.2423
S (1)	0.21410 (8)	0.37222 (8)	0.02556 (9)	O (8)	0.3284 (4)	0.9881 (4)	0.1161 (3)
	0.21523 (15)	0.37238 (15)	0.02501 (18)		0.3265 (8)	0.9908 (8)	0.1132 (7)
	0.21600 (8)	0.37225 (9)	0.02394 (9)		0.3278 (5)	0.9923 (5)	0.1128 (4)
	0.21810 (17)	0.37313 (15)	0.02209 (18)		0.3245 (8)	0.9941 (8)	0.1122 (6)
	0.22678 (8)	0.37504 (8)	0.01178 (9)		0.3216	0.0053	0.0966
S (2)	0.38451 (8)	0.00655 (9)	0.24697 (9)	O (9)	0.0561 (3)	0.0940 (3)	0.3170 (3)
	0.38351 (15)	0.00769 (15)	0.24599 (17)		0.0559 (8)	0.0915 (7)	0.3186 (9)
	0.38339 (8)	0.00793 (9)	0.24467 (9)		0.0540 (6)	0.0951 (5)	0.3184 (5)
	0.38141 (19)	0.00895 (15)	0.24202 (17)		0.0441 (9)	0.0939 (7)	0.3187 (9)
	0.37504	0.01178	0.22678		0.0053	0.0966	0.3216
S (3)	0.00036 (8)	0.21604 (8)	0.36900 (8)	O (10)	0.9031 (3)	0.1885 (4)	0.4702 (3)
	0.00051 (15)	0.21700 (15)	0.36951 (17)		0.9052 (11)	0.1896 (12)	0.4721 (9)
	0.00129 (9)	0.21723 (9)	0.36943 (9)		0.9059 (5)	0.1903 (7)	0.4731 (5)
	0.00030 (17)	0.21873 (14)	0.37001 (19)		0.9089 (9)	0.1979 (12)	0.4736 (9)
	0.01178	0.22678	0.37504		0.9303 (6)	0.2423 (8)	0.4904 (5)
O (1)	0.3056 (3)	0.2628 (3)	0.0336 (3)	O (11)	0.9353 (3)	0.2817 (4)	0.2570 (4)
	0.3037 (10)	0.2626 (8)	0.0294 (10)		0.9339 (9)	0.2828 (9)	0.2603 (9)
	0.3076 (5)	0.2638 (5)	0.0272 (5)		0.9376 (6)	0.2868 (6)	0.2629 (5)
	0.3067 (10)	0.2645 (7)	0.0189 (9)		0.9417 (10)	0.2928 (9)	0.2654 (9)
	0.3165	0.2786	0.9616		0.9616	0.3165	0.2786
O (2)	0.0824 (3)	0.3259 (3)	0.9880 (3)	O (12)	0.1063 (3)	0.2980 (4)	0.4217 (3)
	0.0829 (7)	0.3286 (8)	0.9885 (9)		0.1074 (6)	0.2966 (7)	0.4240 (9)
	0.0832 (4)	0.3274 (5)	0.9899 (5)		0.1091 (4)	0.2955 (5)	0.4235 (5)
	0.0848 (7)	0.3261 (7)	0.9909 (8)		0.1155 (9)	0.2896 (10)	0.4230 (8)
	0.0966 (4)	0.3216 (5)	0.0053 (6)		0.1431	0.2594	0.4178
O (3)	0.2593 (3)	0.4634 (3)	0.9226 (3)				
	0.2607 (7)	0.4640 (7)	0.9266 (8)				
	0.2599 (5)	0.4644 (5)	0.9250 (5)				
	0.2582 (10)	0.4659 (8)	0.9223 (7)				
	0.2423	0.4904	0.9303				

Table 43A-12-004. K₂Cd₂(SO₄)₃. Temperature parameters B_{ij} [$\cdot 10^{-5}$] at the same temperatures as those in Table 43A-12-003 [78Abr]. B_{ij} is defined by Eq. (a) in Introduction.

