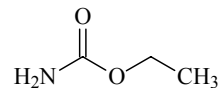


435
MW $\text{C}_3\text{H}_7\text{NO}_2$ **Ethyl carbamate**
Carbamic acid ethyl ester C_s (conformer I)
 C_1 (conformer II)

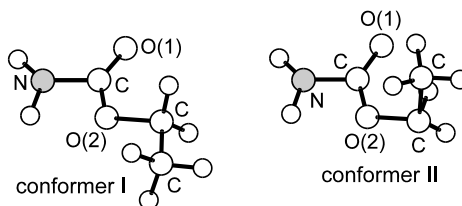
θ_0	deg ^{a)}	
	conformer I	conformer II
$\text{O}(1)=\text{C}-\text{O}(2)-\text{C}^{\text{b)}$	0.0	0.0
$\text{C}-\text{O}(2)-\text{C}-\text{C}^{\text{b)}$	180.0	$180.0 - 98(2)$



The $\text{H}_2\text{NCO}(1)\text{O}(2)\text{CC}$ atoms are coplanar in conformer I with the methyl group *anti* to the C(carbonyl)–O(2) bond. The methyl group is rotated 98° in conformer II from the position it has in I. Conformer I is found to be $0.5(5) \text{ kJ mol}^{-1}$ more stable than conformer II.

^{a)} Uncertainties were not estimated in the original paper.

^{b)} Dihedral angle.



Marstokk, K.-M., Møllendal, H.: Acta Chem. Scand. **53** (1999) 329.