

Table 46A-9-001. (CH₃)₂NH₂Ga(SO₄)₂ · 6H₂O. Fractional coordinates and mean square displacements in phase I [93Pie]. $T = 295$ K. $\overline{u^2}$ is estimated by one third of the trace of the orthogonalized U_{ij} tensor. U_{ij} is defined by Eq. (d) in Introduction. OCP: occupancy parameter.

| Atom | x | y | z | OCP | $\overline{u^2}$ [Å ²] |
|-------|------------|-------------|------------|------|------------------------------------|
| S | 0.0694(1) | 0.1829(1) | 0.2397(1) | | 0.028 |
| Ga | 0.5 | 0 | 0 | | 0.022 |
| O(11) | 0.2880(4) | 0.0780(2) | −0.1268(2) | | 0.036 |
| O(12) | 0.7396(3) | 0.0663(2) | −0.0639(2) | | 0.028 |
| O(13) | 0.5070(3) | 0.1548(2) | 0.0895(2) | | 0.036 |
| O(21) | 0.0595(3) | 0.0475(2) | 0.2197(2) | | 0.037 |
| O(22) | −0.1476(3) | 0.2326(2) | 0.2352(2) | | 0.042 |
| O(23) | 0.1991(4) | 0.2073(2) | 0.3578(2) | | 0.049 |
| O(24) | 0.1629(4) | 0.2433(2) | 0.1471(3) | | 0.052 |
| C(l) | 0.3599(8) | −0.0243(5) | 0.5586(5) | | 0.077 |
| N(1) | 0.4376(14) | −0.0137(7) | 0.4556(7) | 0.38 | 0.051 |
| N(2) | 0.5329(26) | −0.0628(19) | 0.4931(18) | 0.12 | 0.039 |
| H(11) | 0.313(3) | 0.146(3) | −0.171(2) | | 0.064 |
| H(12) | 0.184(3) | 0.047(3) | −0.160(2) | | 0.036 |
| H(21) | 0.780(3) | 0.023(3) | −0.126(2) | | 0.054 |
| H(22) | 0.732(3) | 0.129(3) | −0.088(2) | | 0.051 |
| H(31) | 0.396(3) | 0.186(3) | 0.103(2) | | 0.051 |
| H(32) | 0.621(3) | 0.179(3) | 0.136(2) | | 0.051 |

Table 46A-9-002. (CH₃)₂NH₂Ga(SO₄)₂ · 6H₂O. Hydrogen bond distances and angles in phase I [93Pie]. $T = 295$ K. Hydrogen bonds are those between oxygens O^{Al} in water molecules coordinated by Al³⁺ ions and oxygen atoms O^S in sulphate ions.

| Bond | Distance [Å] | | | Angle [°] |
|-------------------------------------|---------------------------------|--------------------|--------------------|-------------------------------------|
| O ^{Ga} –H...O ^S | O ^{Ga} –O ^S | O ^{Al} –H | H...O ^S | O ^{Ga} –H...O ^S |
| O(11)–H(11)...O(22) | 2.645(3) | 0.92(3) | 1.73(3) | 175(2) |
| O(11)–H(12)...O(21) | 2.636(3) | 0.77(2) | 1.87(2) | 170(3) |
| O(12)–H(21)...O(21) | 2.668(3) | 0.92(3) | 1.78(3) | 162(2) |
| O(12)–H(22)...O(23) | 2.582(3) | 0.72(3) | 1.86(3) | 176(3) |
| O(13)–H(31)...O(24) | 2.584(4) | 0.82(2) | 1.77(2) | 174(3) |
| O(13)–H(32)...O(22) | 2.629(3) | 0.85(2) | 1.77(2) | 178(3) |

Table 46A-9-003. (CH₃)₂NH₂Ga(SO₄)₂ · 6H₂O. Fractional coordinates and mean square displacements in phase II [95Pie]. $T = 125$ K. $\overline{u^2}$ is estimated by one third of the trace of the orthogonalized U_{ij} tensor. U_{ij} is defined by Eq. (d) in Introduction.

