

**Table 32A-1-001.** KIO<sub>3</sub>. Fractional coordinates and equivalent isotropic temperature parameters [ $\text{\AA}^2$ ] in phase I at  $T = 523\text{K}$  [87Byr]. Neutron powder diffraction (Rietvelt method). Space group:  $R3-C_3^4$ . When the structure was analyzed with space group  $R3m-C_{3v}^5$ , fractional coordinates of O atom converged to  $x = y = 0.0408(5)$ .  $B_{\text{eq}} = 4\sum b_{ij}\mathbf{a}_i \cdot \mathbf{a}_j/3$ , where  $b_{ij}$  is defined by Eq. (b) in Introduction.

	$x$	$y$	$z$	$B_{\text{eq}} [\text{\AA}^2]$
K	0.51117(90)	$x$	$x$	3.85(6)
I	0.00000	$x$	$x$	1.79(4)
O	0.02804(150)	0.05498(153)	-0.39094(47)	5.93(24)

**Table 32A-1-002.** KIO<sub>3</sub>. Fractional coordinates and anisotropic temperature parameters [ $\text{\AA}^2$ ] in phase III at RT [78Kal].  $B_{ij}$  is defined by Eq. (a) in Introduction.

Atom	$x$	$y$	$z$	$B_{11}$	$B_{22}$	$B_{33}$	$B_{12}$	$B_{23}$	$B_{13}$
I <sub>1</sub>	0	0	0	0.0063(2)	0.0054(2)	0.0057(2)	0.0052(3)	0.0046(3)	0.0056(3)
I <sub>2</sub>	-0.0132(1)	0.5067(1)	0.5076(1)	0.0046(2)	0.0049(2)	0.0055(2)	0.0033(3)	0.0038(3)	0.0042(3)
I <sub>3</sub>	0.4936(1)	-0.0022(1)	0.5092(1)	0.0062(2)	0.0048(2)	0.0054(2)	0.0038(3)	0.0034(3)	0.0049(3)
I <sub>4</sub>	0.4928(1)	0.4913(1)	0.0163(1)	0.0057(2)	0.0050(2)	0.0058(2)	0.0044(3)	0.0048(3)	0.0057(3)
K <sub>1</sub>	0.0072(5)	0.0208(5)	0.5116(5)	0.014(1)	0.009(1)	0.016(1)	0.007(1)	0.010(1)	0.016(2)
K <sub>2</sub>	0.0047(4)	0.5122(4)	0.0179(4)	0.010(1)	0.011(1)	0.011(1)	0.007(1)	0.005(1)	0.005(2)
K <sub>3</sub>	0.5145(4)	0.0023(5)	0.0211(5)	0.011(1)	0.015(1)	0.011(1)	0.013(1)	0.012(2)	0.008(1)
K <sub>4</sub>	0.4965(4)	0.5084(4)	0.5292(5)	0.012(1)	0.010(1)	0.010(1)	0.010(1)	0.005(1)	0.009(1)
O <sub>11</sub>	0.079(1)	-0.152(2)	-0.154(2)	0.008(2)	0.016(2)	0.023(3)	0.002(3)	-0.029(4)	0.005(4)
O <sub>12</sub>	0.793(1)	-0.199(1)	-0.007(2)	0.017(2)	0.015(2)	0.021(3)	0.012(4)	0.016(4)	0.016(4)
O <sub>13</sub>	0.845(1)	0.046(2)	0.192(1)	0.013(2)	0.029(3)	0.015(2)	0.027(4)	0.026(4)	0.010(3)
O <sub>21</sub>	-0.007(1)	0.303(1)	0.317(1)	0.018(2)	0.014(2)	0.021(2)	0.011(4)	-0.013(4)	0.021(4)
O <sub>22</sub>	0.791(1)	0.343(1)	0.539(2)	0.015(2)	0.015(2)	0.032(3)	0.012(3)	0.033(4)	0.033(4)
O <sub>23</sub>	0.841(2)	0.586(1)	0.341(2)	0.026(3)	0.013(2)	0.023(3)	0.016(4)	0.016(4)	-0.004(5)
O <sub>31</sub>	0.533(2)	-0.156(1)	0.309(1)	0.035(3)	0.016(2)	0.019(2)	0.033(5)	0.011(4)	0.034(5)
O <sub>32</sub>	0.342(1)	-0.203(1)	0.548(2)	0.016(2)	0.012(2)	0.030(3)	0.007(3)	0.028(4)	0.028(4)
O <sub>33</sub>	0.287(1)	0.024(2)	0.346(1)	0.013(2)	0.029(3)	0.020(2)	0.021(4)	0.034(4)	0.011(4)
O <sub>41</sub>	0.345(1)	0.339(1)	0.095(1)	0.015(1)	0.016(2)	0.010(2)	-0.020(3)	0.009(3)	0.004(3)
O <sub>42</sub>	0.286(1)	0.485(1)	-0.188(1)	0.015(2)	0.016(2)	0.021(2)	0.009(4)	0.012(4)	-0.006(4)
O <sub>43</sub>	0.528(2)	0.297(1)	-0.149(1)	0.036(3)	0.012(2)	0.017(2)	0.030(4)	0.007(3)	0.034(4)