Atom	B_{11}	B_{22}	B_{33}	B_{12}	B_{13}	B_{23}
K (1)	396 (7)	408 (7)	411 (7)	−98 (11)	−66 (11)	32 (11)
	471 (14)	454 (13)	468 (17)	−183 (18)	−72 (20)	20 (20)
	473 (8)	519 (8)	594 (9)	−133 (11)	−72 (11)	33 (12)
	500 (20)	594 (16)	636 (23)	−30 (12)	−36 (14)	4 (13)
	775 (23)	775	775	−80 (46)	−80	−80
K (2)	476 (8)	603 (8)	387 (7)	−87 (13)	94 (12)	−80 (13)
	515 (15)	722 (17)	408 (17)	−192 (23)	89 (23)	−96 (24)
	596 (9)	781 (11)	465 (8)	−140 (14)	83 (12)	−91 (13)
	677 (25)	805 (20)	515 (26)	−67 (15)	94 (16)	−2 (15)
	822 (26)	822	822	21 (50)	21	21
Cd (1)	214 (2)	240 (2)	247 (2)	19 (3)	−40 (3)	17 (3)
	252 (3)	259 (3)	245 (3)	17 (4)	−44 (5)	20 (5)
	281 (2)	295 (2)	335 (2)	4 (3)	−50 (3)	6 (3)
	388 (7)	377 (4)	450 (7)	−12 (3)	−39 (4)	−10 (3)
	527 (5)	527	527	−154 (11)	−154	−154
Cd (2)	257 (2)	266 (2)	246 (2)	3 (3)	−52 (3)	−23 (3)
	292 (3)	301 (3)	234 (3)	2 (5)	−62 (5)	−23 (5)
	308 (2)	337 (2)	320 (2)	7 (3)	−61 (3)	−29 (3)
	401 (7)	400 (5)	391 (7)	0 (3)	−38 (4)	−25 (3)
	458 (4)	458	458	−40 (10)	−40	−40
S (1)	180 (6)	194 (6)	229 (6)	50 (9)	8 (9)	−9 (9)
	249 (10)	220 (9)	150 (11)	43 (13)	−26 (14)	−28 (14)
	227 (6)	248 (6)	256 (6)	73 (8)	24 (7)	−34 (8)
	228 (17)	273 (12)	225 (18)	35 (8)	−32 (10)	−24 (9)
	377 (6)	358 (6)	315 (6)	167 (10)	10 (9)	50 (9)
S (2)	213 (6)	217 (6)	181 (6)	−34 (9)	−28 (9)	1 (10)
	272 (10)	218 (9)	178 (11)	−1 (14)	−39 (15)	33 (15)
	267 (6)	254 (6)	257 (5)	−24 (8)	−82 (8)	15 (9)
	312 (18)	219 (12)	300 (21)	−9 (9)	−100 (11)	17 (9)
	358	315	377	50	167	10
S (3)	187 (6)	211 (6)	210 (6)	−22 (9)	−49 (9)	−65 (9)
	207 (10)	209 (9)	206 (11)	−26 (13)	−39 (14)	−93 (14)
	256 (6)	248 (6)	250 (5)	−15 (8)	−60 (8)	−88 (8)
	242 (17)	238 (11)	261 (19)	−8 (8)	−28 (11)	−40 (9)
	315	377	358	10	50	167
O (1)	419 (23)	453 (25)	503 (27)	−511 (38)	246 (43)	−88 (43)
	937 (66)	494 (48)	409 (58)	−703 (78)	281 (97)	5 (76)
	644 (30)	438 (26)	731 (33)	−593 (40)	189 (51)	−21 (46)
	989 (93)	350 (55)	1198 (102)	−310 (52)	496 (77)	−198 (50)
	839	924	2346	817	493	−1181
O (2)	327 (23)	445 (25)	479 (26)	211 (38)	−21 (40)	380 (41)
	344 (42)	563 (48)	473 (57)	269 (67)	145 (68)	340 (76)
	294 (25)	604 (29)	727 (31)	159 (39)	−2 (38)	562 (45)
	313 (63)	728 (68)	1117 (100)	128 (49)	−49 (58)	445 (64)
	438 (32)	1245 (42)	2132 (64)	−514 (62)	569 (82)	−1632 (86)

(continued)

Table 43A-12-004 (continued)