| Atom | x [10 ⁻⁴] | y | z | $\overline{u^2}$ [10 ⁻³ Å ²] |
|-------|----------------------------|----------|-----------|--|
| Ga | 5000 | 2517(1) | 0 | 11(1) |
| S(1) | 5659(2) | 5702(1) | 7401(1) | 15(1) |
| S(2) | 4358(2) | 9338(1) | 2629(1) | 14(1) |
| O(1) | 2938(6) | 3310(3) | -1229(3) | 21(1) |
| O(2) | 7136(6) | 1731(3) | 1259(4) | 18(1) |
| O(3) | 7368(5) | 3190(3) | -649(3) | 17(1) |
| O(4) | 2621(6) | 1818(3) | 631(3) | 15(1) |
| O(5) | 5047(6) | 4089(3) | 920(3) | 17(1) |
| O(6) | 4959(5) | 991(4) | -894(4) | 22(1) |
| O(11) | 5555(5) | 7081(3) | 7135(3) | 19(1) |
| O(12) | 6878(6) | 5488(3) | 8575(4) | 22(1) |
| O(13) | 3443(5) | 5201(3) | 7366(4) | 22(1) |
| O(14) | 6541(6) | 5097(3) | 6444(4) | 24(1) |
| O(21) | 4452(6) | 7983(4) | 2829(3) | 17(1) |
| O(22) | 3026(7) | 9575(4) | 1422(4) | 27(1) |
| O(23) | 6471(6) | 9822(4) | 2653(4) | 23(1) |
| O(24) | 3314(7) | 9998(4) | 3511(4) | 23(1) |
| C(1) | 3744(10) | 2246(7) | 5693(6) | 39(2) |
| C(2) | 6462(11) | 2750(5) | 4455(7) | 33(2) |
| N | 4236(7) | 2382(4) | 4526(4) | 25(1) |
| H(1) | 2749(57) | 2035(53) | 6183(43) | |
| H(2) | 4527(101) | 2945(51) | 5943(91) | |
| H(3) | 4805(72) | 1616(54) | 5915(81) | |
| H(4) | 6703(107) | 3476(33) | 4806(49) | |
| H(5) | 7429(79) | 2160(42) | 4905(45) | |
| H(6) | 6299(84) | 2262(39) | 3820(31) | |
| H(7) | 3285(90) | 3165(49) | 3893(50) | |
| H(8) | 3317(88) | 1508(46) | 4492(59) | |
| H(11) | 1845(84) | 2808(71) | -1311(15) | |
| H(12) | 2963(69) | 4055(30) | -1574(37) | |
| H(21) | 8227(54) | 2120(39) | 1633(39) | |
| H(22) | 6463(73) | 1643(61) | 1850(37) | |
| H(31) | 7790(65) | 2737(37) | -1233(32) | |
| H(32) | 6993(129) | 3974(38) | -828(79) | |
| H(41) | 1523(56) | 2106(50) | 810(49) | |
| H(42) | 2592(79) | 1080(30) | 914(42) | |
| H(51) | 3813(69) | 4447(80) | 924(105) | |
| H(52) | 6073(81) | 4525(59) | 1357(62) | |
| H(61) | 5990(52) | 759(43) | -1223(38) | |
| H(62) | 3783(60) | 987(56) | -1398(44) | |

Table 46A-9-004. (CH₃)₂NH₂Ga(SO₄)₂ · 6H₂O. Bond distances and bond angles in phases II and III [95Pie].

| Phase II, $T = 125$ K | | Phase III, $T = 100$ K | |
|-----------------------|----------|------------------------|----------|
| Distance [Å] | | | |
| Ga–O(6) | 1.916(4) | Ga–O(5) | 1.925(4) |
| Ga–O(1) | 1.940(3) | Ga–O(3) | 1.941(4) |
| Ga–O(3) | 1.942(4) | Ga–O(1) | 1.944(4) |
| Ga–O(4) | 1.948(4) | Ga–O(6) | 1.961(3) |
| Ga–O(5) | 1.973(3) | Ga–O(2) | 1.967(3) |
| Ga–O(2) | 1.984(4) | Ga–O(4) | 1.967(4) |
| S(1)–O(12) | 1.442(4) | S(1)–O(12) | 1.447(4) |
| S(1)–O(14) | 1.465(4) | S(1)–O(14) | 1.468(4) |
| S(1)–O(11) | 1.499(3) | S(1)–O(11) | 1.473(3) |
| S(1)–O(13) | 1.512(3) | S(1)–O(13) | 1.479(3) |
| S(2)–O(23) | 1.445(5) | S(2)–O(24) | 1.461(5) |
| S(2)–O(21) | 1.462(4) | S(2)–O(23) | 1.499(4) |
| S(2)–O(24) | 1.481(5) | S(2)–O(23) | 1.499(4) |
| S(2)–O(22) | 1.505(4) | S(2)–O(21) | 1.519(4) |
| O(11)–O(14) | 2.381(5) | O(11)–O(12) | 2.380(5) |
| O(11)–O(12) | 2.407(5) | O(11)–O(13) | 2.393(4) |
| O(11)–O(13) | 2.458(5) | O(11)–O(14) | 2.406(5) |
| O(12)–O(13) | 2.400(5) | O(12)–O(14) | 2.406(5) |
| O(12)–O(14) | 2.436(6) | O(12)–O(13) | 2.419(4) |
| O(13)–O(14) | 2.411(5) | O(13)–O(14) | 2.366(5) |
| O(21)–O(23) | 2.377(6) | O(21)–O(24) | 2.415(5) |
| O(21)–O(22) | 2.401(6) | O(21)–O(22) | 2.417(5) |
| O(21)–O(24) | 2.439(6) | O(21)–O(23) | 2.469(5) |
| O(22)–O(24) | 2.397(6) | O(22)–O(23) | 2.402(5) |
| O(22)–O(23) | 2.410(6) | O(22)–O(24) | 2.438(6) |
| O(23)–O(24) | 2.407(6) | O(23)–O(24) | 2.455(5) |
| N–C(1) | 1.429(8) | N–C(1) | 1.460(6) |
| N–C(2) | 1.496(9) | N–C(2) | 1.512(6) |
| Angle [°] | | | |
| C(1)–N–C(2) | 116.7(5) | C(1)–N–C(2) | 114.8(3) |