**Table 32A-1-003.** KIO<sub>3</sub>. Fractional coordinates and isotropic temperature parameters [ $\text{\AA}^2$ ] in phase IV ( $T = 100$  K) and phase V ( $T = 10$  K) [85Luc]. Neutron powder diffraction (Rietvelt method). For each atom, the values for phase IV and phase V are listed in the first and the second lines, respectively. Isotropic temperature parameters were constrained to equal values for each element-type atom.  $B$  is defined by Eq. (e) in Introduction.

	$x$	$y$	$z$	$B$ [ $\text{\AA}^2$ ]
K(1)	−0.0063(51) −0.0520(49)	0.0266(45) 0.0389(39)	0.4832(59) 0.5042(47)	0.52(13) 0.34(10)
K(2)	−0.0135(68) −0.0001(58)	0.4926(61) 0.4989(54)	0.0022(70) −0.0119(55)	
K(3)	0.5117(48) 0.5012(44)	−0.0027(34) 0.0172(42)	0.0277(53) 0.0166(47)	
K(4)	0.5014(56) 0.5023(54)	0.5131(62) 0.5281(54)	0.5435(66) 0.5212(62)	
I(1)	0.0000 0.0000	0.0000 0.0000	0.0000 0.0000	0.33(6) 0.07(8)
I(2)	−0.0128(44) −0.0112(41)	0.5109(35) 0.5229(35)	0.5060(46) 0.4890(41)	
I(3)	0.4898(49) 0.4775(47)	0.0002(50) 0.0200(55)	0.5208(50) 0.5092(46)	
I(4)	0.5032(45) 0.4891(37)	0.4879(36) 0.5009(38)	0.0096(42) 0.0035(38)	
O(11)	0.0697(52) 0.0650(44)	−0.1727(41) −0.1699(41)	−0.1360(46) −0.1229(41)	0.94(3) 0.45(3)
O(12)	0.7789(44) 0.7668(41)	−0.1952(39) −0.1819(46)	0.0040(45) −0.0169(40)	
O(13)	0.8567(46) 0.8380(37)	0.0379(47) 0.0304(42)	−0.2163(50) −0.2217(36)	
O(21)	0.0308(40) 0.0216(35)	0.3478(36) 0.3622(37)	0.3028(45) 0.2938(36)	
O(22)	0.7896(46) 0.7687(41)	0.3167(38) 0.3270(37)	0.5014(42) 0.4976(39)	
O(23)	0.8542(41) 0.8298(39)	0.6065(35) 0.6192(39)	0.3518(43) 0.3591(40)	
O(31)	0.4987(45) 0.4910(34)	−0.1746(39) −0.1666(35)	0.2867(45) 0.2914(35)	
O(32)	0.3648(48) 0.3249(48)	−0.1920(45) −0.1852(43)	0.5799(51) 0.5628(42)	
O(33)	0.2909(46) 0.2745(44)	0.0333(38) 0.0601(45)	0.3685(41) 0.3687(47)	
O(41)	0.3535(55) 0.3549(42)	0.3207(48) 0.3229(44)	0.0935(50) 0.0820(39)	
O(42)	0.2839(38) 0.2747(39)	0.4591(31) 0.4706(35)	−0.2000(40) −0.2091(39)	
O(43)	0.5678(44) 0.5778(43)	0.3078(40) 0.3420(42)	−0.1335(45) −0.1366(42)	

**Table 32A-1-004.** KIO<sub>3</sub>. Interatomic distances [Å] and angles [°] in phase I at  $T = 523$  K [87Byr]. The structure was analyzed with space group R3–C<sub>3</sub><sup>4</sup>.

