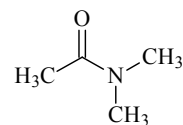


**587**      **C<sub>4</sub>H<sub>9</sub>NO**  
ED, *ab initio* and DFT  
calculations

***N,N*-Dimethylacetamide**

C<sub>s</sub> (skeleton)  
or quasi-C<sub>s</sub> (skeleton)



$r_a$	Å <sup>a)</sup>	$\theta_a$	deg <sup>a)</sup>
C–H	1.101(3)	N–C(1)=O	121.0(7)
C=O	1.226(3)	N–C(1)–C(2)	115.9(9)
N–C(1)	1.368(5)	C(1)–N–C(3)	123.8(13)
N–C(3,4)	1.453(3)	C(1)–N–C(4)	117.8(11)
C(1)–C(2)	1.527(6)	$\Sigma\alpha(\text{N})$ <sup>b)</sup>	354.1(17)
		H–C–H	108.9(7)
		tilt(C–CH <sub>3</sub> ) <sup>c)</sup>	3.4 <sup>d)</sup>
		tilt(N–CH <sub>3</sub> ) <sup>e)</sup>	0.5 <sup>d)</sup>
		$\tau_1(\text{N–C(3)})$ <sup>f)</sup>	19(12)
		$\tau_2(\text{N–C(4)})$ <sup>f)</sup>	–19(12)
		$\tau_3$ <sup>g)</sup>	18(3)
		$\tau_4$ <sup>h)</sup>	–11(2)

The molecule possesses either a planar equilibrium structure of the skeleton with a large-amplitude out-of-plane vibration (a combination of inversion at nitrogen and torsion around the N–C(1) bond) or a pseudoplanar structure of the skeleton with a potential hump of 15 cal mol<sup>–1</sup> lying below the vibrational ground state. Local C<sub>3v</sub> symmetry was assumed for the methyl groups and one C–H bond of the acetyl CH<sub>3</sub> group was assumed to eclipse the C=O bond.

The nozzle temperature was 50 °C.

<sup>a)</sup> Three times the estimated standard errors.

<sup>b)</sup> Sum of the angles at the N atom.

<sup>c)</sup> Tilt angle of the acetyl CH<sub>3</sub> group towards the C=O bond.

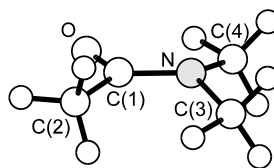
<sup>d)</sup> Assumed at the value from B3PW91/6-311G(2d) calculations.

<sup>e)</sup> Tilt angle of the amino methyl groups away from each other.

<sup>f)</sup> Torsional angles of the methyl groups around the N–C bond, C(1)–N–C–H; zero degree for the *syn* position.  $\tau_1 = -\tau_2$  was assumed.

<sup>g)</sup> C(2)–C(1)–N–C(3) torsional angle from the *syn* position.

<sup>h)</sup> O–C(1)–N–C(4) torsional angle from the *syn* position. The correlation  $\tau_4 = -\alpha\tau_3$  was assumed with  $\alpha$  taken from *ab initio* calculations.



Mack, H.-G., Oberhammer, H.: J. Am. Chem. Soc. **119** (1997) 3567.