

## L2BBL5

1. Black body Radiation. Graph of  $f(x) = dL/d\lambda$ , and  $x = \lambda$
2. Integration over the wavelength range from  $3 \cdot 10^{-6}$  to  $3 \cdot 10^{-5}$  meters to obtain the Radiance.
3. Multiplication with area times solid angle to obtain the Radiant energy.

### 1. Black body Radiation. Graph of $dL/d\lambda$

Constants for wavelength dependence

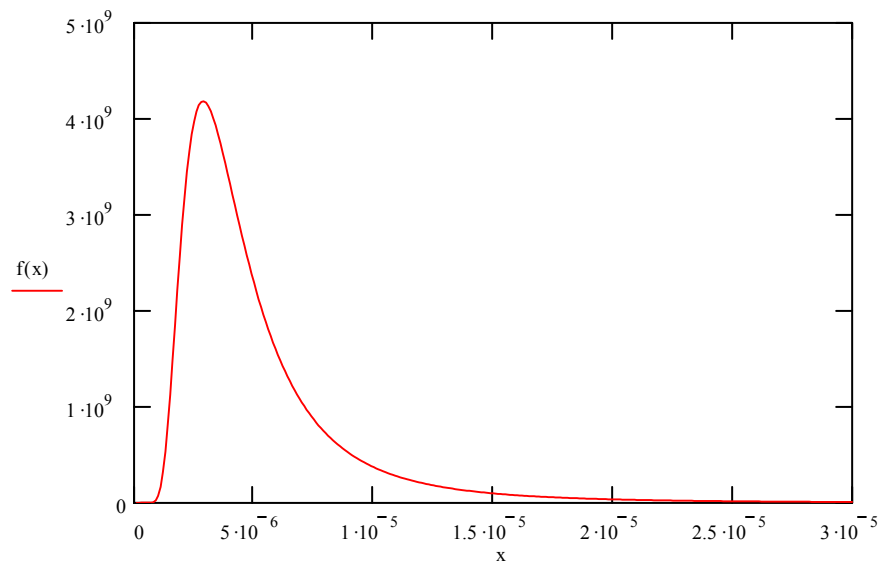
$$c2 := 1.43 \cdot 10^{-2} \quad c1 := 1.18 \cdot 10^{-16} \quad T := 1000$$

Planck's Law depending on wavelength

$$x := 3 \cdot 10^{-5}, 2.99 \cdot 10^{-5} .. 10^{-7}$$

x in meters

$$f(x) := \frac{c1}{x^5 \cdot \left[ e^{\left( \frac{c2}{x \cdot T} \right)} - 1 \right]}$$



### 2. Integration over the wavelength range from $3 \cdot 10^{-6}$ to $3 \cdot 10^{-5}$ meters to obtain the Radiance.

Radiance

$$R := \int_{3 \cdot 10^{-6}}^{3 \cdot 10^{-5}} f(x) dx$$

$$R = 1.316 \times 10^4$$

### 3. Multiplication with area times solid angle to obtain the Radiant energy.

Area A, Solid angle SA

$$A := .1 \quad SA := 4$$

Radiant energy RR

$$RR := A \cdot SA \cdot R$$

$$RR = 5.263 \times 10^3 \text{ Watts}$$

RR has the same value as the corresponding value when integrating over frequency.