IO <sub>3</sub> group		K polyhedra	
I–O	1.775(5)	K...O	3.032(7)
O–I–O	100.8(3)		3.111(8)
			3.280(7)
			3.434(8)

**Table 32A-1-005.** KIO<sub>3</sub>. Interatomic distances [Å] in phase III at RT [78Kal].

IO <sub>3</sub> -group							
I <sub>1</sub> –O <sub>11</sub>	1.785(1)	I <sub>2</sub> –O <sub>21</sub>	1.793(1)	I <sub>3</sub> –O <sub>31</sub>	1.795(1)	I <sub>4</sub> –O <sub>41</sub>	1.757(1)
O <sub>12</sub>	1.780(1)	O <sub>22</sub>	1.768(1)	O <sub>32</sub>	1.782(1)	O <sub>42</sub>	1.810(1)
O <sub>13</sub>	1.789(1)	O <sub>23</sub>	1.816(1)	O <sub>33</sub>	1.792(1)	O <sub>43</sub>	1.782(1)
...O <sub>41</sub>	2.702(3)	...O <sub>11</sub>	2.697(3)	...O <sub>13</sub>	2.727(3)	...O <sub>23</sub>	2.659(3)
...O <sub>33</sub>	2.748(3)	...O <sub>32</sub>	2.748(3)	...O <sub>43</sub>	2.737(3)	...O <sub>31</sub>	2.747(3)
...O <sub>21</sub>	2.802(3)	...O <sub>42</sub>	2.787(3)	...O <sub>23</sub>	2.753(3)	...O <sub>12</sub>	2.826(3)
O <sub>11</sub> –O <sub>12</sub>	2.767(3)	O <sub>21</sub> –O <sub>22</sub>	2.682(3)	O <sub>31</sub> –O <sub>32</sub>	2.747(3)	O <sub>41</sub> –O <sub>42</sub>	2.749(3)
O <sub>13</sub>	2.722(3)	O <sub>23</sub>	2.783(3)	O <sub>33</sub>	2.725(3)	O <sub>43</sub>	2.698(3)
O <sub>12</sub> –O <sub>13</sub>	2.717(3)	O <sub>22</sub> –O <sub>23</sub>	2.786(3)	O <sub>32</sub> –O <sub>33</sub>	2.732(3)	O <sub>42</sub> –O <sub>43</sub>	2.714(3)
K-polyhedra							
K <sub>1</sub> –O <sub>23</sub>	2.788(3)	K <sub>2</sub> –O <sub>13</sub>	3.002(3)	K <sub>3</sub> –O <sub>11</sub>	2.799(3)	K <sub>4</sub> –O <sub>41</sub>	2.791(3)
O <sub>33</sub>	2.834(4)	O <sub>43</sub>	3.062(3)	O <sub>31</sub>	2.840(3)	O <sub>32</sub>	2.834(3)
O <sub>13</sub>	2.930(3)	O <sub>22</sub>	3.070(4)	O <sub>43</sub>	2.962(3)	O <sub>22</sub>	2.939(3)
O <sub>42</sub>	3.002(3)	O <sub>42</sub>	3.102(3)	O <sub>32</sub>	3.031(3)	O <sub>12</sub>	3.003(3)
O <sub>21</sub>	3.018(3)	O <sub>23</sub>	3.120(4)	O <sub>12</sub>	3.059(3)	O <sub>33</sub>	3.099(4)
O <sub>31</sub>	3.037(4)	O <sub>12</sub>	3.179(4)	O <sub>21</sub>	3.089(3)	O <sub>42</sub>	3.127(3)
O <sub>11</sub>	3.242(4)	O <sub>21</sub>	3.220(4)	O <sub>41</sub>	3.240(4)	O <sub>21</sub>	3.209(4)
O <sub>12</sub>	3.297(4)	O <sub>11</sub>	3.247(4)	O <sub>42</sub>	3.294(4)	O <sub>43</sub>	3.364(4)
O <sub>43</sub>	3.359(4)	O <sub>41</sub>	3.294(4)	O <sub>22</sub>	3.348(4)	O <sub>23</sub>	3.397(4)
O <sub>22</sub>	3.406(4)	O <sub>33</sub>	3.294(4)	O <sub>13</sub>	3.444(4)	O <sub>13</sub>	3.463(4)
		O <sub>31</sub>	3.392(4)	O <sub>33</sub>	3.491(4)		
		O <sub>32</sub>	3.409(4)				

**Table 32A-1-006.**  $\text{KIO}_3$ . Interatomic distances [ $\text{\AA}$ ] and angles [ $^\circ$ ] in phase IV at  $T = 100\text{ K}$  (first column) and phase V at  $T = 10\text{ K}$  (second column)  $^{85}\text{Luc}$ .

