

AlBi ₄ Cl ₄	<i>hP</i> 124	(176) <i>P</i> 6 ₃ / <i>m</i> – i ⁶ h ⁵ f ³ e ² d
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Bi₈(AlCl₄)₂ [1]

Structural features: AlCl₄ tetrahedra (partial orientational disorder) and Bi₈ clusters (square antiprism).

Beck J. et al. (1996) [1]

AlBi₄Cl₄

a = 1.7854, *c* = 1.2953 nm, *c/a* = 0.725, *V* = 3.5758 nm³, *Z* = 12

site	Wyck.	sym.	<i>x</i>	<i>y</i>	<i>z</i>	occ.	atomic environment
Cl1	12 <i>i</i>	1	0.028	0.126	0.05	0.5	
Cl2	12 <i>i</i>	1	0.127	0.443	0.124		single atom Al
Bi3	12 <i>i</i>	1	0.2678	0.3212	0.0847		non-coplanar square Bi ₄
Bi4	12 <i>i</i>	1	0.3693	0.2245	0.1312		non-coplanar square Bi ₄
Cl5	12 <i>i</i>	1	0.559	0.218	0.014		non-colinear Al ₂
Bi6	12 <i>i</i>	1	0.576	0.04	0.1304		non-coplanar square Bi ₄
Cl7	6 <i>h</i>	<i>m</i> ..	0.044	0.25	¹ / ₄		single atom Al
Al8	6 <i>h</i>	<i>m</i> ..	0.056	0.36	¹ / ₄		tetrahedron Cl ₄
Bi9	6 <i>h</i>	<i>m</i> ..	0.1992	0.1776	¹ / ₄		non-coplanar square Bi ₄
Bi10	6 <i>h</i>	<i>m</i> ..	0.338	0.4649	¹ / ₄		non-coplanar square Bi ₄
Cl11	6 <i>h</i>	<i>m</i> ..	0.417	0.07	¹ / ₄		single atom Al
Al12	4 <i>f</i>	3..	¹ / ₃	² / ₃	0.034	0.5	trigonal bipyramid AlCl ₄
Cl13	4 <i>f</i>	3..	¹ / ₃	² / ₃	0.203	0.5	single atom Cl
Al14	4 <i>f</i>	3..	¹ / ₃	² / ₃	0.591	0.5	trigonal bipyramid AlCl ₄
Al15	4 <i>e</i>	3..	0	0	0.024	0.5	
Cl16	4 <i>e</i>	3..	0	0	0.164	0.5	single atom Al
Cl17	2 <i>d</i>	-6..	² / ₃	¹ / ₃	¹ / ₄		colinear Al ₂

Transformation from published data: origin shift 0 0 ¹/₂

Experimental: single crystal, diffractometer, X-rays, wR = 0.062, T = 293 K

Remarks: We deduced probable occupancies of the Al and Cl sites from the drawing in fig. 3. Space group (173) *P*6₃ was tested and rejected. Short interatomic distances for partly occupied site(s).

References: [1] Beck J., Brendel C.J., Bengtsson Kloo L., Krebs B., Mummert M., Stankowski A., Ulvenlund S. (1996), Chem. Ber. 129, 1219-1226.