

150 MW	CNZn		Zinc monocyanoide	C _{∞v} Zn–C≡N
	r_0	Å ^{a)}		
	Zn–C	1.9545(20)		
	C≡N	1.1464(20)		
	r_s	Å ^{a)}		
	Zn–C	1.9525(20)		
	C≡N	1.1434(20)		
	$r_m^{(1)}$	Å ^{a)}		
	Zn–C	1.9496(20)		
	C≡N	1.1417(20)		

The most stable form of zinc and the cyanide moiety is the linear cyanide structure.

^{a)} Uncertainties were not estimated in the original paper.

Brewster, M.A., Ziurys, L.M.: J. Chem. Phys. **117** (2002) 4853.