

$$\text{H}_{0.25}(\text{Zn}_{0.15}\text{Fe}_{0.85})_6[\text{SiO}_4]_{0.25}[\text{AsO}_3]_{3.75}[\text{OH}]_3$$
*hP*54

(186) $P6_3mc - d^2c^4b^3$
(Fe,Zn)₁₂(OH)₆(AsO₃)₆(AsO₃,HOSiO₃)₂ [1], ekatite

Structural features: Columns of edge- and face-linked (Fe,Zn)(O,OH)₆ octahedra are interconnected via common vertices, Si(O₃[OH]) tetrahedra and :AsO₃ ψ -tetrahedra (partial substitutional disorder) to form a 3D-framework with large channels parallel to [001].

Keller P. (2001) [1]

 $\text{As}_{3.77}\text{Fe}_{5.07}\text{H}_{3.25}\text{O}_{15.25}\text{Si}_{0.25}\text{Zn}_{0.90}$
 $a = 1.2773$, $c = 0.5051$ nm, $c/a = 0.395$, $V = 0.7137$ nm³, $Z = 2$

site	Wyck.	sym.	<i>x</i>	<i>y</i>	<i>z</i>	occ.	atomic environment
O1	12 <i>d</i>	1	0.0661	0.3441	0.1		non-coplanar triangle AsFe ₂
M2	12 <i>d</i>	1	0.42715	0.08451	0.2565	0.995	octahedron O ₆
As3	6 <i>c</i>	. <i>m</i> .	0.14184	0.85815	0.2876		non-coplanar triangle O ₃
O4	6 <i>c</i>	. <i>m</i> .	0.4755	0.5245	0.441		non-coplanar square Fe ₄
O5	6 <i>c</i>	. <i>m</i> .	0.5968	0.4032	0.386		
O6	6 <i>c</i>	. <i>m</i> .	0.8024	0.1976	0.049		non-coplanar triangle AsFe ₂
Si7	2 <i>b</i>	3 <i>m</i> .	$\frac{1}{3}$	$\frac{2}{3}$	0.0	0.246	
As8	2 <i>b</i>	3 <i>m</i> .	$\frac{1}{3}$	$\frac{2}{3}$	0.058	0.768	
O9	2 <i>b</i>	3 <i>m</i> .	$\frac{1}{3}$	$\frac{2}{3}$	0.329	0.246	
H10	6 <i>c</i>	. <i>m</i> .	0.498	0.502	0.263		
H11	2 <i>b</i>	3 <i>m</i> .	$\frac{1}{3}$	$\frac{2}{3}$	0.5511	0.246	

 $\text{M2} = 0.849\text{Fe} + 0.151\text{Zn}$

Transformation from published data: origin shift 0 0 0.33700

Experimental: single crystal, diffractometer, X-rays, $R = 0.043$, $T = 298$ K

Remarks: Natural specimen from Tsumeb, Namibia. Composition $(\text{Fe}^{3+})_{5.95}(\text{Fe}^{2+})_{5.14}\text{Zn}_{0.81}\text{As}_{7.49}\text{Si}_{0.61}\text{O}_{27.33} \cdot x\text{H}_2\text{O}$ from electron microprobe analysis. When relevant, we changed the last digit of the atom coordinates to respect the symmetry conditions for special positions. Short interatomic distances for partly occupied site(s). Hydrogen atoms are not taken into consideration for Pearson symbol, Wyckoff sequence and atomic environments.

References: [1] Keller P. (2001), Eur. J. Mineral. 13, 769-777.