Table 46A-9-005. (CH₃)₂NH₂Ga(SO₄)₂ · 6H₂O. Hydrogen bond distances and angles in phase II [95Pie]. *T* = 125 K.

| Bond | Distance [Å] | | | Angles [°] |
|--------------------|---------------------|-------------|---------------|-------------------------|
| | <i>D</i> – <i>A</i> | <i>D</i> –H | H... <i>A</i> | <i>D</i> –H... <i>A</i> |
| N–H(7)...O(12) | 2.834(6) | 1.20(5) | 1.70(5) | 156(5) |
| N–H(8)...O(24) | 2.811(6) | 1.00(4) | 1.81(4) | 170(4) |
| O(1)–H(11)...O(21) | 2.682(5) | 0.87(4) | 1.87(8) | 154(10) |
| O(1)–H(12)...O(13) | 2.630(5) | 0.89(3) | 1.78(4) | 159(4) |
| O(2)–H(21)...O(11) | 2.574(5) | 0.86(3) | 1.73(3) | 170(5) |
| O(2)–H(22)...O(23) | 2.662(6) | 0.87(4) | 2.15(4) | 118(5) |
| O(3)–H(31)...O(21) | 2.680(5) | 0.90(3) | 1.81(3) | 162(4) |
| O(3)–H(32)...O(12) | 2.604(5) | 0.88(4) | 1.75(5) | 163(9) |
| O(4)–H(41)...O(11) | 2.624(5) | 0.83(4) | 1.94(5) | 140(5) |
| O(4)–H(42)...O(22) | 2.551(6) | 0.85(4) | 1.71(3) | 169(5) |
| O(5)–H(51)...O(14) | 2.578(5) | 0.88(4) | 1.74(6) | 159(10) |
| O(5)–H(52)...O(13) | 2.598(5) | 0.88(4) | 1.76(4) | 158(4) |
| O(6)–H(61)...O(24) | 2.593(6) | 0.85(3) | 1.77(5) | 162(4) |
| O(6)–H(62)...O(23) | 2.678(5) | 0.86(3) | 1.88(4) | 153(6) |

Table 46A-9-006. (CH₃)₂NH₂Ga(SO₄)₂ · 6H₂O. Fractional coordinates and mean square displacements in phase III [95Pie]. $T = 100$ K. $\overline{u^2}$ is estimated by one third of the trace of the orthogonalized U_{ij} tensor. U_{ij} is defined by Eq (d) in Introduction. Unit cell is chosen with 2₁ axis at 1/4, y , 1/4.

