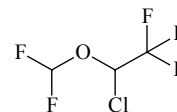


345 **C₃H₂ClF₅O**
ED, *ab initio* and DFT
calculations

2-Chloro-2-(difluoromethoxy)-1,1,1-trifluoroethane

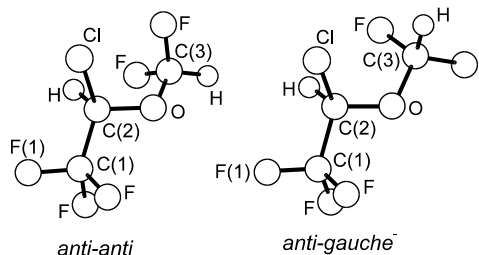
C₁ (*anti-anti*)
C₁ (*anti-gauche*[−])

<i>r_a</i>	Å ^{a)}	<i>θ_a</i>	deg ^{a)}
O–C(2)	1.401(15)	C(1)–C(2)–O	108.5(19)
O–C(3)	1.372 ^{b)}	C(2)–O–C(3)	113.4(28)
C(1)–C(2)	1.536(11)	F–C(1)–F	107.6(7)
C(1)–F	1.332(3)	C(1)–C(2)–Cl	109.9(7)
C(3)–F	1.350 ^{b)}	O–C(2)–Cl	111.4 ^{c)}
C(2)–Cl	1.773(8)	O–C(3)–F	111.8 ^{d)}
C–H (mean)	1.100 ^{e)}	F–C(3)–F	106.4 ^{f)}
		tilt (CF ₃) ^{g)}	1.6 ^{f)}
		τ_1 ^{h)}	−136(5)
		τ_2 ⁱ⁾	170(9)
		τ_3 ^{j)}	−177.2 ^{e)}



The molecule was found to exist as a mixture of *anti-anti* and *anti-gauche*[−] conformers in the ratio of 83(11):17. These conformers possess near *anti* configuration of the C–C–O–C skeleton and *anti* or *gauche*[−] configuration of C–O–C(3)–H unit. Local C_{3v} symmetry was assumed for the CF₃ group and C_{2v} for CHF₂. The structural parameters of the *anti-gauche*[−] conformer were coupled to those of the *anti-anti* conformer using the respective theoretical differences. All angles involving hydrogen atoms were assumed at the calculated values. The structural parameters are listed for the *anti-anti* conformer.

The nozzle was at room temperature.



^{a)} Three times the estimated standard errors.

^{b)} Differences in the O–C and C–F bond lengths were assumed at the values from MP2/6-311G(2d) calculations.

^{c)} Difference between the O–C(2)–Cl and C(1)–C(2)–Cl bond angles was assumed at the value from MP2/6-311G(2d) calculations.

^{d)} Difference between the O–C(3)–F and C(2)–C(1)–F (mean) bond angles was assumed.

^{e)} Assumed.

^{f)} Assumed at the value from MP2/6-311G(2d) calculations.

^{g)} Angle between the C₃ axis of CF₃ group and C(1)–C(2) bond, away from the Cl atom.

^{h)} Torsional angle C(1)–C(2)–O–C(3), $\tau_1 = 0^\circ$ for the *syn* position.

ⁱ⁾ Torsional angle C(2)–O–C(3)–H, $\tau_2 = 0^\circ$ for the *syn* position.

^{j)} Torsional angle F(1)–C(1)–C(2)–O, $\tau_3 = 0^\circ$ for the *syn* position.

Hermann, A., Mack, H.-G., Oberhammer, H.: J. Fluor. Chem. **101** (2000) 223.