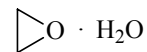
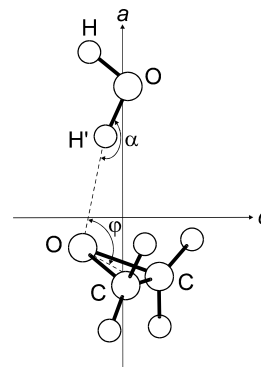


303
MW $C_2H_6O_2$ **Oxirane – water (1/1)**
(weakly bound complex) C_s
(effective symmetry class)
(large-amplitude motion)

r_0	\AA	θ_0	deg
H'...O ^{a)}	1.92(1)	φ ^{b)}	103(1)
R_{cm}	2.84(2) ^{c)}	α ^{b)}	163(2)

Atom ^{d)}	$ a_s $ [\AA]	$ b_s $ [\AA]	$ c_s $ [\AA]
O	2.152	0.0	0.162
H'	1.282	0.15	0.442
H	2.821	0.25	0.456

The water molecule lies in the plane of symmetry of the oxirane perpendicular to the ring; the water hydrogen involved in the hydrogen bond points toward the ring oxygen, while the “free” hydrogen is *entgegen* to the ring.



^{a)} H' denotes the water hydrogen that is involved in the hydrogen bond and O is the oxygen of the oxirane.

^{b)} See figure for the definition.

^{c)} Uncertainty was not estimated in the original paper.

^{d)} Atoms of the water.

Caminati, W., Moreschini, P., Rossi, I., Favero, P.G.: J. Am. Chem. Soc. **120** (1998) 11144.