| Atom | x [10 ⁻⁴] | y | z | $\overline{u^2}$ [10 ⁻³ Å ²] |
|-------|----------------------------|-----------|-----------|--|
| Ga | 4997(2) | 0 | 0(1) | 9(1) |
| S(1) | 5866(2) | 3218(1) | 7505(1) | 11(1) |
| S(2) | 4116(2) | 6794(1) | 2487(1) | 11(1) |
| O(1) | 2928(6) | 703(3) | -1425(4) | 24(1) |
| O(2) | 7108(5) | -715(3) | 1433(3) | 11(1) |
| O(3) | 7509(5) | 602(3) | -573(3) | 12(1) |
| O(4) | 2464(5) | -629(3) | 582(3) | 13(1) |
| O(5) | 5045(5) | 1560(3) | 820(3) | 19(1) |
| O(6) | 4959(5) | -1580(3) | -851(3) | 10(1) |
| O(11) | 5810(5) | 4564(3) | 7368(3) | 11(1) |
| O(12) | 7377(6) | 2896(3) | 8667(4) | 24(1) |
| O(13) | 3598(5) | 2786(3) | 7460(3) | 12(1) |
| O(14) | 6530(6) | 2653(3) | 6452(3) | 21(1) |
| O(21) | 4038(6) | 5408(3) | 2617(3) | 17(1) |
| O(22) | 2578(6) | 7124(3) | 1294(4) | 17(1) |
| O(23) | 6386(6) | 7184(3) | 2412(4) | 19(1) |
| O(24) | 3463(6) | 7349(4) | 3542(4) | 22(1) |
| C(1) | 3717(6) | 9(5) | 5716(4) | 20(1) |
| C(2) | 6638(8) | 447(4) | 4574(4) | 21(1) |
| N | 4585(6) | -263(3) | 4637(3) | 17(1) |
| H(1) | 4544(117) | -473(57) | 6371(42) | |
| H(2) | 3895(154) | 839(31) | 5893(69) | |
| H(3) | 2249(50) | -289(49) | 5485(47) | |
| H(4) | 6297(71) | 1263(27) | 4645(41) | |
| H(5) | 7758(71) | 308(43) | 5354(32) | |
| H(6) | 6915(75) | -5(39) | 3959(32) | |
| H(7) | 3240(100) | 368(55) | 3784(53) | |
| H(8) | 4813(155) | -1437(48) | 4609(91) | |
| H(11) | 2082(72) | 674(67) | -912(43) | |
| H(12) | 3213(66) | 1362(29) | -1677(38) | |
| H(21) | 8272(53) | -287(34) | 1734(38) | |
| H(22) | 6298(97) | -277(74) | 1816(74) | |
| H(31) | 7894(65) | 241(35) | -1208(30) | |
| H(32) | 8279(92) | 1260(42) | -592(80) | |
| H(41) | 2960(353) | -206(88) | 1249(68) | |
| H(42) | 2430(77) | -1384(27) | 777(41) | |
| H(51) | 3755(55) | 1482(61) | 807(64) | |
| H(52) | 6172(69) | 2048(46) | 1074(61) | |
| H(61) | 6078(52) | -1872(40) | -1015(40) | |
| H(62) | 3779(55) | -1629(43) | -1333(37) | |

Table 46A-9-007. (CH₃)₂NH₂Ga(SO₄)₂ · 6H₂O. Hydrogen bond distances and angles in phase III [95Pie]. *T* = 100 K.

| <i>D</i> –H... <i>A</i> | Distance [Å] | | | Angle [°] |
|-------------------------|---------------------|-------------|---------------|-------------------------|
| | <i>D</i> – <i>A</i> | <i>D</i> –H | H... <i>A</i> | <i>D</i> –H... <i>A</i> |
| N–H(7)...O(21) | 3.032(5) | 1.30(5) | 1.84(5) | 150(5) |
| N–H(7)...O(22) | 2.996(5) | 1.30(5) | 1.98(6) | 131(4) |
| N–H(8)...O(24) | 2.886(5) | 1.29(5) | 1.84(8) | 134(6) |
| N–H(8)...O(12) | 3.074(5) | 1.29(5) | 2.38(9) | 110(5) |
| O(1)–H(11)...O(11) | 2.626(5) | 0.87(4) | 2.44(6) | 93(4) |
| O(1)–H(11)...O(4) | 2.734(5) | 0.87(4) | 2.15(6) | 144(5) |
| O(1)–H(12)...O(13) | 2.665(5) | 0.81(3) | 1.87(3) | 170(4) |
| O(2)–H(21)...O(21) | 2.677(5) | 0.86(3) | 1.82(5) | 172(4) |
| O(2)–H(22)...O(5) | 2.799(5) | 0.88(5) | 2.33(9) | 114(7) |
| O(2)–H(22)...O(4) | 2.823(5) | 0.88(5) | 2.48(7) | 104(5) |
| O(3)–H(31)...O(11) | 2.712(5) | 0.89(3) | 1.83(4) | 170(4) |
| O(3)–H(32)...O(12) | 2.632(5) | 0.87(4) | 1.99(5) | 130(5) |
| O(4)–H(41)...O(21) | 2.657(5) | 0.86(5) | 2.08(19) | 123(10) |
| O(4)–H(41)...O(5) | 2.853(5) | 0.86(5) | 2.43(15) | 111(10) |
| O(4)–H(42)...O(22) | 2.567(5) | 0.85(3) | 1.72(3) | 174(4) |
| O(5)–H(51)...O(24) | 2.601(6) | 0.81(3) | 1.95(6) | 137(6) |
| O(5)–H(52)...O(23) | 2.679(5) | 0.87(4) | 1.99(5) | 136(6) |
| O(6)–H(61)...O(14) | 2.576(5) | 0.83(3) | 1.75(3) | 174(5) |