

$[\text{NH}_4]_{0.05}\text{Ca}_{4.65}[\text{PO}_4]_{2.48}[\text{CO}_3]_{0.54}[\text{OH}]_{0.83}[\text{H}_2\text{O}]_{0.38}$	$hP68$	$(176) P6_3/m - i^2h^6fe$
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$(\text{NH}_4)_{0.1}\text{Ca}_{9.5}(\text{PO}_4)_{5.05}(\text{CO}_3)_{0.95}(\text{OH})_{2.05}$  [1], apatite family

Structural features: Infinite columns of base-linked  $\text{CaO}_6$  trigonal prisms (partial vacancies ignored) share vertices with  $\text{PO}_4$  tetrahedra and  $\text{CO}_3$  trigonal units (substitutional disorder) to form a 3D-framework; OH in infinite columns of face-linked  $(\text{Ca},\text{NH}_4)_6$  octahedra parallel to [001] (partial disorder).

Ivanova T.I. et al. (2001) [1]

$\text{C}_{0.54}\text{Ca}_{4.65}\text{H}_{1.41}\text{N}_{0.05}\text{O}_{12.95}\text{P}_{2.47}$

$a = 0.94372$ ,  $c = 0.68881$  nm,  $c/a = 0.73$ ,  $V = 0.5313$  nm<sup>3</sup>,  $Z = 2$

site	Wyck.	sym.	<i>x</i>	<i>y</i>	<i>z</i>	occ.	atomic environment
O1	12i	1	0.3409	0.083	0.0668	0.832	
O2	12i	1	0.362	0.063	0.089	0.187	
C3	6h	<i>m</i> ..	0.0096	0.3618	$\frac{1}{4}$	0.09	
O4	6h	<i>m</i> ..	0.1565	0.4829	$\frac{1}{4}$	0.932	
M5	6h	<i>m</i> ..	0.2457	0.2578	$\frac{1}{4}$		
P6	6h	<i>m</i> ..	0.3991	0.0299	$\frac{1}{4}$	0.824	
C7	6h	<i>m</i> ..	0.438	0.083	$\frac{1}{4}$	0.09	
O8	6h	<i>m</i> ..	0.5879	0.1184	$\frac{1}{4}$	0.945	
Ca9	4f	3..	$\frac{1}{3}$	$\frac{2}{3}$	0.0011	0.848	trigonal prism O <sub>6</sub>
(OH)10	4e	3..	0	0	0.1884	0.602	

$\text{M5} = 0.983\text{Ca} + 0.017\text{NH}_4$

Transformation from published data:  $y, x, -z$

Experimental: powder, diffractometer, X-rays,  $R_p = 0.049$ ,  $T = 298$  K

Remarks: We adjusted the charge balance in the published chemical formula by changing the OH index from 2 to 2.05.  $\text{Ca}_{3.4}[\text{Ca}_{5.9}(\text{NH}_4)_{0.1}][(\text{PO}_4)_{4.95}(\text{CO}_3)_{1.05}(\text{H}_2\text{O})_{0.30}][(\text{OH})_{1.65}(\text{H}_2\text{O})_{0.45}]$  (neutral for OH index 1.75). Refinement considering  $f(\text{Ca})$  for site M5 gave  $\text{occ} = 0.985(2)$ . Short interatomic distances for partly occupied site(s); impossibly short distances occur for published site occupancies. Hydrogen atoms are not taken into consideration for Pearson symbol, Wyckoff sequence and atomic environments.

References: [1] Ivanova T.I., Frank Kamenetskaya O.V., Kol'tsov A.B., Ugolkov V.L. (2001), J. Solid State Chem. 160, 340-349.