

**H<sub>9</sub>Ce<sub>6</sub>Nd<sub>7</sub>(SO<sub>4</sub>)<sub>27</sub>·72.33H<sub>2</sub>O [1]**

Structural features: CeO<sub>6</sub>O<sub>3</sub> and Nd(OH<sub>2</sub>)<sub>6</sub>O<sub>3</sub> tricapped trigonal prisms are interconnected via SO<sub>4</sub> and S(O<sub>3</sub>[OH]) tetrahedra to form infinite slabs perpendicular to [001]; additional H<sub>2</sub>O in channels (partial disorder).

Barnes J.C. et al. (1993) [1]

Ce<sub>6</sub>H<sub>147.64</sub>Nd<sub>7</sub>O<sub>180.32</sub>S<sub>27</sub>

$a = 1.9165$ ,  $c = 2.506$  nm,  $c/a = 1.308$ ,  $V = 7.9713$  nm<sup>3</sup>,  $Z = 2$

site	Wyck.	sym.	<i>x</i>	<i>y</i>	<i>z</i>	occ.	atomic environment
Ce1	12i	1	0.01604	0.345	0.14449		tricapped trigonal prism O <sub>9</sub>
(OH <sub>2</sub> )2	12i	1	0.021	0.1443	0.1994	0.5	single atom (OH <sub>2</sub> )
O3	12i	1	0.0393	0.4484	0.0818		single atom S
(OH <sub>2</sub> )4	12i	1	0.0692	0.1512	0.0814		non-colinear (OH <sub>2</sub> ) <sub>2</sub>
O5	12i	1	0.0731	0.2961	0.2024		single atom S
O6	12i	1	0.0733	0.4505	0.2028		single atom S
(OH <sub>2</sub> )7	12i	1	0.0743	0.0485	0.0287	0.16	non-colinear (OH <sub>2</sub> ) <sub>2</sub>
O8	12i	1	0.0807	0.3141	0.0756		single atom S
(OH <sub>2</sub> )9	12i	1	0.1474	0.065	0.1961	0.5	single atom (OH <sub>2</sub> )
O10	12i	1	0.1629	0.4273	0.1272		single atom S
S11	12i	1	0.16703	0.37605	0.08402		tetrahedron O <sub>4</sub>
(OH <sub>2</sub> )12	12i	1	0.1825	0.6035	0.0891		single atom (OH <sub>2</sub> )
O13	12i	1	0.2019	0.422	0.0353		single atom S
O14	12i	1	0.2108	0.3367	0.1017		single atom S
(OH <sub>2</sub> )15	12i	1	0.2368	0.2266	0.1739		single atom Nd
(OH <sub>2</sub> )16	12i	1	0.2373	0.2295	0.0396		single atom Nd
O17	12i	1	0.2448	0.0473	0.1289		single atom S
O18	12i	1	0.2523	0.097	0.0393		single atom S
S19	12i	1	0.29656	0.10842	0.08864		tetrahedron O <sub>4</sub>
(OH <sub>2</sub> )20	12i	1	0.3011	0.5642	0.1793		single atom Nd
(OH <sub>2</sub> )21	12i	1	0.3282	0.4091	0.1795		single atom Nd
O22	12i	1	0.3285	0.1898	0.1086		single atom S
Nd23	12i	1	0.33318	0.32349	0.10674		tricapped trigonal prism O <sub>3</sub> (OH <sub>2</sub> ) <sub>6</sub>
(OH <sub>2</sub> )24	12i	1	0.3418	0.4168	0.033		single atom Nd
O25	12i	1	0.3628	0.0901	0.0808		single atom S
O26	12i	1	0.3711	0.0874	0.2035		single atom S
(OH <sub>2</sub> )27	12i	1	0.428	0.3191	0.1731		single atom Nd
(OH <sub>2</sub> )28	12i	1	0.4344	0.3235	0.044		single atom Nd
O29	12i	1	0.4755	0.0725	0.138		single atom S
S30	12i	1	0.4911	0.03074	0.09367		tetrahedron O <sub>4</sub>
(OH <sub>2</sub> )31	12i	1	0.5182	0.2401	0.1066	0.82	non-colinear O <sub>2</sub>
O32	12i	1	0.5234	0.0812	0.0463		single atom S
O33	12i	1	0.5438	0.0021	0.1115		single atom S
(OH <sub>2</sub> )34	12i	1	0.6124	0.2489	0.0001	0.24	non-coplanar triangle (OH <sub>2</sub> ) <sub>3</sub>
O35	6h	<i>m..</i>	0.0599	0.5501	<sup>1</sup> / <sub>4</sub>		single atom S
S36	6h	<i>m..</i>	0.0997	0.5027	<sup>1</sup> / <sub>4</sub>		tetrahedron O <sub>4</sub>
S37	6h	<i>m..</i>	0.1224	0.3037	<sup>1</sup> / <sub>4</sub>		tetrahedron O <sub>4</sub>
O38	6h	<i>m..</i>	0.1353	0.2353	<sup>1</sup> / <sub>4</sub>		single atom S
O39	6h	<i>m..</i>	0.1864	0.5532	<sup>1</sup> / <sub>4</sub>		single atom S
O40	6h	<i>m..</i>	0.1969	0.3791	<sup>1</sup> / <sub>4</sub>		single atom S
(OH)41	6h	<i>m..</i>	0.2825	0.1225	<sup>1</sup> / <sub>4</sub>		single atom S

S42	6h	m..	0.3702	0.131	$\frac{1}{4}$	tetrahedron O <sub>3</sub> (OH)
O43	6h	m..	0.4187	0.2104	$\frac{1}{4}$	single atom S
(OH <sub>2</sub> )44	6h	m..	0.5738	0.1053	$\frac{1}{4}$	non-coplanar triangle O(OH <sub>2</sub> ) <sub>2</sub>
(OH <sub>2</sub> )45	4e	3..	0	0	0.1568	non-coplanar triangle (OH <sub>2</sub> ) <sub>3</sub>
Nd46	2c	-6..	$\frac{1}{3}$	$\frac{2}{3}$	$\frac{1}{4}$	tricapped trigonal prism (OH <sub>2</sub> ) <sub>6</sub> O <sub>3</sub>

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Experimental: single crystal, diffractometer, X-rays, R = 0.038, T = 293 K

Remarks: H positions were refined but not published. Short interatomic distances for partly occupied site(s). Hydrogen atoms are not taken into consideration for Pearson symbol, Wyckoff sequence and atomic environments.

References: [1] Barnes J.C., Paton J.D., Seaward K.F. (1993), Acta Crystallogr. C 49, 2057-2060.