Atom	<i>B</i> ₁₁	<i>B</i> ₂₂	<i>B</i> ₃₃	<i>B</i> ₁₂	<i>B</i> ₁₃	<i>B</i> ₂₃
O (3)	536 (27)	337 (25)	463 (26)	– 53 (41)	128 (45)	381 (41)
	381 (43)	405 (43)	575 (53)	– 11 (58)	269 (71)	587 (65)
	577 (31)	501 (28)	623 (27)	– 8 (43)	217 (45)	657 (41)
	1285 (110)	774 (74)	287 (85)	48 (62)	163 (74)	282 (55)
	2261	1208	1981	432	684	2334
O (4)	408 (23)	420 (25)	274 (22)	122 (39)	– 6 (38)	87 (38)
	514 (45)	487 (45)	214 (46)	177 (66)	202 (64)	21 (67)
	625 (30)	541 (29)	325 (25)	283 (42)	246 (35)	205 (38)
	1138 (102)	403 (61)	913 (101)	172 (54)	431 (73)	190 (53)
	2127	1356	513	– 371	– 1043	– 275
O (5)	309 (22)	584 (28)	307 (22)	14 (40)	– 258 (39)	– 224 (39)
	498 (47)	475 (44)	355 (48)	– 49 (69)	– 256 (76)	– 248 (67)
	337 (25)	819 (34)	455 (25)	– 155 (43)	– 179 (36)	– 366 (44)
	492 (70)	1138 (79)	30 (63)	– 336 (51)	– 146 (43)	– 114 (47)
	924 (37)	2346 (73)	839 (39)	– 1181 (81)	817 (54)	493 (83)
O (6)	541 (27)	325 (24)	478 (27)	209 (41)	– 81 (46)	– 274 (42)
	681 (57)	265 (42)	777 (68)	387 (68)	– 190 (94)	– 590 (70)
	740 (37)	311 (28)	858 (36)	48 (47)	– 308 (55)	– 503 (45)
	920 (88)	320 (50)	568 (71)	1 (46)	– 71 (58)	– 57 (37)
	1356 (59)	513 (34)	2127 (72)	– 275 (73)	– 371 (132)	– 1043 (76)
O (7)	196 (21)	529 (25)	686 (28)	333 (37)	278 (42)	96 (52)
	463 (53)	519 (51)	613 (66)	184 (76)	349 (88)	89 (92)
	288 (27)	645 (31)	869 (37)	362 (41)	276 (45)	97 (59)
	366 (76)	856 (67)	1072 (92)	274 (55)	269 (59)	9 (66)
	1208	1981	2261	2334	432	684
O (8)	672 (30)	473 (27)	301 (25)	– 479 (45)	177 (43)	53 (42)
	725 (53)	681 (53)	124 (45)	– 674 (76)	264 (68)	– 170 (69)
	756 (33)	702 (34)	242 (23)	– 537 (50)	175 (38)	– 125 (41)
	628 (72)	954 (68)	129 (66)	– 425 (58)	139 (49)	42 (43)
	1245	2132	438	– 1632	– 514	569
O (9)	492 (24)	329 (23)	455 (26)	341 (39)	67 (42)	– 139 (40)
	545 (49)	273 (44)	628 (63)	276 (63)	– 5 (84)	– 211 (74)
	947 (40)	457 (31)	567 (32)	532 (50)	– 7 (55)	– 246 (46)
	1311 (108)	263 (52)	1160 (104)	245 (54)	124 (82)	– 231 (52)
	2132	438	1245	569	– 1632	– 514
O (10)	538 (27)	626 (30)	450 (28)	– 472 (47)	– 402 (45)	– 80 (48)
	840 (69)	1037 (87)	253 (57)	– 419 (117)	– 666 (84)	– 90 (101)
	576 (31)	1169 (51)	464 (29)	– 610 (58)	– 499 (41)	– 61 (59)
	707 (90)	1909 (144)	451 (89)	– 58 (98)	– 381 (68)	– 123 (89)
	1981 (61)	2261 (92)	1208 (45)	684 (146)	2334 (71)	432 (113)
O (11)	347 (24)	643 (30)	526 (27)	64 (44)	188 (45)	– 483 (48)
	637 (60)	701 (56)	490 (56)	12 (83)	286 (83)	– 813 (75)
	685 (36)	760 (34)	672 (30)	159 (52)	369 (46)	– 868 (42)
	942 (107)	1097 (96)	827 (105)	217 (72)	366 (77)	– 538 (72)
	2346	839	924	493	– 1181	817
O (12)	324 (23)	592 (28)	491 (28)	408 (41)	– 92 (45)	190 (47)
	168 (37)	589 (48)	666 (59)	362 (59)	– 170 (71)	246 (82)
	401 (26)	743 (32)	653 (30)	611 (40)	– 14 (43)	306 (51)
	529 (77)	1134 (101)	829 (102)	359 (64)	– 198 (64)	235 (65)
	513	2127	1356	– 1043	– 275	– 371

Table 43A-12-005. K₂Cd₂(SO₄)₃ (2% Co doped). Deformation parameters ϕ , X , Δ for CdO₆ octahedra vs. T [89Per]. ϕ : rotation of the two oxygen triangles away from the ideal 60° rotation. X : displacement of the metal atom from the center of gravity of the octahedron along the triad axis towards the larger of the two triangles. Δ : variation in the larger of the two triangles of oxygen atoms. The distortion parameter Δ is defined as $\Delta = |A_1 - A_2| / (A_1 + A_2)$ where A_1 and A_2 are the areas of the two triangles.