IO <sub>3</sub> groups											
I(1)–O(11)	1.75(4)	1.66(4)	I(2)–O(21)	1.86(5)	1.76(4)	I(3)–O(31)	1.90(5)	1.87(5)	I(4)–O(41)	1.86(6)	1.82(5)
O(12)	1.88(3)	1.81(3)	O(22)	1.73(5)	1.90(4)	O(32)	1.76(6)	1.89(5)	O(42)	1.78(4)	1.77(4)
O(13)	1.83(4)	1.88(3)	O(23)	1.80(5)	1.81(5)	O(33)	1.74(5)	1.76(5)	O(43)	1.79(5)	1.81(5)
...O(41)	2.67(4)	2.71(3)	...O(11)	2.71(5)	2.85(4)	...O(13)	2.72(5)	2.80(5)	...O(23)	2.66(4)	2.67(4)
...O(33)	2.82(3)	2.68(4)	...O(32)	2.77(5)	2.53(5)	...O(43)	2.63(5)	2.65(5)	...O(31)	2.78(5)	2.77(4)
...O(21)	2.78(3)	2.86(3)	...O(42)	2.86(4)	2.85(4)	...O(22)	2.83(5)	2.72(5)	...O(12)	2.68(5)	2.75(5)
O(11)–O(12)	2.77(6)	2.66(5)	O(21)–O(22)	2.78(5)	2.88(5)	O(31)–O(32)	2.80(6)	2.81(5)	O(41)–O(42)	2.76(5)	2.80(5)
O(13)	2.71(6)	2.75(5)	O(23)	2.75(5)	2.83(5)	O(33)	2.68(5)	2.83(5)	O(43)	2.78(6)	2.78(5)
O(12)–O(13)	2.89(6)	2.66(5)	O(22)–O(23)	2.80(5)	2.75(5)	O(32)–O(33)	2.79(5)	2.79(6)	O(42)–O(43)	2.77(5)	2.77(5)
K polyhedra											
K(1)–O(33)	2.71(6)	3.00(6)	K(2)–O(43)	2.70(6)	2.74(5)	K(3)–O(31)	2.73(6)	2.90(5)	K(4)–O(41)	2.89(6)	2.82(5)
O(23)	2.72(4)	2.73(4)	O(13)	2.95(6)	2.99(5)	O(11)	2.83(5)	2.80(5)	O(32)	2.76(7)	2.92(6)
O(13)	2.84(7)	2.53(5)	O(21)	2.84(7)	2.85(6)	O(12)	2.92(5)	2.95(5)	O(12)	2.95(6)	2.95(5)
O(42)	2.85(5)	2.86(5)	O(22)	3.21(6)	3.13(5)	O(43)	3.00(5)	3.06(5)	O(21)	3.02(5)	3.07(5)
O(21)	3.18(6)	3.37(5)	O(23)	3.15(7)	3.51(6)	O(32)	2.88(5)	2.88(5)	O(22)	3.11(6)	2.95(6)
O(31)	3.15(5)	2.88(5)	O(11)	3.05(7)	2.88(6)	O(22)	3.06(5)	3.08(5)	O(42)	3.03(7)	3.18(6)
O(12)	3.03(5)	3.33(5)	O(12)	3.28(7)	3.47(6)	O(42)	3.41(4)	3.46(4)	O(33)	3.06(5)	3.01(5)
O(22)	3.11(6)	2.96(5)	O(42)	3.20(7)	3.01(6)	O(41)	3.07(6)	2.88(5)	O(23)	3.49(6)	3.15(6)
O(11)	3.69(6)	3.71(5)	O(31)	3.29(6)	3.17(5)	O(21)	3.35(5)	3.34(5)	O(31)	3.56(7)	3.36(6)
O(32)	3.74(6)	3.79(6)	O(41)	3.45(7)	3.39(6)	O(33)	3.56(6)	3.67(6)	O(43)	3.33(7)	
O(41)		3.75(5)	O(33)	3.49(6)	3.63(6)	O(13)	3.73(6)	3.64(5)	O(13)	3.41(6)	3.23(5)
			O(32)	3.70(6)	3.64(6)				O(11)	3.60(6)	3.59(6)

**Table 32A-1-007.** KIO<sub>3</sub>. Static piezoelectric constants [ $\cdot 10^{-12}$  C N<sup>-1</sup>] and piezocoupling coefficients at  $T = 293$  K [95Hau]. The specimen was twinned weakly. Piezoelectric behavior was compatible with point group m even though the point group of phase III is 1. See subsection 2b for orthogonal coordinate system.

$d_{15}$	$d_{16}$	$d_{21}$	$d_{22}$	$d_{23}$	$d_{24}$	$d_{31}$	$d_{32}$	$d_{33}$
1(1)	105(7)	105(6)	-108(7)	5(2)	12(3)	-12(3)	0(2)	35(3)

$k_t = 0.42$  (longitudinal wave propagating along  $Z_{\text{axis}}$ )

$k'_{66} = 0.60$  (quasi-transversal wave propagating along  $X_{\text{axis}}$  with displacement near  $Y_{\text{axis}}$ )

**Table 32A-1-008.** KIO<sub>3</sub>. Elastic stiffnesses [ $\cdot 10^{10}$  N m<sup>-2</sup>] and thermoelastic constants  $T_{\lambda\mu}$  [ $\cdot 10^{-3}$  K<sup>-1</sup>] [95Hau].  $T = 293$  K.  $T_{\lambda\mu} = d \log c_{\lambda\mu} / dT$ . The specimen was twinned weakly. The elastic behavior was well compatible with point group 3m. See subsection 2b for orthogonal coordinate system.

$c_{11}^E$	3.983(6)	$T_{11}^E$	-0.39(2)
$c_{13}^E$	2.00(15)	$ T_{13}^E $	< 0.6
$c_{14}^E$	0.28(2)	$T_{14}^E$	1.8(10)
$c_{33}^D$	3.988	$T_{33}^D$	-0.57(3)
$c_{44}^E$	1.639(7)	$T_{44}^E$	-0.79(5)
$c_{66}^E$	0.578(3)	$T_{66}^E$	-4.85(10)

**Table 32A-1-009.** KIO<sub>3</sub>. Refractive indices [92Yin1].  $T = \text{RT}$ . Principal axes of refractive indices are parallel to orthogonal coordinate system described in subsection 2b.

$\lambda$ [Å]	4358	4861	5461	5893	6563	7065
$n_X$	1.8831	1.8599	1.8419	1.8329	1.8227	1.8172
$n_Y$	1.8867	1.8630	1.8448	1.8358	1.8255	1.8199
$n_Z$	1.7283	1.7138	1.7024	1.6967	1.6899	1.6863

**Table 32A-1-010.** KIO<sub>3</sub>. Nuclear quadrupole resonance frequencies  $\nu$ , the quadrupole coupling constant  $e^2Qq_{zz}/h$ , and the asymmetric parameters  $\eta$  of  $^{127}\text{I}$  [75Bai1].  $n$ : number of lines in the NQR spectrum. Values for  $T = 298\text{ K}$  are from [56Lud].

$T$ [K]	$\nu (\pm \frac{1}{2} \leftrightarrow \pm \frac{3}{2})$ [MHz]	$\nu (\pm \frac{3}{2} \leftrightarrow \pm \frac{5}{2})$ [MHz]	$n$	$e^2Qq_{zz}/h$ [MHz]	$\eta$ [%]
4.2	150.567 149.837 149.786		3		
20	150.548 149.827 149.772		3		
77	150.046 149.943 149.256 149.188	299.925 299.475 298.350 298.200	4	999.83 998.45 994.58 994.10	2.1 3.3 2.1 1.4
116	149.422 149.106 148.513		3		
161	148.654 148.190 147.666	296.930 296.019 295.050	3	989.98 986.92 983.65	3.1 3.1 2.7
252	146.736 146.180 145.646		3		
298	145.38 144.91 144.37	290.71 289.85 288.82	3	969.0 966.2 962.7	
329	144.423 144.016 143.617		3		
344	142.964		1		
366	142.323		1		