Site	Parameter	440 K	540 K	640 K
M1	ϕ [°]	6.98	7.27	8.70
	X [Å]	0.132	0.124	0.121
	Δ	0.074	0.065	0.054
M2	ϕ [°]	−2.16	−1.51	−2.22
	X [Å]	0.198	0.203	0.207
	Δ	−0.123	−0.118	−0.108

Table 43A-12-006. K₂Cd₂(SO₄)₃. Electromechanical coupling factors $k_{i\lambda}$ and piezoelectric constants $d_{i\lambda}$ [81Ant].

T	k_{14}	k_{25}	k_{36}	d_{14}	d_{25}	d_{36}
[°C]	[·10 ^{−12} C N ^{−1}]					
25	0.08	0.08	0.04	3.5(2)	3.4(2)	1.7(2)
200	0.04	–	–	3.8(2)	–	–

Table 43A-12-007. K₂Cd₂(SO₄)₃. Elastic compliance coefficients [·10^{−11} m² N^{−1}] at 25 °C [81Ant].

$s_{11} = 1.8(2)$	$s_{12} = -1.1(2)$	$s_{44}^E = 6.45(6)$
$s_{22} = 1.6(2)$	$s_{13} = -1.2(2)$	$s_{55}^E = 6.51(6)$
$s_{33} = 1.3(2)$	$s_{23} = -0.6(2)$	$s_{66}^E = 5.65(6)$

Table 43A-12-008. K₂Cd₂(SO₄)₃. Temperature dependence of the rotatory power [·10³ m^{−1}] for different directions [96Kam].

T [K]	[100]	[010]	[001]	[0.82 0.57 0]
RT	−1.0(4)	−1.6(2)	3.4(2)	
298	−3.4(4)	−2.0(2)	2.3(2)	−2.5(3)
351			2.6(2)	−2.3(2)
390				−1.5(2)
417	−1.5(4)	−1.0(2)	1.4(2)	−1.3(2)
443	0.27(5)	0.36(5)	0.28(5)	0.30(5)

Table 43A-12-009. K₂Cd₂(SO₄)₃. Dispersion of rotatory power [$\cdot 10^3 \text{ m}^{-1}$] at 298 K [96Kam].

λ [nm]	[100]	[010]	[001]
400	−2.5(5)	−5.5(5)	10(1)
500	−1.7(5)	−3.4(4)	8.1(8)
600	−1.3(5)	−2.0(3)	4.4(4)

Table 43A-12-010. K₂Cd₂(SO₄)₃. Lines of Raman spectra [cm^{-1}] [86Dev]. ν_1 , ν_2 , ν_3 , ν_4 refer to the internal vibrations of the SO₄ ions. vvs: very very strong, vs: very strong, s: strong, m: medium, w: weak, vw: very weak, vvw: very very weak, sh: shoulder.

Description of vibrations	Phase II (orth)	Phase I (cub)	Description of vibrations	Phase II (orth)	Phase I (cub)
ν_3	1206 vvw 1195 vvw 1132 w 1119 w 1104 vw 1093 m 1087 m	1206 vw 1200 vw 1110 w 1093 m 1087 w	External	247 w 228 vw? 218 m 181 vvw 137 vw 109 vw 93 vw	231 sh 209 m 196 sh 172 w? 141 w 132 w 104 vw
ν_1	1022 vvs 1009 s	1020 vs		86 w 79 w	86 wsh 77 w
ν_4	643 s 629 w 620 w 614 m 601 w	635 w 623 m 617 w 606 w		63 w 56 w 46 vw 39 w 26 vw	68 w?
ν_2	477 wsh 471 m 456 m 441 w 437 w	 467 m 453 w 447 w 441 w			

Table 43A-12-011. K₂Cd₂(SO₄)₃. Frequencies of Raman lines [cm⁻¹] [83Moi].

$y (xx) z$	$x (yy) z$	$y (zz) x$	$x (yx) z$	$x (zx) y$	$y (xy) x$
26	26	26	—	—	—
40	40	40	—	—	—
—	—	51	—	55	—
—	63	—	—	—	—
76	78	81	—	81	81
—	125	127	—	—	—
142	139	—	—	—	—
173	—	—	—	—	—
207	207	207	—	207	—
236	236	236	—	236	236
—	428	424	—	434	—
—	—	441	236	—	445
462	463	—	466	—	—
—	593	593	599	—	594
—	607	607	—	—	—
—	615	614	615	615	615
633	—	—	636	—	636
1006	1006	1006	1006	1006	1006
1019	1019	1019	1019	1019	1019
1087	1086	1084	—	—	1086
1092	—	—	—	—	—
1118	1118	1112	—	—	1115
1129	—	1129	—	—	—
—	1142	—	1135	—	1151
—	—	—	—	1188	—